

L 2095-66 EWT(m)

ACCESSION NR: AP5019038

UR/0286/65/000/012/0069/0069
69.057,52810
BAUTHOR: Vorob'yev, A. I.; Ivanovskiy, G. V.; Komarov, A. K.; Tsikhona, V. A.;
Sandomirskiy, G. B.; Rubinshteyn, G. V.TITLE: A device for preparing concrete forms. Class 37, No. 172020¹⁵

SOURCE: Byulleten' izobretений i tovarnykh znakov, no. 12, 1965, 69

TOPIC TAGS: concrete structure, concrete, structural concrete, construction method

ABSTRACT: This Author's Certificate introduces a device for preparing concrete forms. The device is used when the blocks which make up a structure are being joined into a monolithic unit. The apparatus includes a panel which covers the joint, and a clamping attachment. Assembly and disassembly are simplified by making the clamping attachment in the form of a support and pneumatic tubes. The tubes are located between the support and the panel and are drawn together by rods. During setup, the free ends of the rods are connected with support girders located on the other side of the joint. These support girders remain in the structure after the blocks are joined into a single monolithic unit.

Card 1/3

L 20995-66

ACCESSION NR: AP5019038

ASSOCIATION: none

SUBMITTED: 07May63

ENCL: 01

SUB CODE: GO

NO REF SOV: 000

OTHER: .000

Card 2/3

L 20995-66

ACCESSION NR. AP5019038

ENCLOSURE: 01

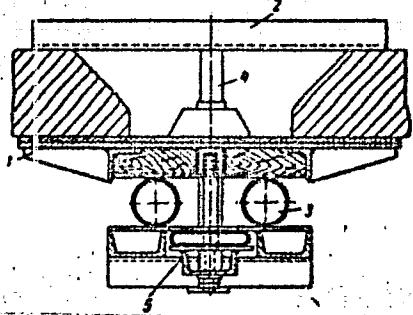


Fig. 1. 1--panel; 2--support;
3--pneumatic tube; 4--rod;
5--support girder

Card 3/3 BK

YUDOVICH, V.G.; KHLEBORODOV, A.D.; SOLONEVICH, Ye.A.; VEYTS, V.L.;
PANOV, P.S.; BELYAYEV, A.N.; ALAD'IN, O.I.; OSIPOV, V.F.;
VOROB'YEV, A.I.; PROKOF'YEV, Yu.V.; SOLOV'YEV, Yu.K.;
KUZ'MIN, A.V.; ZHIDONIS, V.Yu.; ZOLIN, A.V.; YATUN' YE.P.;
DQBROSLAVSKIY, V.L.; TRÓFIMOV, Ye.N.; DRYAGIN, Ye.R.;
KOROLEV, V.F.; KERIMOV, N.B.; KRAVCHENKO, A.S.; RYVLIN, V.A.;
GURCHENKO, A.P.; KRUGLIKOV, T.P.; CHERNYAKOV, F.A.; ARKHIPOV,
N.K.

Authors' certificates and patents. Mashinostroenie no.1:101-
103 Ja-F '65. (MIRA 18:4)

VOROB'YEV, A.I.

Problems of the clonic theory of leukemia. Probl. gemat. i
perel. krovi. no.2:14-22 '65.

1. 3-ya kafedra terapii (zav. - deystvitel'nyy chlen AMN
SSSR prof. I.A.Kassirskiy) TSentral'nogo instituta
usovershenstvovaniya vrachey, Moskva.
(MIRA 18:11)

VOROB'YEV, A.I., inzh.

Eddy currents in the high-speed detecting of rail defects. Vest.
TSNII MPS 24 no.3:40-41 '65.

(MIRA 18:8)

1. Novosibirskiy institut inzhenerov zheleznodorozhnego transporta.

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860820004-2

VOROBIEV, A. I.; KASSIRSKIY, I. A.;

"Le probleme de la remission dans la leucose aigue."

Report presented at the joint meeting of the European Society of Hematology
and the International Society of Blood Transfusion, Lisbon, Portugal, 26-31 Aug 63.

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860820004-2"

VOROB'YEV, A.Kh.

Incubating the eggs of waterfowl at the Shklov Hatchery.
Ptitsevodstvo 9 no.2:18-19 P '59. (MIRA 12:3)

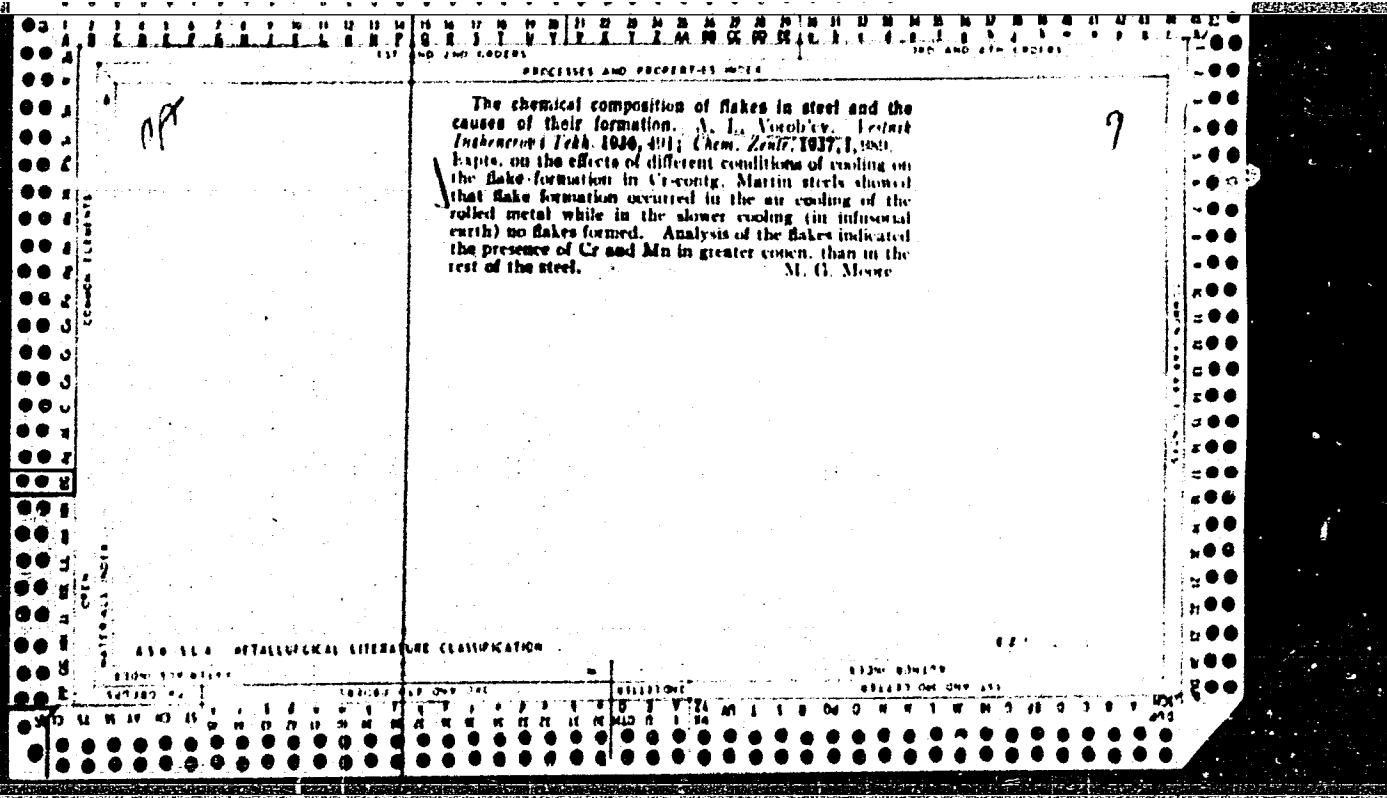
1, Zaveduyushchiy tsekhom inkubatsii Shklovskoy inkubatorno-
ptitsevodcheskoy stantsiyey, Mogilevskoy oblasti, Belorusskoy
SSR.
(Shklov--Incubation)

