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JPRS 36749

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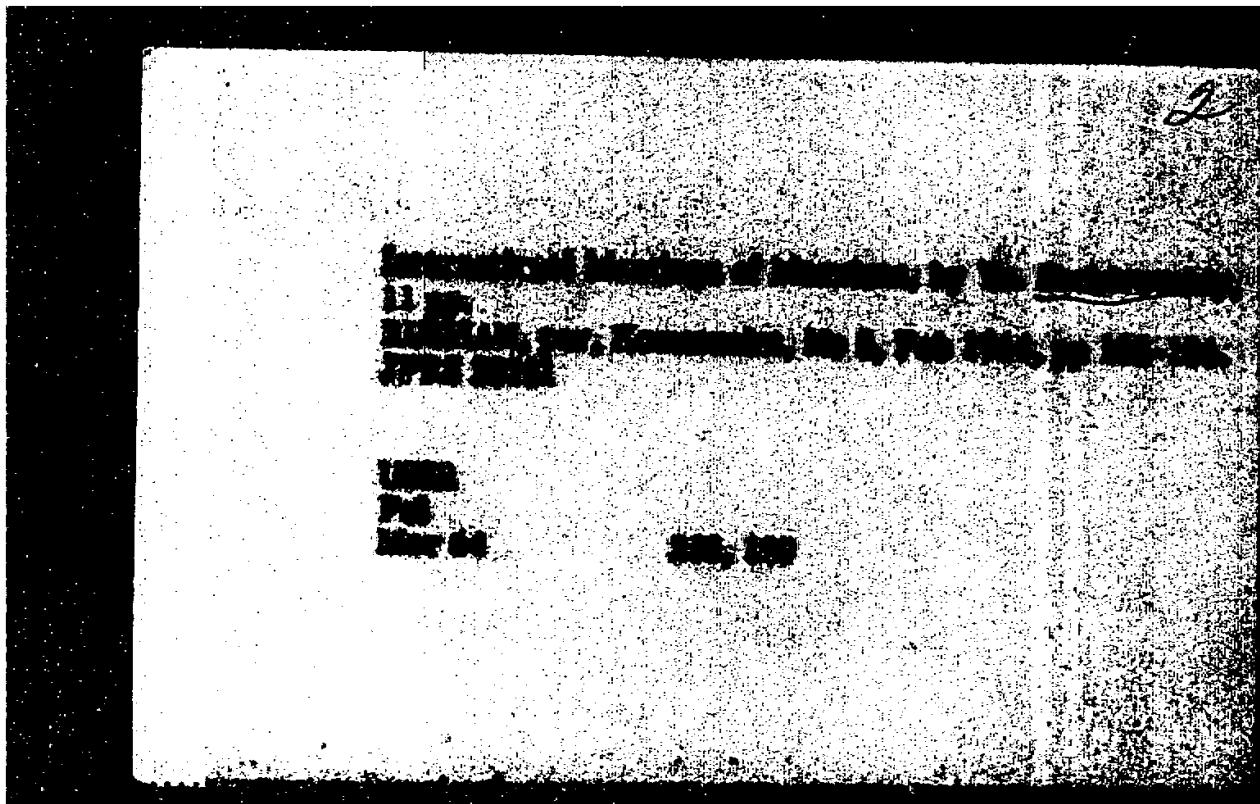
USSR

