

ACCESSION NR: AT4021264

effectiveness of a detector with an organic scintillator. This method proves to be highly effective and acquires high calculation precision with moderate machine time consumption. All interaction processes of neutrons with nuclei of the scintillation substance are taken into consideration, including the marginal effects on the walls of the scintillator. These are presented in a graph. The paper also includes a table of registration efficiency for a 30 X 30 mm crystal. Orig. art. has: 3 formulas, 3 figures, and 1 table.

ASSOCIATION: Moskovskiy inzhenerno-fizicheskiy institut (Moscow Physics and Engineering Institute)

SUBMITTED: 00

DATE ACQ: 06Apr64

ENCL: 00

SUB CODE: MF, PH

NO REF SOV: 002

OTHER: 003

Card 2/2

ZLOTUKHIN, V.G.; MOGIL'NER, A.I.

Distribution of the number of recordings by a neutron detector placed  
in a reactor. Atom. energ. 15 no.1:11-16 J1 '63. (MIRA 16:8)  
(Nuclear counters) (Distribution (Probability theory))

ZOLOTUKHIN, V.G.; DOROSHENKO, G.G.; YEFIMENKO, B.A.

Calculation of pulse amplitude distributions and counting  
efficiencies for a fast neutron scintillation detector.  
Atom. energ. 15 no.3:194-200 S '63. (MIRA 16:10)

(Scintillation counters)

YERMAKOV, S.M.; ZOLOTUKHIN, V.G.; PETROV, E.Ye.

Calculating the passage of neutrons through a plane polyethylene  
layer. Atom. energ. 15 no.3:253-255 S '63. (MIRA 16:10)

(Neutrons—Capture) (Shielding (Radiation))

ACCESSION NR: AT4019045  
 AUTHOR: Zolobakhin, V. G.; Yermakov, S. M.  
 S/0000/63/000/000/0171/0181

TITLE: Application of the Monte Carlo method to the computation of nuclear radiation shielding

SOURCE: Voprosy fiziki zashchity reaktorov; sbornik statey (Problems in physics of reactor shielding; collection of articles). Moscow, Gosatomizdat, 1963, 171-181

TOPIC TAGS: nuclear reactor, reactor shielding, radiation shielding, Monte Carlo method, radiation transfer, scattering, neutron propagation, quadrature formula

ABSTRACT: The article contains a brief summary of the fundamental techniques for increasing the statistical efficiency of the Monte Carlo method with respect to problems of radiation transfer. The most important of the techniques discussed in the article have been proven on the basis of a large number of concrete problems. The author notes, by way of introduction, that considerable mathematical difficulties are encountered while solving the kinetic equation with consideration of the energy dependence of the cross section. Because of this fact, the Monte Carlo method provides the only means of solving the problem.

FOR RELEASE 03/15/2001

ACCESSION NR: AT4019045

The application of this method to problems involving the passage of neutrons and gamma quanta through a substance is possible due to the absence of any interrelation between the particles in real beams. The difficulties that arise in connection with the Monte Carlo method are concerned primarily with the determination of small probabilities. In problems connected with the passage of radiation through a substance, the smallness of the probability  $p$  may be occasioned by the absorption of the particles, their leakage from the medium, energy losses as the result of slowing, etc. It is pointed out that the Neuman series for the solution of the kinetic equation reduces the radiation transfer problem to the computation of multiple intervals, while the Monte Carlo method itself consists essentially in the calculation of the terms of a Neuman series which are the multiple intervals. The use of non-random points, in the opinion of the authors, implies a repudiation of the probabilities of the Monte Carlo scheme, eliminating the possibility of a practical evaluation of the accuracy of the results. In those cases in which interest attaches to a particular functional of the kinetic equation solution, the methods which are described in this article for increasing the statistical effectiveness of the method normally provide accuracy quite satisfactory for practical purposes with a number of histories ranging from

Card

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ACCESSION NR: AT4019045

10<sup>3</sup> to 10<sup>4</sup>. The authors describe the method of conditional probabilities (in American technical literature this method is known as the method of analytical averaging). It is noted that different modifications of this method are possible, all being based on the introduction of the transitional probability  $K(x \rightarrow x')$ , connected with  $K(x \rightarrow x)$  by the formula

$$K(x \rightarrow x') = \frac{K(x \rightarrow x')}{\int_{(D)} K(x \rightarrow x') dx'} \tag{1}$$

where (D) is the region of space T in which the function  $\phi(x)$  assumes the greatest values. The semi-analytical Monte Carlo method is briefly discussed. This method is based on the use of analytical solutions (provided such are possible) for certain ramifications of the basic straying process. The essential idea of the "control variable method" is explained. This technique is sometimes also called the "correlation sampling method". The point is made that the chief difficulty in the use of this method consists in finding random values  $\xi_1, \dots, \xi_n$  of high correlation with  $\xi_0$ , the mathematical expectancies of which are known. The method of local stream calculation is discussed and examples of its use are given. The use of quadrature formulas with random nodes is analyzed, and it is noted that the further development of the general methods for reducing the dispersion, based on the construction of interpolation-quadrature formulas, permits the formulation of

ACCESSION NR: AT4019045

quadrature expressions with random nodes of high accuracy, which are very useful for practical applications. A final section of the article deals with problem-solving through the use of high-speed electronic computers. Orig. art. has: 10 formulas.

ASSOCIATION: none

SUBMITTED: 14Aug63

DATE ACQ: 27Feb64

ENCL: 00

SUB CODE: NP

NO REF SOV: 010

OTHER: 005

x

Card 4/4



DOROSHENKO, G. G.; YEFIMENKO, B. A.; ZOLOTUKHINA, V. G.

"A Method of Calculating Efficiencies for the Investigation of Continuous Spectra of Fast Neutrons."

report submitted for All-Union Conf on Nuclear Spectroscopy, Tbilisi, 14-22 Feb 64.

MIFI (Moscow Engineering Physics Inst)

ZOLOTUKHIN, V.G.; KHAM'YANOV, L.P.; BLYSKAVKA, A.A.

Analyzing the characteristics of many-rotor mechanical  
neutron choppers. Trib. i tekhn. eksp. 9 no.2:36-39 Mr-Ap'64.  
(MIRA 17:5)

MIKHAYLUS, F. F.; ZOLOTUKHIN, V. G.; YERMAKOV, S. M.

"Solution methods of transport equation in inhomogeneous and finite media."  
report submitted for 3rd Intl Conf, Peaceful Uses of Atomic Energy, Geneva,  
31 Aug-9 Sep 64.

MOGIL'NER, A. I.; ZOLOTUKHIN, V. G.

"Space-time correlation relations in multiplying systems."

report submitted for 3rd Intl Cpnf, Peaceful Uses of Atomic Energy, Geneva,  
31 Aug-9 Sep 64.

BUBLIK, Yu.I.; YERMAKOV, S.M.; YEFIMENKO, B.A.; ZOLOTUICHIN, V.G.; PETROV, E.Ye.

Gamma-ray dose from a unidirectional source near the soil-air interface.  
Atom. energ. 18 no.6:628-629 Ja '65. (MIR) 18:7

L 1031-66  
ACCESSION NR: 027960

EWI(II) DIAAP DM

UR/0089/65/0119/001/0052/0056

AUTHOR: Koroshenko, I. G.; Zolotarev, I. G.; Yurimovko, B. I.

TITLE: Matrix treatment of data obtained by fast neutron single crystal scintillation spectrometer

SOURCE: Atomnaya energiya, v. 19, no. 1, 1965, 51-56

TOPIC TAGS: fast neutron, neutron spectrum, mathematic matrix, single crystal, crystal counter, spectrometer, Monte Carlo method

ABSTRACT: Matrices are calculated for the treatment of results of measurements of fast-neutron spectra. The counting efficiencies of a stilbene crystal (height 30 mm and diameter 30 mm) in the energy range 1 to 18 Mev taking into account energy dependence of the absorption coefficient and the shape of the spectrum are calculated. The calculation of the matrix elements is performed by the Monte Carlo method. The calculation of the matrix elements is performed by the Monte Carlo method. The calculation of the matrix elements is performed by the Monte Carlo method.

ASSOCIATION: none  
SUBMITTED: 22Sep64  
NO REF SOV: 007  
Card 1/1

ENCL: (0)  
OTHER: (07)

SUB CODE: NP, MA  
NA

AP5028636

SOURCE CODE: UR/0039/65/019/001/0156/0039

AUTHOR: Zolotukhin, V. G.; Doroshenko, S. G.; Yefimenko, B. A.

CRG: none

TITLE: Analysis of the systematic error in the differentiation of apparatus spectra measured by fast neutron single crystal diffraction

028136

SUB CODE: NP, MA, SS / 3200 1A



L-1164-66 EWT(m)/EWT(n)-2/EWA(h)

ACCESSION NR: AT5023150

UR/2892/65/009/004/0068/0072

AUTHOR: Doroshenko, G G Zolotukhin, V G

THEM E Simple method for finding the continuous spectra of fast neutrons

SOURCE: Moscow, Institute of Physics Institute Voprosy dosimetrii i zaish-

TOPIC TAGS neutron spectrum; fast neutron; mathematic matrix; hydrogen; radiation dosimetry

ABSTRACT: The article gives an evaluation of the accuracy of an approximate solution to the problem of the simple interaction of neutrons with hydrogen nuclei, using a matrix analysis of the results of measurements made with a single crystal scintillation fast neutron spectrometer. In the matrix method for computing the approximate solution, the system of linear equations is set up from

Card 1, 2

L 1164-66

ACCESSION NR: AT5023150

$$N(B_j) = \int_{B_j}^{E_{max}} f(E) \frac{v(E)}{E} (E - B_j) dE \approx \sum_{n=1}^N f(E_n) \frac{v_n(E_n)}{E_n} (k - \eta) (\Delta E)^2, \quad (1)$$

where  $N(B_j)$  is the integral velocity of the counter at the threshold energy  $B_j$ ;  $f(E)$  is the sought differential energy spectrum of the fast neutrons;  $E_{max}$  is the maximum energy of the neutrons;  $v(E)$  is the velocity of the neutrons;  $B_j$  is the spacing of the counter;  $\Delta E$  is the spacing of the elements of the counter;  $k$  and  $\eta$  are the constants of the counter;  $v_n(E_n)$  is the velocity of the neutrons;  $E_n$  is the energy of the neutrons;  $\eta$  is the correction of the energy scale;  $k$  is the constant of the counter;  $\eta$  is the correction of the energy scale. (Fig. 1 and 2)

ASSOCIATION: None

SUBMITTED: 00

ENCL 00

SUB CODE: NP

NR REF SOV: 005

OTHER: 001

L 2898-66

EWT(m)/EPP(c)/ETC/EPP(n)-2/EWS(r)/T

IR 1198/68/007/0001/0009

ACCESSION NR: ATSC...

