

DELIMARSKIY, Yu.K.; OGNYANIK, S.S.

Polarographic investigation of iron and manganese oxides  
with a borosilicate melt as the support. Zhur. neorg. khim. 7  
no.8:1757-1761 Ag '62. (MIRA 16:6)

(Iron oxides) (Manganese oxides)  
(Polarography)

GOLUB, A.M.; OGNYANIK, S.B.

Thiocyanide complexes of tin (II) Ukr.khim.zhur. 27 no.3:283-290  
'61. (MIRA 14:11)

1. Kiyevskiy gosudarstvennyy universitet im. T.G.Shevchenko,  
kafedra neorganicheskoy khimii.

(Tin compounds)  
(Thiocyanic acid)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

OGNYAKOV, V.A.

Tin deposit in dolomites. Trudy VITR no. 4:255-261 '61.  
(MIRA 14:9)  
(Tin ores) (Dolomite)

OGNYAKOV, M. N.

Organization of wages in the construction of the Bratsk Hydroelectric Power Station. Sots.trud 7 no.1:120-123 Ja '62.  
(MIRA 15:4)

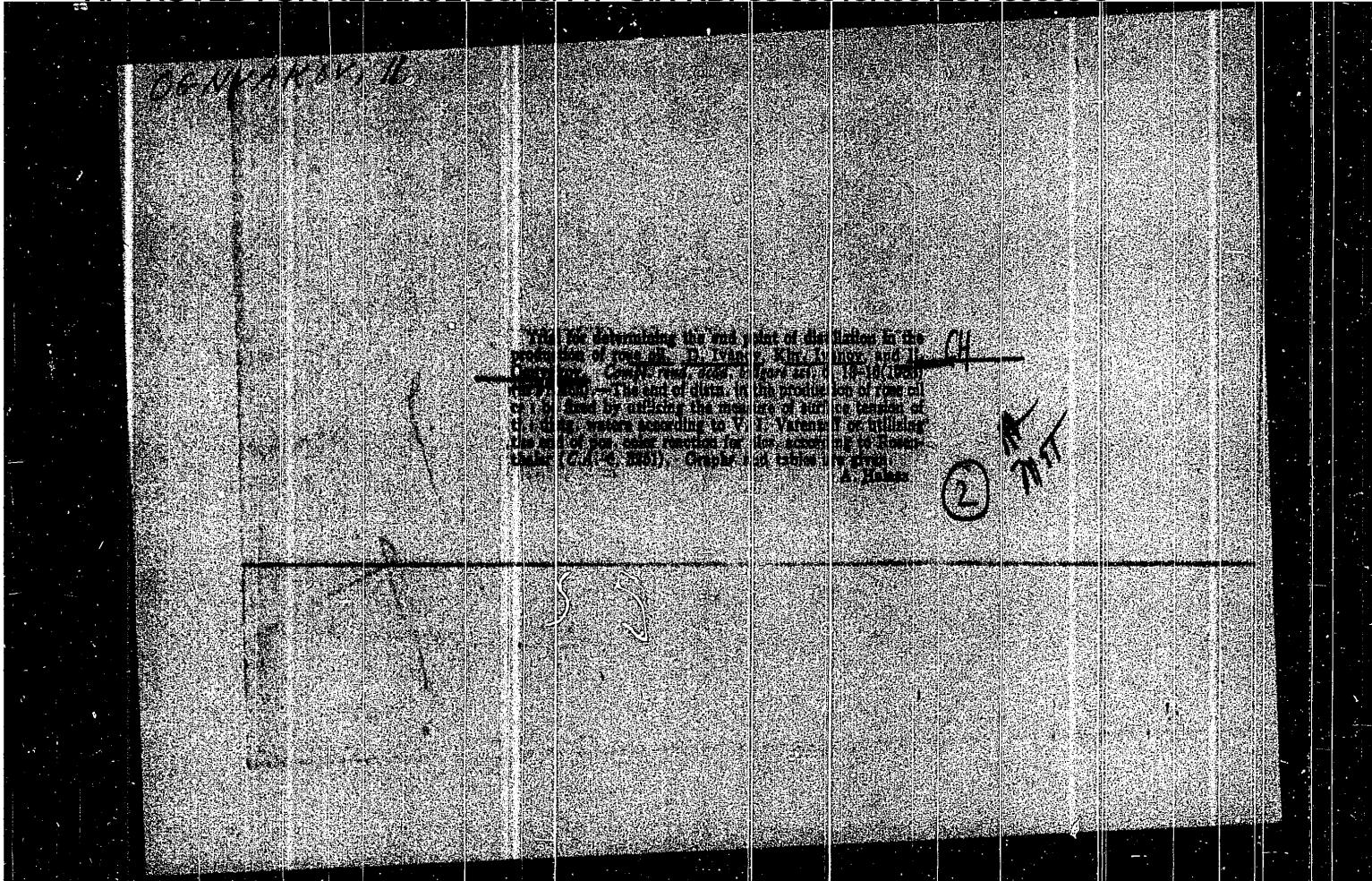
1. Nachal'nik ot dela i zarabotnoy platy stroitel'stva  
Bratskoy gidroelektrostantsii.  
(Bratsk Hydroelectric Power Station—Wages—Construction industry)

OGNYAKOV, M.N.

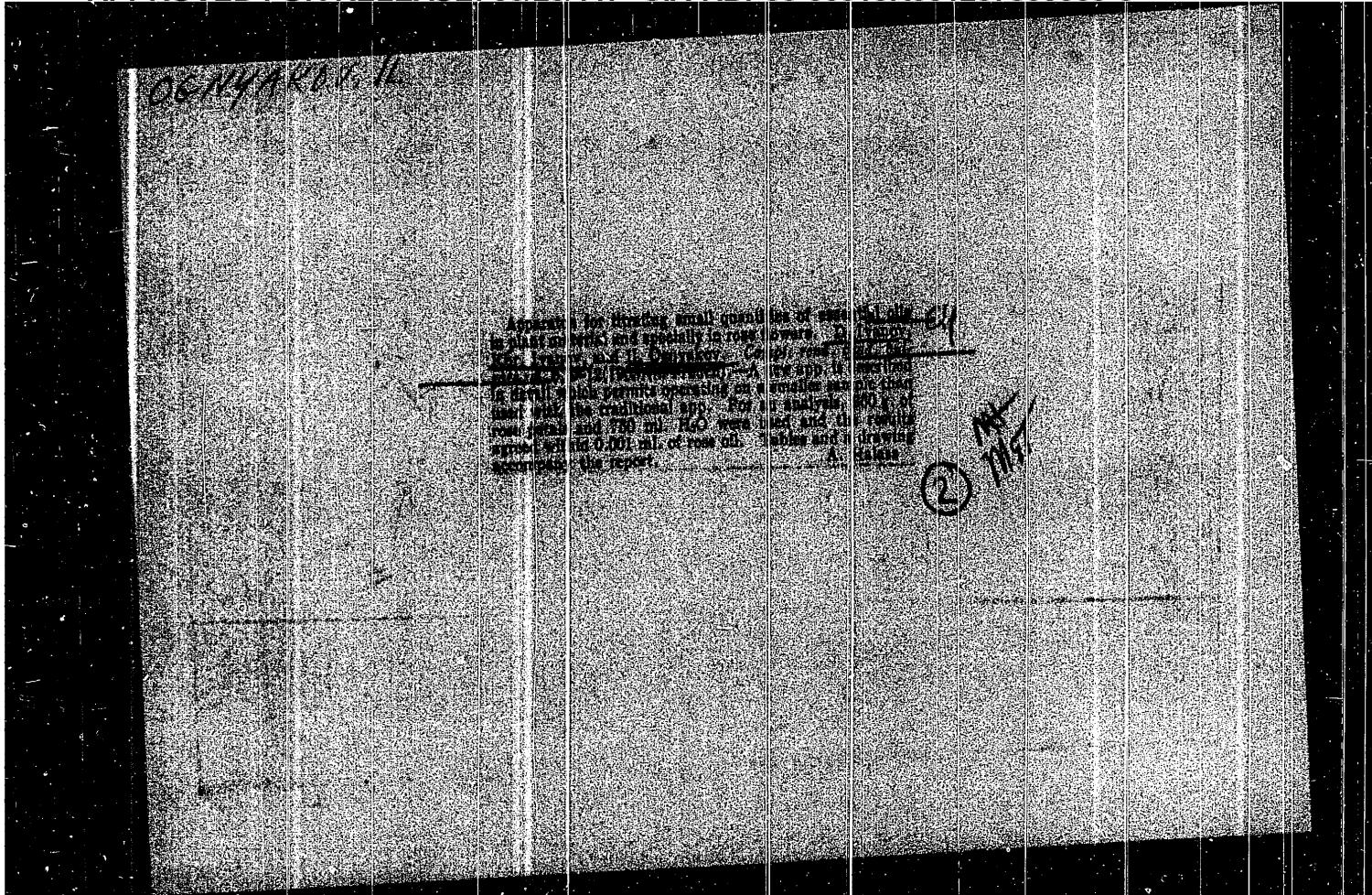
Increase of labor productivity is the basis of the struggle for  
communist labor. Energ.stroi. no.25:7-10 '61. (MIRA 15:4)

1. Stroitel'stvo Bratskoy gidroelektrostantsii.  
(Bratsk Hydroelectric power station--Design and construction)  
(Efficiency, Industrial)

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OGNOV, I.