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Sept 66

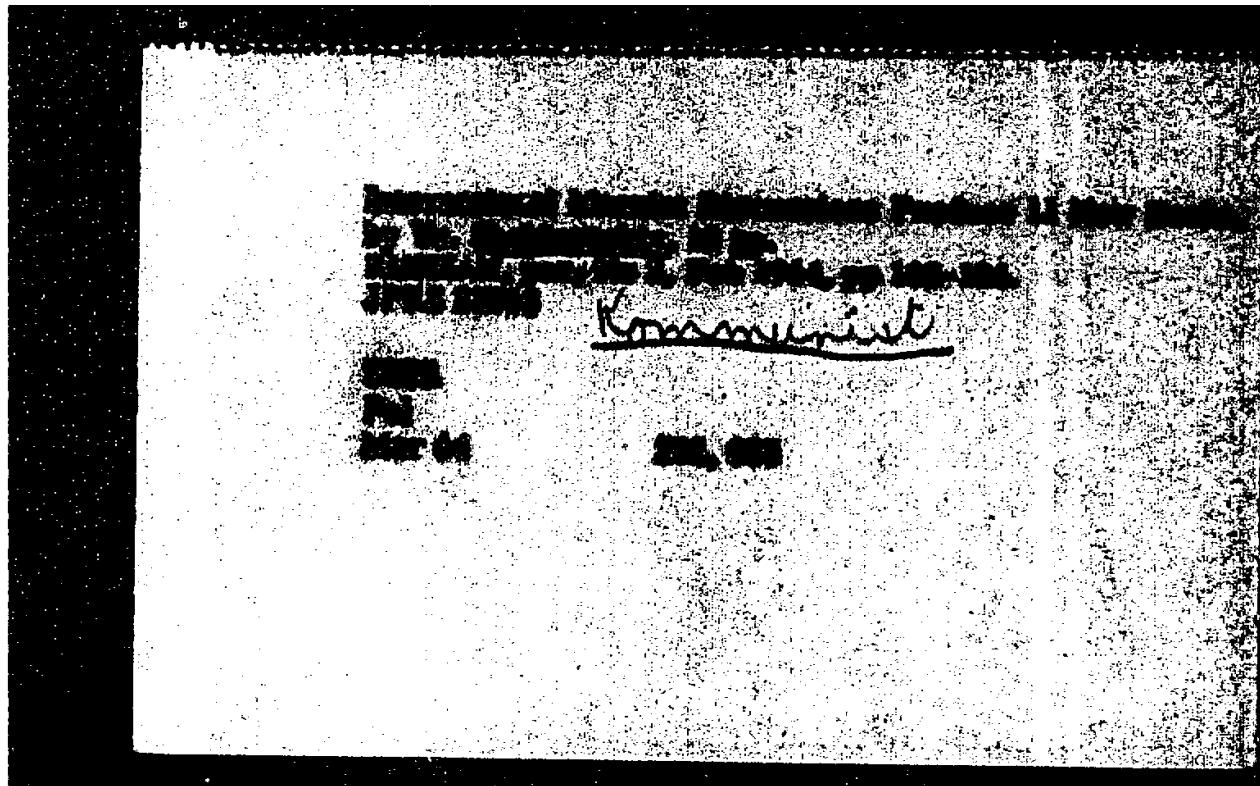
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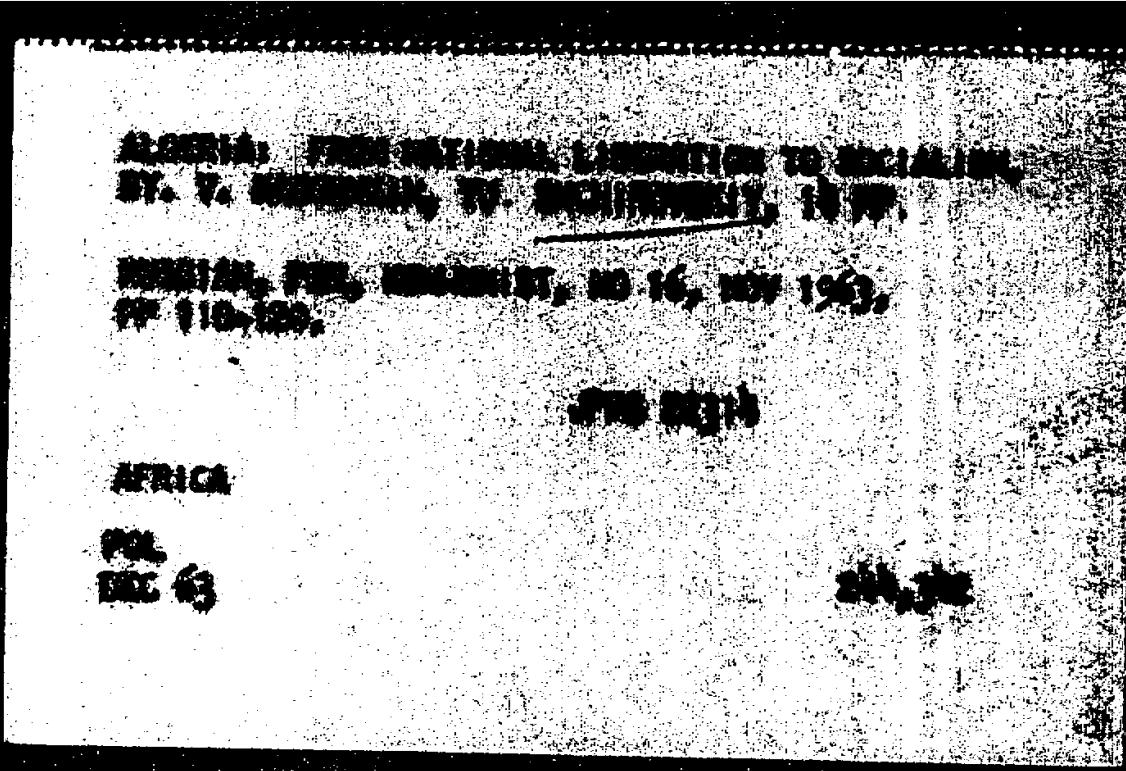
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Order from OTS, SLA, or ETC \$1.10 TT-64-15151

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III. Department of the Interior,  
Washington, D. C.

1083510

Office of Technical Services

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(Ob Isparivani Drukisli Gafniya) F. W. Wood, tr.  
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III. Department of the Interior,  
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\*Chlorides, Bromides, \*Heat of formation, Hydrolysis,  
Heat of reaction.

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An Investigation of the Possibility of Colorimetric Analysis in the Ultraviolet, by S. A. Shchukarev, S. N. Andreev, 12 pp. UNCLASSIFIED.

