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1. Laboratory of Protozoan Cytology, Institute of Cytology of the Academy of Sciences of the U.S.S.R., Leningrad (Head: G. Poljansky)[Polzhanskiy, G.].

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VELKOV, As., ikon.; GUDEVA, Zh., inzh.; SOTIROV, Iv., inzh.; TSONEV, D.,
inzh.; KHRISTOVA, S., inzh.; RAIKOV, Il., inzh.; KOSTADINOV, V., inzh.

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SO:Monthly List of East European Accessions, (EEAL). LC, Vol. 4, No. 11,
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from the Russian.

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SO: Monthly Index of East European Accessions (EEAI) LC, Vol. 7, No. 3,
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SO: East European Accessions List Vol 2 No 6 Aug 1954

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SO: Monthly List of East European Russian Accessions, Library of Congress, Vol. 3, No. 3, 1954
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APPROVED FOR RELEASE: 03/14/2001 **CIA-RDP86-00513R001344020009-8"**
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(Leka Promishlenost, Vol. 5, no. 12, 1956, Bulgaria)

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RAIKOV, P. N.

PROFESSOR P. N. RAIKOV

TECHNOLOGY

Periodical: KHMIIIA I INDUSTRIIA. Vol. 30, no. 5, 1958

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February 1959, Unclass.

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F '63.

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"Compensative regulation of electric-power production."

p. 3 (Elektroenergiia, Vol. 9, no. 1, 1958, Sofia, Bulgaria)

Monthly Index of East European Accessions (EEAI) LC, Vol. 7, no. 9,
September 1958

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(CARDIOSPASM surg)

HAIKOV.

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Sofia, Bulgaria

See: East European Accession, Vol. 6, No. 3, March 1957

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The Sandanska Bistritsa Rapids. Elektroenergiia 15 no.5&2-6
My '64

MIKHAILOV, Mikhail; RAIKOV, Raiko

Possibilities for automatic adjustment of the position of tractor-mounted plows to the relief in the vertically longitudinal plane. Izv mekh selsko stop BAN 4 97-106 '63.

TRINCY, B.

Results from technical standardization in the furnace section of the V. Kolarov
State Cement Plant. p. 55.
Heavy industry in Czechoslovakia during the Second Five-Year Plan. p. 58.
heavy industry in Hungary during the Second Five-Year Plan. p. 60.
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Sofia, Bulgaria

SEARCH: East European Accessions List (EEAL) Library of
Congress, Vol. 6, No. 1, January 1957

RAIKOV, S.

Overexpenditure must not be allowed with the "working wages" fund of the Mining Enterprise for Nonmetallic Minerals. p. 78. Vol. 10, No. 3, May/June 1955. MINNO DELO. Sofiya, Bulgaria.

SOURCE: East European Accessions List, (EEAL) Library of Congress, Vol. 5, No. 1, January, 1956.

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Methods of testing and computing the noble and rare metals and dispersed elements in the nonferrous-metal ores in Bulgaria. Min delo 17 no.7:22-24 JI '62.

1. Upravlenie "Geolozhki prouchvania i okhrana na zemnite nedra".

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P. 16, (Geografiia) Vol 7 no. 6, 1957, Sofia, Bulgaria

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March 1958

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Khidro i meteorolog 13 no. 2:45-50 '64.

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1. Note présentée par L. Krastanov [Krustanov, L.], membre
de l'Académie, membre du Comité de rédaction et rédacteur
responsable, "Doklady Bolgarskoy Akademii Nauk."

RAIKOVA, D.

Celestial omens. Nauka i tekhnolozhiya 16 no.12:19-24 '64.

RAIKOVA, Donka, d-r

The star path. Nauka i tekhnolozhiya mladezh 14 no.12:22-23 '62.

1. Astronomicheska sektiia pri Bulgarskata akademiia na naukite.

MANOLACHE, Mircea, conf. ing.; BODEA, Ion, asistent ing.; RAILEANU, Dumitru,
asistent ing.; SAS, Ion, asistent

Corrosion of aluminum and its alloys. Metalurgia constr mas 8 no.11:
937-950 N '61.

(Corrosion and anticorrosives)
(Aluminum alloys)

R/009/61/000/011/001/001
D282/D303

AUTHORS: Manolache, Mircea, Instructor, Engineer, Bodea, Ion,
Assistant, Engineer, Răileanu, Dumitru, Assistant,
Engineer, and Sas, Ion, Assistant

TITLE: On the corrosion of aluminum and its alloys

PERIODICAL: Metalurgia și construcția de mașini, no. 11, 1961, 937-950 ✓

TEXT: The article presents the results of experiments by the authors on the corrosion of aluminum and aluminum-alloy sheets in the various conditions of the Galați and Constanța harbors. The authors used in their experiments commercial aluminum of the following composition: 0.05% Fe, 0.31% Zn, 0.03% Mg, and the rest aluminum, as well as aluminum alloyed with 5% Cu and 3% Zn. Commercial aluminum was rolled into 1.5 - 2 mm thick sheets, while aluminum alloy into 4 - 6 mm thick sheets. The following

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corrosion media were selected: (1) Danube water; (2) Danube atmosphere; (3) Black Sea water; (4) Black Sea atmosphere; (5) town atmosphere of Galati; and (6) Sea water brought into the laboratory. Since in ship or harbor constructions the aluminum generally comes into contact with other materials, the authors selected the following contact hypotheses: (1) without any contact to other material; (2) in contact with OL 38 steel; (3) in contact with copper; (4) in contact with bronze mixed with tin; (5) in contact with fir-wood; (6) in contact with zinc; and (7) in contact with oak-wood. The samples were tested with or without protection, i.e. (1) without any protection; (2) anodically oxidated; (3) painted, and (4) anodically oxidated and painted. The results obtained by the authors completely verified the modern corrosion theories. Thus, in case of commercial aluminum, an anodic dissolution was produced on the samples. This anodic dissolution was increasingly reduced due to a passivity process. In case of samples made from aluminum alloyed with Cu and Zn, the corrosion velocity permanently increased due to the action of

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On the corrosion ...

the cathodic inclosures. A general passivity of the metal or alloy is only produced if there are some conditions of an anodic passivity of the anodic components. Knowing the appearance mechanism of the anodic passivity, the potential up to which the anode has to be polarized, can be calculated. Preliminarily oxidated aluminum samples were more electronegative; the potentials tended towards a stability, i.e. passivity; and the dissolution current had an increasing tendency. In case of aluminum samples alloyed with Cu and Zn, the potential and the current had a continuously increasing tendency. The powerful corrosion of the alloyed samples which in some cases even led to pitting, was especially due to an increase of the number and size of the cathodic inclosures. The corrosion of the commercial aluminum samples was characterized by a surface corrosion, while that of the aluminum-alloy samples by an intercrystalline corrosion. The most powerful corrosion effect on commercial aluminum samples was exerted by Black-Sea-water, while on aluminum alloy samples



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On the corrosion ...

by Black-Sea and Danube waters. The average corrosion depths in the case of commercial aluminum samples was 60μ , while in case of aluminum alloy samples it was almost 0.5 mm. However, the corrosion process did not vary proportionally with the time. The corrosion velocity increased the longer the aluminum alloy samples were kept in the corroding media, and decreased the longer the commercial aluminum samples were subjected to the activity of the corroding media. The authors draw the following preliminary conclusions: (a) Commercial aluminum is less corroded than aluminum alloyed with Cu and Zn. (b) The most powerful corrosion is produced by Black-Sea water, followed by Danube water. Sea water in the laboratory produced corrosion similar to corrosions produced by the sea-atmosphere. (c) Protecting layers have only delayed the corrosion of all samples submerged in natural waters, but proved to be more efficient in the case of samples subjected to atmospheric corrosion. (d) Generally, the contact materials increased the corrosion effects

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on the samples. (e) Most powerful corrosions were found on samples in contact with copper and bronze. (f) The decreasing order of the influence of the contact material on aluminum and aluminum-alloy samples, independently of the corroding media, was: copper, bronze, fir-wood, oak-wood, steel and zinc. (g) Zinc delayed the corrosion of aluminum and aluminum-alloys. The corrosion of all samples in the atmosphere was generally weak, superficial and uniform, being more powerful under a contact material. Red-lead proved to be a good protecting layer. Anodically oxidated and painted samples were not at all corroded, while painted samples were slightly corroded especially when being in contact with copper and bronze. There are 21 figures, 5 tables and 15 references: 11 Soviet-bloc and 4 non-Soviet-bloc. The reference to the English-language publication reads as follows: J. Sundarjan and T.L. Rama Char: "Inhibition of the Corrosion of Aluminum in Alkaline Solutions", Corrosion prevention and Control, 5, 1958, no. 5, 55-56. ✓

Card 5/5

GIORANESCU, Ecaterina; NAILBANU, Ion

Potential anticancer agents. Pt 2. Rev chimie 8 no.1:21-29
'63.

1. Institute of Chemistry of the Academy of the R.P.R. Bucharest.
2. Corresponding Member of the Academy of the R.P.R. (for Gioranescu).

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Reviewed by Ilie Ion, D. Raileanu. Natura Geografie 15
no.6:91 H-D '63.

RAILEANU, D.; NENITESCU, G.D., *asadarucian*

Synthesis of some 5,6,7-trimethoxy-indole derivatives.
Studii cerc chim il no.1:19-36 '63.

1. Sectia de chimie organica a Centrului de cercetari
chimice al Academiei R.P.R., Bucuresti.

NENITSESKU, K.D. [Nenitescu, C.D.] (Romanian); RAILEANU, D. [Raileanu, D.];
ANGELIDE, N. [Angelide, N.]

New synthesis of the ester of 3-indolylpyrrolic acid. Rev
chimie 8 no.1:59-63 '63.

1. TSentr khimicheskikh issledovaniy Akademii RNR Otdel organicheskoy
khimii, Bukharest.

RAILEANU, G. ; RADALUTA, A.

Study of the fauna of *Aptychus* in the calcaria of the superior Jurassic of the Svinita-Svinicea Mare zone. p. 223.
(ANALELE. SERIA STIINTELOR NATURII, Rumania. Vol. 5, no. 11, 1956)

SO: Monthly List of East European Accessions (EEAL) LC, Vol. 6, no.7, July 1957, Incl.

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Contributions to the knowledge of the Lower Carboniferous from the
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New data on the western tectonic boundary of the Resita
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RAILEANU, Gr.; NASTASEANU, S.; BOLDUR, C.

New data on the west tectonic limit of the Resita zone (Banat).
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1. Comunicare prezentata de academician M.G. Filipescu.

RAILEANU, Gr.; IORDAN, Magdalena

Study on the Liassic brachiopods in the Svinita region. Studii
cerc geol 9 no.1:3-24 '64

1. Chair of Geology of the Faculty of Geology and Geography of the
Bucharest University.

RAILEANU, Gr.; POPESCU, Gh.

Micropaleontologic study of the Lower Cretaceous at Svinita
(Southern Banat). Studii cerc geol 9 no.1:51-60 '64

1. Institute of Geology and Geography of the Rumanian Academy,
Bucharest.

RAILEANU, Gr.; NEGULESCU, V.

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1. Submitted January 1964.

RATLEANI, G. I.; RUSU, A. & RAIBESCU, V.

Tectonic relations of the crystalline of the Muresh Basin Mountains
with sedimentary formations of the Transylvania Basin. *Journal
of Geological Geology*, 1964, 2:251-262.

Geologic Institute of the Geologic Committee and the Institute
of Geology and Geography of the Rumanian Academy. Submitted June
1964.

1963, Gr. 1963, 1963, 1963

... of the Gallia ... in the Resita
... geol. geogr. 9 no. 2: 261-275, 1962.

... Faculty of Geology and Geography, University of Bucharest.
... 1963.

NICULESCU, Ion T.; GAHGI-PARSCHIV, A.; ONICESCU, Doina; RAILEANU, Iona;
GIUGARIU, D.; DEDIU, St.; OPRESCU-LISSIVIEVICI, Elena; TRIFU, P.

Contribution to the study of nerve endings in the pancreas. Rumanian
M. Rev. 1 no.2:5-10 Apr-June 57.