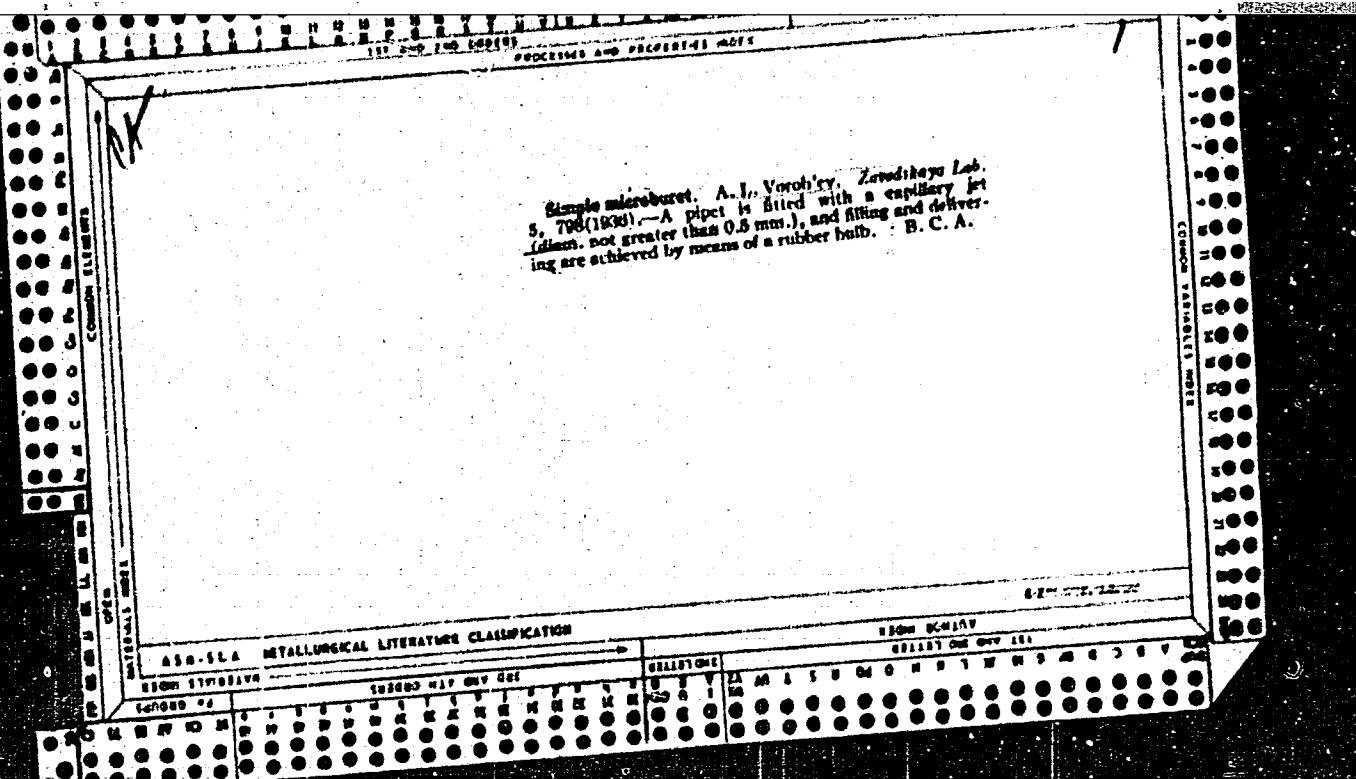
VOROB'YEV, A.L.
CA

Volumetric determination of nickel in steels. A. I
Vorob'yev, Zavodskaya Lab. 4, 11051955. NaD15
The titration of Ni by the method of Oelband (C. A.
29, 789) is prep'd. by dehydrating Na₂NO₃, 12H₂O at 200°
hr. Chas. Blanc

7

140-114 METALLURGICAL LITERATURE CLASSIFICATION





Simplified apparatus for electrolysis with a mercury
electrode. A. L. Vorob'ev. Zal'mskys Lab. 6, N.I.T.
(1937).—App. 1. RECORDED.

CA

8

Dumortierite in Gay (northern Fergana). V. I.
Pupov and A. I. Virul'ev. *Karshtona Nade* 11, No. 2-3,
97-98(1940). The sample of dumortierite contained SiO₂
41.60, TiO₂ 0.4, Al₂O₃ 10.44, FeO₃ 0.00, FeO 0.06, MgO
traces, BaO₃ 4.19, CaO 0.6, K₂O + Na₂O 0.17, and H₂O
0.10-2.24%. A. A. Hochtingk

21

Vanadium and nickel in coals of the Upper Silurian of the Altai and Turkestan Mountains. A. I. Vorob'ev. (compt. rend. acad. sci. U. R. S. S. 28, 237-7 (1930) (in German).

The ash of coals from 7 samples was analyzed for NiO and V₂O₅. Ash and S were also detd. and found to range from 0.23 to 27.77% for ash and 0.78 to 1.69% for S. All of the ash samples showed V₂O₅ from 0.01 to 1.30% but only 3 showed NiO from 0.00 to 1.30%. These results show that the Paleozoic coals of middle Asia have higher V and Ni content than the Mesozoic coals. Various theories of the presence of the V and Ni are mentioned.

Oden E. Sheppard

ABE-166 METALLURGICAL LITERATURE CLASSIFICATION

BROWN BOUNDARY

BBB000 MAP ONE ONE

BROWN BOUNDARY

BBB157 ONE ONE 151

CH

8

Hemihydrate in the desert deposits of Middle Asia.
V. I. Popov and A. I. Voznesenskii, Zapiski Vsesozi. Mineral. Obshchinstva (Mem. soc. russ. mineral.) 121: 76, No. 70 (1947). - $\text{CaSO}_4 \cdot \frac{1}{2}\text{H}_2\text{O}$ was first discovered as an independent mineral in the salt soil deposits of Central Asia. The high desert temp. in the presence of highly concentrated, salt solns, often brings about a weathering of gypsum to a white, powd. product, usually described as anhydrite, but which differs distinctly in its optical properties. These deposits are known in Eastern Turkmenia in the S mine of Gurdak, in the Bardanul Gorge, Uigur-Cayu, Margusar-Cayu of Northern and West Fergana. Pseudomorphs after gypsum crystals are observed. In the oil field of the Changyatah anticline (Southern Fergana), hemihydrate occurs in thin layers in the oil sands. The γ are near 1.893-1.898 while anhydrite has $\gamma = 1.814$; $n = 1.670$; medium birefringence. Analysis: SO_3 54.80; Fe_2O_3 , trace; H_2O 0.78; MgO 4.72; CaO 39.79; MgO 0.35%. The mineral sets with water like plaster of Paris, but there is a great difference of the rate of hydration between the powd. hemihydrate and the same in compact crystals which are only superficially hydrated and remain for a long time in the open air. The numerous observations of anhydrite in Central Asia indicate that often the conditions for the formation of hemihydrate must have existed in such salt deposits. W. Kitel

ASA 1A METALLURGICAL LITERATURE CLASSIFICATION

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860820004-2

VOROB'YEV A-

APPROVED FOR RELEASE: 03/14/2001

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VOROB'YEV, A. L.

