ACC NKI AT 6033694

SOURCE CODE: UR/3231/66/000/002/0095/0103

AUTHOR: Valyus, V. P.; Levshin, A. L.; Sabitova, T. M.

ORG: none

TITLE: Combined interpretation of volume and surface waves for a region in Soviet Central

Asia

SOURCE: AN SSSR. Institut fiziki Zemli. Vychislitel'naya seysmologiya, no. 2, 1966.

Mashinnaya interpretatsiya seysmicheskikh voln (Machine interpretation of seismic waves),

95-103

TOPIC TAGS: seismic wave, data analysis, computer application, shock wave velocity

ABSTRACT: Observational ambiguity can be reduced by integrating the interpretation of different classes of observations, in particular surface and volume waves. Thus, the same set of data may correspond to a large number of different travel-time curves which markedly differ from each other in geophysical significance of the dynamics of the associated wave pattern. In this connection the article describes the solution by means of a computer of a converse problem of this type: The determination of the travel-time curves for the earth's crust in Central

Card 1/2

## ACC NR. AT6033694

Asia from Andizhan to Dushanbe according to the observed time of arrival of P and S waves, over the epicentral distances  $\Delta$ = 50, 100, 150, 250, and 350 km. These waves, recorded by the instruments of the Andizhan and Dushanbe seismic stations, originated from a series of earthquakes in the Kurile-Kamchatka zone. The observational data were converted for loading into a computer by the method of Azbel' et al. (Doklady II simpozium po teorii i vychislitel'nym metodam v geofizike, 1965) so as to correspond to a crystalline two-layer earth's crust overlain by a layer of sedimentary rocks and underlain by a mantle with a weak positive velocity gradients. Two variants of this method were employed: one, based on the premise that only a velocity discontinuity of the pressure gradients is possible in the crust and the other, postulating a velocity discontinuity with gradual acceleration in velocities. In both cases the various possible travel-time curves are successively checked by the Monte Carlo method on computing in every individual case the theoretical curves of the hodographs of surface waves  $t_P$  ( $\Delta$ ) and  $t_S$  ( $\Delta$ ) and the period  $v_{R_2}$  (T) (12-36 sec) and comparing them with curves plotted on the basis of observational data  $^2$  until the optimal curve can be selected. After a sufficiently comprehensive number of possible solutions (500-1000) is thus scanned on the basis of comparison with experimental data (time of arrival of various volume waves, amplitude curves of these waves, curves of variance of the phase and group velocity of any harmonic of Love and Rayleigh waves) and various rriterions of estimation, they can be craily realized in a specially compiled computer routing Orig. art. has: 3 figures, 3 tables.

SUB CODE: 08, 17, 10, 09/ SUBM DATE: none/ ORIG REF: 008/ OTH REF: 003

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LEVSHIN, A. M.

LEVSHIN, A. N. "Occurrence of Elytrosomes in the Leaves of Mosaic Diseased Sugar Beets," in Mosaic Diseases of Sugar Beets, a Collection of Articles, Publishing House of the Variety-Seed Administration of the State All Union Association of Sugar Industries, Kiev, 1930, pp. 177-78. 464.04 Sa2

SO: Sire Si-90-53 15 Dec. 1953

SAVCHENKO, YA. M.; YURTSOVSKIY, M. A.; IAYOK, V. D.; DRYAGINA, I. V.; LEVSHIN, A. N.

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New honey plants, Pchelovodstvo, 29, No. 10, 1952.

9. Monthly List of Russian Accessions, Library of Congress, November 1952. UNCLASSIFIED.

ARALOV, Viktor Aleksandrovich; LEVSHIN, Anatoliy Vladimirovich; KUDRYAVCHIKOVA, V., red.; TROYANOVSKAYA, N., tekhn.red. [Social insurance in the U.S.S.R.] Sotsial noe obespechenie v SSSR. Moskva, Gos.izd-vo polit.lit-ry, 1959. 94 p. (MIRA 12:8) (Insurance, Social)

VLASOV, Sergey Nikolayevich; LEYSHIN, Anatoliy Vladimirovich; LA-PIDUS, M.A., red.; DEYEVA, V.M., tekhn. red.

[Social insurance for collective farmers] Sotsial noe obespechenie kolkhoznikov. Moskva, Gos. izd-vo sel'khoz. lit-ry, 1960. 70 p.

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TRET YAKOV, A.F.; BOGOLEPOV, N.K.; ZIMKIHA, A.M.; SPIVAK, F.N.;

BURBYKO, V.M.; AVERBAKH, A.Ya.; LEVSHIN, A.V.; PANINA, L.G.,

red.; BALDINA, N.F., tekhn.red.

[Principles of disability evaluation; theory, methodology, organization. Guide for physicians of the Medical Experts' Gommission on Workers' Disability, medical and prophylactic and other institutions, teachers and students of medical institutes] Osnovy wrachebno-trudovoi ekspertisy; teoriia, metodika, organizatsiia. Rukovodstvo dlia wrachei VTEK, lechebno-profilakticheskikh i drugikh uchreshdenii, prepodavatelei i studentov meditsinskikh institutov. Moskva, Medgis, 1960. 326 p. (MIRA 14:12)

LEVSHIN B.A.

AID P - 2213

: USSR/Aerodynamics Subject

Card 1/2 Pub. 135 - 14/18

: Not given Author

Readers' suggestions Title .