(PANCREAS, innervation
anat. of nerve endings)

APPROVED FOR RELEASE: 03/14/2001

CIA-RDP86-00513R001344020009-8

RUMANIA/Human and Animal Morphology (Normal)
Nervous System

Abs Jour : Ref Zhur - Biol., No 7, 1958, No 31227

Author : Niculescu Ion T, Hagi-Parschiv A., Enachescu Aurelia,
Bodescu Ruxanda, Raileanu Iona.

Inst : Not Given

Title : Concerning the Nerve Endings of the Digestive Apparatus: the
Stomach and the Duodenum.

Orig Pub : Bul. stiint. Sec. med., 1956, 8, No 1, 101-150

Abstract : The stomach and intestine of dogs, cats, rabbits, rats and
mice were stained with methylene blue; also used were im-
pregnation with silver, Nissle body stain in different cytol-
ogical methods. In the Auerbach system, the networks possess
ganglia which lie under the serous membrane. Especially
numerous are ganglia in the musculature of Oddi's sphincter.
The ganglia are rich in blood vessels. Ganglionic cells of
the Meissner plexus are smaller; they possess thinner synaptic
apparatuses. In the muscle layer, at different pH values, the

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RAILEANU, LEANA
HAGI-PARASHIV, A., Assit. Prof.; ENACHESCU, A.; ONICESCU, Doina; SMOLENSCHI-POTIN,
Ludmila; BOCIRNEA, C.; LISSIEVICI-OPRESCU, Elena; DEDIU, St.; RAILEANU,
Ioana; RADU, Sorina

Contributions to the study of terminal nerve structures in the peritoneum.
Rumanian M. Rev. 2 no.2:9-10 Apr-June 58.

(PERITONEUM, innerv.
terminal nerve structures)

NICULESCU, I.T.; PARASCHIV, Hagi, A.; ENACHESCU, Aurelia.; BADESCU,
Ruxandra.; RAILEANU, Ioanna.

Study of the nerve endings of the liver and bile ducts. Bul. stiint.
sect. med. 7 no.2:609-620 Apr-June 55

(LIVER, innervation
nerve endings, in exper. animals, histol.)
(GALLBLADDER, innervation
nerve endings, in exper. animals, histol.)
(BILE DUCTS, innervation
nerve endings in exper. animals, histol.)

NICULESCU, Ion, T., membru corespondent al Academiei R.P.R.;
HAGI-PARASHIV, A.; ENACHESCU, Aurelia; BADESCU, Ruxanda;
RAILEANU, Ioana

The nerve endings of the digestive tract: stomach and duodenum.
Bul. stiint. sect. med. 8 no.1:101-150 Jan-Mar 56.

(NERVE ENDINGS
of stomach & duodenum, in various animals)
(STOMACH, innervation
nerve endings, in various animals)
(DUODENUM, innervation
neve endings, in various animals)

BERCEANUM, St.; GOCIU, Mariana; RAILEANU, Ioana

Studies of the behavior of hematopoietic cells in tissue culture. Stud. cercet. med. intern. 6 no.1:25-29 '65.

RAILORU, L.

Quality of products in state and cooperative commerce. p. 20.

Vol. 8, no. 1, Jan 1956
STANDARTECARIA
Bucuresti, Rumania

Source: East European Accession List. Library of Congress
Vol. 5, No. 3, August 1956

RAILEANU, M.

✓ Preparation of phenothiazine 5-oxide. C. Bodea and M. Răileanu (Acad. R.P.R., Cluj, Romania). *Acad. rep. populare Romine, Filiala Cluj, Studii cercetări chim.* 11, 129-33(1980).—Optimal conditions for prepg. phenothiazine (Ia) 5-oxide (I) from *N*-benzoylphenothiazine (II) were investigated, and a synthesis of I was achieved. II, obtained from 50 g. Ia and 32.5 cc. BzCl, was treated in 500 cc. AcH with 62.5 cc. HNO₃ (d 1.5), the mixt. agitated 10 min., and poured into H₂O to yield *N*-benzoylphenothiazine 5-oxide (III). III, recrystd. from PhMe was refluxed in 750 cc. EtOH 5 min., 150 cc. 10% aq. NaOH soln. added, the whole refluxed again 15 min., and cooled to give I, m. 257-8°. The yield (on Ia) was 84%. Treating III with Zn powder gave II. I and III were sol. in MeAc, AcH, moderately sol. in EtOH. I was slightly sol. in PhMe. T. Szűll

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JAJ(NB)

RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Author : Bodea, C. and Raileanu, M.
Inst : Rumanian Academy of Sciences
Title : The Nitration of Phenothiazine

Orig Pub: Studii si Cercetari Chim Acad RPR, Fil Cluj, 8, No 3-4, 301-313 (1957) (in Rumanian with French and Russian summaries)

Abstract: For the purpose of studying the biological properties of nitroderivatives of phenothiazine (I), the authors have investigated the nitration of I and of the 5,5-dioxide of I (II). 3,7-dinitrophenothiazine (III) was prepared and oxidized to the 5,5-dioxide (IV) with H₂O₂; IV was also prepared by the nitration of II and

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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Abstract: 90%, mp 172-173° (from alc). A boiling solution of 10 gms IX in 85 ml CH₃COOH is treated with 10 ml 30% H₂O₂, heated for 1.5 hrs (adding 10 ml portions of H₂O₂ every 30 min for a total of 30 ml), and the solution is allowed to stand 14-16 hrs; the 5,5-dioxide of IX is obtained, yield 92%, mp 246-247° (from benzene); hydrolysis of the product obtained with 10% alcoholic KOH gives II, yield 78%, mp 255-256° (from CH₃COOH). A suspension of 5 gms III in 0.1 liter CH₃COOH is treated dropwise with a solution of HNO₃ (d 1.42) in CH₃COOH (1:1) until the solution of III is complete and the solution is diluted with water; V is obtained, yield 65%, mp 309-310° (from aniline). 5 gms V are oxidized

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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Abstract: with H₂O₂ (see IX) and IV is obtained, yield 85%, mp 337-338° (from C₆H₅NO₂ or aniline). 2 gms of powdered II are added gradually to a solution of 50 ml HNO₃ (d 1.5) in 150 ml water, the solution is stirred for 15-20 min, and allowed to stand for 8 hrs with periodic stirring; VI is obtained, yield 70%, mp 344-345° (from C₆H₅NO₂). 5 gms I are gradually added to a mixture of 30 ml HNO₃ (d 1.5) and 15 ml 20% oleum [fuming H₂SO₄], the mixture is heated for 20 min at 100°, and ice water is added to precipitate VII, yield 85%, mp 354-355° (from C₆H₅NO₂). A similar procedure is followed in preparing VIII from II, yield 90%, mp 346-347° (from C₆H₅NO₂). VIII is obtained in equal yield by the oxidation of VII with H₂O₂ or

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RAILEANU, M.

Distr: 4E3d

Nitration of phenothiazine, C. Bodea and M. Raileanu. (*Acad. rep. populare Romine, Fiziola Cluj*). *Studia chim. 8*, 303-13(1957).—Through a systematic study of the nitration of phenothiazine (I) and phenothiazine 5,5-dioxide (II) it was shown that of all nitrated derivs. of I described only 3-nitrophenothiazine 5-oxide (III) corresponded to a pure product. H₂O₂ oxidn. of III produced 3-nitrophenothiazine 5,5-dioxide (IV). IV was also prepd. from II (2 g.) added slowly to a soln. of HNO₃ (50 ml. d. 1.50) and H₂O (150 ml.). After 8 hrs. with intermittent stirring, filtration, and washing with H₂O and alc., 70% IV was obtained, yellow prisms, m. 344-5° (PhNO₂). 3,7-Dinitrophenothiazine (V) was prepd. from 5 g. I by dissolving in CHCl₃ and 10 ml. AcOH and adding to 5 g. NaNO₂ and at 1-hr. intervals adding two 5-ml. portions of AcOH. The brown ppt. was filtered off, washed with glacial AcOH; alc., H₂O and finally with alc. to give 65-80% V, brown needles, m. 238-7° (aniline). Oxidn. of V with 1:1 HNO₃ (d. 1.42) and glacial AcOH produced 65% of 3,7-dinitrophenothiazine 5-oxide (VI), yellow needles, m. 309-10° (aniline). H₂O₂ oxidn. of VI yielded 85% 3,7-dinitrophenothiazine 5,5-dioxide (VII), m. 377-8° (PhNO₂ or PhNH₂). VII was also prepd. in 35% yield from direct nitration of II. 1,3,7,9-Tetranitrophenothiazine 5-oxide (VIII) was prepd. from 5 g. I, introduced in small amts. to a mixt. of HNO₃ (d. 1.50, 30 ml.) and 20% oleum (15 ml.) and then heated on a water-bath 20 min. The mixt. was then poured on ice and the ppt. filtered off, washed with H₂O and then alc. to give 85% VIII, yellow-orange needles, m. 354-5° (PhNO₂). Oxidn. of VIII with 30% H₂O₂ produced 90% 1,3,7,9-tetranitrophenothiazine 5,5-dioxide (IX), m. 348-7° (PhNO₂). Nitration with HNO₃-oleum (as for VIII) produced 90% IX from II and 75% IX from VII.

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III
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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Author : Bodea, C. and Raileanu, M.
Inst : Rumanian Academy of Sciences
Title : The Nitration of Phenothiazine

Orig Pub: Studii si Cercetari Chim Acad RPR, Fil Cluj, 8, No 3-4, 301-313 (1957) (in Rumanian with French and Russian summaries)

Abstract: For the purpose of studying the biological properties of nitroderivatives of phenothiazine (I), the authors have investigated the nitration of I and of the 5,5-dioxide of I (II). 3,7-dinitrophenothiazine (III) was prepared and oxidized to the 5,5-dioxide (IV) with H₂O₂; IV was also prepared by the nitration of II and

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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Abstract: 90%, mp 172-173° (from alc). A boiling solution of 10 gms IX in 85 ml CH₃COOH is treated with 10 ml 30% H₂O₂, heated for 1.5 hrs (adding 10 ml portions of H₂O₂ every 30 min for a total of 30 ml), and the solution is allowed to stand 14-16 hrs; the 5,5-dioxide of IX is obtained, yield 92%, mp 246-247° (from benzene); hydrolysis of the product obtained with 10% alcoholic KOH gives II, yield 78%, mp 255-256° (from CH₃COOH). A suspension of 5 gms III in 0.1 liter CH₃COOH is treated dropwise with a solution of HNO₃ (d 1.42) in CH₃COOH (1:1) until the solution of III is complete and the solution is diluted with water; V is obtained, yield 65%, mp 309-310° (from aniline). 5 gms V are oxidized

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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Abstract: or by the nitration of IV. -- A. Marin

Card 5/5

116

Country : RUMANIA
Category: Organic Chemistry. Organic Synthesis

G

Abs Jour: RZhKhim., No. 17, 1959, No. 60901

Author : Bodea, C.; Kulcsaru, M.

Inst : -

Title : Chloro-Nitro-Derivatives of Phenthiazine
Synthesized by the Direct Chlorination and
Nitration.

Orig Pub: Studii si cercetari chim. Acad. RPR Fil. Cluj,
1958, 9, No 1-4, 159-166

Abstract: The direct chlorination of phenthiazine (I) in
CCl₄ yields 3, 7-dichloro-I (II) and 1, 3, 7
9-tetrachloro-I (III); the products of mono- or
tri-chlorination, thereby, are not formed.