AUTHORS: Bondarenko, I. I. deceased

...to measure directly

...physiological

...space environment

ABSTRACT: A physical experiment is proposed for continuously and directly measuring the n-n interaction cross section in the F-shell. Because of the very high vacuum of outer space, the experiment is proposed at an altitude of 400-500 km with a pulse reactor as the neutron source and an He3-filled ionization chamber as a detector. Starting with an expression for the number of pulses in the given chamber per single burst

$$J = K(t) Q^4 \bar{J}_{nn} \delta \bar{E} \text{ pulses}$$

and the following definition for the pulse width

L 2894-66  
ACCESSION NR: AT5022-18

$$\bar{\epsilon} = 3.5 \frac{\beta}{\delta K_0}$$

a criterion is derived for selecting the most suitable reactor

$$\delta K_0 = \frac{101}{2} (T_{max} - T_0)$$

where  $|\alpha|$  is the thermal coefficient of reactivity  $K_0$ . The best reactor is shown to be a zirconium-hydride one with a beryllium reflector. The number of neutrons emitted from the reactor is a function of  $\beta$ , and the value for the number of

neutrons is investigated, and it is shown that the number of neutrons is a function of  $\beta$  and  $\delta K_0$ . The number of neutrons is a function of  $\beta$  and  $\delta K_0$ . The number of neutrons is a function of  $\beta$  and  $\delta K_0$ .

The number of neutrons is a function of  $\beta$  and  $\delta K_0$ . The number of neutrons is a function of  $\beta$  and  $\delta K_0$ . The number of neutrons is a function of  $\beta$  and  $\delta K_0$ .

See CODE: NP  
Card 2/2

L 27477-66 EWT(1)/T IJP(c)

ACC NR: AT6008420

SOURCE CODE: UR/3158/65/000/021/0001/0012

AUTHOR: Zolotukhin, V. A.; Kukursov, A. A.; Broder, D. I.; Kham'yanov, L. P.;  
Yefimenko, B. A.; Shilkin, A. E.

ORG: None

16  
B

TITLE: Analysis and generalization of the correlation method of measuring the particle lifetime distribution in a physical system

SOURCE: Ozninsk. Fiziko-energeticheskii institut. Doklady, no. 21, 1965, Analiz i obobshcheniye korrelyatsionnogo metoda izmereniya raspredeleniya vremeni zhizni chastits v fizicheskoy sisteme, 1-12

ABSTRACT: The authors present a complete statistical analysis of the correlation method of measuring the distribution of the lifetime of particles in a linear physical system. The method is reduced to a determination of the mutual correlation function between a fixed counter signal and the intensity of the measured signal. The method depends on the counting rate of the detectors. The method is stable with respect to the method depends on the off-diagonal elements of the correlation matrix in the presence of a noise back-

Card 1

2

L 27477-66

ACC NR: AT6008420

ground against which the measurements are made. In particular, it is shown that the conclusions made by T. E. Stern et al. (J. of Nucl. En., p.A/B, 16, 499, 1964) that the use of random (or pseudorandom) excitation can completely reduce the measurement time compared with the classical method (ordinary periodic excitation) is valid only when there is an appreciable background. When there is no background, on the average the statistical accuracy of the classical and correlation methods is approximately the same. A new method of pseudorandom modulation of the particle source is proposed, to take advantage of this fact. If the modulation is made coherent with the background noise, then it can be readily shown that the fast component of the background can be readily eliminated in the same manner as in the classical method, and the slow component can be eliminated by suitable choice of the off-duty factor of the modulating signal. This type of statistical modulation prevents loss of the peak value of the modulated intensity and thus permits the use of the peak power of the source and retain the favorable advantages of the correlation method. (Orig. art. has: 6 figures and 13 formulae.

SUP CODE: 20/ SUBM DATE: 60/ ORIG REF: 001/ OTH REF: 002

Card 2/2

DEMOCHENKO, A.I. (deceased); ROVALEV, V.P.; BUKHARIN, A.O.

Possibility of using a nuclear reactor in outer space for  
direct measurement of the neutronizing cross section, *Izv.*  
*Acad. Sci. USSR Div. Phys. Math. Sci.* 1962, No. 1, p. 105.  
(MIR 1962)

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 REF ID: A100158  
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L 27871-66 INT(m)/3WA(h)

ACCESSION NR: APSC01112

10/15/66

*[Faint, mostly illegible text, possibly bleed-through from the reverse side of the page]*

Card 1/3

0 20 4 15



L 27871-66

ACCESSION NO: AP00111

tion in the work. Orig. authors: Ye. Fedorova and N. Ye. Fedorova for partici-  
pation in the work. Orig. authors: Fedorova, N. Ye., and Fedorova, Ye.

Card 3/12/66

L 04223-57 EWT(1)/EWT(m) GW

ACC NR: AR6031858

SOURCE CODE: UR/0058/66/000/006/V049/V049

AUTHOR: Gudkova, L. Ya. ; Degtyarev, S. F. ; Kukhtevich, V. I. ; Zolotukhin, V. G.

TITLE: <sup>19</sup> Scattered-neutrons field at the interface of earth and water with air

SOURCE: Ref. zh. Fizika, Abs. 6V405

REF SOURCE: Byul Inform. tsentra po yadern. dannym, vyp. 2, 1965, 346-382

TOPIC TAGS: scattered neutron field, earth air boundary, water air boundary, neutron flux, neutron dose rate, spatial variable, initial neutron energy

ABSTRACT: The basic characteristics of the scattered-neutrons field at the interface of earth and water with air have been investigated by both calculation and experimental methods. The dependence of flux and dose rate on spatial variables and on initial neutron energy was studied. A modification of the Monte-Carlo method, known as the method of the local calculation of the flux, was used for computation, and it was assumed that earth is a mixture of dry sand  $\text{SiO}_2$  with a density of  $1.7 \text{ g/cm}^3$  and contains 10 wt % water. The case of water was

Card 1/2

L 04223-67

ACC NR: AR6031858

investigated separately. In the method of calculation consideration was given to all the known interaction processes between neutrons and the nuclei of the substance in the energy range of 1 ev--10 Mev. The results are presented in numerous graphs and tables. [Translation of abstract]

SUB CODE: 18, 20/

Card 2/2 *pld*

L 05011-67 EWT(m)/EMP(t)/ETI IJP(c) JD/WN/JG/JR/GD  
ACC NR: AT6027925 SOURCE CODE: UR/0000/66/000/000/0104/0116

AUTHOR: Broder, D. L.; Zhilkin, A. S.; Zolotukhin, V. G.; Tarasko, M. Z.; Kutuzov, A. A. 47  
B-1

ORG: None

TITLE: Fast neutron spectra in metal-water shielding

SOURCE: Voprosy fiziki zashchity reaktorov (Problems in physics of reactor shielding); sbornik statey, no. 2. Moscow, Atomizdat, 1966, 104-116 19

TOPIC TAGS: fast neutron, radiation shielding, neutron spectrum

ABSTRACT: The authors study the spectra of neutrons in the energy range above 1 mev from sources with energies of 3.35 and 14.9 mev in water and in water behind layers of iron and lead. A scintillation spectrometer with a stilbene crystal was used for the measurements. The sensitivity to  $\gamma$ -quanta was reduced by time division of irradiation. The reactions used for the neutron sources were  $D(d,n)He^3$  and  $T(d,n)He^4$  produced by using deuterons to bombard zirconium-tritium and zirconium-deuterium targets with a thickness of 18 mg/cm<sup>2</sup>. For the measurements in water, the source was located in a paraffin block placed in direct contact to the water tank. The overall dimensions of the shielding were 710x710x600 mm. The scintillation spectrometer was combined with an FEU-13 photomultiplier and an AI-100-1 amplitude analyzer. The results show that

Card 1/2

L 05041-67

ACC NR: AT6027925

the neutron spectrum from a monochromatic source in lead and iron differs considerably from that in water. The iron and lead spectrum shows a stronger concentration of low-energy neutrons (<2 Mev). In the energy range from 2 Mev to the initial energy of the 3.35 Mev source and from 4-5 Mev to the initial energy of the 14.9 Mev source, the spectrum in water contains more neutrons than that in iron and lead. This form of spectrum explains the excellent shielding properties of iron and lead for fast neutrons as well as their poor characteristics for comparatively low-energy neutrons. These data also explain the excellent shielding properties of metal-water shielding throughout the entire energy spectrum. Spectra for neutrons in the energy region below the initial energy in water behind layers of lead and iron approach the shape of spectra in water at a thickness of greater than 20 cm. For thinner water layers, particularly below 2-3 Mev, the spectrum shows high concentrations of neutrons in comparison with the spectrum in water. In this transition region there is also a considerable difference from the spectrum in pure water for the energy range from 2 Mev to the initial energy. Orig. art. has: 10 figures, 1 table, 2 formulas.

SUB CODE:2018/ SUBM DATE: 12Jan66/ ORIG REF: 004/ OTH REF: 004

Card 2/2 *pla*



I. 22419-66 EWT(n)/EWA(h)  
ACC NR: AP6007050

NUMBER: 10/197 (10-11-1964)

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THIS PAGE: neutron spectrometry, neutron detector, coincidence  
detector, pulse height analyzer, nuclear reaction, alpha decay,  
neutronium

ABSTRACT: This is an abstract of article No. 519-01 submitted to  
the source editor but not published. The article improves the  
accuracy of the integral method for determining the derivative,  
... ..

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UDC: 639.16.08:639.195.6

2

L 22419-66  
ACC NR: AP6007950

decreasing the fluctuations that result from differentiation of ax-  
perimental amplitude distributions. For example, the following figure

*[Faint, mostly illegible text, possibly describing a figure or experimental results]*

Card. 2/2 *less*

ZOLOTUKHIN, V.I.

Role of zemstvo physicians in the control of syphilis in Russia.  
Vest. dermat. i ven. 38 no.9:68-70 S '64.