Vest. vozd. flota, 6, 73-79, Je 1955 Periodical:

In this column the four following articles are published,

all related to the evaluation of wind in flight: Abstract 1) "Measuring the drift angle by twice taking the bear-

1) "Measuring the drift angle by twice taking the bearing of a fix in the rear hemisphere of the aircraft" by Lakhtin, M., Lt. Col. Examples, graphs, formulae; by Lakhtin, M., Lt. Col. Examples, graphs, formulae; 2) "How to accelerate the computation of navigational data" by Kurov, V., Guards Maj. Examples, graphs, formulae; 3) "Determination of the drift angle and the formulae; 3) "Determination of the drift angle and the true speed by two slanting ranges and the course angle" true speed by two slanting ranges and the course angle by Levshin. B., Jr. Lt. Examples, graphs, formulae; by Levshin. B., Jr. Lt. Examples, graphs, formulae; by Levshin. B., Jr. Lt., in which the

LEVSHIN G. H.

133-7-3/28

DESCRIPTION OF THE PROPERTY OF

Prikhod'ko, I.P. and Levshin, B.A., Engineers. AUTHOR:

TITIE:

On the Designing of Blast Furnace Skip Hoists (K proyektir-

ovaniyu skipo vykh pod yemnikov domennykh pechey)

PERIODICAL: Stal', 1957, No.7, pp. 584 - 586 (USSR)

ABSTRACT: This is a criticism of the paper by Ya.F. Chel'tsov and G.A. Dubrovin (Stal', 1956, No.9).

There are 2 figures and 2 Slavic references.

ASSOCIATION: Giprostal'

AVAILABLE: Library of Congress

Card 1/1

# Turonian of the Krasnovodsk Peninsula and the central Kara Kum. Inv. AN Turk. SSR. Ser. fiz. - tekh., khim.i geol. nauk no.2:125-126 162. 1. Upravleniye geologii i okhrany nedr pri Sovete Ministrov Turkmenskoy SSR. (Krasnovodsk region--Geology, Stratigraphic) (Kara Kum--Geology, Stratigraphic)

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Upper Cretaceous sediments in the Kizyl-Arvat structural nose. Neftegaz. geol. i geofiz. no.3:27-30 '63. (MIRA 16:8)

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VINOKUROV, A.D., inzh.; DYUBKO, A.P., inzh.; LEVSHIN, B.S., inzh.; L'VITSIN, N.F., inzh.; RESHETIN, I.S., inzh.; KHUDYAKOVSKIY, Yu.K., inzh.; SHAPOVALENKO, M.M., inzh.; SHATSKAYA, E.P., inzh.; MATALASOV, S.F., kand. tekhn.nauk, retsenzent; SHISHLYKOV, Ye.S., inzh., red.; KHITROVA, N.A., tekhn. red.

[Manual on the transportation of perishable goods] Spravochnik po perevozke skoroportiashchikhsia gruzov. [By] A.D. Vinokurov i dr. Moskva, Transzheldorizdat, 1963. 323 p. (MIRA 16:10)

(Railroads-Freight) (Refrigerator cars)

LEVSHIN, B.S., inzh.; MAKARENKO, P.G., inzh. Improve the organization of the transportation of fruit and vegetables. Zhel. dor. transp. 46 no.8:30-34 Ag '64.

(MIRA 17:11)

CIA-RDP86-00513R0009297100 APPROVED FOR RELEASE: Monday, July 31, 2000

no.10:51-55 0 '65.

SERGEYEV. A.M., inzh.; LEVSHIN, B.S., inzh.

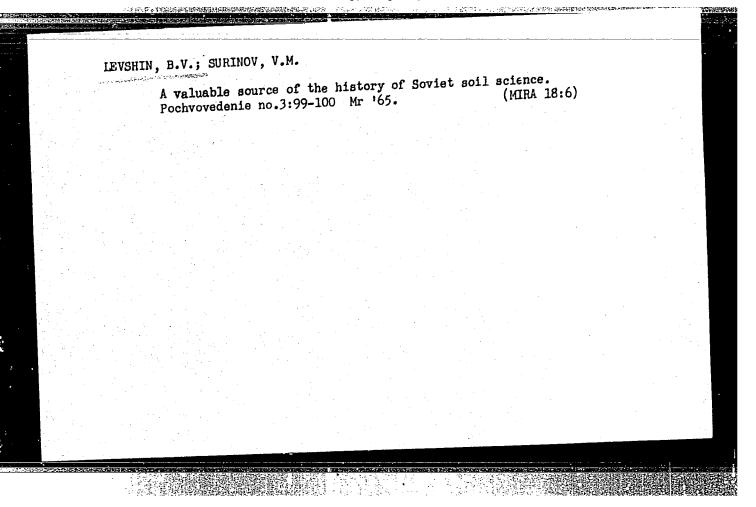
Development of the fleet of refrigerator cars. Zhel.dor.transp. 47

(MIRA 18:10)

ZAYTSEVA, L.L., kand. khim. nauk; LEVSHIN, B.V.; BARANOV, V.I., red.; KHLOPIN, N.G., red.; KNYAZEV, G.A., otv. red.; ARON, G.M., red. izd-va; BOCHEVER, V.T., tekhm. red.