Card : 1/7

Country : RUMANIA
 Category: Organic Chemistry: Organic Synthesis

G

Abs Jour: RZMKhim., No. 17, 1959, No. 60901

nitro-7-c 1-acylenthiazines (VI, VII), and from II are obtained 5-oxides of 1-nitro- and 1, 9-di-nitro-3, 7-dichlorophenthiazines (VIII, IX). VI and VII are oxidized with H_2O_2 into the respective 5, 5-dioxides (VIa, VIIa). In the nitration of 5, 5-dioxide of 3, 7-dichloro-I (X) 5, 5-dioxides of VIII and IX are obtained (VIII a, IX a). In the nitration of IV, 9-nitro-IV (XI) is synthesized. V is oxidized into 5-oxide and 5, 5-dioxide of V (XII, XIII). To a weighed 21 gr I in 0.5 liter CH_3COOH are added, drop by drop and at a temperature $< 20^\circ$, 250 ml CH_3COOH , saturated Cl_2 , the mixture is then

Card : 3/7

"APPROVED FOR RELEASE: 03/14/2001" 10901 CIA-RDP86-00513R001344020009-8"

Country: RUMANIA
 Category: Organic Chemistry: Organic Synthesis

Abs Jour: Organic Chemistry: Organic Synthesis

poored into water, the residue is washed with acetone, thus obtaining III. The solution is then diluted with water and II is separated, yield 33% and of $223-227^\circ$ melting point (from benzene). Through the suspension, of 5 gr sulfone I in 280 ml CH_3COOH , Cl_2 is passed thus synthesizing IV, yield 54%, melting point $235-236^\circ$ (from benzene). 1 gr V is gradually introduced at $15-20^\circ$ into 50 ml HNO_3 ($d = 1.42$), pouring the solution into water and separating VI of $253-254^\circ$ melting point (from CH_3COOH); Analogically from 2 gr II and 50 ml HNO_3 ($d = 1.42$) IX is synthesized, yield 72%, melting point $299-300^\circ$ (from aniline). By the same method, while employ-

Card : 4/7

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Country : RUMANIA
Category: Organic Chemistry. Organic Synthesis.

Abs Jour: RZhKhim., No 17, 1959, No. 60901

alc.). In the nitration of X with HNO_3 ($d = 1.42$) VIIIa is obtained with yield of 80% and melting point of $302-303^\circ$ (from CH_3COOH). By employing HNO_3 ($d = 1.5$), IXa is obtained from X, yielding 82%, melting point $247-248^\circ$ (from CH_3COOH), and XI is derived from IV, yield 58%, melting point $277-278^\circ$ (from $\text{C}_6\text{H}_5\text{NO}_2$). Into a warm solution of 0.6 gr V in 60 gr CH_3COOH , 1 ml H_2O_2 is added, followed by 3-5 minutes heating, filtering, pouring into water, and producing XII, yield 58%, melting point $275-276^\circ$ (from CH_3COOH). To 400 ml CH_3COOH is added a solution containing 5 gr V in 60 ml of conc. H_2SO_4 , followed by heating, addition of 10 ml of 30% H_2O_2 , boiling, while

Card : 6/7

G-26

6-2-may

7 7 7
 Syntheses of heteroauxin, tryptamine, and serotonin.
 Costin D. Neutrescu and Dan Răileanu (Akad. Romanian
 Peoples Republic, Bucharest, Romania). *Chem. Ber.* 91,
 1131-5 (1958).—Powd. NaOAc (80 g.) added with stirring
 to 500 cc. Ac₂O, the mixt. treated rapidly with 100 g. o-
 HO₂CC₆H₄NaCCH₂CO₂H, m. 211°, refluxed 5 min. (CO₂
 evolution), distd. to remove the excess Ac₂O, the residue
 heated with stirring with 760 cc. H₂O to boiling, cooled,
 filtered, the crude 1,3-diacetylindoxyl (about 85 g.) added
 to 120 g. Na₂SO₄·7H₂O in 1.8 l. H₂O at 70°, the mixt.
 stirred 1.5 hrs. at 70° and cooled, and the product filtered
 off yielded 44 g. 1-acetylindoxyl (I), m. 188°; oxime, m.
 177° (decompn.) (50% EtOH); 2,4-dinitrophenylhydrazone,
 m. 202° (decompn.) (pyridine). NH₄OAc (4 g.) in 75 g.
 PhOH warmed, treated with 15 g. NCCCH₂CO₂H, 50 cc.
 xylene, and 15 g. l, refluxed 5 hrs. with the azeotropic re-
 moval of the H₂O, the mixt. steam distd. to remove the
 xylene, and the steam distn. residue allowed to stand over-
 night and recrystd. from aq. EtOH gave 10.5 g. 1-acetyl-3-
 indolylacetoneitrile (II), m. 112° (CCl₄). Crude II (20 g.)
 and 300 cc. 30% aq. NaOH refluxed 6-7 hrs. and acidified
 gave 13.5 g. 3-indolylacetic acid (heteroauxin) (III), m.
 164-5° (decompn.). II (1.2 g.) extd. into 1.0 g. LiAlH₄
 in 80 cc. abs. Et₂O and the soln. of the product in Et₂O
 treated with 1 g. cryst. (CO₂H)₂ in 5 cc. EtOH yielded 1.1 g.
 3-indolylethylamine (tryptamine)-(CO₂H)₂, m. 147°. 2,6-
 Cl(H₂N)C₆H₃CO₂H (20 g.) dissolved in 800 cc. 12.5%
 H₂SO₄ with warming, cooled, diazotized with 8 g. NaNO₂,
 added to 1 l. H₂O of 80°, and the product isolated with Et₂O
 yielded 16 g. 2,5-Cl(HO)C₆H₃CO₂H (IV), m. 172°. IV
 (34 g.) and 8 g. NaOH in 100 cc. H₂O and 300 cc. EtOH
 refluxed 2 hrs. with 70 cc. PhCH₂Cl, treated dropwise with

stirring at reflux during 2 hrs. with 200 cc. 17% aq. NaOH,
 refluxed 1 hr., freed of the EtOH, dild. with 600 cc. H₂O,
 washed with C₆H₆, treated with C, acidified with concd.
 HCl, and the cryst. ppt. recrystd. at -10° from 200 cc.
 EtOH and 100 cc. H₂O gave 38 g. 2,5-Cl(PhCH₂O)C₆H₃-
 CO₂H (V), m. 122° (EtOH). V (5 g.) and 2.5 g. H₂NCH₂-
 CO₂H dissolved with warming in a soln. of 2.5 g. KOH and
 2.5 g. K₂CO₃ in 40 cc. H₂O, the soln. refluxed 4 hrs. with a
 little Cu powder, filtered, dild. with 400 cc. H₂O, and
 acidified with concd. HCl yielded 5.7 g. 3,4-HO₂C(PhCH₂O)-
 C₆H₃NHCH₂CO₂H (VI), m. 221 (decompn.) (EtOH with
 H₂O). VI (15 g.) and 27 g. Na₂CO₃·10H₂O in 300 cc. H₂O
 treated dropwise with shaking with 5.7 g. Ac₂O, shaken 1
 hr., and acidified with concd. HCl to Congo red yielded
 15.5 g. N-Ac deriv. (VII) of VI, m. 110° with previous
 sintering. VII (5 g.) treated in the usual manner with 25
 cc. Ac₂O and 7 g. NaOAc yielded 4 g. 1,3-diacetyl-5-(benzyl-
 oxy)indoxyl (VIII), m. 110° (EtOH). VIII (6 g.) added to
 9.0 g. Na₂SO₄·7H₂O in 180 cc. H₂O and 180 cc. EtOH at
 reflux temp., refluxed 1 hr., and allowed to stand over-
 night yielded 3.5 g. 1-acetyl-5-(benzyloxy)indoxyl (IX), m.
 170° (EtOH). NCCCH₂CO₂H (4.5 g.), 13.5 cc. xylene, and
 1.5 g. IX added to 22.5 g. PhOH and 1.3 g. NH₄OAc and
 processed in the usual manner gave 1.25 g. 5-PhCH₂O
 deriv. (X) of II, m. 188° (iso-PrOH). X (0.82 g.) treated
 with 0.5 g. LiAlH₄ in Et₂O and the Et₂O soln. of the crude
 product treated with 1 g. (CO₂H)₂ in 5 cc. EtOH yielded
 0.76 g. neutral oxalate of the 5-PhCH₂O deriv. (XI) of III, m.
 182° (decompn.) (EtOH); acid oxalate, m. 197° (decompn.);
 (dioxane); HCl salt, 79% yield, m. 265° (EtOH-Et₂O);
 Neutral oxalate (0.2 g.) of XI in 15 cc. abs. EtOH treated

1/2 CG

Costin D. Hontarescu and Dan Răileanu

with 0.1 g. $(CO_2H)_2$ in EtOH, hydrogenated over 0.1 g. 10% Pd-C, and filtered, and the filtrate dild. with dry Et₂O pptd. 0.13 g. acid oxalate of serotonin, m. 198° (decomn.) (abs. EtOH-Et₂O). X (1 g.) in 50 cc. Me₂CO treated with 40 cc. 2% aq. NaOH and 10 cc. 30% H₂O₂, kept 24 hrs. at room temp., refluxed 5 min., evapd. *in vacuo*, and the residue cooled deposited 0.50 g. 5-benzyloxy-3-indolyl-acetamide, m. 148°.

CC
2/2

F. W. Hoffmann

[Handwritten signature]

6
2-may

RAILEANU, T., dr.; LONESCU, R., dr.; IEN, A.C., dr.; FARCAS, Gh., dr.;
ROZOR, Florica, dr.

Gastric polyps. Med. ... Ag '63.

1. Lucrare efectuata in Spitalul MTC din Oradea.
(STOMACH NEOPLASMS) (POLYPI)

RAILEANU, Valentin, ing.

Influence of limestone granulation on the physicochemical properties of self-founding agglomerates. Metalurgia constr mas 14 no.4:289-293 Ap '62.

1. Combinatul siderurgic Hunedoara.

IFTIME, P., ing.; ROSCA, P., ing.; RAILEANU, V., ing.

Devices for protection from dangerous axial move in steam turbines. Energetica Rum 11 no.12:593-600 D'63.

FOLTICSKA, Maria, ing.; RAILEANU, Vasile, ing.;

Increasing the productivity of the agglomerating machines
and widening of the fuel basis. Metalurgia Rum 15 no.5:
352-354 My '63.

TOLUBINSKIY, V.I. [Tolubyns'kyi, V.I.]; VOROB'YEV, P.I. [Vorobiov, P.I.];
RAILKO, G.A. [Railko, H.O.]; KOZLYUK, V.N. [Kozliuk, V.M.]

Pilot plant in Aleksandriya for studying the utilization of
lignite for power fuel production. Zbir.prats' Inst.tepl.AN
URSR no.23:49-56 '61. (MIRA 15:2)

(Aleksandriya--Coke)
(Lignite)

RAILKO, G.A.; BAZEYEV, Ye.T.

Industrial cyclone furnace. Energ. i elektrotekh. prom. no.1:
18-20 Ja-Mr '63. (MIRA 16:5)

(Furnaces)

RAILLO, A.

Griby Roda Fizarina (Fizarina Mushrooms)

410 p. 2.00

SO: Four Continent Book List, April 1954

RAILLO, A. I.

Fungi of the Genus Fusarium, State Publishers of Agricultural Literature, Moscow,
1950 415 pages. 464 07 R13.

SO: SIRA SI 90-53, 15 Dec 1953

RAILLO, A. I.

"Taxonomy of the Genus *Fusarium* and a Method for the Determination of the Species Belonging to It", Trudy Botanicheskogo Instituta Akademii Nauk SSSR, Seriya 2: Sporovye Rasteniia, no 3, 1936, pp 803-857 541 Sa 21P.

SO: SIRA SI-90-53, 15 Dec 1953

RAILLO, A. I.

"Diagnostical Estimation of Morphological and Cultural Characters of Species in the Genus *Fusarium*" Trudy po Zashchite Rastenii, Seria 2, #7, 1935, pp 5-99, 423 92.

SO: SIRA SI 90-53, 15 Dec 1953

AM

RAILLO (Mme A. I.). Диагностическая оценка морфологических и культуральных признаков у видов рода *Fusarium*. [Diagnostic evaluation of the morphological and cultural characters in the genus *Fusarium*.]—*Bull. Pl. Prot. Leningr.*, 1935, Ser. ii (Phytopath.), 7, pp. 5-100, 11 pl., 7 figs., 1935. [English summaries, pp. 37, 78-79. Received June, 1936.]