Chromium-bearing sedimentary formations in Central Asia. Trudy Inst.
geol. AN Uz.SSR no. 9:218-219 '53. (MIRA 12:1)
(Soviet Central Asia--Chromium)

VOROB'YEV, A.L.

Distribution of minor elements in soils of cotton growing areas
of Surkhan-Dar'ya and Kashka-Dar'ya Provinces (geological premises).
Trudy SAGU no.39:47-53 '53. (MLRA 10:5)
(Surkhan-Dar'ya Province--Trace elements)
(Kashka-Dar'ya Province--Trace elements)

~~VOROB'YEV A.I.~~

Accessory ore minerals of igneous rocks. Trudy SAGU no.39:55-56 '53.
(MLRA 10:5)
(Rocks, Igneous)

VOROB'YEV, A. I., AND BADALOV, S. T.

Data on Hemihydrate

The authors distinguish the following genetic types of deposits of hemihydrate: hemihydrates of sedimentary origin (in marine and lake-brackish deposits); hemihydrates formed in soil under desert conditions in gypsum-thennardites, carbonate, and halite saliniferous crustations; hemihydrates formed during the dehydration of gypsum on the surface yields or under the action of sulfuric acid in sulfur deposits; hemihydrates formed during hydration of sedimentary or hydrothermal anhydrite; hemihydrates as intermediate variety in the transition of gypsum into anhydrite at great depths. Deciphering of the Debye-grams and thermograms of the hemihydrates indicate the presence in them of admixtures of gypsum and anhydrite. (RZhGeol, No. 5, 1955) Tr. Sredneaz. un-ta. Geol. na. bk. 5, 1954, 29-34.

SO: Sum. No. 744, 8 Dec 55 - Supplementary Survey of Soviet Scientific Abstracts (17)

V. P. O. B'YEV, A. L.

USSR/Cosmochemistry - Geochemistry. Hydrochemistry, D

Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 741

Author: Popov, V. I., and Vorob'yev, A. L. [redacted]

Institution: None

Title: Concerning Some Mineralogical and Geochemical Peculiarities of Arid
Continental Formations

Original

Periodical: Zap. Uzbekist. otd. Vses. mineralog. o-va, 1955, No 8, 231-239

Abstract: A survey with a bibliography of 65 items.

Card 1/1

15-1957-10-14164

Translation from: Referativnyy zhurnal, Geologiya, 1957, Nr 10,
p 127

AUTHOR: Vorob'yev, A. L.

TITLE: The Value of Investigating the Isotopic Composition of
Minerals (Znacheniye issledovaniya izotopnogo sostava
mineralov)

PERIODICAL: Zap. Uzbekist. otd. Vses. mineralog. o-va, 1956, Nr 10,
pp 69-72

ABSTRACT: The method, based on determining the isotopic composition of minerals, furnishes objective criteria for defining the nature of many of the geological processes which occurred in the earth's crust in the most remote periods of its development. The determination of the content of Pb and He in ancient geological objects (minerals and rocks) makes it possible to ascertain their absolute age; and the determination of the content of the radiogenic isotope of carbon, C¹⁴, leads to age determinations of younger geological formations and arche-

Card 1/2

15-1957-10-14164

The Value of Investigating the Isotopic Composition of Minerals

ological discoveries. The determination of the isotopic composition of the non-radiogenic elements O₂ and H₂ in water leads, above all, to clarification of the problem of the origin of the water, the area of supply, etc. Study of the isotopic composition of C in the carbonate rocks of the Precambrian may aid in solving the problem of the possible existence of life during Precambrian time. The isotope C¹³ is more highly concentrated in chemically precipitated limestones than it is in limestones of organic origin. It has been found that the ratio C¹²:C¹³ in Precambrian carbonate rocks is similar to that in organic limestones. The origin of iron-ore skarns may be solved by study of the isotopic composition of oxygen in magnetite, inasmuch as magnetites which form by metasomatic replacement of carbonate rocks contain a considerably higher quantity of O¹⁸ than do iron ores of primary-sedimentary origin, in which the isotopic ratio of oxygen is approximately normal.

Card 2/2

K. N. Ryabicheva

BULANOV¹, I.D.; VOROB'YEV, A.M.

Extraction of protactinium from hydrochloric solutions by tributyl phosphate. Radichimija 6 no. 5:621-623 '64.

(MIRA 18:1)

Extraction of protactinium from hydrochloric solutions by methyl isobutyl ketone. Ibid. 6:623-626

KLYUYEV, Yu.A.; VOROB'YEV, A.M.

Study of the 4197 cm^{-1} absorption band of potassium ferricy ride
under high pressure. Dokl. AN SSSR 158 no.6;1996-1998 O '64.
(MIRA 17:12)

I. Institut fiziki vysokikh davlenii AN SSSR. Predstavлено
академиком А.Н. Терениным.

VOROB'YEV, A.M.; FOKICHEVA, V.I.

Analytical determination of americium, plutonium, and uranium
by means of the AMP anion exchanger. Radichimia 7 no.6:
728-729 '65. (MIRA 19:1)

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VOROB'IEV, A.M.

Irrigation by the use of long furrows. Gidr. i mel. 12 no.6;20-24
Je '60. (MIRA 13:7)

1. Groznenskaya optyno-meliorativnaya stantsiya.
(Checheno-Ingush A.S.S.R.--Irrigation)

SHVETSOV, I.K.; VOROB'YEV, A.M.

[Methods used for the separation of neptunium and plutonium]
K voprosu o metodakh razdeleniya neptuniia i plutoniia. Moskva,
1955. 6 p.
(Neptunium) (Plutonium)

VOROB'YEV, A. M., and SHVETSOV, I. K.

"On Methods of Separation of Neptunium from Plutonium," a paper presented at
the Atoms for Peace Conference, Geneva, Switzerland, 1955.

KUCHIN, I.P., dotsent, kand. istorich. nauk, kapitan 1-go ranga; GAVRILYUK,
V.K., dotsent, kand. pedagog. nauk, podpolkovnik; BARANOV, G.A.,
kapitan 1-go ranga; VOROB'IEV, A.M., gvardii kapitan 3-go ranga;
CHERNAVSKIY, V.A., podpolkovnik

Reviews and bibliography. Mor. sbor. 48 no.1:87-93 Ja '65.
(MIRA 18:4)

L 35847-66 EXT(m)/EXP(t)/DTI IJP(c) JD

ACC NR: AP6014724 (N) SOURCE CODE: UR/0186/65/007/006/0728/0729

37

AUTHOR: Vorob'yev, A. M.; Fomicheva, V. I.