[Letters from V.G.Khlopin to V.I.Vernadskii; 1916-1943] Pis'ma V.G. Khlopina k V.I.Vernadskomu, 1916-1943. Sost. L.L.Zaltseva i B.V.Lev-shin. Pod obshchei red. V.I.Baranova i N.G.Khlopina, 1961. 88 p. (MIRA 14:8)

1. Akademiya nauk SSSR. Arkhiv. (Khlopin, Vitalii Grigor'evich, 1890-1950)

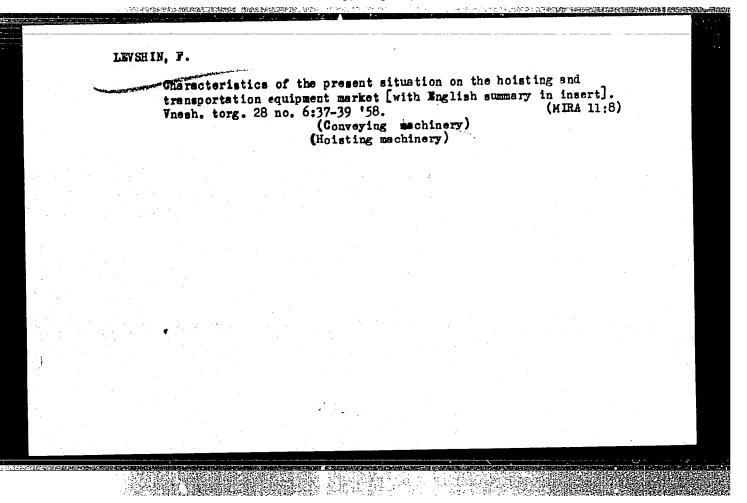


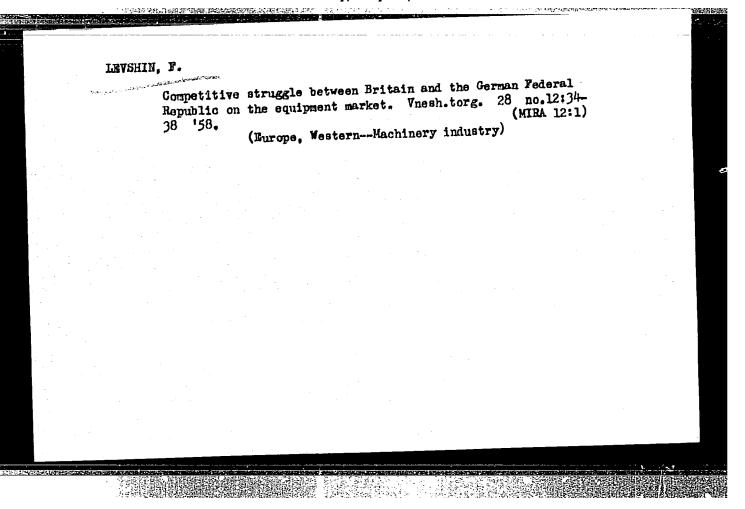
KHLOPIN, Vitaliy Georgiyevich (1890-1950); ZAYTSEVA, L.L.;

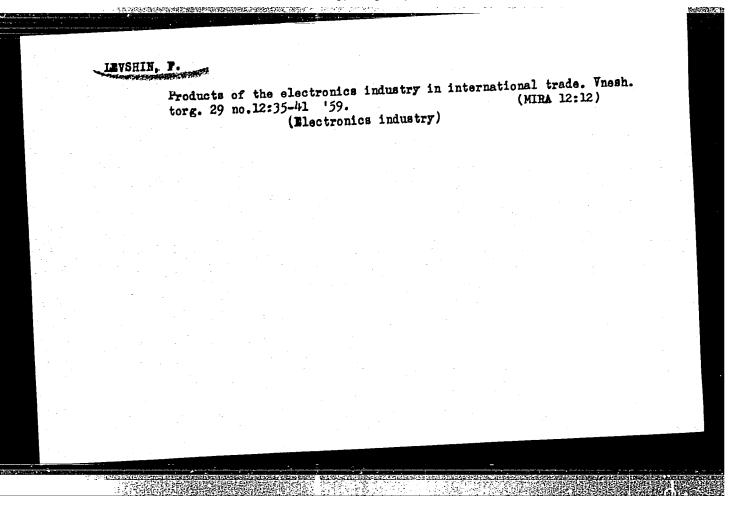
LEVSHIN, B.V., KNYAZEV, G.A., otv. red.; BARANOV, V.I.,

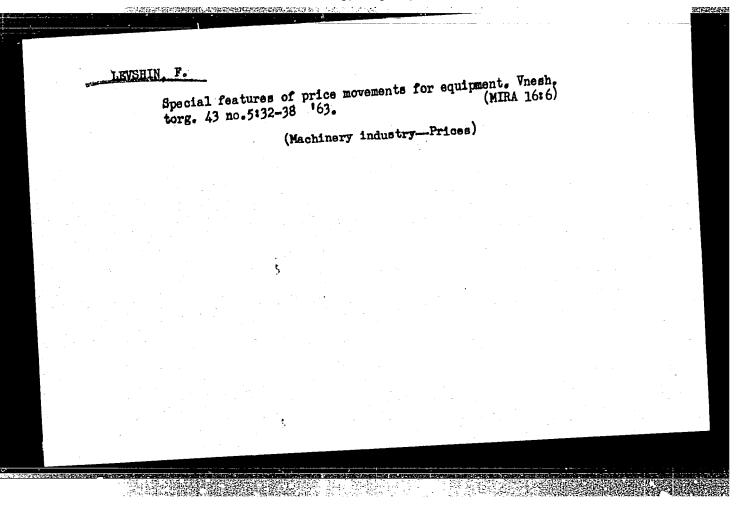
red.