Microorganisms, Nitrogen-Fixing

Effect of an azotobacter preparation in increasing the yield of grain and vegetable crops. Dost. sel'khoz. No. 3, 1953.

Monthly List of Russian Accessions, Library of Congress, June 1953. Unclassified.

OGNjanovic, Marija

Respiratory diseases and preschool and school children in  
Novi Sad. Med. pregl. 18 no. 3: 89-98 ' 65.

1. Skolski Antituberkulozni dispanzer, Novi Sad (Upravnik:  
Prim. dr. Marija Ognjanovic).

OGNJANOVIC, Marija

Results of ambulatory therapy of tuberculosis in school children  
and adolescents in the city of Novy Sad during the period of  
1953-1957 inclusively. Tuberkuloza, Beogr. 12 no.1:120-129 '60.

1. Antituberkulozni dispanzer Skolske poliklinike NOO Novi Sad,  
(TUBERCULOSIS PULMONARY in inf. & child)

OGNJANOVIC, Marija, dr. ; MILOSEVIC, Angela, dr.

The problem of tuberculosis in school children of Novi Sad and its  
suburbs. Med. pregl., Novi Sad 7 no.4:292-299 1954.

1. Decji antitub. dispanzer Skolske poliklinike, Novi Sad, sef  
dr. Marija Ognjanovic.  
(TUBERCULOSIS, statist.  
Yugosl., in school child.)

STRASIMIROV, D.; OGNJANOV, M.

The effect of the summation of stress factors of different forces on the quantity of steroid hormones in human urine.  
Dokl. Bolg. akad. nauk 16 no.4:425-427 '63.

1. Vorgelegt von Akademiemitglied D. Orahovats.  
(17-KETOSTEROIDS) (URINE) (STRESS)

MOLLOV, N.; HAIMOVA, M.; TSCHERNEVA, N.; PECICARGOVA, N.; OGNJANOV, I.;  
PANOV, P.

On alkaloids of Aconitum ranunculaceafolium. Dokl. Bolg. akad.

nauk 17 no.1:251-254 '64.

1. Vorgelegt von B. Kurtev, korr. Mitglied der Akademie.

OONIVENKO, V.V. [Ohnivenko, V.V.], inzh.

Frame for the attachment of harrows. Mekh. sii'. hosp. 14  
no. 5:16 My '63. (MIRA 16:10)

1. Yelitno-nasinnits'koye gospodarstvo, Sinel'nikiv's'kogo  
rayonu, Dnipropetrov's'koi oblasti.

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

BEL'KOVA, L.N.; OGNIKOV, V.N.

Stratigraphy of Paleozoic strata of southwestern Altai. Mat. VSEGEI  
no. 9:65-69 '55. (MLRA 9:9)  
(Altai Mountains--Geology, Stratigraphic)

OZNIAOV, St., inzh., dots.

Tackle ropes for depth boring, and their rational utilization.  
Godishnik Min geol. inst 7 no.1:415-440 '60/'61.

1. Chlen na redaktsionnata kolegiia, "Godishnik na minno-geolozhkia institut."

OCHNIANOV, ST., dots. inzh.

Lowering of drive pipes in the pits. Godishnik Min geol inst  
7:143-168 '60/'61 [publ. '62].

1. Chlen na Redaktsionnata kolegiia, "Godishnik na Minno-  
geologhkiia institut."

OGNIANOV, S.

Basic function of clayey solutions in drilling. (To be contd.) p. 91.

Great Development of Soviet economics. p. 97.

MINNO DELO, Sofiya, Vol. 10, no. 1, Jan./Feb. 1955.

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, no. 10, Oct. 1955,  
Uncl.

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

OGNYANOV, ST.

Basic Functions of Argillaceous Solutions in the Sounding Process.  
Minno Delo (Mining), #2:73:Feb 55

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

OGNIANOV, St., dots. Inzh.

Cementation of drive pipes. Min delo 17 no.12018-21 D '62.

1. Minno-geologichki institut.

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

KHLEBAROVA, M.; OGMIANOV, M.

Patho-biochemistry of atherosclerosis. Suvr. med. (Sofia)  
15 no.6:23-36 1964

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

GAGOV, St.; OGNIANOV, M.

Blood cholesterol and lipoprotein level in hypertension and  
their modification in decreased blood pressure by means of  
low-frequency impulse currents. Izv. inst. fiziol. (Sofia)  
8:209-215 '64

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

STRASBIMIROV, D.; GAGOV, SU.; OGNIANOV, M.

Effect of the stimulation of the perirenal zone using low-frequency currents on the urinary excretion of steroids in hypertensive patients. Izv. Inst. fiziol. (Sofija) 7:29-35 '64.

STRASIMIROV, D. [Strashimirov, D.]; OGNJANOV, M. [Ognianov, M.]

Stress factors variable in force, and influence of their summation on the amount of steroid hormones in human urine.  
Doklady BAN 16 no. 4: 425-427 '63.

1. Physiologisches Institut der Bulgarischen Akademie der Wissenschaften. Vorgelegt von Akademiemitglied D. Orahovats [Orakhovats, D.].

OCHNIAOV, L.; LESEVA, I.

Method for the quantitative determination of azulene in Matricaria chamomilla oil. In English. p. 33.  
(DOKLADY, Vol. 9, no. 3, July/Sept. 1956, Sofia, Bulgaria.)

SO: Monthly List of East European Accessions (EEAL) LC, Vol. 6, no. 12, December 1957 Uncl.

OGNIANOV, Karl; RACHEVA, Boriana

Theory and practice of exchange blood transfusion in newborn  
infants with hemolytic disease. Suvar. med. 13 no.3:17-25 '62.

(EXCHANGE TRANSFUSION)  
(ERYTHROBLASTOSIS, METAL)

OGNIANOV, K.; RACHEVA, B., Predvaritelno suobshchenie

Supplementary therapy of hemolytic disease in newborn with  
periston N. Khirurgiia, Sofia 9 no.3:258-262 1956.

1. Vtori Gradski Hodilen Dom, Glaven lekar: Iv. Doganov.  
(ERYTHROBLASTOSIS, FETAL, therapy,  
polyvinylpyrrolidone (Bul))  
(POLYVINYL PYRROLIDONE, therapeutic use,  
erythroblastosis fetalis (Bul))

OONIANOV, K.; GANEV, P.V.

Roentgenologic localization and diagnosis of placenta praevia.  
Khirurgija, Sofia 7 no.9:528-538 1954.

1. I Gradski rodilen dom Tina Kirkova. Glaven lekar: prof.  
G.Boiadzhiev.  
(PLACENTA PRAEVIA, diagnosis,  
x-ray)

MOLLOV, N.; HAIMOVA, M. [Khaimova, M.]; TSCHERNEVA, P. [Cherneva, P.];  
PECIGARGOVA, N. [Pechigargova, N.]; OGNJANOV, I. [Ognjanov, I.];  
PANOV, P.

Alkaloids of Aconitum ranunculaceum. Doklady BAN 17 no.3:  
251-254 '64.

1. Vorgelegt von B.Kurtev, korr. Mitglied d. Akademie.

VLAHOV, R. [Vlakhov, R.]; OGNYANOV, I. [Ognianov, I.]

Oxygen terpene compounds in Bulgarian peppermint oil.  
Doklady BAN 17 no.2:145-147 '64.

1. Institute of Organic Chemistry at the Bulgarian  
Academy of Sciences. Submitted by Academician  
D.Ivanov.

OCHYANOV, I.; DALEV, P.; DUTSCHEVSKA, H.; MOLLOV, N.

A new alkaloid from Vinca herbacea W.K. Dokl. Bokl. akad.  
nauk 17 no.1:153-156 '64

1. Vorgelegt von B.Kurtev korrig. Mitglied der Akademie.

OGNIANOV, I., k. kh. n., st. n. sutr.

Third International Congress on Essential Oils. Priroda Bulg  
13 no.5:111-113 S-O '64.

1. Institute of Organic Chemistry of the Bulgarian Academy  
of Sciences. *Институт по органическа химия*.

OGNYANOV, I. [Ognianov, I.]; MIHAIEV, M. [Mikhailov, M.]; BOURGOUDJIEV,  
Z. [Burgudzhiev, Z.]

Method for quantitative determination of germacrone in Geranium  
macrorrhizum L. oil. Doklady BAN 16 no.5:517-520 '63.

I. Institute of Organic Chemistry, Bulgarian Academy of Sciences,  
Sofia. Submitted by Academician D. Ivanov.

IVANOV, N.; OGNYANOV, I. [Ognianov, I.]

On the composition of neutral extractive substances in Bulgarian  
tobacco. Doklady BAN 16 no.3:293-296 '63.

1. Submitted by Corresponding Member B. Kourtev [Kurtev, B].

OGNYANOV, I. [Ognianov, I.]

Chromatographic separation of aromatic hydrocarbons on thin-layer  
non-fixed aluminium oxide. Doklady BAN 16 no.3:265-267 '63.

1. Submitted by Corresponding Member B. Kourtev [Kurtev, B.].

• . OGNYANOV, I. [Ognianov, I.]

Qualitative reaction for the detection of unsaturated compounds.  
Doklady BAN 16 no.2:161-162 '63.