1. Respublikanskiy kozhno-venerologicheskiy dispanser (MIRA 18:4)  
vrach V.P.Konenkova) Udmurtskoy ASSR; nauchnyy rukovoditel' -  
chlen-korrespondent AMN SSSR prof. P.V.Kozhevnikov.

ZOLOTOUKHIN, V.I. (Izhevsk)

Incidence of syphilis in the population of Udmurtia in the  
postwar period. Vest.derm.i ven. no.11:61-62 '61. (MIRA 14:11)  
(UDMURT A.S.S.R.---SYPHILIS)

ZOLOTUKHIN, V.I.

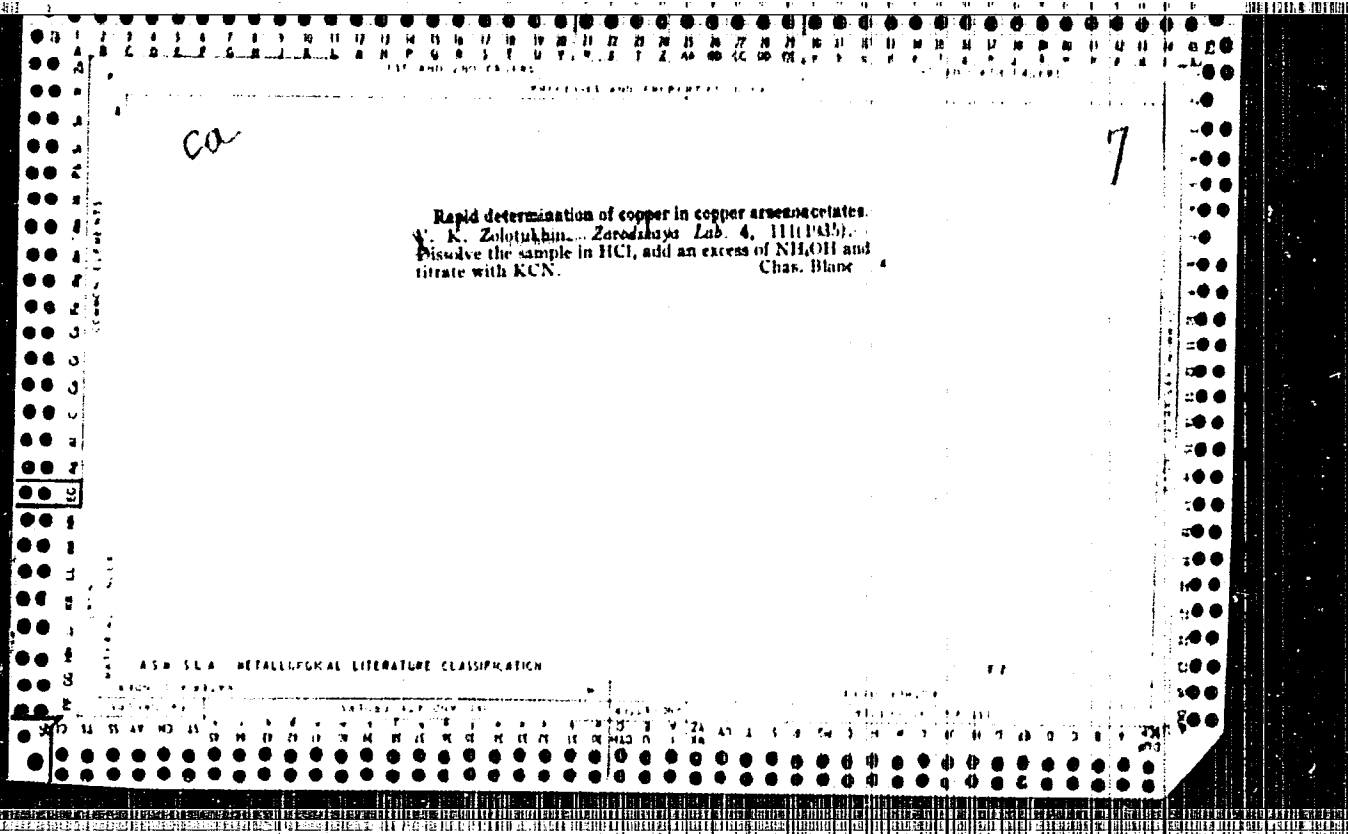
Concept of dispensary services for syphilitic patients by  
"zemstvo" physicians in the Vyatka Government. Vest. dermat.  
i ven. 37 no.4:65-69 ap '63. (MIRA 17:5)

1. Respublikenskiy kozhno-venerologicheskiy dispanser Udmurtskoy  
ASSR (glavnyy vrach V.P. Kosenkova).

ZOLOTKHIN, V.I.

Overall mechanization of the assembling and painting shop.  
Biol. tekh.-ekon. inform. Gos. nauch.-issl. inst. nauch. i  
tekh. inform. 17 no.12:29-31 D '64.

(MIRA 18:3)



BC ~~SECRET~~ ~~CONFIDENTIAL~~ ~~SECRET~~ a. v.

Determination of potassium carbonate in waste  
Dewar ash. V. K. KOTAROVNIK (J. Appl. Chem.,  
Russ., 1936, 9, 522-523) 20 g. of ash is extracted  
with 150-200 ml. of boiling H<sub>2</sub>O for 10 min., the  
cooled solution is diluted to 250 ml., and 50 ml. of the  
filtered solution are titrated with 0.1N-HCl. A deduc-  
tion of 0.3 ml. of acid per 100 ml. of solution is made,  
to correct for solubility of CaCO<sub>3</sub>. R. T.

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION	ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION	ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION	ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION
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25	26	27	28
29	30	31	32
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45	46	47	48
49	50	51	52
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73	74	75	76
77	78	79	80
81	82	83	84
85	86	87	88
89	90	91	92
93	94	95	96
97	98	99	100



PROPERTIES AND PREPARATION

The carrying down of barium, calcium and magnesium salts by ferric and aluminum hydroxides precipitated by ammonia. V. K. Zolotarev. *Acta Univ. Tomskensis* 9, No. 3, (1951) German 22, 30, 31, 32; cf. C. A. 44, 14939. When  $Fe(OH)_3$  or  $Al(OH)_3$  is pptd. with excess  $NH_4OH$  in the presence of Ba, Ca or Mg salts, these cations are carried down with the ppt. and cannot be washed out. Ba and Ca are partly pptd. as  $BaCO_3$  and  $CaCO_3$ , owing to absorption of  $CO_2$  from the air by  $NH_4OH$ , and are partly adsorbed by the hydroxides. With Mg, adsorption alone is responsible. The impurities can be almost entirely removed by washing the ppt. with  $NH_4Cl$  soln. However, if pptn. occurs at  $pH$  less than 7, the carbonates are not formed and the ppt. has a pos. charge and cannot adsorb cations. Hence, if Al and Fe are pptd. by neutral reagents, such as  $NaNO_3$  or nitroprusside, or if only enough  $NH_4OH$  is used to keep methyl red on the acid side, ppt. entirely free from Ba, Ca and Mg are obtained.

H. M. Leicester

ASB-55A METALLURGICAL LITERATURE CLASSIFICATION

PROCESSES AND PROPERTIES AND  
LIST AND JAC. SERIES

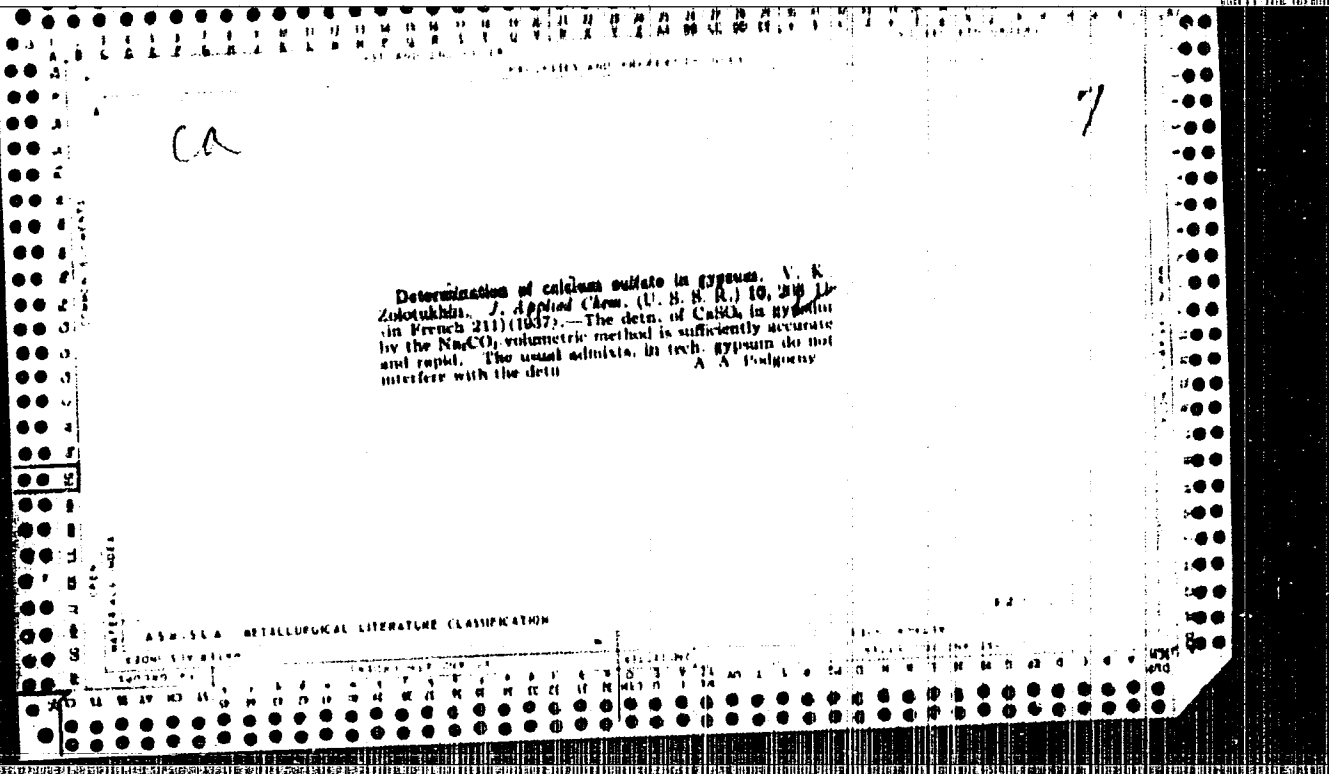
7

The carrying down of barium, calcium and magnesium salts by the sulfides of the third-group elements precipitated by ammonium sulfide. V. K. Zolotarev. *Dokl. Akad. Nauk SSSR*, No. 3, 84, 7 in German 1957(1957), 5.