B

ORG: none

TITLE: Analytical determination of americium, plutonium, and uranium
using AMP anion exchange resin 27 27 27

SOURCE: Radiokhimiya, v. 7, no. 6, 1965, 728-729

TOPIC TAGS: americium, plutonium, uranium, quantitative analysis, ion
exchange resin

ABSTRACT: The method described for the separation of americium, plutonium, and uranium using AMP ion exchange resin is based on the difference in the degree of sorption of ions of uranium (VI), plutonium (IV) and (III), and americium (III) from hydrochloric acid of different concentrations. The column used was a glass tube 6 cm long and 5 mm in diameter, with a drawn out end. In the determinations, a small amount of sulfuric acid does not interfere with the separation, but nitric acid must be eliminated, since it can promote the reduction of plutonium to the trivalent state. To this end, the solution being analyzed was evaporated to dryness, 10 ml of concentrated HCl was added to the

UDC: 543.541.3:546.791:546.799.4-5

Card 1/2

L 35847-66

ACC NR: AP6014724

residue and the solution was again evaporated to dryness. Analytical results of the method are said to be completely satisfactory. Orig. art. bss: none.

SUB CODE: 07/ SUBM DATE: 20Feb65/ ORIG REF: 002

ms
Card 2/2

ZHDANOV, G.P.; VOROB'YEV, A.N.

With the same equipment twice as much mineral wool. Stroi. mat.
11 no.4-4-5 Ap '65. (MIRA 18;6)

1. Glavnyy inzhener Voronezhskogo zavoda silikatnogo kирпича
(for Zhdanov). 2. Nachal'nik konstruktorskogo byuro Voronezh-
skogo zavoda silikatnogo kирпича (for Vorob'yev).

AFANASOV, I.A.; VOROB'YEV, A.N.

Effectiveness of the biomass of methane bacteria (vitamin B₁₂ concentrate) in feeding swine and hens. Vit. res. i
ikh isp. no.6:111-118 '63. (MIRA 17:1)

1. Checheno-Ingushskaya nauchno-issledovatel'skaya veteri-
narnaya stantsiya, Groznyy.

VOROB'YEV, A.N., kand.veter. nauk; NAYDENOV, K.I., mladshiy nauchnyy sotrudnik

Prophylaxis of helminthiases and intoxication in ducks. Veterinariia no.
12:4/-48 D '63. (MIRA 17:2)

1. Checheno-Ingushskaya nauchno-issledovatel'skaya veterinarnaya stantsiya.

L 15938-66 EMT(m)/ITC(f)/EFF(n)-2/EWS(m) MM
ACC NR: AP6005940 (A)

SOURCE CODE: UR/0097/66/000/002/0011/0013

AUTHOR: Vorob'ev, A. N. (Engineer); Dubrovskiy, V. B. (Candidate of technical sciences); Ibragimov, Sh. Sh. (Doctor of technical sciences); Ladygin, A. Ya. (Engineer); Fergamenshchik, B. K. (Engineer)

ORG: none

TITLE: Radiation resistance of the portland cement-based chromite concrete

SOURCE: Beton i zhelezobeton, no. 2, 1966, 11-13

TOPIC TAGS: concrete, construction material, nuclear reactor shield, irradiation resistance, radiation damage

ABSTRACT: The effect of neutron irradiation has been studied on samples of chromite concrete with portland cement binder to supply data on radiation resistance of this material. The material was recognized as a potential substitute for expensive and scarce materials, such as steel, graphite, boron graphite, etc., which are presently used for construction of a heat-shield around nuclear reactors. The briquetted samples were made from a mixture of chromite, portland cement, and phosphoric acid and were irradiated with $2.37 \times 10^{21}/\text{cm}^2$ neutron flux in a BP-5

Card 1/2

UDC: 666.974.2:621.039.58

L 15938-66

ACC NR: AP6005940

reactor for a period of time at temperature fluctuating in the 200—550C range. The irradiated samples maintained the original form and dimensions. Compressive strength of irradiated samples decreased to 60% of the strength of non-irradiated samples kept at room temperature and up to 39% of the strength of non-irradiated samples but exposed to the same temperature fluctuations as irradiated samples. The effect of radiation accounted for a 26% decrease in compressive strength, which indicated that the use of this material in construction of the heat shield for nuclear reactors may be possible. Orig. art. has: 2 figures and 2 tables.

[JK]

SUB CODE: 11/ SUBM DATE: none/ ORIG REF: 010/ ATD PRESS: 4202

18/

FW

Card 2/2

18.820025380
S/089/61/011/001/009/010
B102/B214

AUTHORS: Ibragimov, Sh. Sh., Vorob'yev, A. N.

TITLE: Hardening of molybdenum on irradiation by neutrons

PERIODICAL: Atomnaya energiya, v. 11. no. 1, 1961, 65 - 66

TEXT: The effect of a fast-neutron irradiation on the hardness of molybdenum, and the kinetics of annealing of the radiation defects are investigated in this paper. The samples were irradiated from a 5P-2 (BR-2) reactor at a temperature of 40 - 70°C. With increase of the irradiation dose the microhardness of 99.92%-pure molybdenum increases exponentially, the range studied being 10^{17} - 10^{20} n/cm². The kinetics of the hardening was investigated for 99.9%-pure samples (with 0.003 Ni, 0.004 Fe, 0.002 Cu, 0.001 SiO₂, and 0.01 R₂O₃ in wt %) at 150-220°C in a channel of the 5P-5 (BR-5) reactor, the integral dose being $1.9 \cdot 10^{20}$ n/cm². The hardness was measured by an apparatus of the type "Vickers" with diamond pyramid at a load of 5 kg. The initial hardness was 197 kg/mm², and after the radiation it became 268 kg/mm². The irradiation samples were then exposed to heat

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8/089/61/011/001/009/p10
B102/B214

Hardening of molybdenum ...

treatment at 805, 835, 865, and 897°C. The hardness was found to decrease with increasing temperature and with increasing duration of the heat treatment. The curves at 835, 865, and 897°C show a horizontal part, which indicates the occurrence of two processes with different activation energies. The second process is not noticeable at 805°C even on heating for 6.5 hours. It can be assumed that the increase of hardness on irradiation is related with the formation of two types of lattice defects. In this case the hardness of the irradiated sample may be expressed by Eq. (1), and the decrease in hardness as a function of the holding time τ , at a given temperature by Eq. (2):

$$H_{\text{irr}} = H_0 + A_1 C_1^0 + A_2 C_2^0 \quad (1)$$

$$\Delta H = A_1 C_1^0 (1 - e^{-\frac{\tau}{\tau_1(T)}}) + A_2 C_2^0 (1 - e^{-\frac{\tau}{\tau_2(T)}}) \quad (2)$$

H_0 is the hardness before the irradiation, C_1^0 and C_2^0 the concentrations of the defects of the first and the second kind after the irradiation, A_1 and A_2 are the proportionality factors, and $\tau_1(T)$ and $\tau_2(T)$ the average times in which the defects of the first and the second kind vanish when held at Card 2/3

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8/089/61/011/001/009/010
B102/B214

Hardening of molybdenum ...
the temperature T. The values obtained for the activation energies of the annealing of the defects are compared in the table with the values for iron:

Defects	Activation energy, cal/mole	
	Fe	Mo
First kind	16,500	45,000
Second kind	28,700	76,000
Ratio of the activation energies	0.58	0.59

From the similarity of the ratios it may be concluded that the radiative hardening for Mo is due to defects of the two kinds just as in case of Fe. Since these defects do not affect the resistivity of the metal, the Frenkel' type of defect is not involved. There are 3 figures, 1 table and 4 Soviet-bloc references.