1. Institute of Organic Chemistry at the Bulgarian Academy  
of Sciences. Sofia. Submitted by Corresponding Member B.  
Kourtev [Kurtev, B.].

OGNIANOV, I., kand. na khim. nauki

Some tasks of the Institute of Organic Chemistry, Bulgarian  
Academy of Sciences. Priroda Bulg 12 no. 5: 106-107 S-0 '63.

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OGNIANOV, I.

International Congress on Industrial Chemistry,  
Spisanie BAN 8 no. 4: 64-66 '63.

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OGNIAOV, I.; DALEV, P.

An apparatus for the concentration of extracts. Khim i industriia 34  
no.2:76 '62.

OGNYANOV, I. [Ognianov, I.]; VLAHOV, R. [Vlakhov, R.]

Nonterpene compounds in the terpene fraction of Bulgarian peppermint oil. Doklady BAN 15 no.4:395-397 '62.

1. Submitted by Academician D. Ivanoff [Ivanov, D.]. Chlen Redaktsionnoy kollegii, "Doklady Bolgarskoy akademii nauk."

OGNYANOV, I. [Ognianov, I]

Origin of  $\beta$ -elemenone in the oil of Bulgarian "zdravets" (*Ceranium macrorrhizum L.*). Doklady BAN 15 no.4:393-394 '62.

1. Submitted by Academician D. Ivanov. Chlen Redaktsionnoy kollegii, "Doklady Bolgarskoy akademii nauk."

II/081/63/000/001/037/061  
B144/B186

AUTHORS: Ognyanov, I., Dalev, P.

TITLE: A laboratory apparatus for concentrating extracts

PERIODICAL: Referativnyy shurnal. Khimiya, no. 1, 1963, 149-150,  
abstract 1D65 (Khimiya i industriya (Bulg.), no. 2, 1962,  
76 [Bulg.] )

TEXT: The apparatus is designed for concentrating organic extracts in vacuo. It comprises an evaporator (designed as ball condenser), a condenser, and two vessels to collect the extract and the solvent. The extract is fed into the top part of the heated evaporator, from where the evaporating solvent is led off into the condenser and collected in the container. The non-evaporating extract flows along the walls of the evaporator into a separate container. [Abstracter's note; Complete translation.]

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OGNYANOV, J.[Ognianov, I.]; VLAHOV, R.[Vlakhov, R.]

Terpene hydrocarbons in the Bulgarian peppermint oil. Doklady BAN 14,  
no.5:459-462 '61.

1. Institute of Organic Chemistry, Bulgarian Academy of Sciences.  
Submitted by Academician D. Ivanov.

(Terpenes) (Peppermint oil)

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Synthetic Analogs.

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Abs Jour: Ref Zhur-Khim., No 11, 1959, 38775.

is obtained, bp 129-131°/1.1 mm, mp 64-66° (from alcohol). When a mixture of 10 gms I, 5 ml. H<sub>2</sub>SO<sub>4</sub>, and 20 ml. alcohol is heated to 60°, XVI is obtained, yield 7.75 gms, bp 130.5-138.5°/2.5 mm, n<sup>20</sup><sub>D</sub> 1.5270, d<sup>20</sup><sub>40</sub> 0.9840; XVI forms two 2,4-dinitrophenylhydrazone derivatives melting at 195-196° and 184-185°. The hydrogenation of 2 gms XVI over Pt (from 50 mg PtO<sub>2</sub>) in glacial CH<sub>3</sub>OH gives XVII, yield 1 gm, mp 107.5-108.5° (from alc). The oxidation of 150 mg XVII by CrO<sub>3</sub> in glacial CH<sub>3</sub>COOH, followed by chromatography of the product on Al<sub>2</sub>O<sub>3</sub> (active towards I-III), gives XVIII, n<sup>20</sup><sub>D</sub> 1.4894, d<sup>20</sup><sub>40</sub> 0.9648. The dehydrogenation of 300 mg XVII by heating in 5 ml 100% HCOOH

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Abs Jour: Ref Zhur-Khim., No 11, 1959, 38775.

mp 183-183.5°. The hydrogenation of XIV (700 mg) in glacial CH<sub>3</sub>COOH by the same procedure as that used for I, followed by oxidation of the reaction product by CrO<sub>3</sub> in glacial CH<sub>3</sub>COON and purification by chromatography on Al<sub>2</sub>O<sub>3</sub> (active towards I-II) gives IV, bp 120-121°/5 mm, n<sup>25</sup>D 1.4760, d<sup>25</sup>17.5 0.9222. The hydrogenation of XIV (500 mg) over Pt (from PtO<sub>2</sub>) in alcohol (of procedure used for I), followed by reduction of the reaction product by LiAlH<sub>4</sub> in ether and chromatography on Al<sub>2</sub>O<sub>3</sub> (active towards I-II) gives 30 mg XII, bp 130-132°/20 mm, n<sup>20</sup>D 1.4720, d<sup>20</sup> 0.8576. When 2 gms XIV are reduced by LiAlH<sub>4</sub> (0.4 g) in ether solution, XV

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Abs Jour: Ief Zhur-Khin., No 11, 1959, 38775.

by chromatography of the product obtained on  $\text{Al}_2\text{O}_3$   
(active towards I-II) and elution with petroleum  
ether, gives XII, yield 320 mg, bp 132-135°/22 mm,  
 $n^{20}_{D}$  1.4717,  $d_{4}^{20}$  0.8538; elution with alcohol gives  
130 mg of III. The dehydration of 900 mg III with  
1.0 gm  $\text{KHSO}_4$  at 180-200° (30 min) gives XIII, yield  
750 mg, bp 114-115°/7 mm,  $n^{25}_{D}$  1.4799,  $d_{4}^{25}$  0.8594;  
the hydrogenation of the latter product (230 mg) over  
Pt (from 30 mg  $\text{PtO}_2$ ) in glacial  $\text{CH}_3\text{COOH}$ , gives XII,  
bp 115-117°/8 mm,  $n^{25}_{D}$  1.4704,  $d_{4}^{25}$  0.8562. When  
15 gms I are treated with  $\text{C}_2\text{H}_5\text{ONa}$  (from 2 gms Na  
and 50 ml alc, 12 hrs) gives XIV, yield 9.5 gms bp  
131-134°/1mm, mp 50-52°, 2,4-dinitrophenylhydrazone

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Abs Jour: Ref Zhur-Khim., No 11, 1959, 38775.

The hydrogenation of 312 mg VI in 5 ml glacial CH<sub>3</sub>COOH or freshly reduced PtO<sub>2</sub> gives VIII, bp 124°/14 mm, n<sup>20</sup><sub>D</sub> 1.4681, d<sup>20</sup><sub>D</sub> 0.8538. The hydrogenation of 390 mg V under the same conditions as described in the last case, followed by chromatography on Al<sub>2</sub>O<sub>3</sub> (active towards I-II) gives IX, bp 129-132°/14 mm, n<sup>20</sup><sub>D</sub> 1.4817, d<sup>20</sup><sub>D</sub> 0.8939. The hydrogenation of 3 gms I over Pt (from PtO<sub>2</sub>) in alcohol (7 days) with 4 mols of H<sub>2</sub>, followed by chromatography on Al<sub>2</sub>O<sub>3</sub>, gives X, yield 2.6 gms, bp 108°/0.8 mm, n<sup>20</sup><sub>D</sub> 1.5038, d<sup>20</sup><sub>D</sub> 0.9549. The reduction of 0.90 gm X by LiAlH<sub>4</sub> (0.15 gm) gives XI, bp 123-125°/2.5 mm. The hydrogenation of 500 mg XI in glacial CH<sub>3</sub>COOH over Pt (from PtO<sub>2</sub>) followed

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Abs Jour: Ref Zhur-khim., No 11, 1959, 38775.

duction of 2 gms I by aluminum isopropylate  
(from 1.0 gm Al) gives (after chromatography) 1.35  
gms V, bp 108-110°/0.15 mm. The reduction of 0.6  
gm I by LiAlH<sub>4</sub> gives V, bp 128°/0.8 mm, n<sup>25</sup>D  
1.5297. When 3 gms V and 7 ml of 100% HCOOH are  
heated quickly to boiling, VI is obtained, yield  
1.5 gm, bp 124.5-127°/11 mm, n<sup>25</sup>D 1.5250, d<sub>42</sub><sup>20</sup>  
0.9115. When 0.4 gms V is heated to boiling and  
chromatographed on Al<sub>2</sub>O<sub>3</sub> (active towards I-II)  
0.25 gm VI is obtained. When 1 gm V (or VI) and 0.25 gm  
S are heated for 30 min at 180°, followed by chroma-  
tography on Al<sub>2</sub>O<sub>3</sub> (active towards II), 0.25 gm of VII  
is obtained, trinitrobenzoate derivative mp 151-152°.