When sulfides of the 3rd group are pptd. in alk. soln., Ba, Ca and Mg are carried down by adsorption and as carbonates. This can be prevented by pptg. at pH, but in this case only Co, Zn and Ni are pptd. completely. H. M. Leckner.

METALLURGICAL LITERATURE CLASSIFICATION

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
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1ST AND 2ND PAGES

PROCESSES AND REPERFORM INDEX

BC

Gravimetric determination of iron and aluminum by precipitation as hydroxide with ammonia, in presence of calcium, barium, and magnesium. V. E. Zolotarevskii (*J. Appl. Chem. Russ.*, 1937, 10, 1181-1180). Fe and Al hydroxides are pptd. by a slight excess of  $NH_3$ , and the ppt. is washed with 1% aq.  $NH_4NO_3$ ; Ca, Ba, and Mg do not interfere. R. T.

a-1

Common Element

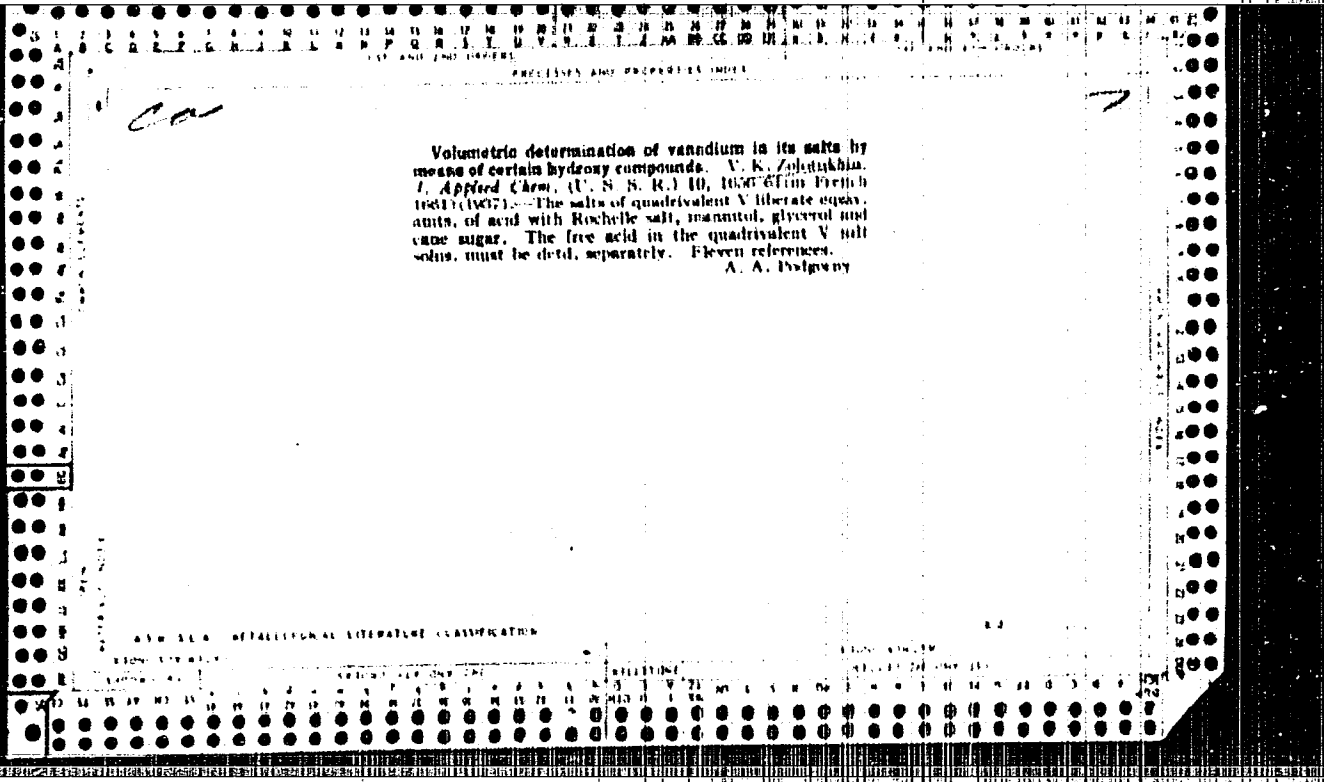
45A-31A METALLURGICAL LITERATURE CLASSIFICATION

STANDARD	SYMBOL	DESCRIPTION	REMARKS
...	...	...	...
...	...	...	...
...	...	...	...

ca

Acidimetric determination of vanadium in quinquevalent compounds by means of hydroxy compounds. V. K. Zolotukhin. *J. Applied Chem. (U. S. S. R.)* 10, 1651-5 (in French 1935) (1937).—Reduce the V to the quadrivalent state and carry out the method of Zolotukhin, *J. Applied Chem. (U. S. S. R.)* 6, 1676 (1933). A. A. P.

ALU SLA METALLOGICAL LITERATURE CLASSIFICATION



Reaction of thallic salts (of mineral acids) with alkali tartrate and its utilization in volumetric analysis. V. R. Zolotarev. *Acta Univ. Voronezensis* 11, No. 2, 111-116 (1939); *Khim. Referat. Zhur.* 1940, No. 2, 60.---When the thallic salt of a mineral acid reacts with a neutral soln. of alkali tartrate, a complex thallic tartrate is formed and from 4  $Tl^{3+}$  there are 3  $H^+$  liberated. The complex thallic salt is fairly stable provided a considerable excess of alkali tartrate is present (about 10 times the theoretical quantity). To analyze a thallic salt, first det. any free acid present (a) by adding NaCl or KCl and titrating directly with base to a methyl orange end point, or (b) by adding excess alkali oxalate +  $MgCl_2$  and titrating with base to a methyl red end point or (c) by titrating the solution until a slight, permanent ppt. of reddish brown  $Tl(OH)_3$  is formed. Then, in another sample det. the free acid + the mineral acid formed by reaction with neutral, alkali tartrate in the presence of phenol red as indicator. With  $TlCl_3$ , the complete analysis can be made with a single sample, first titrating with NaOH until a permanent turbidity of  $Tl(OH)_3$  appears, then adding a 10-fold excess of alkali tartrate and titrating as above described. Mannitol, glycerol, alkali citrate, etc., do not have a corresponding effect upon solns. of thallic salts. W. R. Henn

ASB-11A METALLURGICAL LITERATURE CLASSIFICATION

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

PROCEDURES AND PROPERTIES INDEX

7

*Ch*

Determination of zirconium by the acidometric method with hydroxy compounds. *Vestn. Zhelezn. A. to U.S.S.R. (U. S. S. R.) 11, No. 2, 37-38 (1948); Akad. Referat. Zhur. 1948, No. 2, 61.*—The method is based on the detn. of acid sep'd. in the reaction of the Zr salt with Na tartrate. To a known amt. of the Zr salt soln. add an excess of Na tartrate and titrate the soln. in the cold with NaOH (thymolphthalein indicator). In a sep. portion of the soln. det. the free acid according to the KP method, for which to a known amt. of the Zr salt add a soln. of KP (neutral to phenol red) soln. with KCl and KNO<sub>3</sub>, add to the soln. KCl or KNO<sub>3</sub> to satn., let stand for 15-20 min. and titrate with NaOH in the presence of phenol red. The Zr content is calcd. from the amt. of NaOH used for the titration of the acid equiv. to Zr. This is obtained from the difference between the NaOH used for detg. the total acidity (after the addn. of Na tartrate) and the free acid. W. R. Benn

438-35A METALLURGICAL LITERATURE CLASSIFICATION

COMMON ELEMENTS

COMMON AND RARE EARTH ELEMENTS

COMMON AND RARE METALS

COMMON AND RARE ALLOYS

COMMON AND RARE COMPOUNDS

COMMON AND RARE PROCESSES

COMMON AND RARE EQUIPMENT

COMMON AND RARE TOOLS

COMMON AND RARE MATERIALS

COMMON AND RARE METHODS

COMMON AND RARE STANDARDS

COMMON AND RARE TESTS

COMMON AND RARE MEASUREMENTS

COMMON AND RARE CALCULATIONS

COMMON AND RARE DIAGRAMS

COMMON AND RARE TABLES

COMMON AND RARE REFERENCES

COMMON AND RARE INDEXES

COMMON AND RARE NOTES

COMMON AND RARE APPENDICES

COMMON AND RARE GLOSSARIES

COMMON AND RARE ABBREVIATIONS

COMMON AND RARE SYMBOLS

COMMON AND RARE UNITS

COMMON AND RARE CONVERSIONS

COMMON AND RARE FORMULAS

COMMON AND RARE EQUATIONS

COMMON AND RARE REACTIONS

COMMON AND RARE PROCESSES

COMMON AND RARE METHODS

COMMON AND RARE STANDARDS

COMMON AND RARE TESTS

COMMON AND RARE MEASUREMENTS

COMMON AND RARE CALCULATIONS

COMMON AND RARE DIAGRAMS

COMMON AND RARE TABLES

COMMON AND RARE REFERENCES

COMMON AND RARE INDEXES

COMMON AND RARE NOTES

COMMON AND RARE APPENDICES

COMMON AND RARE GLOSSARIES

COMMON AND RARE ABBREVIATIONS

COMMON AND RARE SYMBOLS

COMMON AND RARE UNITS

COMMON AND RARE CONVERSIONS

COMMON AND RARE FORMULAS

COMMON AND RARE EQUATIONS

COMMON AND RARE REACTIONS



*B. Also*

Some of various Chlorm as indicators in alkalimetry and acidimetry. V. K. Zolotareva (Zvezd. Lab. 1940, 8, 133-134). The colour change is so abrupt that it can be used in titration, especially in presence of oxidising substances which destroy org. indicators. The solution must contain much  $\text{Cl}^-$ , so that  $\text{NaCl}$  is added to  $\text{H}_2\text{SO}_4$  or  $\text{HNO}_3$  before titration. Examples are given for solutions containing  $\text{Cl}_2$ ,  $\text{Br}$ , and  $\text{NO}_2$ . J. J. H.