In the first part of this paper the author indicates the difficulties inherent in Wollenweber's [*R.A.M.*, x, p. 626] and Wollenweber's and Reinking's [*ibid.*, xv, p. 321] classification of the genus *Fusarium*, claiming that these are chiefly due to the fact that most of the characters used by them in establishing their system are casual and insufficient for diagnostic purposes. She then gives a detailed and fully tabulated account of her own studies of the morphological and cultural properties of 17 species of *Fusarium*. All the fungi were studied under standardized conditions on ordinary potato (var. Centifolia) or acid potato agar at 22° to 23° C. in diffuse light, and their pigmentation on

rice or potato slices. The records were made according to a standard method and the results, which were tested statistically, indicated that the only morphological character of diagnostic value distinguishing the species is the shape of the apical cell of the conidia derived from pionnotes, pseudopionnotes, or sporodochia, as the conidia borne on aerial mycelium are too variable to be of any use. Such characters as the curvature of the conidia, the number of septa, width of the conidia, and

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ADD 114 METALLURGICAL LITERATURE CLASSIFICATION

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the length of the apical cell can only serve to distinguish subspecies, varieties, or subvarieties, while cultural characters such as pigment, presence of sclerotia, and mode of spore formation are only indicative of forms. The data thus obtained are used in an attempt to outline a theoretical structure of the species in the genus *Fusarium* comparable with that already introduced into the classification of the phanerogams by Vavilov and other Russian authors, and which in the author's opinion should considerably simplify the present classification of the genus.

In the second part, the author describes her studies on the variability of the morphological and cultural behaviour of different single-spore isolates from single-spore cultures of the different species, the results of which showed that the shape of the apical conidial cell, the prevailing number of septa, and the shape of the curvature of the conidia remain constant in all the isolates derived from single-spore cultures. The length and width of the conidia, on the other hand, as well as the cultural characters, varied considerably in the isolates. On the whole, these results are considered to confirm the relative taxonomic value of the different characters as indicated in the first part of the paper.

AM

RAILLO (Mme A. I.). Систематика и методика определения видов рода *Fusarium*. [Taxonomy of the genus *Fusarium* and a method for the determination of the species belonging to it.]—*Acta Inst. Bot. Acad. Sci. U.R.S.S., Sér. II (Pl. Cryptogamae)*, 1936, 3, pp. 803-857, 6 pl., 5 figs., 1936. [English summary.]

In this paper the author gives a detailed description of the standardized method suggested by her for the systematic study of the fungi belonging to the genus *Fusarium* [*R.A.M.*, xv, p. 684], as well as instructions for the preparation of the standard culture media (potato agar and acidified potato agar) which she considers as best adapted to maintain a typical development of the organisms. She further recommends that the measurement of macroconidia be made at regular intervals of 15 days from the inception of the cultures to ensure that none is over 15 days old, since her observations showed that these organs are most characteristic of the given species on the 15th day of growth.

In the second part dichotomous keys are given for the determination of the sections and species of the genus, based on the work of Wollenweber, though the recent joint monograph of this author and Reinking [*ibid.*, xiv, p. 708] was received by the writer too late to be taken into consideration in this paper.

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

FROM SOURCE

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION										FROM SOURCE														
SUBJECT					SUBJECT WITH ONLY ONE					SUBJECT					SUBJECT WITH ONLY ONE									
A	B	C	D	E	F	G	H	I	J	K	L	M	N	O	P	Q	R	S	T	U	V	W	X	Y

ACC NR: AP6021699 SOURCE CODE: CZ/0032/66/016/001/0035/0040

AUTHOR: Raiman, J.--Rayman, Yu. (Engineer)

31
B

ORG: 'Victorious February' Works, Hradec Kralove (Zavody vitezneho unora)

TITLE: Ferritic structural steels for high-pressure vessels operating at low temperatures

SOURCE: Strojirnstvi, v. 16, no. 1, 1966, 35-40

TOPIC TAGS: structural steel, ferritic steel, welding technology

ABSTRACT: Experience is reported on the manufacture of large high-pressure vessels for chemical plants where they are exposed to low temperature. Czechoslovakia has 3 brands of ferritic steel suitable for this purpose (11 368 for temperatures to -50°C; 11 413, to -40°C; and 16 222, to -70°C) and each brand requires special technology and treatment. Particular attention is devoted to welding problems, composition of electrodes, heat treatment of welds, and the reliability of testing methods. Orig. art. has: 15 figures and 4 tables. [Based on author's Eng. abst.] [JPRS]

SUB CODE: 11, 13 / SUBM DATE: none / ORIG REF: 005 / OTH REF: 002

Card 1/1 *BB*

UDC: 620.165:669.15-194.57:669.14.018.41

RHEIMBEKOV, YE. S.

29

The Second All-Union Conference on Rhenium, sponsored by the Institute of Metallurgy imeni A. A. Baykov, Academy of Sciences USSR, and the State Institute of Rare Metals, was held in Moscow 19-21 November 1962. A total of 335 representatives from 83 scientific institutions and industrial establishments participated. Among the reports presented were the following: autoclave extraction of Re from Cu concentrates (A. P. Zelikman and A. A. Peredereyev); Re extraction from the gaseous phase (V. P. Savrayev and N. L. Peysakhov); recovery of Re by sorption and ion interchange (V. I. Bibikova, V. V. Il'ichenko, K. B. Lebedev, G. Sh. Tyurekhodzhayeva, V. V. Yermilov, Ye. S. Raimbekov, and M. I. Filimonov); production of carbonyl Re (A. A. Ginzburg); electrolytic production of high-purity Re and electroplating with Re (Z. M. Sominskaya and A. A. Nikitina); Re coatings on refractory metals produced by thermal dissociation of Re chlorides (A. N. Zelikman and N. V. Baryshnikov); plastic deformation and thermomechanical treatment of Re (V. I. Karavaytsev and Yu. A. Sokolov); growth of Re single crystals and effect of O₂ on their properties (Ye. M. Savitskiy and G. Ye. Chuprikov); Re-Mo, Re-W, and Re-precious-metal alloys (Ye. M. Savitskiy, M. A. Tylkina, and K. B. Povarova); synthesis of Re nitrides, silicides, phosphides, and selenides (G. V. Samsonov, V. A. Obolonchik, and V. S. Neshpor); weldability of Re-Mo and Re-W alloys (V. V. D'yachenko, B. P. Morozov, and G. N. Klobanov); new fields of application for Re and Re alloys (M. A. Tylkina and Ye. M. Savitskiy); and Re-Mo alloy for thermocouples (S. K. Danishevskiy, Yu. A. Kocherzhinskiy, and G. B. Lapp). [WW]

Tsvetnyye metally, no. 4, Apr 1963, pp 92-93

L 23876-65 EWT(m)/EPR/EWP(t)/EWP(b) Ps-4 IJP(c) JD/MLK
ACCESSION NR: AT5002755 S/0000/64/000/000/0040/0043

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B+1

AUTHOR: Lebedev, K. B.; Ageyev, S. A.; Okhotnikova, N. A.; Yermilov, V. V.;
Raimbekov, Ye. S.; Filimonov, M. I.

TITLE: Recovery of rhenium from copper concentrates by alkaline leaching, 4

SOURCE: Vsesoyuznoye soveshchaniye po probleme reniya. 2d, Moscow, 1962. Reniy
(Rhenium); trudy soveshchaniya. Moscow, Izd-vo Nauka, 1964, 40-43

TOPIC TAGS: rhenium, rhenium extraction, copper concentrate, alkaline leaching,
rhenium cementation, potassium perrhenate

ABSTRACT: The authors propose a method for recovering rhenium in which the concentrate (about 30% copper, 3% lead, 2% zinc, and 0.003% rhenium) is leached with sodium hydroxide, rhenium and lead go into solution, and their cementation is then carried out on zinc. A complete flow diagram of the process is given, and the procedure is described in detail. The method is applicable to both copper and copper-lead rhenium-containing concentrates. The final recovery of the metals is tentatively estimated as follows: rhenium in potassium perrhenat, 50-55%; lead in crude lead, 20-25%; zinc in sheet zinc, up to 2%. Orig. art. has: 1 figure

Card 1/2

L 23876-65

ACCESSION NR: AT5002755

and 1 formula.

ASSOCIATION: None

SUBMITTED: 05Aug64

NO REF SOV: 011

ENCL: 00

SUB CODE: MM

OTHER: 000

Card 2/2

ARIFKHODZHAYEV, S.A.; RAIMBOKOV, Z.

Using electronic digital computers in construction management.
Izv. AN Uz. SSR. Ser. tekhn. nauk 9 no.6:14-18 '65 (MIRA 19:1)

1. Institut mekhaniki i Vychislitel'nyy tsentr AN UzSSR. Submitted June 25, 1965.

RAIMOV, Kh.S.

Biological characteristics of the grain leaf beetle. Zashch.
rast. ot vred. i bol. 3 no.5:57 S-0 '58. (MIRA 11:10)
(Beetles) (Grain--Diseases and pests)

VERETENNIKOV, Leonid Porfir'yevich, kand. tekhn. nauk, dotsent;

~~RAIMOV, Mikhail Porfir'yevich, kand. tekhn. nauk, dotsent;~~ ~~CIA-RDP86-00513R001344020009-8"~~

~~nauchnyy sotrudnik~~

Use of an analog computer in studying the stability of a syn-
chronous generator in the small. Izv. vys. ucheb. zav.;
elektromekh. 5 no.7:796-809 '62. (MIRA 15:10)

(Electric generators)
(Electronic analog computers)

VEREJENNIKOV, Leonid Porfir'yevich; VITALEV, Viktor Ivanovich,
RAJNOV, Mikhail Mikhaylovich; VERNIKOV, V.A., doktor tekhn.
nauk, prof., laureat Leninskoy premii, retirement;
SHIROKHOV, Ye.I., nauchn. red.; (SHEVCHUK, G.V., red.


[Modeling, computer techniques, and transient processes
in electric ship propulsion systems] Modelirovanie, vy-
chislitel'naya tekhnika i perekhodnye protsessy v sudov-
nykh elektroenergeticheskikh sistemakh. Leningrad, Su-
dostroenie, 1964. 383 p. (K114 13:1)

SOV/144-59.2-2/19

AUTHORS: Ryabinin, I.A., Candidate of Technical Sciences. Lecturer
and Raimov, M.M., Senior Scientific Worker

TITLE: The Use of Electronic Simulators for Investigating the
Stability of Parallel Operation of Alternators of
Comparable Power

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy, Elektromekhanika,
1959. Nr 2, pp 11 - 26 (USSR)

ABSTRACT: The transient behaviour of coupled alternators is
strictly described by non-linear differential equations
which are not solvable in their most general form. Even
if the simpler problem of so-called static stability is
resorted to, the high order of the system is a great
disadvantage. It is, however, necessary to make a
complete assessment of static stability in connection with
present developments on the voltage- and frequency-
control of alternators taking hunting into account.
Refs 4, 9 and 10 describe how electronic computers, both
analogue and digital, have been used to solve similar
problems. The particular example studied is shown in 

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Figure 1 as two coupled alternators feeding a load. The equations used were first proposed by A.A. Gorev (Ref 2). They are the set of ten denoted by (A). The power balance equations (active and reactive) are (6) and the voltage-regulator equations for a phase-compounded machine are (B). The complete system is tenth order and has 22 non-linearities. If the influence of transient oscillations of current on the rotor motion is neglected the order may be reduced to 6 and the non-linearities to 14. The number of function generators necessary for a conventional solution was not available and the investigation was thus limited to "small" scale. The equation system must first be linearized as Eq (1). This process takes into account: 1 - losses in the stator circuit; 2 - the saliency of the alternator poles; 3 - transient electromagnetic processes in the field and stator windings; 4 - the torque-speed mechanical characteristic of the prime movers; 5 - the static characteristics of the load for voltage and frequency; 6 - the automatic voltage

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SOV/144-59-2-2/19

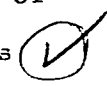
The Use of Electronic Simulators for Investigating the Stability of Parallel Operation of Alternators of Comparable Power

regulator. On the other hand, the influence of: a) the dynamic characteristics of the load; b) the dynamic characteristics of the prime mover speed regulator - are ignored. If it is supposed that each alternator is rated at 30 kVA and is loaded 50%, the insertion of numerical values gives Eq (2). The computer chosen is the MPT-9 containing 48 operational amplifiers with automatic zero setting. Each amplifier can sum up to 12 quantities. 48 constants and 48 variable coefficients can be set in. The variable is a direct voltage in the computing range ± 100 V. The maximum computing error does not exceed $\pm 0.5\%$. Drift is less than $100 \mu\text{V}$ over 10 minutes. The form of the equations as fed to the machine is Eq (5). The scale is chosen from a consideration of the peak values likely to be obtained. Each "machine" second represents 10 synchronous "alternator" seconds. Two coordinates, ΔS_{ω} and ΔU , have been excluded as less important. As written, Eq (5) would require 70 voltage

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dividers but the block diagram adopted (Figure 2) reduces this requirement to 64. Besides these, 11 summing, 10 integrating and 10 inverting amplifiers are used. The specific topics examined are: stability in the absence of voltage control; stability with voltage control; the effect of the slope of the torque-speed characteristic of the prime mover on the transient response; the effect of stator resistance on system behaviour. In each experiment, the constants have the same values and the disturbance is a unit function. Figure 3 was taken without voltage control. Figure 4 shows the effect of the excitation parameter K on parallel working. With $K > 1$ the system is unstable. Figure 5 shows the effect of damping torque when $K = 1.2$. The features specially stressed in this study are: relative simplicity of the method; comparatively little difficulty encountered; the direct, graphic appeal of the results; possibility of introducing non-linear effects. The disadvantage of relative inaccuracy is 

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The Use of Electronic Simulators for Investigating the Stability
of Parallel Operation of Alternators of Comparable Power

not important in this application.