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Abs Jour: Ref Zhur-Khim., No 11, 1959, 38775.

latter product yields sclinanone (XVIII) [sic]. The dehydration of XVII followed by hydrogenation of the product obtained gives IX. Ic, mp 56-57° (from CH<sub>3</sub>OH),  $[\alpha]_D^{20}$  D 0° (chloroform). The hydrogenation of I (278 mg) in glacial CH<sub>3</sub>COOH over 30 mg of freshly reduced PtO<sub>2</sub> gives II, bp 134-135°, n<sup>14</sup>D 1.4774, d<sup>14</sup> 0.9162. The reduction of 2 gms II by LiAlH<sub>4</sub> in ether followed by chromatography on Al<sub>2</sub>O<sub>3</sub> (active towards I-II) [sic] gives III, yield 1.8 gm, bp 108.5-109.5°/0.05mm. The oxidation of 2 gms II with an excess of CrO<sub>3</sub> in glacial CH<sub>3</sub>COOH for 20 min at about 20° followed by chromatography on Al<sub>2</sub>O<sub>3</sub> (active towards I-II) gives IV, bp 131-136°/7 mm, n<sup>15</sup>D 1.4770, d<sup>15</sup> 0.9225. The

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Abs Jour: Ref Zhur-Khim., No 11, 1959, 38775.

tion of III gives a hydrocarbon (XIII). The ozonation of X yields 1 mol of acetone, while the ozonation of I gives 1.5-1.6 mol of acetone. The oxidation of I by  $\text{KMnO}_4$  in acetone gives  $(\text{COOH})_3$  and levulinic acid. The action of  $\text{C}_6\text{H}_5\text{ONa}$  on I in the cold gives crystalline isogermacrone (XIV). The hydrogenation of XIV in glacial  $\text{CH}_3\text{COOH}$  over Pt (from  $\text{PtO}_2$ ) gives IV; when the hydrogenation is carried out in alcohol, tetrahydrogermacrone (XV) is obtained. The acid isomerization of I yields a liquid ketone (XVI) which on hydrogenation over Pt (from  $\text{PtO}_2$ ) in glacial  $\text{CH}_3\text{COOH}$  with subsequent reduction by LiAlH<sub>4</sub> gives selinane (XVII) [sic: nomenclature]; oxidation of the

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Abs Jour: Ref Zhur-Khim., No 11, 1959, 38775.

or of VI with S gives guaiazulene (VII). The hydrogenation of VI in glacial  $\text{CH}_3\text{COOH}$  over Pt (from  $\text{PtO}_2$ ) with 4 moles of  $\text{H}_2$  gives elemene (VIII); the hydrogenation of V under the same conditions gives selinane (IX). The hydrogenation of I over Pt (from  $\text{PtO}_2$ ) in alcohol gives  $\alpha, \beta$ -unsaturated tetrahydrogermacrone (X). The reduction of X by LiAlH<sub>4</sub> gives tetrahydrogermacrol (XI) which on hydrogenation in glacial  $\text{CH}_3\text{COOH}$  over Pt (from  $\text{PtO}_2$ ) gives the saturated hydrocarbon germacrane (XII) and XIII. The IR spectrum of XII differs from the spectra of saturated sesquiterpenic hydrocarbons and resembles the spectra of humulane and farnesane (see RZhKhim, 1953, 8544). The dehydra-

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Synthetic Analogs.

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Abs Jour: Ref Zhur-Khim., No 11, 1959, 38775.

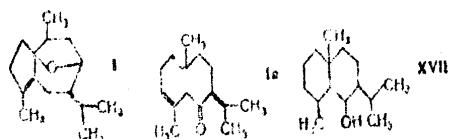
previously assumed, but that of the ketone (Ia).  
The authors propose the name Germacrone for Ia.  
The structure of Ia has been proved spectroscopically and by chemical reactions. The hydrogenation of I in glacial  $\text{CH}_3\text{COOH}$  over a Pt (from  $\text{PtO}_2$ ) catalyst gives a liquid product (II) which on reduction with  $\text{LiLi}_4$  gives hexahydrogermacrol (III). The oxidation of II with  $\text{CrO}_3$  in glacial  $\text{CH}_3\text{COOH}$  gives hexahydrogermacrone (IV). IV like Ia does not form a semicarbazone and 2,4-dinitrophenylhydrazone derivative. The reduction of I by aluminum isopropylate or by  $\text{LiLi}_4$  gives germacrol (V) which on dehydration yields a hydrocarbon (VI). The dehydrogenation of V

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Als Jour: Ref Zhar-Khim., No 11, 1959, 33775.



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CZECHOSLOVAKIA/Organic Chemistry. Natural Compounds and Their  
Synthetic Analogs.

G

Abstr Jour: Ref Zhur-Khim., No 11, 1959, 3677.

Author : Gmyanov, I., Ivanov, D., Hercut, V., Horek, M., Kliva,  
J., and Sohn, F.

Inst :

Title : Chemistry of the Terpenes. LXXXVII. Structure of  
Germacrene, the Crystalline Component of Bulgarian  
Medicinal Volatile Oil.

Orig Pub: Chem Listy, 52, No 6, 1163-1173 (1958) (in Czech)

Abstract: The authors have shown that the principal component  
of Bulgarian medicinal essential oil (*Geranium macro-*  
*rhizum L.*), previously designated *germacrol*, does not  
have the oxide structure (I) [see inset below], as

Card : 1/12

OGNIAKOV, I., AND OTHERS.

"Terpenes. LXXXVIII. Structure of germacrone, the crystalline constituent of the Bulgarian z davets oil." (In German)

COLLECTION OF CZECHOSLOVAK CHEMICAL COMMUNICATIONS, Praha, Czechoslovakia,  
Vol. 23, no. 11, Nov. 1958

Monthly list of EAST EUROPEAN ACQUISITIONS (EEA), LC, Vol. 8, No. 7, July 1959, Unclassified.

OZNIAOV, I.

"The structure of the germacrone (germacrol), a crystal substance in the Bulgarian Geranium macrorrhizum L. oil."

IZVESTIJA, Sofia, Bulgaria, Vol. 6, 1958.

Monthly List of East European Accessions Index (EEAI), The Library of Congress, Volume 8, No. 8, August 1959.

Unclassified

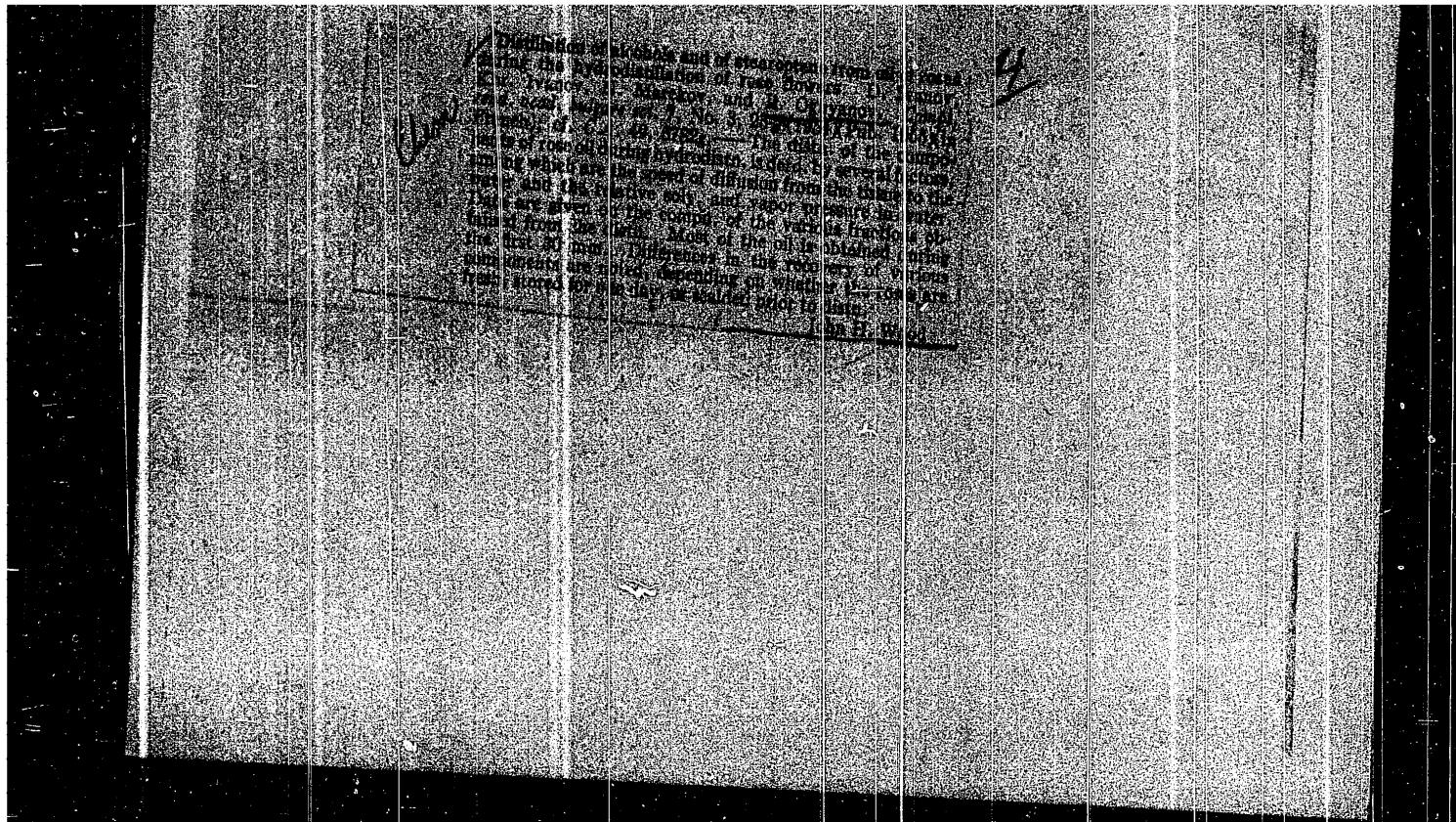
Ognyanov; I.