[Letters written to V.I. Vernadskii, 1916-1943] Pis'ma k V.I. Vernadskomu, 1916-1943. Sost.: L.L. Zaitseva i B.V. Levshin. Pod obshchei red. V.I. Baranova i N.G. Khlopina. Moskva, Akad. nauk 1961. 88 p. (MIRA 15:9) (Vernadskii, Vladimir Ivanovich, 1863-1945)





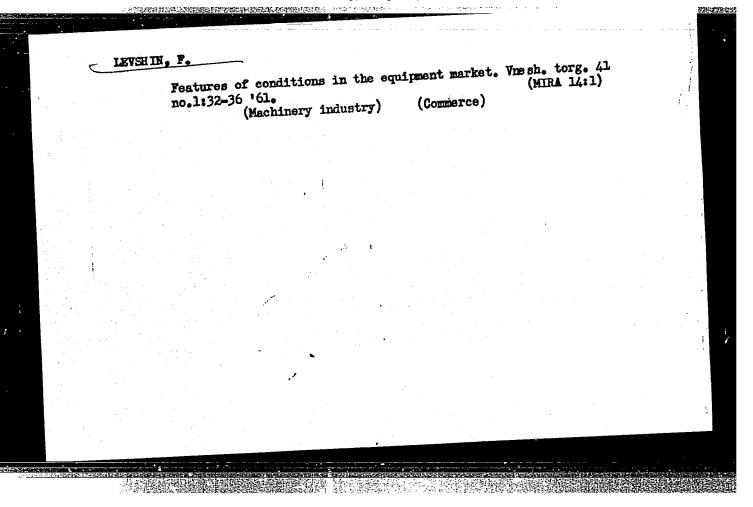


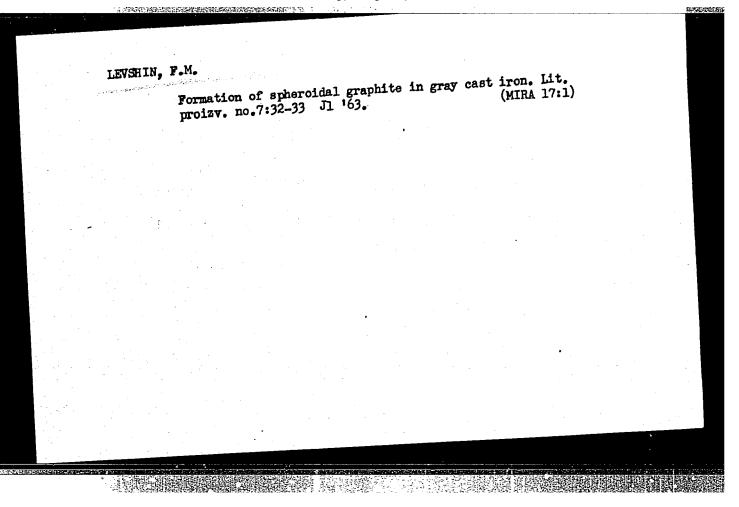


VORONOV, Konstantin Gordeyevich; LEVSHIN Filipp Mikhaylovich; GORDEYEV,

B.S., red.; KAKHOYSKAYA, O.G., red.izd-va; TTSHEKVICH, Z.V.,
tekhn.red.

[brganisation and technique of Soviet foreign trade in equipment]
Vopromy organisatsii i tekhniki yneshnei torgovli SSSR oborudovaniema. Moskva, Vneshtorgizdat, 1960. 66 p. (MIRA 13:10)
(Russia--Commerce) (Machinery industry)





LEVSHIN, Filipp Mikhaylovich; SMIRNOV, G.V., red.; SHLENSKAYA,

V.A., red.izd-va; PAVLOVSKIY, A.A., tekhn. red.

[Products of the electric equipment industry; market of the capitalist countries] Elektrotekhnicheskie izdeliia; rynok capitalisticheskikh stran. Moskva, Vneshtorgizdat, 1963.

kapitalisticheskikh stran. Moskva, Vneshtorgizdat, 17:1)

218 p.

**APPROVED FOR RELEASE: Monday, July 31, 2000 CIA-RDP86-00513R000929710(** 

PETROV, A.P., doktor tekhn. nauk, prof.; TULUPOV, L.P., kand. tekhn.
nauk; KRYUKOV, N.D., kand. tekhn.nauk; GUNDOBIN, V.N., inzh.;
VASIL'YEV, G.S., kand. tekhn. nauk; GRISHIN, M.S., kand.
tekhn. nauk; MOROZOVA, K.N., inzh.; ROZE, V.A., inzh.; LEVSHIN.
G.L., inzh.; BERNGARD, K.A., doktor tekhn. nauk, prof.;
BIKCHENTAY, M.A., inzh.; BUYANOV, V.A., inzh.; ILOVAYSKIY,
N.D., inzh.; MUKHAMEDOV. G.A., kand. tekhn.nauk; MIROSHNICHENKO,
A.P., inzh.; ANDRIANOV, V.P., inzh.; BUTS, V.D., inzh.; KAZIMOV,
A.A., inzh.; KIREYEV, O.P., inzh.; DYUFUR, S.L., kand. tekhn.
nauk; USTINSKIY, A.A., kand. tekhn. nauk; MIKHAYLOV, S.M., inzh.;
NESTEROV, Ye.P., kand. tekhn. nauk, retsenzent; LIVSHITS, V.N.,
inzh., retsenzent; PREDE, V.Yu., inzh., red.; VOROTNIKOVA, L.F.,
tekhn. red.