SUBMITTED: October 29, 1960

Card 3/3

IBRAGIMOV, Sh.Sh.; VOROB'YEV, A.N.

Hardening of molybdenum as a result of neutron irradiation.
Atom energ. 11 no.1:65-66 J1 '61. (MIRA 14:7)
(Metals, Effect of radiation on) (Neutrons)

VOROB'YEV, A. N. (Candidate of Veterinary Sciences, City of Groznyi).

"Causes of barrenness of cows on some farms of the Checheno-Ingush Autonomous SSR"...

Veterinariya, vol. 39, no. 8, August 1962 pp. 53

ZHERTOVSKIY, A.N., elektromekhanik; KONURIN, I.M., starshiy
elektromekhanik; VOROB'YEV, A.N.; GORODETSKIY, N.P.,
elektromekhanik

Efficiency experts suggest. Avtom., telem. i sviaz' 4
no.1:32-33 Ja '60. (MIRA 13:4)

1. Kremenchugskaya distantsiya signalizatsii i svyazi Yuzhnay
dorogi (for Zhertovskiy). 2. Yaroslavskaya distantsiya signalizatsii
i svyazi Severnoy dorogi (for Konurin). 3. Starshiy inzhener
Moskovsko-Okrushnay distantsii signalizatsii i svyazi Moskovskoy
dorogi (for Vorob'yev). 4. Krasnoarmeyskaya distantsiya
signalizatsii i svyazi D'onetskoy dorogi (for Gorodetskiy).
(Railroads--Electronic equipment) (Radio--Repair)

VOROB'YEV, A. N., shestikratnyy champion mira, zasluzhenny master
sporta.

Regimen and will. Zdorov'e 5 no.11:24 N '59.
(MIRA 13:3)
(Physical education and training)

VOROB'YEV, A.N., inzhener.

Amplifier for cable test sets using transistors. Avtom.elem.
i sviaz' no.7:29 J1 '57. (MLRA 10:8)

1.Distantsiya signalizatsii i svyazi Moskovsko-Okruzhnaya dorega.
(Electric cables--Testing)

VOROB'IEV, A.N.

Radio signaling in switching operations. Avtom., telem.
sviaz' 4 no.6:26-27 Je '60. (MIRA 13:7)

1. Starshiy inzhener Moskovsko-Okruzhnoy distantsii signalizatsii
i svyazi Moskovskoy dorogi.
(Railroads--Signalizing) (Railroads--Switching)

VOROB'YEV, A. N. Cand Vet Sci -- (diss) *\ Mineral- /* ~~the vitamin-mineral metabolism
of ruminants~~
and the milk productivity ^W of cows during acute primary ~~acute~~ atony ((Treatment
and Prophylaxis). Novocherkassk, 1958. 15 pp (Min of Agr USSR. Novocherkassk
Zootech Vet Inst of First Mounted Army), 130 copies (KL, 52-58, 105)

- 95 -

VOROB'YEV, A.N.

Equipping diesel locomotives with ZhR-4 transmitter-receiver sets.
Avtom., telem. i sviaz' 4 no.2:29-30 F '60. (MIRA 13:6)

1. Starshiy inzhener Moskovsko-Okrushnoy distantsii signalizatsii i
i svyazi Moskovskoy dorogi.
(Railroads--Communication systems)

VOROB'YEV, A.N., inzh.

Electric power supply for locomotive radio stations. Avtom., telem. i sviaz' 2 no.10:36-37 O '58. (MIRA 11:10)

1.Distantsiya signalizatsii i svyazi Moskovsko-Okrughnoy dorogi.
(Railroads--Radio)

VOROB'YEV, A.N., inzh.

Remote switching of amplifiers. Avtom., telem. i svias' 2 no.11:
32-33 N '58. (MFA 11:12)

1. Distantsiya signalizatsii i svyazi Moskovsko-Okruzhnyy doregi.
(Remote control) (Amplifiers, Electron-tube)

VOROB'YEV, A. N.

Motorboats

Motorboats for maintenance workers of the inland waterway system, Rech. transp., 12, no. 4, 1952.

9. Monthly List of Russian Accessions, Library of Congress, October 1952. Unclassified.

VOROB'YEV, A.P.

Constructing integral curves in the region of the origin of coordinates for a system of differential equations on a plane. Dokl. AN BSSR 3 no.8:325-330 Ag '59. (MIRA 12:11)

1. Predstavлено академиком AN BSSR N.P. Yeruginym.
(Differential equations)

ANASHKIN, I.A., kapitan 1 ranga; BARABOLYA, P.D., polkovnik yuridicheskoy sluzhby; VOLKOV, A.S., inzh.-kapitan 1 ranga; VOROB'YEV, A.P., kapitan 1 ranga; VASIL'YEV, I.V., kapitan 1 ranga zapasa; VYUNENKO, N.P., kand.voyenno-morskikh nauk, kapitan 1 ranga; GENKIN, A.L., dotsent, kand.tekhn.nauk, inzhener-kontr-admiral; YEREMENKO, B.Ia., kapitan 1 ranga; ZVEREV, B.I., kand.istor.nauk, mayor; KAZANKOV, A.A., kapitan 1 ranga; KOZIN, K.K., kapitan 1 runga zapasa; KOLYADA, N.I., kapitan 1 ranga zapasa; KULINICH, D.D., inzh.-kapitan 1 ranga; LOBACH-ZHUCHEMKO, M.B., dotsent, inzhener-kapitan 2 ranga zapasa; MASHAROV, A.I., polkovnik zapasa; MYASISHCHEV, V.I., inzhener kontr-admiral; PETROV, L.G., kapitan 1 ranga v otstavke; PROKOF'YEV, V.M., kapitan 1 ranga; POZNAKHIRKO, A.S., kapitan 1 ranga zapasa;

(Continued on next card)

ANASHKIN, I.A.----(continued) Card 2.

PYASKOVSKIY, G.M., polkovnik; SHITSYN, N.I., polkovnik. Prinimali uchastiye: ANDREYEV, V.V., kapitan 1 ranga; IVANOV, V.P., inzhener-kapitan 2 ranga; CHERNOUS'KO, L.D., inzhener-kapitan 1 ranga; SHIKANOV, Ye.P., inzhener-kapitan 2 ranga. FADEYEV, V.G., vitse-admiral zapasa, slavnnyy red.; GERNGROSS, V.M., kapitan 1 ranga zapasa, red.; STAROV, N.M., kapitan 1 ranga v otstavke, red.; SOKOLOVA, G.Y., tekhn.red.

[Marine dictionary] Morskoi slovar'. Moskva, Voen.izd-vo M-va obor. SSSR. Vol.2. O - IA. 1959. 440 p. (MIRA 12:12)

(Naval art and sciences--Dictionaries)
(Merchant marine--Dictionaries)

16,3400

S/250/62/006/005/001/007
I027/I227AUTHOR: Vorob'yev, A. P.