OGNYANOV, I.; ZAGOROVA, A.

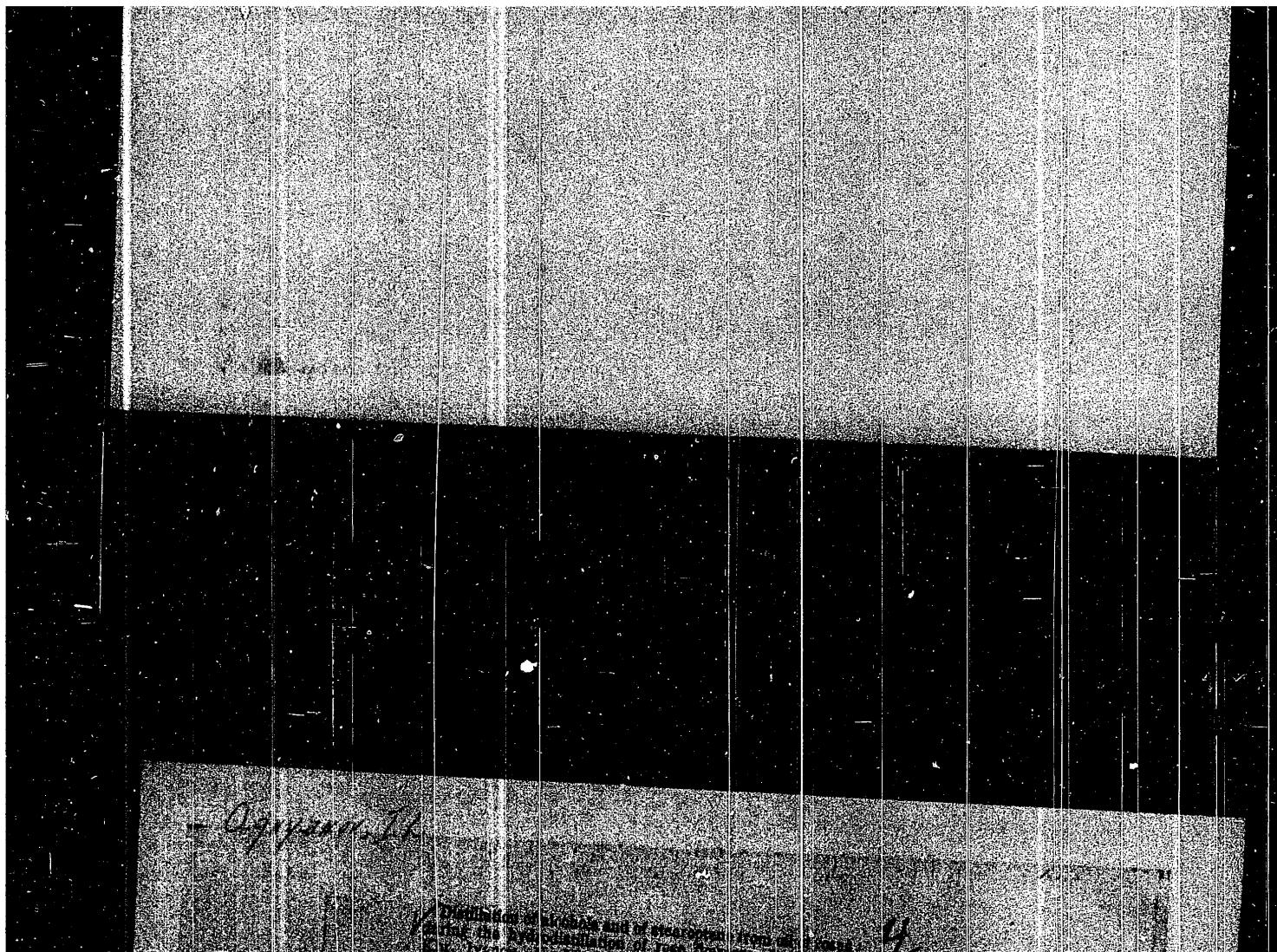
Preparation of n-toluenesulfonyl chloride. Zhur.prikl.khim. 29  
no.8:1299-1301 Ag '56.  
(MIRA 10:10)

1.Kafedra organicheskoy khimii Fiziko-matematicheskogo fakulteta  
Sofiyskogo universiteta, Bolgariya.  
(Toluenesulfonyl chloride)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

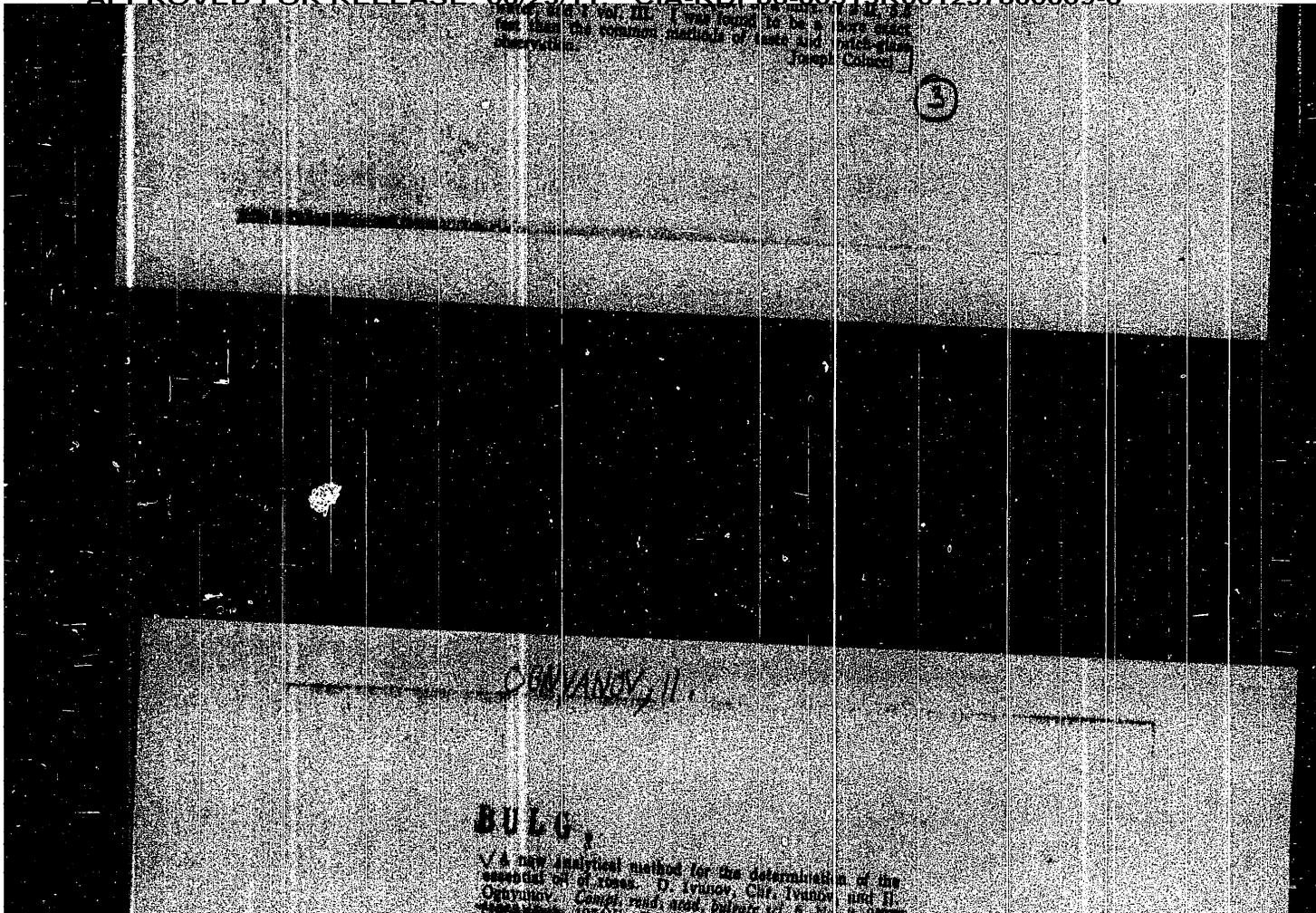


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SECRET  
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S E C R E T