[Control of transportation processes using electronic digital computers] Upravlenie perevozochnym protsessom s primeneniem elektronnykh tsifrovykh vychislitel nykh mashin. Pod obshchei red. A.P.Petrova. Moskva, Transzheldorizdat, 1963. 207 p. (MIRA 16:8)

1. Chlen-korrespondent AN SSSR (for Petrov).
(Railroads--Management) (Electronic digital computers)

APPROVED FOR RELEASE: Monday July 31, 2000 Zas GIA-RDP86 to 00513R0009297:

Potato wart in Khmel'niverly 160.

vred. i bol. 5 no. 8:47-48 & 160.

1. Machal'nik Khmel'nitskoy karantinnoy inspektsii (for Kurbalev).

(Khmel'nitskiy Province--Potato wart)

BOROVOY, Natan Yefimovich; LEVSHIN, Ivan Konstantinovich; OBORIH, Nikolay Nikolayevich; PANOV, V.I., redaktor; MANDYKIN, A.Ye., tekhnicheskiy redaktor

[Innovation in the practice of a level shunting station; the experience of the Uzlovaya station no.1] Novoe v tekhnologii bezgorochnoi stantsii; opyt stantsii Uzlovaia I. Moskva, Gos. transportnoe zhel-dor. izd-vo. 1955. 17 p.

(Uzlovaya--Railroads--Stations)

(MLRA 816)

3/058/61/000/006/052/053 A001/A101

6,9200

Levshin, I.P., Slobodenyuk, G.I.

TITLE

AUTHORS:

A device for measuring correlation coefficients in distant communica-

tion lines using ultrashort waves

Referativnyy zhurnal. Fizika, no. 6, 1961, 359, abstract 6Zh290 ("Sb. tr. Nauchno-tekhn. o-vo radiotekhn. i elektrosvyazi im. A.S. Popova", PERIODICAL:

1959, no. 3, 140 - 151)

The authors describe a device designed for measuring coefficients of correlation between two random processes which characterize signal fading at distant tropospheric propagation of ultrashort waves. They give recommendations on using this device, correlation meter, in those regions which are connected with studying statistical properties of random processes. The description of the device is preceded by a short substantiation of the measurement method.

[Abstracter's note: Complete translation]

Card 1/1

**APPROVED FOR RELEASE: Monday, July 31, 2000** 

CIA-RDP86-00513R000929716

ARMAND, N.A.; VVEDENSKIY, B.A.; GUSYATINSKIY, I.A.; IGOSHEV, I.P.;
KAZAKOV, L.Ya.; KALININ, A.I.; KOLOSOV, M.A.; LEVSHIN, I.P.;
LOMAKIN, A.N.; NAZAROVA, L.G.; NEMIROVSKIY, A.S.; PROSIN,
A.V.; RYSKIN, E.Ya.; SOKOLOV, A.V.; TARASOV, V.A.; TRASHKOV,
P.S.; TIKHOMIROV, Yu.A.; TROITSKIY, V.N.; FEDOROVA, L.V.;
CHERNYY, F.B.; SHABEL'NIKOV, A.V.; SHIREY, R.A.; SHIFRIN, Ya.S.;
SHUR, A.A.; YAKOVLEV, O.I.; ARENBERG, N.Ya., red.

[Long-distance tropospheric propagation of ultrashort radio waves] Dal'nee troposfernoe rasprostranenie ul'trakorotkikh radiovoln. Moskva, Sovetskoe radio, 1965. 414 p. (MIRA 18:9)

22727 s/108/61/016/005/001/005 B104/B205

9,9100 AUTHORS:

Prosin, A. V., Igoshev, I. P., Levshin, I. P.

TITLE:

Automation of the statistical evaluation of radio signals

by electronic computers

Radiotekhnika, v. 16, no. 5, 1961, 64 - 70

TEXT: A description is given of a method for the statistical evaluation of experimental data by digital electronic computers. This method was developed for computers of the types M-2 (M-2) and 5XM-2 (BESM-2) of the Institut elektronnykh upravlyayemykh mashin AN SSSR (Institute of Electronic Control Machines, AS USSR) by the Institut radiotekhniki i elektroniki AN SSSR (Institute of Radio Engineering and Electronics, AS USSR) in a laboratory under the supervision of V. I. Siforov, Corresponding Member AS USSR, and the apparatus required was also built. The proper conversion of experimental data to be processed by electronic computers is discussed first. Fig. 1 shows the code of the M-2 machine; a signal and its conversion into a digital code are illustrated in Fig. 2. For the purpose of feeding data given in the code of the M-2 machine into