TITLE: On the periods of solutions in the case of a center

PERIODICAL: Akademiya nauk Belaruskay SSR. Doklady. v. 6, no. 5, 1962, 281-284

TEXT: In case the origin (0,0) is a center of the system

$$\frac{dx}{dt} = -y + \sum_{i+j=2}^{\infty} a_{ij} x^i y^j, \quad \frac{dy}{dt} = x + \sum_{i+j=2}^{\infty} b_{ij} x^i y^j \quad (1)$$

(a_{ij}, b_{ij} – constants), Lyapunov (Ref. 2: Sibbr. soch. [collected works] vol. 2, Izd. AN SSSR, M – L, 1956, p. 120), using polar coordinates ρ, θ , represented the period of the solutions in the neighborhood of the origin in the form

$$T(c) = 2\pi \left(1 + \sum_{i=1}^{\infty} h_i^2 c^{2i} \right) \quad (2)$$

where $\rho(\theta_0 c) = c$ and h_i are polynomials of degree $2i$ in a_{ij}, b_{ij} . Hence a necessary and sufficient condition for constant period solutions is $h_{2i} = 0$ ($i = 1, 2, 3 \dots$). A detailed study is given here for the special case

JA

Card 1/2

On the periods of solutions...

S/250/62/006/005/001/007
I027/I227

$$\frac{dx}{dt} = -y - bx^2 - (2c + \beta)xy - dy^2, \quad \frac{dy}{dt} = x + ax^2 + (2b + \alpha)xy + cy^2 \quad (3)$$

and the origin is a center. Six conditions on the coefficients are given, the disjunction of which is necessary and sufficient for constant period solutions. Moreover: a) In the finite part of the plane, either there are no singularities or there is a second center, around which the periods are also constant. b) The integral curves forming center are curves of the second or 4th order. Another result contains a condition that the period $T(c)$ is a monotonic function of c .

✓A

ASSOCIATION: Institut matematiki i vychislitel'noy tekhniki. AN BSSSR (Institute of Mathematics and calculated technics AS BSSR)

PRESENTED: by I. P. Yerugin, Academician

SUBMITTED: September 29, 1961

Card 2/2

ACC NR: AP7002010

SOURCE CODE: UR/0043/66/000/004/0075/0080

AUTHOR: Vorob'yev, A. P.

ORG: none

TITLE: Free point-mass motion including random perturbation of medium and nonspherical earth

SOURCE: Leningrad. Universitet. Vestnik. Seriya matematiki, mekhaniki i astronomii, no. 4, 1966, 75-80

TOPIC TAGS: earth gravitational field, particle mechanics, ordinary differential equation, earth gravity

ABSTRACT: The free motion of a point-mass system in a noncentral terrestrial gravitational field is analyzed. The field potential is given by

$$U(r, \theta) = fM \left[\frac{1}{r} + a \frac{R^2}{r^3} \left(\frac{1}{3} - \cos^3 \theta \right) + \right. \\ \left. + b \frac{R^4}{r^5} \left(\frac{3}{35} - \frac{6}{7} \cos^3 \theta + \cos^6 \theta \right) + \dots \right].$$

It is assumed that a resistive force is acting on the particle, proportional to the position and velocity of the particle. Furthermore, small random perturbations are superimposed on the particle motion to take into account inhomogeneities in the

UDC: 531. 353.

Card 1/2

ACC NR: AP7002010

medium and in the gravitational field. The resulting equations of motion for the particle are then simplified by assuming $b \ll a$ and written in a universal form. This is given by

$$\frac{dm_i}{dt} + \frac{dx_i}{dt} = X_i(m_i + x_i) + F_i(m_i + x_i; t),$$

$(i, j = 1, 2, \dots, 6; k = 2, 4, 6).$

The solution of this equation is discussed briefly for a finite time interval in the motion of the particle. Orig. art. has: 12 equations and 1 figure.

SUB CODE: 20⁰⁸¹ SUBM DATE: 29Jun65/ ORIG REF: 003/ OTH REF: 003

Card 2/2

VOROB'YEV, A.P.

Sufficient conditions for the isochronism of the canonical
systems of two differential equations. Dif. u:av. 1 no.5:
582-584. My '65. (MIRA 18:7)

1. Institut matematiki AN BSSR.

8/250/63/007/003/002/006
ACSN [A124]

AUTHOR: Voponevich, A. F.

TITLE: The system of differential equations

TEXT: The system of differential equations

$$\frac{dx}{dt} = -y + P(x, y), \quad \frac{dy}{dt} = x + Q(x, y) \quad (1)$$

is called linear, where $P(x, y)$ and $Q(x, y)$ are holomorphic functions in the complex plane C^2 . It is known that the system (1) has a unique solution for any initial conditions (x_0, y_0) if $P(x, y)$ and $Q(x, y)$ are bounded in some neighborhood of the point (x_0, y_0) .

Card 1, 4

8/250/63/007/003/002/006
A059/A126

On isochronous systems of two differential equations

$$\dot{x} = P(x, y) + \sum_{i=1}^n a_i x^i$$

The system (1) is called isochronous if there exists a function $\varphi(x)$

φ'

$$T(x, y; 2\varphi) = 1 + \sum_{i=1}^n b_i x^i$$

is always isochronous (2) for the 'isochronism' of the system (1). It is necessary and sufficient to make the coordinate transformation

$$\begin{aligned} \xi &= x + \sum_{i=1}^n a_i x^i \varphi, \quad \eta = y + \sum_{i=1}^n b_i x^i \varphi \end{aligned}$$

should exist converting (1) to

$$\frac{dy}{dx} = -\frac{(x+1)}{y}$$

Card 2/4

8/25/63/007/003/002/006

A059/A126

On isochronous systems of two differential equations

3) if the general integral of the canonical system (1) is represented in polar coordinates in the form:

$$\rho^4 + 2F(\rho \cos \theta, \rho \sin \theta) = c^4.$$

it is necessary and sufficient for the identity

$$\frac{1}{2} \int_0^{2\pi} \rho^2 d\theta = c^2$$

to hold; 3') if the general integral of the canonical system (1) is written in the form (2), it is necessary and sufficient for the isochronism of (1) that the curve of the closed-curve radius average of the family (2) should be determined from the equation: $\Psi(x, y) = 0$, where if the general integral of (1) is written in the form (2), then $\Psi(x, y) = \psi(x)$ and the period of the

$$2F(x, y) + \sum_{i=1}^n \frac{h_i H^{i+1}(x, y)}{i+1} = 0.$$

Card 3/4

8/250/63/007/003/002/006
A059/A126

On isochronous systems of two differential equations

This suggestion solves completely the problem concerning the possibility of the transformation of the variables of the family of integral curves of the system (1) into a family of regions in the case of a center by holomorphic change, suggested by the head of the training course for differential equations N.P. Yerugin. Thanks are due to Yu.S. Bogdanov for attention.

ASSOCIATION: Institut matematiki i vychislitel'noy tekhniki AN BSSR (Institute of Mathematics and Computing Engineering of the AS BSSR)

PRESENTED: by N.P. Yerugin, Academician AS BSSR

SUBMITTED: September 22, 1962

Card 4/4

VOROB'YEV, A.P.