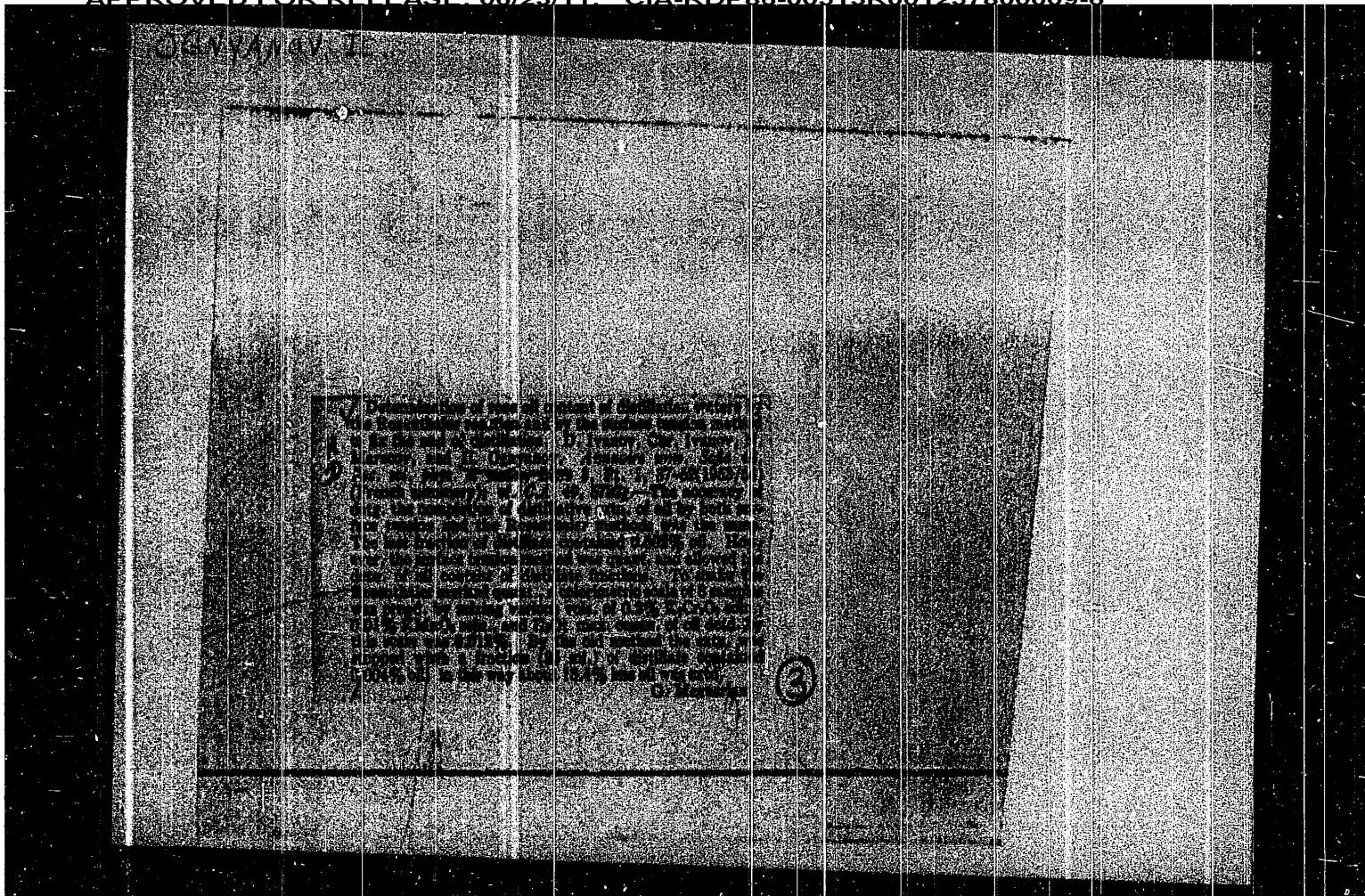
APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

DEPARTMENT OF DEFENSE

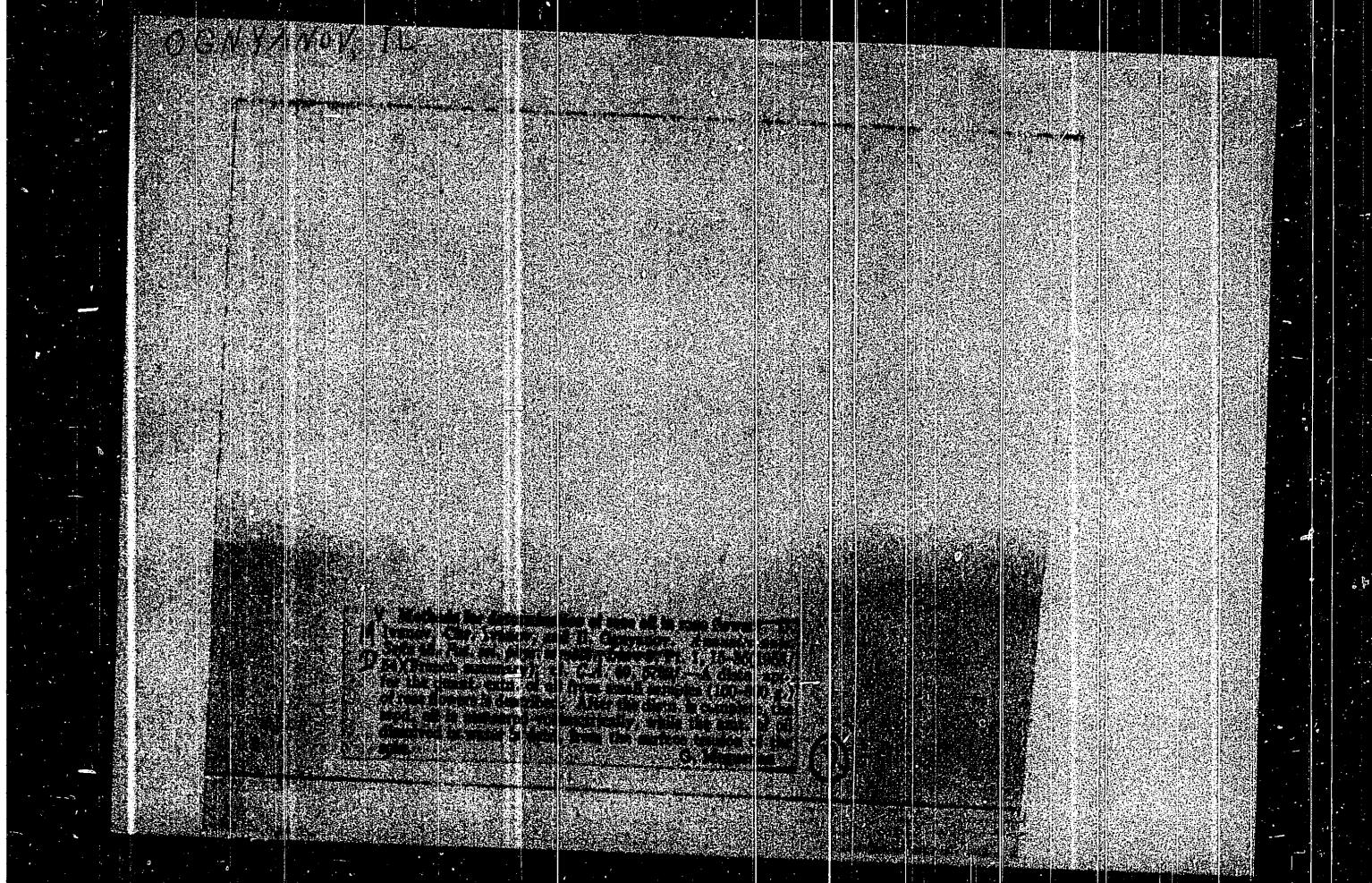
BULG

The composition of information from all sources available  
pertaining to the following subject(s) is requested:  
1. All information concerning the  
activities of the Bulgarian Communist Party  
and its armed forces.  
2. All information concerning the  
activities of the Bulgarian government  
and its armed forces.

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6



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APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

OGNIANOV, I.

Ognianov, I., Karakolev, G., Marinov, M., "Some Peculiarities of Pectin in the Jelling Process; the Influence of the Degree of Esterification on the Jelling Process." p.91  
(GODISCHIK, KHEMIIA, Vol. 47, 1952, Sofiya.)

SO: Monthly List of East European Accessions, Vol. 3, No. 3, Library of Congress,  
March, 1954, Uncl.

OGNYANOV, IL.

C.A. V-48

Jan 10, 1954

Food

Gelation of pectin. I. Irreversible change in pectin during gelation. II. Ognyanov, G. Karakolev, and M. Marinov, *Annuaire fac. sci.-phys. et math. (Sofia)*, *Chemis.* 47, 69-90 (1952) (German summary).—Gel (0.5 g.) from apple pectin (I) was dissolved in 20 ml. warm H<sub>2</sub>O, the soln. boiled to initial wt., the surface covered with liquid paraffin and left to gel; 24 hrs. later the gel strength and the relative viscosity of its aq. soln. were detd. This process was repeated several times. The results show that strength and viscosity decrease with repetition of gelation. Besides hydrolysis of I, homopolar bonding through lactone and ester formation could be responsible for the thermally irreversible change. II. Effect of degree of esterification upon gelation. *Ibid.* 91-103.—Above 80% esterification (II) the gel strength (III) reaches a max. at 64%; the gelation rate (IV) decreases between 8% and 61% and increases above that, and the optimum acidity increases with II. Below 80% II; III and IV depend upon Ca content; below 35% II pectin cannot be esterified in the absence of Ca.

G. Meguerian

Chem  
④

VLAHOV, R. [Vlahov, R.]; OGNYANOV, I. [Ognianov, I.]; TSANKOVA, E.

Composition of Bulgarian essential oil from *Foeniculum dulce* Mill  
fruit. Doklady BAN 17 no.6:569-572 '64.

I. Institute of Organic Chemistry at the Bulgarian Academy of  
Science, Sofia - Bulgaria. Submitted by Academician D. Ivanov.

SEMERDZHIEV, Bolan, dr.; GANILOV, D.; MAKAVREVA-SIMOVA, Ek.