ZOLOTKHIN, V.K.

Acidimetric determination of beryllium by certain organic hydroxy  
compounds. Nauk.zap.L'viv.un. 9:55-63 '48. (MINIA 10:5)

1.Kafedra analiticheskoy khimii.  
(Beryllium) (Titration)

PROCESS AND PROPERTY INDEX

7

*Handwritten:* 4

Determination of free acid in solutions of the salts.  
 A. S. Mokohtova and V. K. Zolotareva, *Zavodskaya Lab.*  
 14, 46-7(1948).—Mix a definite vol. of the test soln. of  
 Sn salt with a 1.5-2-fold amt. of  $K_2Fe(CN)_6$  (of that re-  
 quired stoichiometrically) in a 100- or 200-ml. flask until  
 a ppt. forms, dil. to the mark with a mixt. of equal vols.  
 of EtOH and  $NH_4NO_3$ , stir, filter, and titrate (13 or 20)  
 ml. of the filtrate with NaOH soln. in the presence of 2-3  
 drops of methyl red. B. Z. Kuzich

ASB-51A METALLURGICAL LITERATURE CLASSIFICATION

MATERIALS INDEX

GROUP	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
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ZOLOTOUKHIN, V. K.

35840. O permanganatometricheskom opredelenii formal'degida i murev'inox kisloty.  
Zavodskaya laborotoriay, 1949, No. 11, S. 1284-86.-Bibliogr: 5 nazv.

SO: Letopis' Zhurnal'nykh "tatey, Vol. 39, Moskva, 1949

CA

7

Permanganate method of determination of formalde-  
hyde and formic acid. A. S. Mokhtova and V. K. Evsi-  
tukhin (State Univ., Lvov). *Zovshchyni Lab.* 15, 1281 (1949). -- Detn. of  $\text{CH}_2\text{O}$  with  $\text{KMnO}_4$  in  $\text{Na}_2\text{CO}_3$  soln. is unsatisfactory because of the uncertain endpoint, but in strongly acid  $\text{H}_2\text{SO}_4$  soln. is worse from decompn. of  $\text{KMnO}_4$ .  $\text{KMnO}_4$  is stable in  $\text{Na}_2\text{CO}_3$  solns. for only 1 hr. in the cold. The best procedure is to add excess  $\text{KMnO}_4$  to the  $\text{Na}_2\text{CO}_3$  soln. of the unknown, let stand 20-30 min., acidify with 7N  $\text{H}_2\text{SO}_4$ , and immediately back titrate with 0.1 N Mohr's salt soln. with an excess of the latter; the excess of Mohr's salt is then back titrated with 0.1-0.5N  $\text{KMnO}_4$ . Results check within 0.01%.  $\text{MeOH}$  interferes with the analysis. G. M. Kuznetsov

6

The reactions of tin salts with the tartrates of the  
alkaline metals. V. K. Zolotarev (U.S.S.R. State Univ.).  
V. Gen. Chem. U.S.S.R. 19, 673-6 (1940) (Engl. transla-  
tion). See C.A. 44, 604. B. J. C.

CA

Reactions of tin salts with alkali metal tartrates. A. K. Zdobychkin, *Zh. Obshch. Khim.* (J. Gen. Chem.) 19, 983 (1946). - By titration with alkali, to phenolphthalein or thymolphthalein, it was proved that mixing of solns. of  $\text{SnCl}_4$  with  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$  liberates 4 equivs. acid per Sn. This ratio remains unchanged when the proportions are varied from 3:1 to 1:20. At all these proportions, the solns. remained clear at the neutralization point, but became turbid with time at 3:1 and 2:1. In alk. solns., Sn tartrates are pptd., unless there is at least a 4:1 excess of  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$ . Clear solns. obtained at the molar proportions 3:1, 2:1, 1:1, and 1:2, became turbid on addn. of HCl or KCl, probably owing to coagulation of colloidal dissolved Sn tartrates.  $\text{H}_2\text{S}$  ppts.  $\text{SnS}_2$  even in the pres-

ence of a 70-fold excess of  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$ . Heating causes pptn. through hydrolysis. That the total acid is liberated in the very act of mixing, not gradually during the titration, was demonstrated by detns. of the acidity of the filtrate from the Sn tartrate ppt. formed in a weakly acid soln. (0.03 N). These facts prove that salts of Sn<sup>IV</sup> react with  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$  with substitution of SnCl for H of the OH groups of 1 or of 2 mols. of the tartrate. Liberation of free acid was found also on mixing solns. of  $\text{SnCl}_4$  with  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$  in the molar proportions 2:1 to 1:20. The amt. of acid liberated varies with the mol. proportion, from 50% of the theoretical amt. at 1:1, 54% at 1:2, 58% at 1:4, to 75% at 1:8 to 1:12; in the latter case, no ppt. was formed during the titration up to the end point. With  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$  added to a strongly acid soln. of  $\text{SnCl}_4$ , no ppt. is formed at any proportion. In a weakly acid soln., a ppt. is formed unless  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$  is used in at least 3-fold excess. The stability of the Sn tartrates towards NaOH,  $\text{Na}_2\text{HPO}_4$ , and  $\text{K}_2\text{Cr}_2\text{O}_7$  increases with increasing excess of the  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$ . At  $\text{SnCl}_4:\text{Na}_2\text{C}_2\text{H}_4\text{O}_6 = 1:8$ , the solns. remain unprecipitated by any of these reagents even on heating.  $\text{H}_2\text{S}$  ppts.  $\text{SnS}$  even in the presence of 20-fold excess of  $\text{Na}_2\text{C}_2\text{H}_4\text{O}_6$  in a neutral medium, and a 75-fold excess in an acid medium. The  $\text{SnC}_2\text{H}_4\text{O}_6$  described in the literature can be formed only in strongly acid solns.

N. Thon

7

*Acidimetric determination of beryllium. V. N. Zolotarev (Lvov State Univ.). Zhur. Anal. Khim. 6 (1951)-60 (1951).—The reaction  $\text{Be(OH)}_2 + 4\text{KF} \rightarrow \text{K}_2\text{BeF}_4 + 2\text{KOH}$  proceeds to completion. However, the  $\text{Be(OH)}_2$  obtained by pptn. of Be in a neutral soln. with  $\text{NaOH}$  or  $\text{KOH}$  occluded basic salts or ions of Be and the subsequent titration of  $\text{KOH}$  or  $\text{NaOH}$  with  $\text{HCl}$  gave lowered results. Direct titration of neutral Be solns. with  $\text{NaOH}$  or  $\text{KOH}$  with phenolphthalein as indicator also gave low results.  $\text{Be(OH)}_2$  obtained from the reaction  $\text{BeSO}_4 + \text{KIO}_3 + \text{KI} \rightarrow \text{Be(OH)}_2 + \text{K}_2\text{SO}_4 + 2\text{I}$  was also contaminated. Better results were obtained by titrating neutral  $\text{BeSO}_4$  soln. with  $\text{Be(OH)}_2$ . However, this method was applicable only to  $\text{BeSO}_4$  and other Be salts had to be converted first into  $\text{BeSO}_4$ . The  $\text{Be(OH)}_2$  thus obtained was free of occlusions. M. Horsch.*



1039. ACIDIMETRIC DETERMINATION OF TIN BY MEANS OF SODIUM TARTRATE.

V. K. Zolotukhin (J. anal. Chem. USSR, 1951, 6, 300-302). - The reaction between  $\text{Sn}^{4+}$  and tartrate ( $\text{Sn}^{4+} + \text{C}_4\text{H}_4\text{O}_6 + \text{H}_2\text{O} = \text{C}_4\text{H}_3\text{SnO}_6 + 4\text{H}^+$ ) (Zolotukhin, A., 1950, I, 640), can be used for acidimetric determination of  $\text{Sn}^{4+}$  by titration to the phenolphthalein end-point. Presence of free acid ( $\text{C}_4\text{H}_4\text{O}_6$ ) is essential, but this can be allowed for by adding  $\text{K}_2\text{Fe}(\text{OH})_6$  (Kolotkova and Zolotukhin, Izv. Akad. Nauk SSSR, 1949, 14, 46) to precipitate  $\text{Sn}^{4+}$  and then alkali titration of the filtrate. With mixtures of  $\text{Sn}^{4+}$  and  $\text{Sn}^{2+}$  the  $\text{Sn}^{4+}$  can be determined by titration with I, leaving the acidity unchanged, and portions of the oxidized solution can be used for determination of free acid and total Sn, respectively.

G. S. Smith.

ZOLOTUKHIN, V.K., dotsent.

Volumetric determination of aluminum, Dop.ta pov.L'viv.un. no.  
3 pt.2:31-32 '52. (MLRA 9:11)

(Volumetric analysis) (Aluminum)

~~ZOLOFUKHIN V F.~~

Acidimetrid determination of aluminum. Nauk.zap.L'viv.un.  
21:145-151 '52. (MIRA 10:7)

1. Kafedra analiticheskoy khimii.  
(Volumetric analysis) (Aluminum)

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USSR/Analytical Chemistry - Analysis of Inorganic Substances G-2

Abs Jour : Referat Zhur - Khimiya, No 3, 1957, 8434

Author : Zolotukhin, V. K.

Inst : Lvov University

Title : On the Determination of Beryllium and Aluminum by the Phosphate Method.

Orig Pub : Dopovidi ta povidomlennya. L'vivsk. un-t, 1955, Vbl 2, No 6, 134-137

Abstract : The accuracy of methods based on the determination of Be and Al as  $\text{BeNH}_4\text{PO}_4$  and  $\text{AlPO}_4$  and the basicity of the precipitates have been investigated. It is shown that when Be is precipitated as  $\text{BeNH}_4\text{PO}_4$ , the results are lower than those obtained with the ammonia method by  $\sim 3.8\%$ . The basicity of the beryllium phosphate precipitate, precipitated from a 0.1 M solution of  $\text{BeSO}_4$  by heating, corresponds to 37.8-38.8%; the acidity of the filtrate is 33.2-39.6% of the starting Be salt (in equiv.). The basicity of the beryllium phosphate precipitated by heating a 0.05 M solution and the acidity of the filtrate are 9-10% higher than the corresponding values obtained with the ammonia method; when the precipitation is

Card 1/2

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ZOLOTUKHIN, V. K.