Construction of isochronous systems of two differential equations.
Dokl. AN BSSR 7 no.8:513-515 Ag '63. (MIRA 16:10)

1. Institut matematiki i vychislitel'noy tekhniki AN BSSR.
Predstavлено академиком AN BSSR N.P. Yeruginym.

VOROB'YEV, A. P.

Plane motion of a ship exposed to wind. Vest, LCU 18 no.1:
90-95 '63. (MIRA 16:1)

(Ship propulsion) (Hydrodynamics)

VOROB'YEV, A.P.

Methods of probability theory applied to studying the nonlinear
lateral rolling of a ship. Vest.GLU 17 no.7 101-104 '62.
(MIRA 15:5)

(Probabilities) (Stability of ships)

VOROB'YEV, A.P.

Solution periods in the case of a center. Dokl. AN BSSR
6 no. 5:281-284 My '62. (MIRA 15:6)

1. Institut matematiki i vychislitel'noy tekhniki AN BSSR.
Predstavлено академиком AN BSSR N.P. Veruginym.
(Differential equations)

VOROB'YEV, A.P.

Qualitative study of integral curves in the large of isochronous
systems of two differential equations. Dif. urav. 1 no.4:439-441
Ap '65. (MIRA 18:5)

1. Institut matematiki AN BSSR.

VOROB'YEV, A.P.

Cycles around a special point of the "node" type. Dokl. AN BSSR
4 no.9:369-371 S '60. (MIRA 13:9)

1. Institut matematiki i vychislitel'noy tekhniki AN BSSR.
Predstavлено акад. AN BSSR N.P. Yeruginym.
(Differential equations)

VOROB'YEV, A.P.

Data on the flora of the Kurile Islands. Trudy Dal'nevost.fil.
AN SSSR. Ser.bot. vol.3:3-79 '56. (MLRA 9:8)
(Kurile Islands--Botany)

VOROG'YEV, A. P.

"Several Investigations of the Lateral Rolling of a Ship by the Methods of Probability Theory." Cand Phys-Math Sci, Leningrad State U, Leningrad, 1953. Dissertation (Referativnyy Zhurnal--Matematika Moscow, Feb 54)

SO: SU 136, 19 Aug 1954

15-1957-3-3060

Translation from: Referativnyy zhurnal, Geologiya, 1957, Nr 3,
p 90 (USSR)

AUTHORS: Vorob'yev, A. P., Yenkeyev, M. R.

TITLE: Hydrous Phosphates and Silicates of Aluminum in
Carboniferous-Siliceous Shales (O vodnykh
fosfatakh i silikatakh alyuminiya v formatsiyakh
uglerodisto-kremnistykh slantsev)

PERIODICAL: Tr. Sredneaz. un-ta, 1956, Nr 82, pp 25-27

ABSTRACT: A network of veins of a colloform mineral, suggestive
in its outward aspect of allophane, has been
recognized in the Middle Cambrian carbonaceous-
siliceous shales of southern Kazakhstan. The
mineral is an opaline deposit which is milky white
in color, with faint greenish tints. Its fracture
is conchoidal to irregular; it is brittle and is
easily broken down into fine sharp-edged fragments.

Card 1/2

15-1957-3-3060

Hydrous Phosphates and Silicates of Aluminum

The luster is generally dull but may be slightly waxy. It has a hardness of 3.5, a specific gravity of 2.16, and a refractive index of 1.475. The chemical composition is SiO₂ 8.05%; Al₂O₃ 21.93%; CaO 3.26%; MgO 1.01%; P₂O₅ 25.82%; V₂O₅ 1.18%; SO₃ 0.83%; Cl 1.11%; H₂O 35.8%; total 99.5%. Very small quantities of Na, Fe, Ti, Mo, Sr, and Cu have been identified by spectral analysis. The thermal curve shows an endothermic effect with a maximum at 160° and an exothermic effect at 775°. The author believes the mineral to be a mixed type, a combination of hydrous phosphate, silicate and, in part, sulfate and chloride. The mineral was formed by the action of ground waters on the carbonaceous-siliceous and interbedded argillaceous shales.

G.A.G.

Card 2/2

32445

S/044/61/000/010/009/051
C111/C222

16.3410

AUTHOR:

Vorob'yev, A.P.

TITLE:

On the question on the cycles around a singular point of
 the type "knot"

PERIODICAL: Referativnyy zhurnal. Matematika, no. 10, 1961, 23-24.
 abstract 10 B 111. ("Dokl. AN BSSR", 1960, 4, no. 9,
 369 - 371)

TEXT: The author considers the differential equation

$$\frac{dr}{d\zeta} = \frac{F(r, \zeta)}{\Phi(r, \zeta)}$$

(1)

for which the conditions of existence and uniqueness are satisfied in
 the whole plane $r\zeta$ (r and ζ are polar coordinates, $F(0, \zeta) \neq 0$,
 $\Phi(0, \zeta) \neq 0$). The author proves the following lemma: If there
 exists a branch Γ_1 of the isochinal line of the infinity the one end
 of which goes in infinity and which lies in the sector $\zeta_1 < \zeta < \zeta_2$

Card 1/2

32445

S/044/61/000/010/009/051
C111/0222

On the question on the cycles ...

$(\zeta_2 - \zeta_1 < 2\pi)$, and if for the transition over Γ_1 , $\phi(r, \zeta)$ changes the sign then no closed characteristic of (1) which runs around the origin has common points with Γ_1 . Herefrom it follows the conclusion

that if the isoclinal line of infinity $\phi(r, \zeta) = 0$ has a single branch Γ_1 which ends with one end in the infinity and with the other end in the origin, and which lies in the above mentioned sector, then the equation (1) has no closed characteristics around the origin. Starting from this statement the author proves that the equation

$\frac{dy}{dx} = \frac{Q_2(x, y)}{P_2(x, y)}$, where Q_2 and P_2 are polynomials of second degree, has

no limit cycles if the coordinate origin is a knot.

[Abstracter's note : Complete translation.]

Card 2/2

MIROFOL'SKIY, Yu.A., inzh.; VOROB'YEV, A.P., inzh.

New design of the transfer mechanism on automatic nut-upsetters.
[Nauch. trudy] ENIKMASHA 6:52-59 '63. (MIRA 16:9)
(Forging machinery)
(Mechanisms—Design and construction)

VOROB'IEV, A.P.

Isochronous systems of two differential equations. Dokl. AN BSSR
7 no.3:155-156 Mr '63. (MIRA 16:6)

1. Institut matematiki i vychislitel'noy tekhniki AN BSSR.
Predstavлено академиком AN BSSR N.P.Yeruginym.
(Differential equations)

VOROB'YEV, A. S.

57/49T28

USSR/Chemistry - Quantitative May/Jun 49
Analysis
Chemistry - Chlorine

"Method of Quantitative Determination of Chlorine
Ions in Iodides," A. S. Vorob'yev, ½ p

"Zhur Anal Khim" Vol IV, No 3

Corrects errors found in Berg's method by increasing
the concentration of sulfuric acid in the
solution to 2.5 N. This permits a sufficient
concentration of hydrogen ions and causes a
complete reaction between iodine and acetone.
Determines chlorine ions nephelometrically.