There are 8 figures and 10 references; 9 of which
are Soviet and 1 English.

ASSOCIATION: Voyenno-Morskoy akademii korablestroyeniya i
voczuzheniya imeni A.N. Krylova (The A.N. Krylov Naval
Academy of Ship Building and Armaments)

SUBMITTED: December 15, 1958



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trudy TashGU no. 213 Geography no. 24:97-110 '63.
(MIRA 17:5)

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of a large city. Nauch. trudy TashG' no.251. Trudy Nauch.-issl.
otd. Geog. fak. no.3:10-17 '64. (MIRA 18:3)

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A.I., CHULKOV, M.P.; BORISENOK, V.G.; RAIMOVA, S.S.; KULIK,
O.A.; UDALOVA, L.I.; KAZACHKOV, S.S., otv. red.; ZHDANOVA,
L.P., red.

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1. Laboratory of Protozoan Cytology, Institute of Cytology of the Academy of Sciences of the U.S.S.R., Leningrad (Head: G. Poljansky)[Polzhanskiy, G.].

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PROFESSOR P. N. RAIKOV

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MANOLACHE, Mircea, conf. ing.; BODEA, Ion, asistent ing.; RAILEANU, Dumitru,
asistent ing.; SAS, Ion, asistent

Corrosion of aluminum and its alloys. Metalurgia constr mas 8 no.11:
937-950 N '61.

(Corrosion and anticorrosives)
(Aluminum alloys)

R/009/61/000/011/001/001
D282/D303

AUTHORS: Manolache, Mircea, Instructor, Engineer, Bodea, Ion,
Assistant, Engineer, Răileanu, Dumitru, Assistant,
Engineer, and Sas, Ion, Assistant

TITLE: On the corrosion of aluminum and its alloys

PERIODICAL: Metalurgia și construcția de mașini, no. 11, 1961, 937-950 ✓

TEXT: The article presents the results of experiments by the authors on the corrosion of aluminum and aluminum-alloy sheets in the various conditions of the Galați and Constanța harbors. The authors used in their experiments commercial aluminum of the following composition: 0.05% Fe, 0.31% Zn, 0.03% Mg, and the rest aluminum, as well as aluminum alloyed with 5% Cu and 3% Zn. Commercial aluminum was rolled into 1.5 - 2 mm thick sheets, while aluminum alloy into 4 - 6 mm thick sheets. The following

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corrosion media were selected: (1) Danube water; (2) Danube atmosphere; (3) Black Sea water; (4) Black Sea atmosphere; (5) town atmosphere of Galati; and (6) Sea water brought into the laboratory. Since in ship or harbor constructions the aluminum generally comes into contact with other materials, the authors selected the following contact hypotheses: (1) without any contact to other material; (2) in contact with OL 38 steel; (3) in contact with copper; (4) in contact with bronze mixed with tin; (5) in contact with fir-wood; (6) in contact with zinc; and (7) in contact with oak-wood. The samples were tested with or without protection, i.e. (1) without any protection; (2) anodically oxidated; (3) painted, and (4) anodically oxidated and painted. The results obtained by the authors completely verified the modern corrosion theories. Thus, in case of commercial aluminum, an anodic dissolution was produced on the samples. This anodic dissolution was increasingly reduced due to a passivity process. In case of samples made from aluminum alloyed with Cu and Zn, the corrosion velocity permanently increased due to the action of

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D282/D303

On the corrosion ...

the cathodic inclosures. A general passivity of the metal or alloy is only produced if there are some conditions of an anodic passivity of the anodic components. Knowing the appearance mechanism of the anodic passivity, the potential up to which the anode has to be polarized, can be calculated. Preliminarily oxidated aluminum samples were more electronegative; the potentials tended towards a stability, i.e. passivity; and the dissolution current had an increasing tendency. In case of aluminum samples alloyed with Cu and Zn, the potential and the current had a continuously increasing tendency. The powerful corrosion of the alloyed samples which in some cases even led to pitting, was especially due to an increase of the number and size of the cathodic inclosures. The corrosion of the commercial aluminum samples was characterized by a surface corrosion, while that of the aluminum-alloy samples by an intercrystalline corrosion. The most powerful corrosion effect on commercial aluminum samples was exerted by Black-Sea-water, while on aluminum alloy samples



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On the corrosion ...

by Black-Sea and Danube waters. The average corrosion depths in the case of commercial aluminum samples was 60μ , while in case of aluminum alloy samples it was almost 0.5 mm. However, the corrosion process did not vary proportionally with the time. The corrosion velocity increased the longer the aluminum alloy samples were kept in the corroding media, and decreased the longer the commercial aluminum samples were subjected to the activity of the corroding media. The authors draw the following preliminary conclusions: (a) Commercial aluminum is less corroded than aluminum alloyed with Cu and Zn. (b) The most powerful corrosion is produced by Black-Sea water, followed by Danube water. Sea water in the laboratory produced corrosion similar to corrosions produced by the sea-atmosphere. (c) Protecting layers have only delayed the corrosion of all samples submerged in natural waters, but proved to be more efficient in the case of samples subjected to atmospheric corrosion. (d) Generally, the contact materials increased the corrosion effects

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on the samples. (e) Most powerful corrosions were found on samples in contact with copper and bronze. (f) The decreasing order of the influence of the contact material on aluminum and aluminum-alloy samples, independently of the corroding media, was: copper, bronze, fir-wood, oak-wood, steel and zinc. (g) Zinc delayed the corrosion of aluminum and aluminum-alloys. The corrosion of all samples in the atmosphere was generally weak, superficial and uniform, being more powerful under a contact material. Red-lead proved to be a good protecting layer. Anodically oxidated and painted samples were not at all corroded, while painted samples were slightly corroded especially when being in contact with copper and bronze. There are 21 figures, 5 tables and 15 references: 11 Soviet-bloc and 4 non-Soviet-bloc. The reference to the English-language publication reads as follows: J. Sundarjan and T.L. Rama Char: "Inhibition of the Corrosion of Aluminum in Alkaline Solutions", Corrosion prevention and Control, 5, 1958, no. 5, 55-56. ✓

Card 5/5

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1. Institute of Chemistry of the Academy of the R.P.R. Bucharest.
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1. Chair of Geology of the Faculty of Geology and Geography of the
Bucharest University.

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(PANCREAS, innervation
anat. of nerve endings)

APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R001344020009-8

RUMANIA/Human and Animal Morphology (General)
Nervous System

Abs Jour : Ref Zhur - Biol., No 7, 1958, No 31227

Author : Niculescu Ion T, Hagi-Parschiv A., Enachescu Aurelia,
Bodescu Ruxanda, Raileanu Iona.

Inst : Not Given

Title : Concerning the Nerve Endings of the Digestive Apparatus: the
Stomach and the Duodenum.

Orig Pub : Bul. stiint. Sec. med., 1956, 8, No 1, 101-150

Abstract : The stomach and intestine of dogs, cats, rabbits, rats and
mice were stained with methylene blue; also used were im-
pregnation with silver, Nissle body stain in different cytol-
ogical methods. In the Auerbach system, the networks possess
ganglia which lie under the serous membrane. Especially
numerous are ganglia in the musculature of Oddi's sphincter.
The ganglia are rich in blood vessels. Ganglionic cells of
the Meissner plexus are smaller; they possess thinner synaptic
apparatuses. In the muscle layer, at different pH values, the

Card : 1/2

RAILEANU, LEANA
HAGI-PARASHIV, A., Assit. Prof.; ENACHESCU, A.; ONICESCU, Doina; SMOLENSCHI-POTIN,
Ludmila; BOCIRNEA, C.; LISSIEVICI-OPRESCU, Elena; DEDIU, St.; RAILEANU,
Ioana; RADU, Sorina

Contributions to the study of terminal nerve structures in the peritoneum.
Rumanian M. Rev. 2 no.2:9-10 Apr-June 58.

(PERITONEUM, innerv.
terminal nerve structures)

NICULESCU, I.T.; PARASCHIV, Hagi, A.; ENACHESCU, Aurelia.; BADESCU,
Ruxandra.; RAILEANU, Ioanna.

Study of the nerve endings of the liver and bile ducts. Bul. stiint.
sect. med. 7 no.2:609-620 Apr-June 55

(LIVER, innervation
nerve endings, in exper. animals, histol.)
(GALLBLADDER, innervation
nerve endings, in exper. animals, histol.)
(BILE DUCTS, innervation
nerve endings in exper. animals, histol.)

NICULESCU, Ion, T., membru corespondent al Academiei R.P.R.;
HAGI-PARASHIV, A.; ENACHESCU, Aurelia; BADESCU, Ruxanda;
RAILEANU, Ioana

The nerve endings of the digestive tract: stomach and duodenum.
Bul. stiint. sect. med. 8 no.1:101-150 Jan-Mar 56.

(NERVE ENDINGS
of stomach & duodenum, in various animals)
(STOMACH, innervation
nerve endings, in various animals)
(DUODENUM, innervation
neve endings, in various animals)

BERCEANUM, St.; GOCIU, Mariana; RAILEANU, Ioana

Studies of the behavior of hematopoietic cells in tissue culture. Stud. cercet. med. intern. 6 no.1:25-29 '65.

RAILORU, L.

Quality of products in state and cooperative commerce. p. 20.

Vol. 8, no. 1, Jan 1956
STANDARTECARIA
Bucuresti, Rumania

Source: East European Accession List. Library of Congress
Vol. 5, No. 3, August 1956

RAILEANU, M.