USSR/Analytical Chemistry - Analysis of Inorganic Substances, G-2

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61822

Author: Zolotukhin, V. K.

Institution: None

Title: Acidimetric Determination of Beryllium by Means of Sodium Salicylate

Original

Periodical: Nauk. zap. L'vivs'k. un-tu., 1955, 34, 115-117

Abstract: To determine Be in neutral and acid media and in basic salts a definite volume of Be salt is treated with excess 0.4N-solution of Na-salicylate and neutralized with 0.1N NaOH to thymolphthalein. Then is added 1.25-3-fold amount (on the basis of theoretical) of 0.1N HCl and 2-3-fold excess of KF solution neutralized to the orange transition point of phenol red (I). After 3-5 minutes excess acid is titrated with 0.1N NaOH in the presence of I, to the control (KF solution neutralized to I). Optimal results obtained with molar ratio  $BeSO_4 : C_6H_4OHCOONa = 1:3.2$  and 0.25 excess

Card 1/2



USSR/Analytical Chemistry - Analysis of Inorganic Substances, G-2

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61822

Abstract: HCl. On increasing the excess of salicylate the amount of HCl is increased.

Card 2/2

ZOLOTUKHIN, V. K.

USSR/Analytical Chemistry - Analysis of Inorganic Substances, G-2

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61830

Author: Molotkova, A. S., Zolotukhin, V. K.

Institution: None

Title: On Acidi-Alkalimetric Determination of Zirconium and Titanium

Original

Periodical: Nauk. zap. L'vivs'k. un-tu, 1955, 34, 125-133

Abstract: Zr is determined in its salts by titration with NaOH in the presence of salts of tartaric, citric and salicylic acid (concentration of solutions of Zr  $\geq$  0.1 N) or without them (concentration of solutions  $\leq$  0.025 N). With an excess of anions of hydroxy acids even large amount of  $KNO_3$  and  $K_2SO_4$  are without effect; cations reacting with caustic alkalies interfere. In basic salts Zr is determined from the sum of basicity of solutions of salt and amount of caustic alkali used up for its neutralization, the basicity being determined on the basis of the reaction  $ZrO^{2+} + 6F^- + H_2O \rightarrow (ZrF_6)^{2-} + 2OH^-$ . In neutral tartrate solutions Zr cannot be determined by the

Card 1/2

USSR/Analytical Chemistry -- Analysis of Inorganic Substances, 4-2

Abst Journal: Referat Zhur -- Khimiya, No 19, 1956, 61830

Abstract: potassium fluoride method, in salicylate solutions the error of determination  $\approx 0.8\%$ . Ti cannot be determined by the above described method since the fluoride complex of Zr are more stable toward alkalies than the fluoride complex of Ti.

Card 2/2

ZOLOTUKHIN, V. K.

USSR/Analytical Chemistry - Analysis of Inorganic Substances, G-2

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61880

Author: Zolotukhin, V. K.

Institution: None

Title: Determination of Free Acid in Thorium Salts

Original

Periodical: Nauk. zap. L'vivs'k. un-tu, 1955, 34, 138-140

Abstract: Free acid in solutions of the salts cannot be determined directly by titration with alkali due to the low pH of precipitation of basic salts of Th and  $\text{Th}(\text{OH})_4$ .  $\text{Th}^{4+}$  is combined with KF or NaF (4-fold excess) or 1 N solution of oxalate of K or Na (10-fold excess) and free acid is titrated with 0.1 N NaOH to phenol red or phenolphthalein. Basic salts of Th are detected by means of their reaction with fluorides or oxalates of alkali metals, as a result of which alkali is liberated which is revealed by suitable indicators.

Card 1/1

ZOLOTUKHIN, V.K.; MOLOTKOVA, A.S.

Photometric and colorimetric determination of thallium in the  
form of thallic hydroxide. Zhur.anal.khim. 11 no.2:248-249  
Mr-Ap '56. (MLBA 9:8)

1. L'vovskiy gosudarstvennyy universitet.  
(Thallium)

ZOLOTUKHIN, Y.K.

Salicylate compounds of beryllium. Dop. ta pov. L'viv. un.  
no.7 pt. 3:209-213 '57. (MIRA 11:2)  
(Beryllium compounds)  
(Salicylic acid)

ZOLOTUKHIN, V.K.

Scientific technical conferences at the Kharkov Aeronautics  
Institute. Izv.vys.ucheb.zav.;av.tekh. no.4:123-127 '58.  
(MEBA 11:12)

1. Khar'kovskiy aviatsionnyy institut.  
(Kharkov—Aeronautics)

ZOIOTUKHIN, V.K.

Effect of the anion and cation composition of beryllium containing  
mixtures on the reaction of beryllium ions with hydroxyl ions.  
Nauk.sop.L'viv.un. 46:124-132 '58. (MIRA 12:7)  
(Beryllium) (Hydroxyl group)



ZOLOTUKHIN, V.K.; KOMLEV, O.I.; GALANETS, Z.G. [Halane'ts', Z.H.]

Investigation of the tartaric acid compounds of copper and  
cadmium. Nauk.zap.L'viv.un. 46:133-140 '58. (MIRA 12:7)  
(Tartaric acid) (Copper compounds) (Cadmium compounds)

MOLOTKOVA, G.S. [Molotkova, H.S.]; ZOLOTUKHIN, V.K.

Reaction of thallium trichloride with sodium tartrate. Nauk.  
zap.L'viv.un. 46:155-160 '58. (MIRA 12:7)  
(Thallium chlorides) (Sodium tartrate)

S/147/59/000/04/020/020  
E031/E413

AUTHOR: Zolotukhin, V.K.

TITLE: The Scientific-Technical Conference at Khar'kov  
Aviation Institute

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy, Aviatsionnaya  
tekhnika, 1959, Nr 4, pp 161-165 (USSR)

ABSTRACT: In May 1959, the 16th Conference of Professorial and Teaching Staff took place. At a plenary session the following reports were read: "The XXI Congress of the Communist Party of the Soviet Union on the Further Development of the Two Forms of Socialist Ownership" by N.N.Aleksandrov, Director of the Chair Marxism-Leninism; "The Contemporary State of Rocket Technology" by Docent, Candidate of Technical Sciences I.P.Goldavay; "Efforts to Produce the First Aircraft Wholly Manufactured in China" by Docent, Candidate of Technical Sciences S.I.Kuz'min. The work of the Conference continued in twelve sections. The following papers were read: Social Sciences Section. "Contemporary Bourgeois Philosophy" by Senior Instructor S.I.Epshteyn; "Discussion on Trade Unions in the

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S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute

Khar'kov Party Organization" by Assistant A.G.Levchuk;  
"The Solution of the Housing Problem Under Socialism"  
by Senior Instructor in the Department of Economical  
Sciences, V.A.Yakovenko; "The Final and Complete  
Victory of Socialism in the USSR" by Senior Instructor  
V.A.Kravets; "The Problems of Socialist Competition at  
the XII Congress of the Trade Unions of the USSR" by  
Assistant Yu.N.Doroshenko.

Foreign Languages Section. The following papers were read:

"Foreign Languages in the Life and Work of V.I.Lenin"  
by Docent, Candidate of Philological Sciences

G.G.Pochentsov; "The Organization and Work of the  
Departmental Section of Foreign Language Teachers at  
Colleges which are not Specifically Language Colleges"  
by Senior Instructor N.S.Shneye; "Work on Translations

at Higher Technical Colleges" by Assistant V.I.Kryzhenko;  
"On the Principles of Constructing a Handbook of Technical  
Texts - Educational Assistance for III-rd Course at  
Aviation Colleges" by Assistants A.M.Gurevich and

Card 2/11

L.A.Litovskaya. ✓

S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Kharkov Aviation Institute  
Mathematics and Mechanics Section. The following papers  
were read: "A Spectral Representation of the Theory  
of Axisymmetric Turbulence" by Candidate of Physical  
and Mathematical Sciences G.N.Taranova; "Some  
Evaluations for Functions with Positive Real Parts" by  
Assistant G.S.Shpak; "Existence, Uniqueness and  
Correctness Theorems for Mixed Systems of Functional  
Equations" by Docent, Candidate of Physical and  
Mathematical Sciences M.N.Tikhov; "On the Application  
of Bohl and Chebyshev Points to the Solution of Some  
Problems in the Synthesis of Four Bar Linkages" by  
Docent, Candidate of Physical and Mathematical Sciences  
Ya.L.Geronimus; "The Influence of the Structural  
Properties of Functions on the Convergence Almost  
Everywhere of their Conjugate Fourier Series" by  
Docent, Candidate of Physical and Mathematical Sciences  
B.L.Golinskiy.

General Technological Section. The following papers were  
read: "The Relation Between the Compton Length of Waves,  
the Length of de Broglie Waves and the Acceleration ✓

Card 3/11

S/147/59/000/04/020/020  
EO31/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute

Potential for High Energy Particles" by Docent,  
Candidate of Physical and Mathematical Sciences  
I.Ya.Mints; "The Problem of Determining the Heat  
Transfer Coefficient of Conductors" by Senior Instructor  
P.P.Bezuglov; "An Electron-Graphical Method of  
Investigating the Structure of Matter" by Assistant  
I.Ya.Surovtsev; "On the Results of the VIII th  
Mendeleev Congress of Chemists of the USSR" by  
Docent, Candidate of Chemical Sciences E.I.Krech.  
Electrical and Radio Technology Section. The following  
papers were read; "On the Problem of the Optimum  
Passage of Transients in an Electric Drive with a  
Controlling Exciter" by Docent, Candidate of Technical  
Sciences M.M.Perel'muter; "The Experimental Determination  
of the Reactances in Synchronous Machines" by Senior  
Instructor S.V.Khmel'nitskiy; "An Experimental Method  
of Investigating Electric Fields" by Assistant  
A.I.Lopatin; "A Discrete Transformer of Current into  
Code Signals with Magneto-Electric Comparison Units" by  
Docent, Candidate of Technical Sciences G.M.Butayev; ✓

Card 4/11

S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute

"The Application of Infrared Instruments in Aviation"  
by Docent, Candidate of Technical Sciences I.D.Artamonov.  
General Engineering Section.