Isolation of a virus agent of the psittacosis-ornithosis group  
from pneumoniae calves. Izv Vet inst virus 2:5-8 '63

1. Otgovoren redaktor i chlen na Redaktsionnata kolegia, "Iz-  
vestija na Veterinarski institut po virusologija" (for  
Semerdzhiev).

OGNIANOV, D.; ZHELEV, Vl.; SEMERIEV, Boian, d-r.; PAVLOV, N.; MAKAVEEVA  
SIMOVA, Ek.

Isolation of the virus, and some studies on the ovine abortion  
virus in Bulgaria. Izv Vet inst virus 1:37-51 '62

OGNIANOV, D.

Isolation of a virus agent of the psittacosis-ornithosis group  
pneumonic sheep. Izv Vet inst virus 2:25-30 '63

Some virusological studies on the abortion virus in sheep. Pts.  
1-2. Ibid. 31-45

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

SEMERZHIEV, Boian, dr.; GANTANOV, B.; MAKAVEEVA-SIMOVA, Ek.

Culture of the ovine abortion virus in experimental animals.  
Izv Vet Inst virus 1:27-35 '62

OGNITANOV, D.

Serological studies on the ovine abortion virus. Izv Vat inst virus 1:53-59 '62

Postvaccinal pareses in hens after the vaccination with E-vaccine, and their relationship with calcium-phosphate metabolism. Ibid. 8:105-109

Studies on an enzooty of the Q-fever in goats. Ibid. 8:159-163.

DELIMARSKIY, Yu.K.; OGNYANIK, S.S.

Polarographic study of slags. Part 1: Solutions of oxides of chromium, molybdenum, and tungsten. Ukr. khim. zhur. 29 no.9: 932-939 '63. (MIRA 17:4)

1. Institut obshchey i neorganicheskoy khimii AN UkrSSR i Institut elektrosvarki im. Ye.O.Patona AN UkrSSR.

GAITANDZHIEV, Georgi; KOLEV, Kol'o; OGNIANOV, Dimitur, KHRISTOFOROV,  
Liubomir.

Quality of the anthrax vaccine produced in Bulgaria, and  
results of its application after the Max Sterne method.  
Selskostop nauka 1 no.10:1131-1140 '62.

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

OGNEVSKIY, L.A.

The study of stero-regulated rubber, CKB(SKV) properties and their use  
in automobile tires.

Report submitted for the 4th Scientific research conference on the chemistry  
and technology of synthetic and natural rubber Yaroslavl, 1962

Properties of combined systems of ...

29457  
S/081/61/000/017/164/166  
B117/B110

the vulcanizates. If the formaldehyde aniline resin content is increased to 120 parts by weight, the swelling of the SKS-30-vulcanizates in the gasoline is reduced to one third. At the same time the brittle point is increased only by 10°. With increased filling the critical strain up to which the vulcanizate follows Hooke's law, increases. A substitution of the aniline portion by n-toluidine increased the strengths of the vulcanizates, the moduli and the resistances to wear, and also increased the plasticity of the mixture. [Abstracter's note: Complete translation.]

Card 2/2

15 9201

29457  
S/081/61/000/017/164/166  
B117/B110

AUTHORS: Morozov, A. D., Epshteyn, V. G., Ognevskiy, L. A., Gracheva, G. N.

TITLE: Properties of combined systems of different rubbers with resins obtained in the condensation of aromatic amines and formaldehyde

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 17, 1961, 596, abstract 17/316 (Uch. zap. Yaroslavsk. tekhnol. in-ta, v. 5, 1960, 115 - 125)

TEXT: The properties of combined systems of CKC-30 (SKS-30), CKH-18 (SKN-18) and CKH-26 (SKN-26) and formaldehyde aniline resins obtained by rolling lie between those of rubbers and those of plastics: increased modulus and strength, reduced relative and permanent extension, high hardness with good impact strength, high resistance to gasoline and frost. The SKS-30-vulcanizates with formaldehyde aniline resin are sufficiently temperature stable and resistant to thermal ageing. An increase in the HCOH-to-aniline ratio in the resin causes a higher gasoline resistance of

Card 1/2

OGNEVSKIY, A.; KUZ'MIN, V.

At a synthetic rubber plant. Pozh. delo 7 no. 2:7 F '61.

(MIRA 14:2

1. Glavnyy inzhener Yaroslovskogo zavoda sinteticheskogo kauchuka  
(for Ognevskiy). 2. Nachal'nik pozharnoy okhrany Yaroslovskogo  
zavoda sinteticheskogo kauchuka (for Kuz'min).  
(Yaroslavl--Rubber industry--Fires and fire prevention)

Polymerization of the bivinyl recovered from gases passed through the polymerizers. A. P. Ognevskii, I. P. Shashkin and A. P. Kryuchkov. *Sintet. Kaučuk* 6, No. 2, 23(1935).—The condensate, obtained from products passed once through the polymerizer in the usual manner yielded, on being passed again through the polymerizer, a semiliquid product. The gases were passed upward through 2 polymerizers with the usual Na catalyst, whereby an increase of the temp. was noticed after 5 days. The polymers obtained after 7 and 8 days amounted to 816 and 680 kg. of products of plasticities of 0.42 and 0.51, resp. A. A. Bochtingk

87918

3/138/60/000/008/004/015  
A051/A029

The Modification of the Properties of Synthetic Rubbers, Containing Active Functional Groups, Using Resorcin-Formaldehyde Resin in the Latex Stage

assumed, the increase in physico-mechanical properties caused by resorcin-formaldehyde is due to the formation of additional bonds between the copolymer chains containing active functional groups capable of interaction. The conclusion is drawn that the observed strength of the rubbers under investigation can be used in the production of highly-stable vulcanizates, with elevated elasticity and low heat formation. There are 7 figures, 1 table and 8 references: 5 Soviet, 1 French 2 English.

ASSOCIATION: Yaroslavskiy shinniy zavod (Yaroslavl' Tire Plant)

Card 4/6

87918

S/138/60/000/008/004/015  
A051/A029

The Modification of the Properties of Synthetic Rubbers, Containing Active Functional Groups, Using Resorcin-Formaldehyde Resin in the Latex Stage

and methylvinylpyridine rubbers (from 52 - 56 kg/cm). The optimum dosage of the resin would depend on the type of rubber being filled and the chemical nature of its functional groups and molecular chain (Fig. 3). It is assumed that the strengthening effect on rubbers with active functional groups from resorcin-formaldehyde resin is due to both the formation of chain-like structures from resin particles, adsorbed at the surface of the latex globules and by the substantial increase in the interaction between the rubber molecules and the filler particles. It was noted that further improvement of the physico-mechanical properties of the resin-filled rubbers could be accomplished by combining the resorcin-formaldehyde resin with carbon black. The latter also increases the rubber-filler gel. The wear resistance is increased when using two fillers (resin and carbon black). The latter exceed rubber filled only with resin by 12 - 20% according to laboratory findings. The following ratios of the resin and carbon black are assumed by the authors to be the optimum values (in weight parts to 100 weight parts of rubber): for SKS-30A, 15 resin, 15 - 20 carbon black; for SIS - 30 - 1, 10 resin, 10-15 carbon black; for SKMVP, 5 resin, 15-20 carbon black. As to the softener used in all the resin-filled rubbers, the most suitable was found to be pine tar. It is

Card 3/6

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87918

S/138/60/000/008/004/015  
A051/A029

The Modification of the Properties of Synthetic Rubbers, Containing Active Functional Groups, Using Resorcin-Formaldehyde Resin in the Latex Stage

-15) rubbers. The plasticity according to Carriere was 0.40 - 0.50. The presence of copolymers of active polar substitutes with acidic or basic properties in the molecular chain could affect the condensation process of the thermoreactive resin and thus affect the properties of the rubber-resin mixtures. The process of condensation took 22 - 24 hours at a normal temperature. The effect of the different ratios of the resorcin to the formaldehyde is shown in Figure 1. The optimum molar ratio of the resorcin to the formaldehyde in the strengthening of the methyl-vinylpyridine rubber was found to be 1 : 0.8, and for the butadiene-styrene and carboxyl-containing rubbers, it was found to be within the range of 1 : 1.5 to 1 : 1.8. Apparently the condensation of the SKMVP-15A rubber upon introducing lacquer resins, to the resol stage is activated on the surface of the globules by the pyridine groups having basic properties. In filling the carboxyl-containing and methyl-vinylpyridine rubbers, vulcanizates can be obtained with a tensile strength of 220 - 280 kg/cm<sup>2</sup> contrary to those of natural and butadiene-styrene rubbers. The tear-resistance of the resin-filled butadiene-styrene rubbers is found to be rather low (25 - 30 kg/cm), contrary to that of the carboxyl-containing

Card 2/6

15.9200 2109.2209

11.2211

87918

S/138/60/000/002/004/515  
A051/A029

AUTHORS: Boguslavskiy, D.B.; Epshteyn, V.G.; Ognevskaya, T.Ye.; Lyapina,  
L.A.; Lyubeznikov, V.K.