Submitted 9 Jan 48

FDD

57/49T28

VOROBIEV, A.S.,

S. IVANOV, Chem. Umschau Fette, Oele, Wachtse Harze 37, 349-
54, (1930)

VOROB'YEV, A.S.

Effect of initial chromium oxide on the properties of magnesium-chromium ferrites. Izv. vys. ucheb. zav.; fiz. no.5:135-138 '64.

(MIRA 17:11)

1. Moskovskiy ordena Lenina energeticheskiy institut.

(A) VOROB'YEV, A-S.

Determination of chloride ion in iodides. A. S. Novikov, Zhur. Anal. Khim. 4, 231 (1949); cf. Chem. Abstr. 43, 7042g. — The accuracy of the Berg method is greatly increased if the H-ion concn. of the soln. is raised. It is recommended to have the soln. 2.5 N H_2SO_4 at the start and upon addn. of I^- it should not drop below 1 N.
M. Hösch

VOROB'YEV, A.S.

C

Quantitative determination of chloride ion in the presence of bromine and iodine ions by the Berg method.
 A. S. Vorob'yev, *Zhur. Anal. Khim.*, 1, 187-97 (1947).—
 The purpose of this investigation was to test the Berg method for detg. Cl⁻ in the presence of Br⁻ and I⁻; by this method Br and I are fixed with MgSO₄ and Cl is detd. by the Volhard method. In one series of expts. the interaction of monobromo- and moniodoacetone with Ag as affected by concn., temp., and time was studied. When the concn. of monobromoacetone in soln. was 0.01 N none of it reacted with Ag at 0-50°. At the boiling temp. it formed a ppt. with Ag within 5 min.; the quantity of reacting Ag increased with the vol. of the monobromoacetone taken. Ag reacted with 0.10 N monobromoacetone at 30° only after 24 hrs. and with 0.2 N to 0.5 N solns. after 5 min. The quantity of reacting Ag increased with temp., and concn. of monobromoacetone. The interaction of Ag and moniodoacetone was similar to its interaction with monobromoacetone but more intense. In a 2nd series of expts. the oxidation of Cl by KBrO₄ was studied. The latter did oxidise Cl. The quantity of Cl oxidised increased with its concn., time of interaction, and temp. Still another source of error in this Berg method is the interaction of Ag with (NH₄)₂FeSO₄ to form a ppt. Ag₂SO₄ and the reduction of Ag ion by FeSO₄. The errors in this method are of two kinds: (a) the interaction of Ag with Br, I, and FeSO₄ tends to give higher results, and (b) the oxidation of Cl by BrO₄⁻ tends to give low results. By properly balancing the two, accurate results can be obtained.

M. Hirsch

ASA-514 METALLURGICAL LITERATURE CLASSIFICATION

12001 SUBJECTS

SEARCHED

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"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860820004-2

BLOKH, I.M.; VOROB'YEV, A.S.; KROLENKO, N.G.

Electric field of a pattern mapping unit above the contact of
two media. Prikl. geofiz. no.40:101-119 - 64 (MIRA - 18:1)

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860820004-2"

VOROB'YEV, A.S.; KOLCHIN, V.V.

Effect of small additions on the temperature dependence of
the initial magnetic susceptibility of magnesium zinc
ferrites. Izv. vys. ucheb. zav.; fiz. no.4:180-183 '64

(MIFRA 178)

1. Moskovskiy energeticheskiy institut.

Vorob'yev, A.S.

USSR /Electricity

G

Abs Jour : Ref Zhur - Fizika, No 4, 1957, No 9657

Author : Vorob'yev, A.S.

Inst : Not given

Title : Mechanical Losses in Piezoelectric Ceramics

Orig Pub : Sb. stately nauch. stud. o-va. Mosk. Energ. in-t, 1955, vyp. 8, 325-334

Abstract : Using a Q-meter, the author measured the mechanical losses in specimens of ceramic BaTiO₃ with inclusions of oxides of lead and tin. The temperature range of the investigations was from -25 to 120°. It was found that in piezo-ceramic resonators, the mechanical losses predominate over the dielectric losses and amount to 99 — 70% of the total losses. The losses must therefore be determined in such materials at the resonant frequencies, when the losses are considerably higher than at the non-resonant frequencies.

Card : 1/2

USSR / Electricity

G

Abs Jour : Ref Zhur - Fizika, No 4, 1957, No 8657

Abstract : The mechanical losses are proportional to the magnitude of the piezo-modulus, and increase with increasing modulus. The mechanical losses duplicate to a certain extent the temperature behavior of the piezo-modulus, and the dielectric losses of the unpolarized ceramic increase with the temperature.

Card : 2/2

VOROB'YEV, A.S., dotsent

Obtaining bivinyl by the S.V. Lebedev method. Khim. v shkole
15 no.3:59-60 My-Je '60. (MIRA 14:7)

1. Udmurtskiy pedagogicheskiy institut, g. Izhevsk.
(Boutadiene) (Chemistry--Experiments)

VOROB'YEV, A.S.

Titrimetric method for the determination of chlorine, bromine and iodine ions in their mixture [with summary in English]. Zhur. analkhim. 12 no.3:395-397 My-Je '57.
(MLRA 10:7)

1. Udmurtskiy gosudarstvennyy pedagogicheskiy institut, Izhevsk.
(Halogens)

"APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860820004-2

L 6777-65

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860820004-2"

"APPROVED FOR RELEASE: 03/14/2001

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CIA-RDP86-00513R001860820004-2

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001860820004-2"

LORBERG, M.G., inshener; MINAYEV, A.F. (Leningrad); SOTNIKOV, B.I.;
ENGEL', B.V.; RADOSTAYEV, N.I.; VOROB'YEV, A.S.; MINASYAN,
I.S.; BAKSHAYEVA, S.I. (Moskva); KOROCHANSKIY, V.K. (Moskva).

Combined work teams as an untapped resource in raising labor
productivity. Stroi. prom. 33 no.11:5-14 N '55. (MLRA 9:2)

1.GPI Leningradskiy Promstroyprojekt (for Lorberg).2.Magnito-
stroy (for Sotnikov).3.Lishkimpromstroy (for Engel').4.Tagil-
stroy (for Radostayev).5.Trest Kaspmorstroy (for Vorob'yev).
6.Stroitel'noye upravleniye No.3 tresta Azbeftezavodstroy
(for Minayan).

(Construction industry)

VOROB'YEV, A.T., glav. red.; POLYAKOV, L.N., zam. glav. red.; BORISOV,
Ye.G., red.; IVASYSHIN, S.N., red.; IMANALIYEV, Sh.I., red.; LYA-
SHENKO, I.V., red.; OLEZNIK, A.K., red. Prinimali uchastiye: BEK-
BOYEV, D.B., spets. red.; KIRKIN, M.F., spets. red.; TEIEVIN, G.P.,
spets. red.; YUDAKHIN, N.P., red.; YEFIMOV, N.A., tekhn. red.

[Agriculture of Kirghizistan] Sol'skoe khoziaistvo Kirgizi; kratkii
spravochnik. Frunze, Ob-vo po raspr. polit. i nauchn. znanii Kirgiz-
skoi SSR, 1961. 199 p.
(Kirghizistan—Agriculture) (MIRA 14:10)