"The Adaptation of a Thermobaric Chamber to the  
Simulation of the Sinking of a Mine Shaft in Quicksand  
and Certain Results of Investigations to Determine the  
Mechanical Characteristics of Sand at Different  
Temperatures and Humidities" by Docent, Candidate of  
Technical Sciences S.V.Blyashenko; "Friction and  
Abrasion in Cermets" by Docent, Candidate of Technical  
Sciences O.I.Goldayeva; "The Construction of Multi-  
Satellite Planetary Gears" (by Assistant V.A.Tkachenko;  
"The Influence of Work Hardening on the Fatigue of  
Threaded Connections" by Assistant V.M.Rydchenko;  
"Investigation of Cermets Slide Bearings" by Assistant  
A.S.Efoyan.

Strength of Aircraft Section.

"On the Theory of Bending of Thin-Walled Columns" by  
Docent, Candidate of Technical Sciences L.P.Vinokurov;

"The Simulation of Static Experiments on Thin-Walled

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S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute

Structures" by Candidate of Technical Sciences  
L.A.Kolesnikov and Senior Instructor V.K.Zolotukhin;

"The Bending of Beams Framing an Opening" by

Candidate of Technical Sciences L.A.Kolesnikov;

"The Influence of the Rigidity of Ribs and Beams on  
their Bending" by Assistant N.A.Shelomov;

"The Calculation of the Bending of Rectangular Plates by  
the Discrete Method" by Assistant Yu.F.Petrov;

"The Calculation of Cylindrical Shells" by the Method  
of Discrete Variables" by Aspirant N.I.Gur'yev.

Engine Construction Technology Section.

"The Choice of a Scheme for a Hydraulic Servo-System  
for the Automation of Welding Processes" by Assistant

V.V.Balatskiy; "An Investigation of the Process of  
Polishing by an Abrasive Belt" by Senior Instructor,

Candidate of Technical Sciences V.N.Verezub; "The

Investigation of the Operation of a Pneumatic-  
Hydraulic Plant" by Assistant V.I.Basteyev;

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S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute

"A Static Analysis and Calculation of the Accuracy of the Technological Processes of Machining" by O.M.Parkhomenko; "The Automatic Welding of Long Panels" by Candidate of Technical Sciences L.F.Kamukov; "Prospects in the Use of Specialised Computers for the Determination of the Optimum Geometry of Cutting Tools" by Docent, Candidate of Technical Sciences V.P.Kosharnovskiy; "The Spreading of the Experience of Innovators and the Classification of Organizational-Technical Measures in Machine Construction" by Senior Instructor M.M.Apanovich; "Features of Measurable Abrasion of a Cutting Tool in Fine Sharpening" by Assistant V.N.Malikov; "An Investigation of the Process of Compression at High Velocities of Deformation" by Docent, Candidate of Technical Sciences A.K.Bayev; "The Standardization of Vibration Effects" on the Human Organism in Aircraft Production" by Senior Instructor V.D.Ivanov.  
Theory and Construction of Aircraft Engines and Propellor-Driven Machines Section. "The Investigation" ✓

Card 7/11

S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute

of the Flow Between the Inlet and Outlet Valves of a Turbine" by Instructor, Candidate of Technical Sciences V.N.Yershov; "The Variation in the Stage Parameters of an Axial Compressor in Accordance with the Size of the Radial Clearance" by Assistant A.N.Anyutin; "On the Problem of Non-Stationary Heat Transfer" by Assistant S.D.Frolov; "The Influence of an Electric Field on the Flame of a Burner" by Senior Engineer P.P.Kostenko; "Calculation of the Temperature Compensation of Capacitance Pressure Pick-Ups" by Assistant L.Ya.Astaf'yev.  
Aerohydrodynamics Section.

"Ideal Hypersonic Flow Round a Body" by Assistant V.I.Kholyavko; "The Control of the Boundary Layer of a Wing by Perforation of the Leading Edge" by Assistant Ye.P.Vachasov; "The Gas-Hydraulic Analogy and its Application" by Senior Instructor D.A.Munshtukov; "The Aerodynamic Investigation of Wing Profiles for Small Reynolds Numbers" by Engineer Yu.F.Usik.  
The Technology of Aircraft Construction and Metal Working Section. "A New Model of the Plasticity of Metals" by ✓

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S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute

Instructor, Candidate of Technical Sciences  
Yu.N.Alekseyev; "The Forging Extrusion<sup>8</sup> of Large  
Components from Sheet Metal" by Aspirant A.P.Barsukov;  
"On the Problem of Constructing Second Order Curves in  
Aircraft Construction" by Senior Instructor  
M.A.Zaydenverg; "The Electric Contact Welding<sup>8</sup> of Thin  
Pieces of Metal" by Assistant N.M.Tarasov; "The Influence  
of Plastic Deformation on the Properties of Austenitic  
Stainless Steel at Various Temperatures" by Assistant  
N.V.Pisareva; "The Deformation of Non-Ferrous Metals  
and Alloys at Low Temperatures" by Assistant  
N.N.Lyulicheva; "The Investigation of Phase Changes in  
Austenitic Steels Previously Deformed at Below Freezing  
Point Temperatures" by Candidate of Technical Sciences  
A.N.Chukhleb and Aspirant V.P.Martynov; "The Influence  
of the Temperature and Velocity of Deformation on the  
Phase Changes of Austenitic Steels" by Candidate of  
Technical Sciences A.N.Chukhleb and Fellow V.P.Martynov;  
"The Determination of Optimum Technical Grouping in the  
Design and Production of Aircraft" by Assistant

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S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute

Yu.A.Boborykin; "On the Use of Explosives in the Technology of Drop Forging" by Assistant K.I.Zaytsev; "Welding by Friction" by Assistant N.P.Ostrovskiy.  
Structure of Aircraft Section.

"On the Problem of Protecting the Structure of Aircraft from Aerodynamic Heating" by Docent F.V.Dybskiy;

"Passive Methods of Protection from Aerodynamic Heating" by Candidate of Technical Sciences F.G.Yasinskiy;

"The Influence of the Parameters of a Thermally Isolated Packet on Heat Transfer Characteristics" by Assistant

A.A.Kobylyanskiy; "Aircraft Structures Made from

Fibreglass Sheets" by Docent, Candidate of Technical Sciences S.I.Kuz'min; "An Apparatus for Investigating

Repeated Static Loading and High Temperatures" by Assistant L.A.Malashenko; "The Approximate Calculation of the Weight Taking into Account the Technical Features

of the Aircraft Structure" by Candidate of Technical Sciences L.D.Arson; "The Determination of Stresses in a Shell as a Result of Riveting" by Assistant

Card 10/11 Yu.G.Fursa; "The Ultrasonic Altimeter (Sounding Device)" ✓

S/147/59/000/04/020/020  
E031/E413

The Scientific-Technical Conference at Khar'kov Aviation Institute  
and "The Radio-Control and Autopilot of an Experimental  
Model" by Engineer L.F.Teplov. ✓

SUBMITTED: June 27, 1959

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S/147/60/000/004/016/016  
E073/E335

AUTHOR: Zolotukhin, V.K.

TITLE: Scientific-technical Conference at the Khark'khov  
Aviation Institute

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy,  
Aviatsionnaya tekhnika, 1960, No. 4, pp. 155--157

TEXT: The Seventeenth Conference of the Professional-  
lecturing Personnel of Khar'kovskiy aviatsionnyy institut  
(Khar'kov Aviation Institute) was held in April, 1960.  
81 papers were read and discussed. At the plenary meetings  
the following papers were read:

Docent, Candidate of Historical Sciences N.A. Bystrov -  
"International Importance of Leninism";

Docent, Candidate of Technical Sciences V.G. Kononenko -  
"Explosive Forming of Metals";

Candidate of Military Sciences A.I. Chernovolenko -  
"V.I. Lenin - In Defence of the Socialist System".

The work of the conference proceeded in eleven sections.

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S/147/60/000/004/016/016  
E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Section on Mathematics and Mechanics:

Professor, Doctor of Physicomathematical Sciences

Ya.L. Geronimus - "Some Properties of the Focal Curve",

"Evaluation for Coefficients of Limited Functions";

Candidate of Physicomathematical Sciences B.L. Golinskiy -

"On approximating Two Conjugate Functions by Whole Functions  
of the Exponential Type";

Candidate of Physicomathematical Sciences G.M. Taranova -

"Hydrodynamic Equations in Stresses";

Professor, Doctor of Physicomathematical Sciences

A.D. Myshkis - "On the Stability of Motion in the Presence of  
Jerks";

Candidate of Physicomathematical Sciences Ya.I. Zhitomirskiy -

"The Cauchy Problem for Equations with Variable Coefficients

Which Are Concrete in Accordance With the Definition of

I.G. Petrovskiy";

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E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Candidate of Physicomathematical Sciences L.I. Ronkin -

"On Quasi-analytical Classes of Functions";

Docent, Candidate of Physicomathematical Sciences M.N. Tikhov

- "On Solving Certain Nonlinear Integral Equations";

Docent, Candidate of Physicomathematical Sciences W.S. Shun -

"Approximation of Functions and Their Derivatives";

Docent, Candidate of Physicomathematical Sciences I.Ya. Mints -

"Relativistic Correction to the  $3/2$  Law".

General Technical Section:

Docent, Candidate of Chemical Sciences E.I. Krech - "On the  
Modern Definition of "Atom"";

Postgraduate G.G. Lomakina - "Spectrophotometric Investigation  
of Complex Compounds of Elements of the Second Group of the  
Periodic System of D.I. Mendeleev";

Candidate of Physicomathematical Sciences N.A. Khizhnyak -

"Electrodynamic Methods of Acceleration of Plasmoids";

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S/147/60/000/004/016/016  
E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Postgraduate Shapovalova, L.Ya - "On Obtaining Metallic Powders  
of Fe and Cu From Mixed Oxylates Fe and Cu";

Postgraduate L.Ya. Kurilenko - "Study of the Interaction of  
Aluminium with Carbontetrachloride";

Senior Lecturer P.P. Bezuglov - "Determination of the Blackness  
Coefficient of Thin Steel Sheets";

Postgraduate L.Ya. Kheyfets - "Radioactive Precipitates and  
Radiochemistry".