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Thermal Dissociation of Gold Chlorides, by S.A.  
Shchukarev, M.A. Oranskaya, 7 pp.

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AEC-Tr-4586

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Thermodynamic Investigation of Certain Chlorine Derivatives of Tungsten, by S. A. Shchukarev, G. I. Novikov, 6 pp.

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AEC-Tr-4584

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Enthalpy of formation of niobium pentabromide and  
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The Oxychloride of Trivalent Uranium, by S. A.  
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The Periodic Law of D. I. Mendeleev as a Basic Concept  
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Full translation.

RUSSIAN, no per, Zhur Obshch Khim, Vol XXIV, No 4,  
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Effect of the Concentration of Solutions of Acids on the  
Heat of Their Reaction with Zinc and Barium Hydroxides,  
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4 pp.

RUSSIAN, no per, Zhur Obshch Khim, Vol XXV,  
No 8, Aug 1955, pp 1444-1447. CIA C 42156.

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Sci - Chemistry

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Missing Perissads and Artiads that have No Stable  
Odd Subelements\* (\* Read at a meeting of the  
Leningrad Section of the D. I. Mendeleyev All-Un  
Chem Society on 25 Mar 1948.)

Zhur Obshch Khim, Vol XIX, No 3, 1949 (Mar), pp 345-  
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Consultants Bureau Translation

A - 1051

SHCHUKAREV, S. A., Chair of Gen and Inorg Chem,  
University of Leningrad

So-called Anomalies and Degenerated Anomalies of  
Element (Atomic) Weights\* (\*Presented Feb 6, 1947 at  
a meeting of the Chemical Society in Commemoration  
of the 40th Anniversary of the death of D. I. Men-  
deleyev

Zhur Obshch Khim, Vol XIX, No 3, 1949 (Mar), pp 329-  
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Consultants Bureau Translation

A 10152

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Stable Subelements between the Artiads and the  
Perissads\* (\*Read before a scientific meeting of  
Leningrad State University on January 9, 1947.)

Zhur Obshch Khim, Vol XIX, No 3, 1949 (Mar), pp ~~22~~-  
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A-10853

The Heats of Formation of  $\text{UO}_2\text{Cl}_2\text{aq}$ ,  $\text{UO}_2\text{Br}_2\text{aq}$ ,  
 $\text{UO}_2\text{Cl}_2 \cdot \text{H}_2\text{O}$ ,  $\text{UO}_2\text{Cl}_2 \cdot 3\text{H}_2\text{O}$ ,  $\text{UO}_2\text{Br}_2 \cdot \text{H}_2\text{O}$  and  
 $\text{UO}_2\text{Br}_2 \cdot 3\text{H}_2\text{O}$ , by S. A. Shechukarev, I. V. Vasili'-  
Nova, V. M. Drodova, K. E. Frantseva.

RUSSIAN, per, Zhur Neorgan Khim, Vol IV, No 1,  
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**DESCRIPTORS:** \*Perchloric acid, Gases, Water, Solutions, Organic solvents, \*Spectra (Infrared), Absorption spectra.

A study is made of vibrational frequencies of aqueous HClO<sub>4</sub> in a broad range of its concentrations, and of the vibrational spectra of this acid in gas phase and in organic, non polar solutions.

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ANHYDROUS RUTHENIUM TRIBROMIDE. [1962] [7]p.  
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no. 4, Ser[iya] Fiz[ika] i Khim[ii] no. 1, p. 100-104.

DESCRIPTORS: \*Ruthenium compounds, \*Bromides,  
Anhydrides, Synthesis, Physical properties,  
Dissociation.

Anhydrous ruthenium tribromide was prepared by  
directly combined ruthenium and bromine in sealed  
tubes at a bromine pressure of 15 to 20 atm and a tem-  
perature of 450° to 500°C. The density and the lattice  
plane spacing of RuBr<sub>3</sub> prepared in the dry way were  
determined. A study was made of the thermal dis-  
sociation of ruthenium tribromide, and the thermo-  
(Chemistry--Inorganic. TT, v. 10, no. 4) (over)

63-13401

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A THERMODYNAMIC INVESTIGATION OF CERTAIN  
CHLORINE DERIVATIVES OF MOLYBDENUM AND  
TUNGSTEN. [1962] 15p. 26 refs.  
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Trans. of Leningrad U. Vestnik (USSR) 1961 [v. 16]  
no. 4 [Seriya Fizika i Khimii no. 1] p. 87-99.

DESCRIPTORS: \*Molybdenum compounds, \*Tungsten  
compounds, \*Chlorides, \*Oxychlorides, Evaporation,  
Dissociation, Disproportionation, Transition  
temperature, Measurement, Chemical reactions,  
\*Thermodynamics.

The processes of evaporation, dissociation and  
disproportionation in the system  $\text{MoCl}_5 - \text{MoO}_3$  and  
 $\text{WCl}_6 - \text{WO}_3$  by means of membrane and optico-  
densimetric methods were studied. Some thermodynamic  
functions of these processes were obtained. (Author)  
(Chemistry--Physical, TT, v. 10, no. 3)

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