✓ Preparation of phenothiazine 5-oxide. C. Bodea and M. Raileanu (Acad. R.P.R., Cluj, Romania). *Acad. rep. populare Romine, Filiala Cluj, Studii cercetari chim.* 11, 129-33(1980).—Optimal conditions for prepg. phenothiazine (Ia) 5-oxide (I) from *N*-benzoylphenothiazine (II) were investigated, and a synthesis of I was achieved. II, obtained from 50 g. Ia and 32.5 cc. BzCl, was treated in 500 cc. AcH with 62.5 cc. HNO₃ (d 1.5), the mixt. agitated 10 min., and poured into H₂O to yield *N*-benzoylphenothiazine 5-oxide (III). III, recrystd. from PhMe was refluxed in 750 cc. EtOH 5 min., 150 cc. 10% aq. NaOH soln. added, the whole refluxed again 15 min., and cooled to give I, m. 257-8°. The yield (on Ia) was 84%. Treating III with Zn powder gave II. I and III were sol. in MeAc, AcH, moderately sol. in EtOH. I was slightly sol. in PhMe. T. Szall

3
JAJ(NB)

RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Author : Bodea, C. and Raileanu, M.
Inst : Rumanian Academy of Sciences
Title : The Nitration of Phenothiazine

Orig Pub: Studii si Cercetari Chim Acad RPR, Fil Cluj, 8, No 3-4, 301-313 (1957) (in Rumanian with French and Russian summaries)

Abstract: For the purpose of studying the biological properties of nitroderivatives of phenothiazine (I), the authors have investigated the nitration of I and of the 5,5-dioxide of I (II). 3,7-dinitrophenothiazine (III) was prepared and oxidized to the 5,5-dioxide (IV) with H₂O₂; IV was also prepared by the nitration of II and

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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Abstract: 90%, mp 172-173° (from alc). A boiling solution of 10 gms IX in 85 ml CH₃COOH is treated with 10 ml 30% H₂O₂, heated for 1.5 hrs (adding 10 ml portions of H₂O₂ every 30 min for a total of 30 ml), and the solution is allowed to stand 14-16 hrs; the 5,5-dioxide of IX is obtained, yield 92%, mp 246-247° (from benzene); hydrolysis of the product obtained with 10% alcoholic KOH gives II, yield 78%, mp 255-256° (from CH₃COOH). A suspension of 5 gms III in 0.1 liter CH₃COOH is treated dropwise with a solution of HNO₃ (d 1.42) in CH₃COOH (1:1) until the solution of III is complete and the solution is diluted with water; V is obtained, yield 65%, mp 309-310° (from aniline). 5 gms V are oxidized

Card 3/5

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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Abstract: with H₂O₂ (see IX) and IV is obtained, yield 85%, mp 337-338° (from C₆H₅NO₂ or aniline). 2 gms of powdered II are added gradually to a solution of 50 ml HNO₃ (d 1.5) in 150 ml water, the solution is stirred for 15-20 min, and allowed to stand for 8 hrs with periodic stirring; VI is obtained, yield 70%, mp 344-345° (from C₆H₅NO₂). 5 gms I are gradually added to a mixture of 30 ml HNO₃ (d 1.5) and 15 ml 20% oleum [fuming H₂SO₄], the mixture is heated for 20 min at 100°, and ice water is added to precipitate VII, yield 85%, mp 354-355° (from C₆H₅NO₂). A similar procedure is followed in preparing VIII from II, yield 90%, mp 346-347° (from C₆H₅NO₂). VIII is obtained in equal yield by the oxidation of VII with H₂O₂ or

Card 4/5

RAILEANU, M.

Distr: 4E3d

Nitration of phenothiazine, C. Bodea and M. Raileanu.
 (*Acad. rep. populare Romine, Fiziola Cluj*). *Studia chim. rom.* 8, 303-13(1957).—Through a systematic study of the nitration of phenothiazine (I) and phenothiazine 5,5-dioxide (II) it was shown that of all nitrated derivs. of I described only 3-nitrophenothiazine 5-oxide (III) corresponded to a pure product. H₂O₂ oxidn. of III produced 3-nitrophenothiazine 5,5-dioxide (IV). IV was also prepd. from II (2 g.) added slowly to a soln. of HNO₃ (50 ml. d. 1.50) and H₂O (150 ml.). After 8 hrs. with intermittent stirring, filtration, and washing with H₂O and alc., 70% IV was obtained, yellow prisms, m. 344-5° (PhNO₂). 3,7-Dinitrophenothiazine (V) was prepd. from 5 g. I by dissolving in CHCl₃ and 10 ml. AcOH and adding to 5 g. NaNO₂ and at 1-hr. intervals adding two 5-ml. portions of AcOH. The brown ppt. was filtered off, washed with glacial AcOH; alc., H₂O and finally with alc. to give 65-80% V, brown needles, m. 238-7° (aniline). Oxidn. of V with 1:1 HNO₃ (d. 1.42) and glacial AcOH produced 65% of 3,7-dinitrophenothiazine 5-oxide (VI), yellow needles, m. 309-10° (aniline). H₂O₂ oxidn. of VI yielded 85% 3,7-dinitrophenothiazine 5,5-dioxide (VII), m. 377-8° (PhNO₂ or PhNH₂). VII was also prepd. in 35% yield from direct nitration of II. 1,3,7,9-Tetranitrophenothiazine 5-oxide (VIII) was prepd. from 5 g. I, introduced in small amts. to a mixt. of HNO₃ (d. 1.50, 30 ml.) and 20% oleum (15 ml.) and then heated on a water-bath 20 min. The mixt. was then poured on ice and the ppt. filtered off, washed with H₂O and then alc. to give 85% VIII, yellow-orange needles, m. 354-5° (PhNO₂). Oxidn. of VIII with 30% H₂O₂ produced 90% 1,3,7,9-tetranitrophenothiazine 5,5-dioxide (IX), m. 348-7° (PhNO₂). Nitration with HNO₃-oleum (as for VIII) produced 90% IX from II and 75% IX from VII.

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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Author : Bodea, C. and Raileanu, M.
Inst : Rumanian Academy of Sciences
Title : The Nitration of Phenothiazine

Orig Pub: Studii si Cercetari Chim Acad RPR, Fil Cluj, 8, No 3-4, 301-313 (1957) (in Rumanian with French and Russian summaries)

Abstract: For the purpose of studying the biological properties of nitroderivatives of phenothiazine (I), the authors have investigated the nitration of I and of the 5,5-dioxide of I (II). 3,7-dinitrophenothiazine (III) was prepared and oxidized to the 5,5-dioxide (IV) with H₂O₂; IV was also prepared by the nitration of II and

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114

RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Abstract: 90%, mp 172-173° (from alc). A boiling solution of 10 gms IX in 85 ml CH₃COOH is treated with 10 ml 30% H₂O₂, heated for 1.5 hrs (adding 10 ml portions of H₂O₂ every 30 min for a total of 30 ml), and the solution is allowed to stand 14-16 hrs; the 5,5-dioxide of IX is obtained, yield 92%, mp 246-247° (from benzene); hydrolysis of the product obtained with 10% alcoholic KOH gives II, yield 78%, mp 255-256° (from CH₃COOH). A suspension of 5 gms III in 0.1 liter CH₃COOH is treated dropwise with a solution of HNO₃ (d 1.42) in CH₃COOH (1:1) until the solution of III is complete and the solution is diluted with water; V is obtained, yield 65%, mp 309-310° (from aniline). 5 gms V are oxidized

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RUMANIA / Organic Chemistry--Synthetic Organic Chemistry G-2

Abs Jour: Ref Zhur-Khimiya, No 8, 1959, 27510

Abstract: or by the nitration of IV. -- A. Marin

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Country : RUMANIA
Category: Organic Chemistry. Organic Synthesis

G

Abs Jour: RZhKhim., No. 17, 1959, No. 60901

Author : Bodea, C.; Kulacanu, M.

Inst : -

Title : Chloro-Nitro-Derivatives of Phenthiazine
Synthesized by the Direct Chlorination and
Nitration.

Orig Pub: Studii si cercetari chim. Acad. RPR Fil. Cluj,
1958, 9, No 1-4, 159-166

Abstract: The direct chlorination of phenthiazine (I) in
CCl₄ yields 3, 7-dichloro-I (II) and 1, 3, 7
9-tetrachloro-I (III); the products of mono- or
tri-chlorination, thereby, are not formed.

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Country : RUMANIA
 Category: Organic Chemistry: Organic Synthesis

G

Abs Jour: RZMKhim., No. 17, 1959, No. 60901

nitro-7-c 1-acylenthiazines (VI, VII), and from II are obtained 5-oxides of 1-nitro- and 1, 9-di-nitro-3, 7-dichlorophenthiazines (VIII, IX). VI and VII are oxidized with H_2O_2 into the respective 5, 5-dioxides (VIa, VIIa). In the nitration of 5, 5-dioxide of 3, 7-dichloro-I (X) 5, 5-dioxides of VIII and IX are obtained (VIII a, IX a). In the nitration of IV, 9-nitro-IV (XI) is synthesized. V is oxidized into 5-oxide and 5, 5-dioxide of V (XII, XIII). To a weighed 21 gr I in 0.5 liter CH_3COOH are added, drop by drop and at a temperature $< 20^\circ$, 250 ml CH_3COOH , saturated Cl_2 , the mixture is then

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Country: RUMANIA
 Category: RZMKhim., No. 17, 1959, No. 60901
 "APPROVED FOR RELEASE: 03/20/2001" CIA-RDP86-00513R001344020009-8"

Abs Jour: Organic Chemistry: Organic Synthesis

poored into water, the residue is washed with acetone, thus obtaining III. The solution is then diluted with water and II is separated, yield 33% and of $223-227^\circ$ melting point (from benzene). Through the suspension, of 5 gr sulfone I in 280 ml CH_3COOH , Cl_2 is passed thus synthesizing IV, yield 54%, melting point $235-236^\circ$ (from benzene). 1 gr V is gradually introduced at $15-20^\circ$ into 50 ml HNO_3 ($d = 1.42$), pouring the solution into water and separating VI of $253-254^\circ$ melting point (from CH_3COOH); Analogically from 2 gr II and 50 ml HNO_3 ($d = 1.42$) IX is synthesized, yield 72%, melting point $299-300^\circ$ (from aniline). By the same method, while employ-

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Country : RUMANIA
Category: Organic Chemistry. Organic Synthesis.

Abs Jour: RZhKhim., No 17, 1959, No. 60901

alc.). In the nitration of X with HNO_3 ($d = 1.42$) VIIIa is obtained with yield of 80% and melting point of $302-303^\circ$ (from CH_3COOH). By employing HNO_3 ($d = 1.5$), IXa is obtained from X, yielding 82%, melting point $247-248^\circ$ (from CH_3COOH), and XI is derived from IV, yield 58%, melting point $277-278^\circ$ (from $\text{C}_6\text{H}_5\text{NO}_2$). Into a warm solution of 0.6 gr V in 60 gr CH_3COOH , 1 ml H_2O_2 is added, followed by 3-5 minutes heating, filtering, pouring into water, and producing XII, yield 58%, melting point $275-276^\circ$ (from CH_3COOH). To 400 ml CH_3COOH is added a solution containing 5 gr V in 60 ml of conc. H_2SO_4 , followed by heating, addition of 10 ml of 30% H_2O_2 , boiling, while

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7 7 7
 Syntheses of heteroauxin, tryptamine, and serotonin.
 Costin D. Neutrescu and Dan Răileanu (Akad. Romanian
 Peoples Republic, Bucharest, Romania). *Chem. Ber.* 91,
 1131-5 (1958).—Powd. NaOAc (80 g.) added with stirring
 to 500 cc. Ac₂O, the mixt. treated rapidly with 100 g. o-
 HO₂CC₆H₄NaCCH₃CO₂H, m. 211°, refluxed 5 min. (CO₂
 evolution), distd. to remove the excess Ac₂O, the residue
 heated with stirring with 760 cc. H₂O to boiling, cooled,
 filtered, the crude 1,3-diacetylindoxyl (about 85 g.) added
 to 120 g. Na₂SO₄·7H₂O in 1.8 l. H₂O at 70°, the mixt.
 stirred 1.5 hrs. at 70° and cooled, and the product filtered
 off yielded 44 g. 1-acetylindoxyl (I), m. 188°; oxime, m.
 177° (decompn.) (50% EtOH); 2,4-dinitrophenylhydrazone,
 m. 202° (decompn.) (pyridine). NH₄OAc (4 g.) in 75 g.
 PhOH warmed, treated with 15 g. NCCCH₂CO₂H, 50 cc.
 xylene, and 15 g. l, refluxed 5 hrs. with the azeotropic re-
 moval of the H₂O, the mixt. steam distd. to remove the
 xylene, and the steam distn. residue allowed to stand over-
 night and recrystd. from aq. EtOH gave 10.5 g. 1-acetyl-3-
 indolylacetoneitrile (II), m. 112° (CCl₄). Crude II (20 g.)
 and 300 cc. 30% aq. NaOH refluxed 6-7 hrs. and acidified
 gave 13.5 g. 3-indolylacetic acid (heteroauxin) (III), m.
 164-5° (decompn.). II (1.2 g.) extd. into 1.0 g. LiAlH₄
 in 80 cc. abs. Et₂O and the soln. of the product in Et₂O
 treated with 1 g. cryst. (CO₂H)₂ in 5 cc. EtOH yielded 1.1 g.
 3-indolyethylamine (tryptamine)-(CO₂H)₂, m. 147°. 2,6-
 Cl(H₂N)C₆H₃CO₂H (20 g.) dissolved in 800 cc. 12.5%
 H₂SO₄ with warming, cooled, diazotized with 8 g. NaNO₂,
 added to 1 l. H₂O of 80°, and the product isolated with Et₂O
 yielded 16 g. 2,5-Cl(HO)C₆H₃CO₂H (IV), m. 172°. IV
 (34 g.) and 8 g. NaOH in 100 cc. H₂O and 300 cc. EtOH
 refluxed 2 hrs. with 70 cc. PhCH₂Cl, treated dropwise with