TITLE: The Modification of the Properties of Synthetic Rubbers, Containing  
Active Functional Groups, Using Resorcin-Formaldehyde Resin in the  
Latex Stage

PERIODICAL: Kauchuk i Rezina, 1960, No. 8, pp. 13 - 18

TEXT: The strengthening effect of resorcin-formaldehyde resin in synthetic  
rubbers was studied using the usual processing methods, such as coagulation, roll-  
ing and mixing, etc. It has been previously shown that in filling butadiene-  
styrene rubbers in the latex stage using resorcin-formaldehyde resin, the rubber  
mixtures produced are satisfactorily processed and the vulcanizates have sufficient-  
ly high physico-mechanical properties (Ref. 4). The properties of the filled rub-  
bers depend to a great extent on the amount of resin, the molar ratio of resorcin  
and formaldehyde and on several colloidal-chemical factors. The rubbers investi-  
gated were regulated carboxyl-containing (KC-30-1 (SKS-30-1) butadiene-styrene  
rubbers with 1.2% methacrylic acid, and 2-methyl-5-vinylpyridine (KMB-15 (SKMVP-

Card 1/6

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OCHIEVCHUK, N.A.; SOROKIN, F.G.; KALMANOV, N.Ye.; KIR'YANOV, Yu.A.

Horizontal slicing and filling mined areas with rubble concrete.  
Biul. TSIIN tavet. met. no.6:2-5 '58. (MIRA 11:7)  
(Mining engineering)

ZHUKOVSKIY, Yefim Semenovich; IVANOV, Nikolay Vasil'yevich,  
kand. ekon. nauk; KUPERMAN, Yakov Mironovich, kand.  
ekon. nauk; Prinimal uchastiye BUKSHTEIN, D.I.;  
VARENIK, Ye.I., prof., doktor tekhn. nauk, retsenzent;  
OGNEVAYA, N.V., kand. ekon. nauk, st. prepod., retsen-  
zent; USPENSKIY, V.V., kand. ekon. nauk, retsenzent;  
VERESHCHAGINA, V.Ya., red.

[Organization of procurement in construction] Organizatsiya  
snabzheniya stroitel'stva. Moskva, Vysshiaia shkola, 1965.  
283 p. (MIRA 18:8)

1. Zaveduyushchiy kafedroy "Ekonomiki i organizatsii  
stroitel'stva" Moskovskogo inzhenerno-ekonomiceskogo insti-  
tuta im. S.Ordzhonikidze (for Varenik). 2. Kafedra "Ekonomiki  
i organizatsii stroitel'stva" Moskovskogo inzhenerno-ekonomi-  
cheskogo instituta im. S.Ordzhonikidze (for Ognevaya).

BATUTEV, A.S. (Leningrad); OGNEVA, Ye.P. (Leningrad)

Teaching physiology in institutions for higher education.  
Fisiol.zhur. 51 no.11:1384-1385 N 465.

(MIRA 18:11)

OGNEVA, Y.I., inzh.

New technological processes for copper and nickel plating.  
Mashinostroenie no.6:58-61 N-D '62. (MIRA 16:2)

1. Proyektno-konstruktorskiy tekhnologicheskiy institut  
Kiyevskogo soveta narodnogo khozyaystva.  
(Copper plating) (Nickel plating)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

OGNEVA, T.A.

Errors in determining expenditures of heat to evaporation and  
the turbulent heat flux according to data of observations at  
a station network. Trudy GGO no.174:88-100 '65.

Diurnal variation of evaporation from the land surface. Ibid.:  
193-203 (MIRA 19:1)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

OGNEVA, T.A.

Methodology of wind measurements over inland waters. Trudy  
GGO no. 112:203-210 '63.  
(MIRA 17:5)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

OGNEVA, T.A.; PROKOF'YEVA, L.I.; FAN,A.A.

Possibilities for using gradient observations at network  
stations in determining the components of the heat balance  
of the active surface. Trudy GG no. 112:142-159 '63.  
(MIRA 17:5)

ACCESSION NR: AT4004718

true for all bodies of water but the one for humidity holds true only for very deep and large bodies of water. Consequently, only bodies of water at least 5 kilometers in size and 10 kilometers in depth influence the meteorological conditions of a surrounding area. The article mentions a number of previous researches in this field and relies heavily on their conclusions.

Orig. art. has: 3 figures and 3 tables.

ASSOCIATION: GGO, Leningrad

SUBMITTED: 00

DATA ACQ: 27Dec63

ENCL: 00

SUB CODE: ES

NO REF SOV: 008

OTHER: 000

ACCESSION NR: AT4004718

S/2922/63/007/000/0116/0124

AUTHOR: Ogneva, T. A. (Leningrad)

TITLE: Special features of the meteorological conditions of the layer of air above water

SOURCE: Vses. nauchn. meteorologich. soveshch. Trudy\*, v. 7, Fizika prizemnogo  
sloya. Leningrad, 1963, 116-124

TOPIC TAGS: meteorology, heat balance, radiation, water body radiation, turbulence,  
evaporation, humidity, thermal current, heat absorption, wind, wind velocity

ABSTRACT: The latest findings on the radiation and heat balance of water surfaces  
enable one to explain the fundamental features underlying the formation of meteorological  
conditions over bodies of water which are not very large, as well as to state the quantita-  
tive physical characteristics defining the meteorological conditions of the layer of air  
above the water. The basic conclusion is that the deviation in temperature and air  
humidity over water surfaces, in relationship to the features of the surrounding dry land,  
increases with an increase in depth and size of the bodies of water. Temperature and  
air humidity also increase somewhat in the direction north to south, as well as showing a  
seasonal variation. It should be noted that the seasonal variation for temperature holds

The Wind Profile in the Water-near Layer  
Above Lake Ladoga

S/020/61/136/003/014/027  
B019/B056

without taking the thermal stratification into account. The latter was carried out with a temperature difference of  $\Delta T$  between water and air. The small size of the island warranted conditions that were not influenced by land, as a comparison of the temperature measurements carried out on the island Khankhipasi and near it shows. In Table 1 the mean values of the vertical wind velocity profile and of the roughness parameter with and without taking the thermal stratification into account, were given. There are 3 figures, 1 table, and 7 references: 5 Soviet, 1 German, and 1 British.

ASSOCIATION: Laboratoriya Ozerovedeniya Akademii nauk SSSR (Laboratory of Lake Science of the Academy of Sciences, USSR)

PRESENTED: June 18, 1960, by D. V. Nalivkin, Academician

SUBMITTED: June 16, 1960

Card 2/~~A~~2

S/020/61/136/003/014/027  
B019/B056

AUTHORS: Izotova, A. F., Ogneva, T. A., and Smirnova, N. P.

TITLE: The Wind Profile in the Water-near Layer Above Lake Ladoga

PERIODICAL: Doklady Akademii nauk SSSR, 1961, Vol. 136, No. 3,  
pp. 587-590

TEXT: From July 7 to August 16, 1959, the vertical wind velocity profile above lake Ladoga, and its dependence upon the stratification temperature was studied, and also the roughness of the wind was determined. The observations were carried out by means of a remote anemometer with electric contact which had been designed at the Glavnaya geofizicheskaya observatoriya im. A. I. Voyeykova (Geophysical Main Observatory imeni A. I. Voyeykova). These anemometers were installed on a mast on the south side of the island Khankhipasi in altitudes of 6.15, 3.15, 1.65 and 0.75 m reckoned from the mast fundament. From the tape recordings the values averaged for one hour were used for the analysis. These analyses were carried out in consideration of the direction of the wind with and

Card 1/4 2

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AYZENSHTAT, B.A.; ANTROPOVA, U.I.; GRACHEVA, V.P.; OGNEVA, T.A.; SEROVA, N.V.

Thermal balance of the active surface. Trudy GGO no.107:34-43 '61.  
(MIRA 14:10)

(Solar radiation)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

VORONTSOV, P.A.; OGNEVA, T.A.; SEROVA, N.V.

Formation of the temperature regime of soil and air. Trudy GGO  
no.107:21-33 '61. (MIRA 14:10)  
(Soil temperature) (Atmospheric temperature)

VORONTSOV, P.A.; MISHCHERSKAYA, A.V.; SLEZNEVA, Ye.S.; CHESNAYA, I.I.;  
AYNBUND, M.M.; MIRILLOVA, T.V.; NESINA, L.V.; OGNEVA, T.A.;  
SEROVA, N.V.; TIMOFEEV, M.P., kand.fiz.-mat.nauk; ZEDANOVA, L.P.,  
red.; BRAYNINA, M.I., tekhn.red.

[Meteorological regime of Lake Sevan] Meteorologicheskii rezhim  
ozera Sevan. Pod red. M.P.Timofeeva. Leningrad, Gidrometeor.  
izd-vo, 1960. 310 p. (MIRA 14:3)

1. Leningrad. (Glavnaya geofizicheskaya observatoriya.  
(Sevan Lake region--Meteorology)

OGNEVA, T.A.; MKHITARYAN, A.M.; GAFAYAN, A.A.

Characteristics of turbulent exchange in the boundary layer next  
to the surface of lake Sevan. Izv.AN Arm.SSR.Ser.tekh.nauk. 12  
no.1:37-44 '59. (MIRA 12:4)

1. Vodno-energeticheskiy institut AN Arm.SSR.  
(Sevan, Lake) (Evaporation)

APPROVED FOR RELEASE: 06/23/11: CIA-RDP86-00513R001237800009-6

At the Final Meeting of the Scientific Council of the SOV/50-59-5-19/22  
Main Geophysical Observatory imeni A. I. Voevukov

of the actinometric net apparatus. D. P. Bespalov spoke about  
a net device for soil-temperature measurements in the depth.

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