Electro-Radio Engineering Section:

Docent, Candidate of Technical Sciences M.M. Ferel'muter -  
"Methods of Determining the Volumetric Quantities of Material";

Docent, Candidate of Technical Sciences I.D. Artamonov -  
"Development of Illumination System/Runways";

Senior Lecturer S.V. Khmel'nitskiy - "Investigation of the  
Temperature Rise of Synchronous Machines";

Docent, Candidate of Technical Sciences G.M. Butayev -  
"On a Circuit for Transforming Coded Signals";

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S/147/60/000/004/016/016  
E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Docent, Candidate of Physicomathematical Sciences P.V. Bliokh -  
"Parametric Resonance in a System of Coupled Circuits";

Candidate of Physicomathematical Sciences F.G. Bass -

"Determination of the r.m.s. Permeability of a Medium with  
Random Nonuniformities";

Postgraduate M.A. Savchenko - "Thermal Radiation of Magnetised  
Ferrites";

Postgraduate Yu.G. Yagola - "Application of Galvanomagnetic  
Effects for Measuring the Potential of Magnetic Fields".

Section on Strength of Aircraft:

Postgraduate N.A. Shelomov - "Experimental Investigation of  
the Stress States of Thin-walled Rods";

Docent, Candidate of Technical Sciences S.V. Blyashenko -

"Investigation of the Strength and the physicomathematical  
Properties of a glacier-ice cylinder from rocks under  
Creep Conditions Subject to the Effect of External Hydrostatic  
Pressure";

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E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Candidate of Technical Sciences L.A. Kolesnikov - "Bending  
of Longitudinal Elements Bounding a Free Edge of Open Thin-  
walled Lattice Rods. ✓

Section on Engine Manufacturing Technology:

Candidate of Technical Sciences V.A. Zemlyanskiy and  
Engineer Yu.F. Granin - "Kinematics of the Process of Cutting  
with Circular Self-rotating Cutting Tools";

Engineer Yu.F. Granin - "Machineability of the Steel WX-15  
(ShKh-15) Using Circular Self-rotating Cutters";

Docent, Candidate of Technical Sciences A.I. Nikishov -  
"Some Problems of Milling Gears";

Docent, Candidate of Technical Sciences V.P. Kosharnovskiy -  
"On Formulating the Problem of Science of Metal-cutting" and  
"Basic Assumptions of the Theory of Working Out an Optimum  
Geometry of Cutting Tools";

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S/147/60/000/004/016/016  
E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Senior Lecturer M.M. Apanovich - "Certain Problems of  
Planning of Organisation-technical Measures";

Candidate of Technical Sciences O.M. Parkhomenko - "Statistical  
Analysis and Norms for Calculating the Accuracy of Machining in  
Engine Manufacture".

Section on the Theory and Design of Aviation Engines and  
Bladed Machinery:

Senior Lecturer A.P. Pershin - "On Experimental Investigation  
of the Thermal Effectiveness of Gas Flows with Very High Thermal  
Parameters";

Postgraduate G.V. Pavlenko - "Rotating Discontinuity in the  
Elementary Stage of an Axial Compressor";

Postgraduate A.N. Anyutin - "Influence of the Radial Gap on the  
Flow Past Blades of the Individual Stages of an Axial  
Compressor";

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S/147/60/000/004/016/016  
E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Senior Lecturer D.A. Munshtukov - "Application of Gas-  
hydraulic Analogy for Solving Nonsteady-state Problems in  
Gas Dynamics";

Senior Lecturer I.V. Aslanov - "New Type of Gas Expanders";  
Postgraduate D.F. Simbirskiy - "On Measuring Pulsating  
Parameters";

Docent, Candidate of Technical Sciences N.T. Ozhgikhin -  
"Comparative Evaluation of Air-breathing Jet Engines";

Postgraduate S.K. Zaika - "Two-circuit Pulse Jet Engines.  
Experimental Results".

Section on Aerohydrodynamics:

Senior Lecturer O.V. Florinskiy - "Jet Wing";

Postgraduate V.I. Kholyavko - "On Calculating Aerodynamic  
Characteristics of a Thin Profile" and "Approximate method,  
position and the shape of an outgoing shock wave in the Case  
of Symmetric Flow Past Bodies of Simple Configuration";

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S/147/60/000/004/016/016  
E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Engineer L.P. Koval'skiy - "Influence of Elastic wing deformations  
on its Aerodynamic Characteristics";

Postgraduate Ye.P. Vachasov - "Influence of the Suction of the  
Boundary Layer on the Aerodynamic Characteristic of a Thin Wing";

Docent, Candidate of Technical Sciences Ya.Ye. Tkachenko -  
"Boundary-layer Control" (Review paper).

Section on Aircraft Technology of Metals:

Docent, Candidate of Technical Sciences V.G. Kononenko -  
"Explosive loss-free Cutting of Hot Metal in Metallurgy";

Docent, Candidate of Technical Sciences D.A. Lyukevich -  
"Contact Welding in Aircraft Construction";

Postgraduate Yu.A. Boborykin - "Selection of Optimum Assembly  
of the "Glider" of an Aircraft";

Postgraduate K.I. Zaytsev - "Investigation of the Process of  
Explosive Punching of Holes";

Postgraduate V.L. Karpin - "On Applying Plastics for Assembly  
Work and Store Equipment";

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S/147/60/000/004/016/016  
E073/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Docent, Candidate of Technical Sciences Yu.N. Alekseyev -  
"Plastic Flow of Metals";

Candidate of Technical Sciences V.Ye. Semenov - "On the  
Possibility of Separate Manufacture of Dies and Punches for  
Slotting and Hole-punching";

Senior Lecturer A.N. Zaytsev - "Explosive Stamping of  
Components from Sheet"

Postgraduate N.M. Tarasov - "Electrocontact-welding of metals of  
small thicknesses";

Postgraduate L.Ya. Astaf'yev - "Thermal Machine for Slab-cutting";

Postgraduate V.N. Malikov - "Fine Turning with Large Rates of  
Feed".

Section on Aircraft Construction:

Candidate of Technical Sciences F.G. Yasinskiy - "On Planning  
Conditions";

Postgraduate V.N. Romanenko - "On the Selection of the Parameters  
of the Control Process";

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EO73/E335

Scientific-technical Conference at the Khar'kov Aviation  
Institute

Senior Engineer L.F. Teplov - "Apparatus for Proportional  
Control";

Postgraduate V.N. Revinov - "On Calculating the Take-off of  
a Pilotless Glider with a Damper";

Docent N.V. Dybskiy, - "Protective Sheathing";

Candidate of Technical Sciences F.G. Yasinskiy - "Solution  
of One Problem of Heat Conductivity";

Candidate of Technical Sciences L.D. Arson - "Tensile  
Strength Taking Into Consideration Technological Factors";

Docent, Candidate of Technical Sciences S.I. Kuz'min -  
"Application of SVAM in Aircraft Design";

Senior Lecturer B.A. Cherapennikov - "Behaviour of Metals  
in the Case of Impact Loads".

Card 11/11



69023

5.26.20  
AUTHORS:

Zolotukhin, V. K., Molotkova, A. S.

S/07B/60/005/04/018/040  
B004/B007

TITLE:

The Relative Stability of the Tartrate-, Citrate-, Malate-, and Salicylate Complexes of Iron and Aluminum

PERIODICAL:

Zhurnal neorganicheskoy khimii, 1960, Vol 5, Nr 4, pp 879 - 881 (USSR)

ABSTRACT:

The authors mention the fact that published data on the constitution of complex salts of organic oxy acids are insufficient and contradictory. They chose the compounds mentioned in the title, because they play a part in analytical chemistry. Relative stability was determined according to I. V. Pyatnitskiy's method (Ref 2). A mixture of potassium-aluminum-alum and iron-ammonium-alum was treated with the sodium salt of the organic acid, after which Al and Fe were precipitated - partly after addition of NaOH - with oxyquinoline. In the solution of the organic acid the non-precipitated Fe and Al, the degree of binding  $\alpha = \frac{c_{\text{complex}}}{c_{\text{Me}^{n+}}}$  as well as the ratio  $\frac{\alpha_{\text{Fe}}}{\alpha_{\text{Al}}}$  was determined.

The results obtained are given in a table. The tartrate-, citrate-, malate-, and salicylate-complexes of iron are more stable than the corresponding complexes of aluminum. The ratio

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The Relative Stability of the Tartrate-, Citrate-,  
Malate-, and Salicylate Complexes of Iron and  
Aluminum

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B004/B007

$\alpha_{Fe}/\alpha_{Al}$  of the degrees of binding of the complexes of these oxy-  
acids increases with the pH of the solution. The stability of  
the investigated complex compounds of Fe and Al is in a reci-  
procal ratio to each pH, at which the precipitation of Fe and  
Al as hydroxide begins. There are 1 table and 7 Soviet ref-  
erences. 4

ASSOCIATION: L'vovskiy gosudarstvennyy universitet (L'vov State University)

SUBMITTED: January 30, 1959

Card 2/2

ZOLOTOUKHIN, V.K.

Solubility of cadmium, lead, and zinc hydroxides in sodium tartrate solutions, and that of aluminum hydroxide in sodium citrate solutions. Zhur. neorg. khim. 5 no.8:1886-1888 Ag '60.

(MIRA 13:9)

1. L'vovskiy gosudarstvennyy universitet.

(Cadmium hydroxide)

(Lead hydroxide)

(Zinc hydroxide)

(Aluminum hydroxide)

S/073/60/026/004/014/018/XX  
B023/B064

AUTHORS: Zolotukhin, V. K. and Oshchapovskiy, V. V.

TITLE: The Interaction of Trivalent Cerium With the Salts of Tartaric Acid

PERIODICAL: Ukrainskiy khimicheskiy zhurnal, 1960, Vol. 26, No. 4, pp. 510-513

TEXT: The interaction of trivalent cerium with the salts of tartaric acid has, in the authors' opinion, hitherto been insufficiently investigated (Refs. 1, 2). This paper describes the investigation of cerium tartrate compounds formed in weakly acid or weakly alkaline medium, as well as their acidic and basic properties and their resistance to several reagents and to heat. It was found that an addition of alcohol to the cerium tartrate solutions causes precipitation. The salt has the following composition:  $\text{Ce}_2(\text{C}_4\text{H}_4\text{O}_6)_3 \cdot 2\text{H}_2\text{O}$  with impurities of tartrate compounds containing sodium. Mixtures of cerium nitrate and sodium tartrate were titrated either potentiometrically or with phenol phthalein as indicator. In these investigations, the amount of  $\text{Na}_2\text{C}_4\text{H}_4\text{O}_6$  was up to 25 and more moles per 1 mole

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