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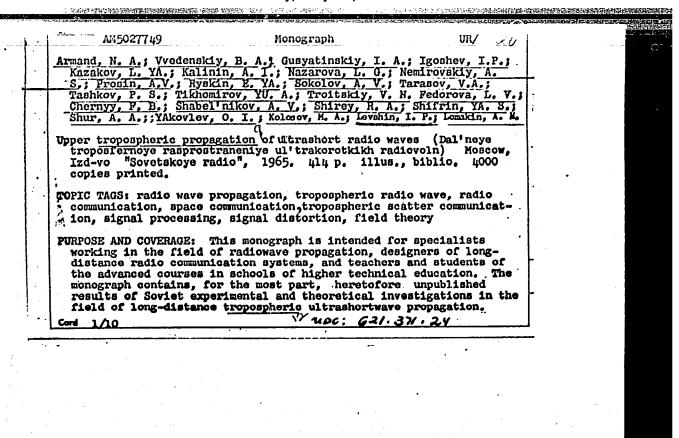
CIA-RDP86-00513R0009297100 APPROVED FOR RELEASE: Monday, July 31, 2000

## "APPROVED FOR RELEASE: Monday, July 31, 2000

CIA-RDP86-00513R000929710

28503-66 EEC(k)-2/EWT(d)/EWT(1)/FCC GW/WS-2 SOURCE CODE: UR/0108/66/021/002/0002/0011 ACC NR: AP6007149 AUTHOR: Levshin, I. P. (Active member); Prosin, A. V. (Active member) ORG: Scientific and Technical Society of Radio Engineering and Electrocommunication (Nauchno-tekhnicheskoye obshchestvo radiotekhniki i elektrosvyazi) TITLE: Digital-computer simulation of a multipath channel with long-distance UHF tropospheric propagationd SOURCE: Radiotekhnika, v. 21, no. 2, 1966, 2-11 TOPIC TAGS: UHF wave propagation, multipath communication, computer simulation, digital computer, tropospheric radio wave, communication channel ABSTRACT: The development of a discrete mathematical simulator of the tropospheric channel is reported. The simulator permits reproducing random characteristics of such a channel which may be useful in planning multipath radiocommunication lines. The simulator describes a fluctuating quadripole whose random amplitude-frequency and phase-frequency characteristics are statistically UDC: 621.371.176 Card 1/2

L 28503-66 ACC NR: AP6007149 O close to those of a real tropospheric channel. The received signal, as a combination of many random-amplitude, random-phase waves, is described by: From this formula, a complex transfer factor  $u_{\kappa}(l)$ , of the quadripole is derived. Physically, the received signal comprises these three components: coherent scatter, reflection from various layers, and incoherent scatter. The average period of variation of the coherent scatter is assumed to be 1.5-2 hrs or longer; of the reflection, 2-12 min; of the incoherent scatter, from a fraction to a few seconds. A simplified scheme of the machine algorithm of the tropospheric-channel simulator is shown. Simulated amplitude-frequency and phase-frequency characteristics and also group delay time determined for a 300-km 1000-Mc line agreed almost exactly with the experimental characteristics measured on such a line. Orig. art. has: 6 figures, 33 formulas, and I table. SUB CODE: 17, 09 / SUBM DATE: 15May64 / ORIG REF: 004 Card 2/2



## ACC NRI AM5027749 Problems of investigating the troposphere by means of refractometers, the mean level of signals, meteorological conditions and topography, fluctuation of arrival angles and distortions of antenna-

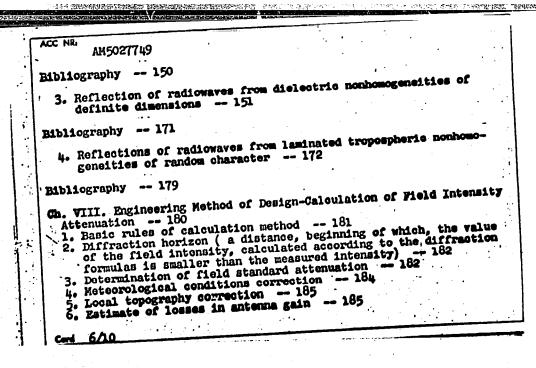
topography, fluctuation of arrival angles and distortions of antennadirectivity patterns, losses in antenna gain, and quick and slow
fadings of signal levels are discussed. The statistical characteristics of the signals at diversity reception in time, space, frequency and angle as well as the distortion of signals in the communication systems are also investigated. The long-distance propagatication systems are also investigated. The long-distance propagatintensity at long-distance tropospheric propagation is given. At
intensity at long-distance tropospheric propagation is given. At
present, there is no theory of Long-Distance Tropospheric Propagation which can be applied effectively enough in practice. Thus, in
the investigation of that propagation, considerable attention has
the investigation of that propagation, considerable attention has
to be paid to experiments. The special characteristics of geographical conditions of the territory involved should be taken into consideration during the analysis of experimental data and in their
practical application because the conditions of propagation in
arctic and tropical climates differ from those existing over seas arctic and tropical climates differ from those existing over seas and continents. A considerable part of the monograph deals with the investigation of long-distance tropospheric propagation carried out over dry land routes, 800 km long, in the central part of the USSR under the general supervision of B. A. Vvedenskiy and A. G. Arenberg (up to 1957). V. I. Siforov investigated problems con-