stirring at reflux during 2 hrs. with 200 cc. 17% aq. NaOH,
 refluxed 1 hr., freed of the EtOH, dild. with 600 cc. H₂O,
 washed with C₆H₆, treated with C, acidified with concd.
 HCl, and the cryst. ppt. recrystd. at -10° from 200 cc.
 EtOH and 100 cc. H₂O gave 38 g. 2,5-Cl(PhCH₂O)C₆H₃-
 CO₂H (V), m. 122° (EtOH). V (5 g.) and 2.5 g. H₂NCH₂-
 CO₂H dissolved with warming in a soln. of 2.5 g. KOH and
 2.5 g. K₂CO₃ in 40 cc. H₂O, the soln. refluxed 4 hrs. with a
 little Cu powder, filtered, dild. with 400 cc. H₂O, and
 acidified with concd. HCl yielded 5.7 g. 3,4-HO₂C(PhCH₂O)-
 C₆H₃NHCH₂CO₂H (VI), m. 221 (decompn.) (EtOH with
 H₂O). VI (15 g.) and 27 g. Na₂CO₃·10H₂O in 300 cc. H₂O
 treated dropwise with shaking with 5.7 g. Ac₂O, shaken 1
 hr., and acidified with concd. HCl to Congo red yielded
 15.5 g. N-Ac deriv. (VII) of VI, m. 110° with previous
 sintering. VII (5 g.) treated in the usual manner with 25
 cc. Ac₂O and 7 g. NaOAc yielded 4 g. 1,3-diacetyl-5-(benzyl-
 oxy)indoxyl (VIII), m. 110° (EtOH). VIII (6 g.) added to
 9.0 g. Na₂SO₄·7H₂O in 180 cc. H₂O and 180 cc. EtOH at
 reflux temp., refluxed 1 hr., and allowed to stand over-
 night yielded 3.5 g. 1-acetyl-5-(benzyloxy)indoxyl (IX), m.
 170° (EtOH). NCCCH₂CO₂H (4.5 g.), 13.5 cc. xylene, and
 1.5 g. IX added to 22.5 g. PhOH and 1.3 g. NH₄OAc and
 processed in the usual manner gave 1.25 g. 5-PhCH₂O
 deriv. (X) of II, m. 188° (iso-PrOH). X (0.82 g.) treated
 with 0.5 g. LiAlH₄ in Et₂O and the Et₂O soln. of the crude
 product treated with 1 g. (CO₂H)₂ in 5 cc. EtOH yielded
 0.76 g. neutral oxalate of the 5-PhCH₂O deriv. (XI) of III, m.
 182° (decompn.) (EtOH); acid oxalate, m. 197° (decompn.);
 (dioxane); HCl salt, 79% yield, m. 265° (EtOH-Et₂O);
 Neutral oxalate (0.2 g.) of XI in 15 cc. abs. EtOH treated

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Costin D. Hontarescu and Dan Răileanu

with 0.1 g. $(CO_2H)_2$ in EtOH, hydrogenated over 0.1 g. 10% Pd-C, and filtered, and the filtrate dild. with dry Et₂O pptd. 0.13 g. acid oxalate of serotonin, m. 198° (decompn.) (abs. EtOH-Et₂O). X (1 g.) in 50 cc. Me₂CO treated with 40 cc. 2% aq. NaOH and 10 cc. 30% H₂O₂, kept 24 hrs. at room temp., refluxed 5 min., evapd. *in vacuo*, and the residue cooled deposited 0.50 g. 5-benzoyloxy-3-indolyl-acetamide, m. 148°.

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F. W. Hoffmann

JH

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2-may

RAILEANU, T., dr.; LONESCU, R., dr.; IERU, A.C., dr.; FARCAS, Gh., dr.;
ROZOR, Florica, dr.

Gastric polyps. Med. ... Ag '63.

1. Lucrare efectuata in Spitalul MTC din Oradea.
(STOMACH NEOPLASMS) (POLYPI)

RAILEANU, Valentin, ing.

Influence of limestone granulation on the physicochemical properties of self-founding agglomerates. Metalurgia constr mas 14 no.4:289-293 Ap '62.

1. Combinatul siderurgic Hunedoara.

IFTIME, P., ing.; ROSCA, P., ing.; RAILEANU, V., ing.

Devices for protection from dangerous axial move in steam turbines. Energetica Rum 11 no.12:593-600 D'63.

FOLTICSKA, Maria, ing.; RAILEANU, Vasile, ing.;

Increasing the productivity of the agglomerating machines
and widening of the fuel basis. Metalurgia Rum 15 no.5:
352-354 My '63.

TOLUBINSKIY, V.I. [Tolubyns'kyi, V.I.]; VOROB'YEV, P.I. [Vorobiov, P.I.];
RAILKO, G.A. [Railko, H.O.]; KOZLYUK, V.N. [Kozliuk, V.M.]

Pilot plant in Aleksandriya for studying the utilization of
lignite for power fuel production. Zbir.prats' Inst.tepl.AN
URS SR no.23:49-56 '61. (MIRA 15:2)

(Aleksandriya--Coke)
(Lignite)

RAILKO, G.A.; BAZEYEV, Ye.T.

Industrial cyclone furnace. Energ. i elektrotekh. prom. no.1:
18-20 Ja-Mr '63. (MIRA 16:5)

(Furnaces)

RAILLO, A.

Griby Roda Fizerium (Fizerium Mushrooms)

410 p. 2.00

SO: Four Continent Book List, April 1954

RAILLO, A. I.

Fungi of the Genus Fusarium, State Publishers of Agricultural Literature, Moscow,
1950 415 pages. 464 07 R13.

SO: SIRA SI 90-53, 15 Dec 1953

RAILLO, A. I.

"Taxonomy of the Genus *Fusarium* and a Method for the Determination of the Species Belonging to It", Trudy Botanicheskogo Instituta Akademii Nauk SSSR, Seriya 2: Sporovye Rasteniia, no 3, 1936, pp 803-857 541 Sa 21P.

SO: SIRA SI-90-53, 15 Dec 1953

RAILLO, A. I.

"Diagnostical Estimation of Morphological and Cultural Characters of Species in the Genus *Fusarium*" Trudy po Zashchite Rastenii, Seria 2, #7, 1935, pp 5-99, 423 92.

SO: SIRA SI 90-53, 15 Dec 1953

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RAILLO (Mme A. I.). Диагностическая оценка морфологических и культуральных признаков у видов рода *Fusarium*. [Diagnostic evaluation of the morphological and cultural characters in the genus *Fusarium*.]—*Bull. Pl. Prot. Leningr.*, 1935, Ser. ii (Phytopath.), 7, pp. 5-100, 11 pl., 7 figs., 1935. [English summaries, pp. 37, 78-79. Received June, 1936.]

In the first part of this paper the author indicates the difficulties inherent in Wollenweber's [*R.A.M.*, x, p. 626] and Wollenweber's and Reinking's [*ibid.*, xv, p. 321] classification of the genus *Fusarium*, claiming that these are chiefly due to the fact that most of the characters used by them in establishing their system are casual and insufficient for diagnostic purposes. She then gives a detailed and fully tabulated account of her own studies of the morphological and cultural properties of 17 species of *Fusarium*. All the fungi were studied under standardized conditions on ordinary potato (var. Centifolia) or acid potato agar at 22° to 23° C. in diffuse light, and their pigmentation on

rice or potato slices. The records were made according to a standard method and the results, which were tested statistically, indicated that the only morphological character of diagnostic value distinguishing the species is the shape of the apical cell of the conidia derived from pionnotes, pseudopionnotes, or sporodochia, as the conidia borne on aerial mycelium are too variable to be of any use. Such characters as the curvature of the conidia, the number of septa, width of the conidia, and

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ADD 114 METALLURGICAL LITERATURE CLASSIFICATION

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
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the length of the apical cell can only serve to distinguish subspecies, varieties, or subvarieties, while cultural characters such as pigment, presence of sclerotia, and mode of spore formation are only indicative of forms. The data thus obtained are used in an attempt to outline a theoretical structure of the species in the genus *Fusarium* comparable with that already introduced into the classification of the phanerogams by Vavilov and other Russian authors, and which in the author's opinion should considerably simplify the present classification of the genus.

In the second part, the author describes her studies on the variability of the morphological and cultural behaviour of different single-spore isolates from single-spore cultures of the different species, the results of which showed that the shape of the apical conidial cell, the prevailing number of septa, and the shape of the curvature of the conidia remain constant in all the isolates derived from single-spore cultures. The length and width of the conidia, on the other hand, as well as the cultural characters, varied considerably in the isolates. On the whole, these results are considered to confirm the relative taxonomic value of the different characters as indicated in the first part of the paper.

AM

RAILLO (Mme A. I.). Систематика и методика определения видов рода *Fusarium*. [Taxonomy of the genus *Fusarium* and a method for the determination of the species belonging to it.]—*Acta Inst. Bot. Acad. Sci. U.R.S.S., Sér. II (Pl. Cryptogamae)*, 1936, 3, pp. 803-857, 6 pl., 5 figs., 1936. [English summary.]

In this paper the author gives a detailed description of the standardized method suggested by her for the systematic study of the fungi belonging to the genus *Fusarium* [*R.A.M.*, xv, p. 684], as well as instructions for the preparation of the standard culture media (potato agar and acidified potato agar) which she considers as best adapted to maintain a typical development of the organisms. She further recommends that the measurement of macroconidia be made at regular intervals of 15 days from the inception of the cultures to ensure that none is over 15 days old, since her observations showed that these organs are most characteristic of the given species on the 15th day of growth.

In the second part dichotomous keys are given for the determination of the sections and species of the genus, based on the work of Wollenweber, though the recent joint monograph of this author and Reinking [*ibid.*, xiv, p. 708] was received by the writer too late to be taken into consideration in this paper.

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

6-27-36

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION										6-27-36									
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ACC NR: AP6021699 SOURCE CODE: CZ/0032/66/016/001/0035/0040

AUTHOR: Raiman, J.--Rayman, Yu. (Engineer)

ORG: 'Victorious February' Works, Hradec Kralove (Zavody vitezneho unora)

TITLE: Ferritic structural steels for high-pressure vessels operating at low temperatures

SOURCE: Strojirnstvi, v. 16, no. 1, 1966, 35-40

TOPIC TAGS: structural steel, ferritic steel, welding technology

ABSTRACT: Experience is reported on the manufacture of large high-pressure vessels for chemical plants where they are exposed to low temperature. Czechoslovakia has 3 brands of ferritic steel suitable for this purpose (11 368 for temperatures to -50°C; 11 413, to -40°C; and 16 222, to -70°C) and each brand requires special technology and treatment. Particular attention is devoted to welding problems, composition of electrodes, heat treatment of welds, and the reliability of testing methods. Orig. art. has: 15 figures and 4 tables. [Based on author's Eng. abst.] [JPRS]

SUB CODE: 11, 13 / SUBM DATE: none / ORIG REF: 005 / OTH REF: 002

Card 1/1

UDC: 620.165:669.15-194.57:669.14.018.41

RHIMBEKOV, YE. S.