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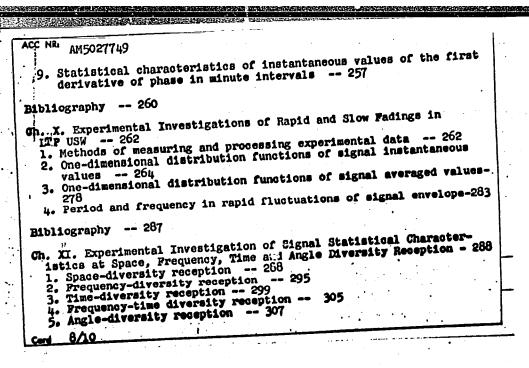
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•		ACC NRI AM5027749
		nected with distortions and fluctuations of signals. References follow each chapter.
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Ch. V. Fluctua	tion of Radiowave Ar Antennas Directiviti of measuring radiowave meous antenna direction	a arrival angles an	d recording of		
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2. Experimental investigation of distortion in LTP -- 384

3. Distortions appearing during TV transmission over tropospheric radio links -- 389

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Cod 9/10
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### LEVSHIN, I.Ye.

Methods for developing the dentritic structure of carbon steel.

Zav.lab. no.ll:1333-1334 59. (MIRA 13:4)

1. Dneprovskiy metallurgicheskiy savod. im. Dzerzhinskogo. (Steel - Metallography)

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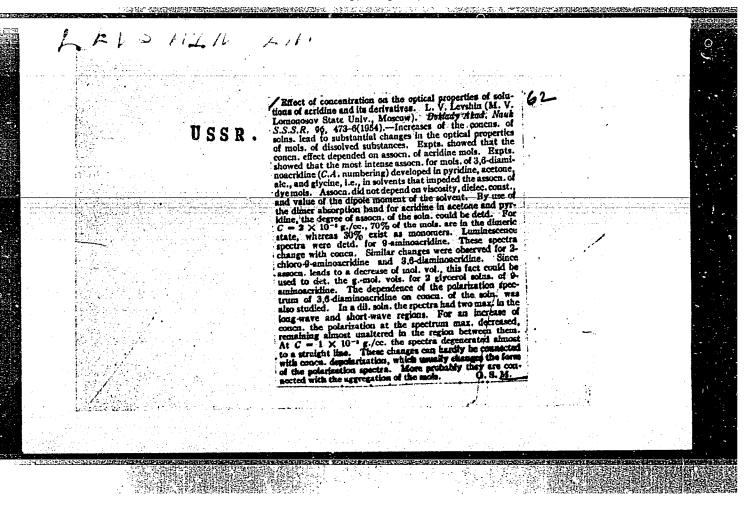
THE PROPERTY OF THE PROPERTY O

ZHDANOV, Andrey Andreyevich; LEVSHIN, Lev Vasil'yevich; KAZAKOVA, L.A., red.; BYKOVA, V.V., tekhn.red.

[Protection of forest and water resources in the U.S.S.R.]

Okhrana lesnykh i vodnykh bogatstv v SSSR. Moskva, Gos.izd-vo iurid.lit-ry, 1958. 49 p. (MIRA 12:2)

(Forests and forestry) (Hunting) (Fisheries)



#### "APPROVED FOR RELEASE: Monday, July 31, 2000

CIA-RDP86-00513R000929710 (4) - 42-4) 全国共和国的政策及中国国际的国际的

Leushin, L.V

USSR/Physics - Luminescence

FD-1824

Card 1/1

Pub 146-9/25

Author

Levshin, L. V.

Title

Influence of concentration upon the optical properties of solutions of

3, 6-diaminoacridine

Periodical: Zhur. eksp. i teor. fiz. 28, 201-212, February 1955

Abstract

: The author investigates the influence of the concentration of a solution upon the absorption spectra, luminescence spectra, illumination intensity, average duration of the excited state, and polarization spectra of 3, 6diaminoacridine. He demonstrates that the variations in optical properties during increasing concentration of the solution are caused by the dimerization of the molecules of this compound. He establishes the optical properties of the dimers and determines the degree of association of the solutions of various concentration. He investigates the influence of the solvent and temperature upon the association of the molecules of 3, 6diaminoacridine. He thanks Prof. P. A. Bazhulin, who guided the present work; prof. A. M. Grigorovskiy, who advised on chemical problems; M. D. Galanin, fluorometer expert; N. D. Zhevandrov, who helped on the polarization measurements. Fifteen references; e.g. V. L. Levshin, Zhur. fiz.

khimii, 6. 1, 1935, and ZS. f. Phys. 43, 1927.

Institution: Moscow State University

Submitted :

March 2, 1954

USSR/Physics - Luminescence

<u>17D-1825</u>

Card 1/1

Pub 146-10/25

Author

L. V. Levshin

Title

Influence of concentration upon the optical properties of solutions of

acridine compounds

Periodical: Zhur. eksp. i teor. fiz. 28, 213-222, February 1955

Abstract

The author investigates the influence of concentration upon the abscrption spectra (in the region from 500 to 220 millimicrons), luminescence spectra, polarization spectra, light yield, and average duration of the excited state of ten acridine compounds. He demonstrates that many concentration effects can be explained by the association of the molecules of the dissolved substance. Association of the molecules of certain compounds annihilates the luminescent capacity, and association of the molecules of other compounds varies the spectral composition of radiation. He thanks Prof. P. A. Bazhulin, who guided the present work; Prof. A. M. Grigorovskiy, expert on chemical problems; M. D. Galanin, fluorometer expert; and N. D. Zhevandrov,

who helped on polarization measurements. Ten references.