29

The Second All-Union Conference on Rhenium, sponsored by the Institute of Metallurgy imeni A. A. Baykov, Academy of Sciences USSR, and the State Institute of Rare Metals, was held in Moscow 19-21 November 1962. A total of 335 representatives from 83 scientific institutions and industrial establishments participated. Among the reports presented were the following: autoclave extraction of Re from Cu concentrates (A. P. Zelikman and A. A. Peredereyev); Re extraction from the gaseous phase (V. P. Savrayev and N. L. Peysakhov); recovery of Re by sorption and ion interchange (V. I. Bibikova, V. V. Il'ichenko, K. B. Lebedev, G. Sh. Tyurekhodzhayeva, V. V. Yermilov, Ye. S. Raimbekov, and M. I. Filimonov); production of carbonyl Re (A. A. Ginzburg); electrolytic production of high-purity Re and electroplating with Re (Z. M. Sominskaya and A. A. Nikitina); Re coatings on refractory metals produced by thermal dissociation of Re chlorides (A. N. Zelikman and N. V. Baryshnikov); plastic deformation and thermomechanical treatment of Re (V. I. Karavaytsev and Yu. A. Sokolov); growth of Re single crystals and effect of O₂ on their properties (Ye. M. Savitskiy and G. Ye. Chuprikov); Re-Mo, Re-W, and Re-precious-metal alloys (Ye. M. Savitskiy, M. A. Tylkina, and K. B. Povarova); synthesis of Re nitrides, silicides, phosphides, and selenides (G. V. Samsonov, V. A. Obolonchik, and V. S. Neshpor); weldability of Re-Mo and Re-W alloys (V. V. D'yachenko, B. P. Morozov, and G. N. Klobanov); new fields of application for Re and Re alloys (M. A. Tylkina and Ye. M. Savitskiy); and Re-Mo alloy for thermocouples (S. K. Danishevskiy, Yu. A. Kocherzhinskiy, and G. B. Lapp). [WW]

Tsvetnyye metally, no. 4, Apr 1963, pp 92-93

L 23876-65 EWT(m)/EPR/EWP(t)/EWP(b) Ps-4 IJP(c) JD/MLK
ACCESSION NR: AT5002755 S/0000/64/000/000/0040/0043

1
3
B+1

AUTHOR: Lebedev, K. B.; Ageyev, S. A.; Okhotnikova, N. A.; Yermilov, V. V.;
Raimbekov, Ye. S.; Filimonov, M. I.

TITLE: Recovery of rhenium from copper concentrates by alkaline leaching, 4

SOURCE: Vsesoyuznoye soveshchaniye po probleme reniya. 2d, Moscow, 1962. Reniy
(Rhenium); trudy soveshchaniya. Moscow, Izd-vo Nauka, 1964, 40-43

TOPIC TAGS: rhenium, rhenium extraction, copper concentrate, alkaline leaching,
rhenium cementation, potassium perrhenate

ABSTRACT: The authors propose a method for recovering rhenium in which the concentrate (about 30% copper, 3% lead, 2% zinc, and 0.003% rhenium) is leached with sodium hydroxide, rhenium and lead go into solution, and their cementation is then carried out on zinc. A complete flow diagram of the process is given, and the procedure is described in detail. The method is applicable to both copper and copper-lead rhenium-containing concentrates. The final recovery of the metals is tentatively estimated as follows: rhenium in potassium perrhenat, 50-55%; lead in crude lead, 20-25%; zinc in sheet zinc, up to 2%. Orig. art. has: 1 figure

Card 1/2

L 23876-65

ACCESSION NR: AT5002755

and 1 formula.

ASSOCIATION: None

SUBMITTED: 05Aug64

NO REF SOV: 011

ENCL: 00

SUB CODE: MM

OTHER: 000

Card 2/2

ARIFKHODZHAYEV, S.A.; RAIMBOKOV, Z.

Using electronic digital computers in construction management.
Izv. AN Uz. SSR. Ser. tekhn. nauk 9 no.6:14-18 '65 (MIRA 19:1)

1. Institut mekhaniki i Vychislitel'nyy tsentr AN UzSSR. Submitted June 25, 1965.

RAIMOV, Kh.S.

Biological characteristics of the grain leaf beetle. Zashch.
rast. ot vred. i bol. 3 no.5:57 S-0 '58. (MIRA 11:10)
(Beetles) (Grain--Diseases and pests)

VERETENNIKOV, Leonid Porfir'yevich, kand. tekhn. nauk, dotsent;

~~RAIMOV, Mikhail Porfir'yevich, kand. tekhn. nauk, dotsent;~~ **APPROVED FOR RELEASE: 03/20/2001** **CIA-RDP86-00513R001344020009-8"**

~~nauchnyy sotrudnik~~

Use of an analog computer in studying the stability of a syn-
chronous generator in the small. Izv. vys. ucheb. zav.;
elektromekh. 5 no.7:796-809 '62. (MIRA 15:10)

(Electric generators)
(Electronic analog computers)

VEREJENNIKOV, Leonid Porfir'yevich; VITALEV, Aleksandr Ivanovich,
RAJNOV, Mikhail Mikhaylovich; VERNIKOV, V.A., doktor tekhn.
nauk, prof., laureat Leninskoy premii, retirement;
SHIROKHOV, Ye.I., nauchn. red.; (RAJNOV), S.V., red.

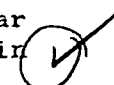
[Modeling, computer techniques, and transient processes
in electric ship propulsion systems] Modelirovanie, vy-
chislitel'naya tekhnika i perekhodnye protsessy v sudov-
nykh elektroenergeticheskikh sistemakh. Leningrad, Su-
dostroenie, 1964. 383 p. (K114 13:1)

SOV/144-59.2-2/19

AUTHORS: Ryabinin, I.A., Candidate of Technical Sciences. Lecturer
and Raimov, M.M., Senior Scientific Worker

TITLE: The Use of Electronic Simulators for Investigating the
Stability of Parallel Operation of Alternators of
Comparable Power

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy, Elektromekhanika,
1959. Nr 2, pp 11 - 26 (USSR)

ABSTRACT: The transient behaviour of coupled alternators is
strictly described by non-linear differential equations
which are not solvable in their most general form. Even
if the simpler problem of so-called static stability is
resorted to, the high order of the system is a great
disadvantage. It is, however, necessary to make a
complete assessment of static stability in connection with
present developments on the voltage- and frequency-
control of alternators taking hunting into account.
Refs 4, 9 and 10 describe how electronic computers, both
analogue and digital, have been used to solve similar
problems. The particular example studied is shown in 

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Figure 1 as two coupled alternators feeding a load. The equations used were first proposed by A.A. Gorev (Ref 2). They are the set of ten denoted by (A). The power balance equations (active and reactive) are (6) and the voltage-regulator equations for a phase-compounded machine are (B). The complete system is tenth order and has 22 non-linearities. If the influence of transient oscillations of current on the rotor motion is neglected the order may be reduced to 6 and the non-linearities to 14. The number of function generators necessary for a conventional solution was not available and the investigation was thus limited to "small" scale. The equation system must first be linearized as Eq (1). This process takes into account: 1 - losses in the stator circuit; 2 - the saliency of the alternator poles; 3 - transient electromagnetic processes in the field and stator windings; 4 - the torque-speed mechanical characteristic of the prime movers; 5 - the static characteristics of the load for voltage and frequency; 6 - the automatic voltage

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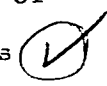
The Use of Electronic Simulators for Investigating the Stability of Parallel Operation of Alternators of Comparable Power

regulator. On the other hand, the influence of: a) the dynamic characteristics of the load; b) the dynamic characteristics of the prime mover speed regulator - are ignored. If it is supposed that each alternator is rated at 30 kVA and is loaded 50%, the insertion of numerical values gives Eq (2). The computer chosen is the MPT-9 containing 48 operational amplifiers with automatic zero setting. Each amplifier can sum up to 12 quantities. 48 constants and 48 variable coefficients can be set in. The variable is a direct voltage in the computing range ± 100 V. The maximum computing error does not exceed $\pm 0.5\%$. Drift is less than $100 \mu\text{V}$ over 10 minutes. The form of the equations as fed to the machine is Eq (5). The scale is chosen from a consideration of the peak values likely to be obtained. Each "machine" second represents 10 synchronous "alternator" seconds. Two coordinates, ΔS_{ω} and ΔU , have been excluded as less important. As written, Eq (5) would require 70 voltage

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dividers but the block diagram adopted (Figure 2) reduces this requirement to 64. Besides these, 11 summing, 10 integrating and 10 inverting amplifiers are used. The specific topics examined are: stability in the absence of voltage control; stability with voltage control; the effect of the slope of the torque-speed characteristic of the prime mover on the transient response; the effect of stator resistance on system behaviour. In each experiment, the constants have the same values and the disturbance is a unit function. Figure 3 was taken without voltage control. Figure 4 shows the effect of the excitation parameter K on parallel working. With $K > 1$ the system is unstable. Figure 5 shows the effect of damping torque when $K = 1.2$. The features specially stressed in this study are: relative simplicity of the method; comparatively little difficulty encountered; the direct, graphic appeal of the results; possibility of introducing non-linear effects. The disadvantage of relative inaccuracy is 

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not important in this application.

There are 8 figures and 10 references; 9 of which
are Soviet and 1 English.

ASSOCIATION: Voyenno-Morskoy akademii korablestroyeniya i
voczuzheniya imeni A.N. Krylova (The A.N. Krylov Naval
Academy of Ship Building and Armaments)

SUBMITTED: December 15, 1958



Card 5/5

RAIMOV, T.

Tashkent as an industrial and transportation center. Nauch.
trudy TashGU no. 213 Geography no. 24:97-110 '63.
(MIRA 17:5)

RAIMV, T.

Economic and geographical study of the construction industry
of a large city. Nauch. trudy TashG' no.251. Trudy Nauch.-issl.
otd. Geog. fak. no.3:10-17 '64. (MIRA 18:3)

GOLUBEVA, Z., strakhovoy delegat; RAIMOVA, N., strakhovoy delegat

Help for sick co-workers. Okhr. truda i sots. strakh. 5 no.9:
19 S '62. (MIRA 16:5)

1. Chulochno-trikotazhnaya fabrika, Smolensk.
(SMOLENSK--KNIT GOODS INDUSTRY--HYGIENIC ASPECTS)

RAIMOVA, N.I., assistant

Scalp forceps in individual forms of obstetrical pathology. Kaz.
med.zhur. 40 no.6:76-78 N-D '59. (MIRA 13:5)

1. Iz kafedry akusherstva i ginekologii (zav. - prof. A.B. Gillerson)
Omskogo meditsinskogo instituta im. M.I. Kalinina.
(FORCEPS, OBSTETRIC)

RAINOVA, N. I.

Card Med Sci - (diss) "Nearest (in time) and remote results of operation on the position of obstetrical forceps." Kazan', 1961. 18 pp; (Ministry of Public Health RSFSR, Kazan State Med Inst); 200 copies; free; (KI, 7-61 sun, 261)

MALKIN, S.I.; KULKOVA, N.K.

Effect of experimental hypercholesterolemia on the partial
degeneratization in frogs on the phenomenon of Bekkenov's
inhibition. Nauch. trudy Kaz. gos. med. inst. 14:221-222
1964. (MIRA 18:9)

1, ul. Poora normal'noy fiziologii (nav. - prof. I.N.Volkova)
Kazanskogo meditsinskogo instituta.

SHUMANOVA, A. A.; SOKOLOV, B.S.; CHERKASHENINA, Ye.F.; GARSKOVA,
A.I., CHULKOV, M.P.; BORISENOK, V.G.; RAIMOVA, S.S.; KULIK,
O.A.; UDALOVA, L.I.; KAZACHKOV, S.S., otv. red.; ZHDANOVA,
L.P., red.

[Agroclimatic manual on Omsk Province] Agroklimaticheskii
spravochnik po Omskoi oblasti. Leningrad, Gidrometeoizdat,
1959. 227 p. (MIRA 17:7)

1. Russia (1923- U.S.S.R.) Glavnoye upravleniye gidrometeo-
rologicheskoy sluzhby. Omskoye upravleniye. 2. Gidrometeoro-
logicheskaya observatoriya Omskogo upravleniya gidrometeorologicheskoy
sluzhby (for all except Kazachkov, Zhdanova).