Institution: Moscow State University

Submitted :

March 2, 1954

LEVSHIN, L.V

USSR / Physical Chemistry. Molecule. Chemical Bond.

B-4

Abs Jour

: Ref Zhur - Khimiya, No 8, 1957, 25772

Author

: L.V. Levshin

Inst

: Academy of Sciences of USSR

Title

2 Luminescence and Absorption of Concentrated Solutions of 3,6-Diaminoacridine.

Orig Pub

: Izv. AN SSSR, ser. fiz., 1956, 20, No 4, 419-423

Abstract

A strong concentration dependence of absorption spectra of 3,6 diaminoacridine (I) solutions was observed (RZhKhim, 1955, 36658) earlier and this dependence was explained by the association of molecules of I. The hypothesis of molecule association at the concentration rise met an essential objection, because 1) the heating of the concentrated solution to 50 to 70° altered the spectrum little, and 2) the effects of concentration were more noticeable in such solvents (diozane, pyridine, acetone, alcohol), in which ordinary dyes did not a ssociate in general, or associated very weakly. The observed phenometric diagram of the concentration were did not a ssociated very weakly.

Card

: 1/2

- 23 -

#### "APPROVED FOR RELEASE: Monday, July 31, 2000

CIA-RDP86-00513R000929710

KEVSHIN, LV

USSR/Optics - Physical Optics.

K-5

Abs Jour

: Referat Zhur - Fizika, No 3, 1957, 7749

Author

: Levshin, L.V.

Inst

: Moscow State University, USSR.

Title

Nature of the Concentration Effects in Solutions of

Acridine Compounds.

Orig Pub

: Dokl. AN SSSR, 1956, 108, No 2, 228-231

Abstract

Measurements of the luminescence spectra were made with a photoelectric method employing the ISP-51 spectroscope together with the PS-381 attachment; the receiver employed was a FEU-17 photographic paper. The excitation was by means of the 436 and 365 m/s lines from a mercury spectrum. The pH of the solution was determined with the aid of a vacuum tube potentiometer LP-5. By changing the pH of the solution, it was possible to maintain in it the predominance of one of the two forms of 3.6-diamino-acridine. The absorption and luminescence spectra of these

Card 1/3

- 45 .

Card 3/3

- 41

USSR/Optics - Physical Optics.

K-5

Abs Jour

: Referat Zhur - Fizika, No 3, 1957, 7749

forms differ in intensity and position of the maxima. If all the molecules are in the second form, the previously-noted dependence of the spectrum on the wavelength of exciting light is lacking. The frequencies of the electron transitions are  $\nu$   $\approx 21050$  cm<sup>-1</sup> for the first form and  $\nu$  = 22100 cm<sup>-1</sup> for the second form. The second form of 3.6-diamino-acridine which appears when the pH of the solution is increased, and darkens rapidly upon illumination; therefore concentrated solutions of 3.6-diamino-acridine become dark when exposed to light, while dilute solutions do not react to illumination. The concentration effects are quite pronounced also in 9-amino-acridine the glow of which changes from blue into azure-green when the concentration increases. It is obvious that here, too, the pH of the solution determines the existence of the particular form of molecules. By artificially changing the pH of a solution of

Card 2/3

- 46 -

Add the continues of the state	PRIKHOT'KO, A.F.  24(7) b.3 PHASE I BOOK EXPLOITATION SOV/136  L'vov. Universytet  teterialy X Vsescyuzzogo soveshchaniya po spektroskopii Molekulyarnaya spektroskopiya (Papers of the 10th Al' Conference on Spectroscopy. Vol. 11 Molecular Spect [L'vov] Izd-vo L'vovakogo univ-ta, 1957. 499 p. 4,00 printed. (Series: Its: Pizychnyy sbirnyk, vyp. 3/4	. t. 1:	
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	ard 1/30		-
Rai	akov, A.V. Dependence of the Line Width in Raman Spectrum on the Aggregate State of the Sub- stance	A Part of the Control	
Res	Rayev, M.I. Photoelectric Study of the Form and Width of Reman Lines in Liquids and Solids	229	
Pay	iviovakaya, T. Ye., and A.G. Fasynskiy. Variation in Absorption Spectra of Protein Solutions Due to Ionizing Radiation in Air and in Vacuum	230	
Lev	vahin. L.V., and A.F. Khovanakiy. Spectroscopic Study of the Ionization of Molecules of Acridine Compounds	235	
Kar	ryakin, A.V., and A.V. Shablya. Infrared-spectro- graphic Study of the Sensitization of the Photo- oxidation of Organic Compounds by Means of Anthraquinone Derivatives	240	
\	N 16/20	243	

LEVSHIN, L.V.

51-6-9/26

AUTHORS:

and Khovanskiy, A. P. Levshin, L. V.

TITLE:

Study of the Ionisation of Molecules of Acridine and its Derivatives using Luminescence Spectra. (Issledovaniye ionizatsii molekul akridina i yego proizvodnykh po spektram lyuminestsentsii)

PERIODICAL:

Optika i Spektroskopiya, 1957, Vol.II, Nr.6, pp. 747-754.

ABSTRACT:

In acid solutions molecules of acridine compounds are ionised and, since their alkalinities are not the same, the degree of ionisation for a given value of pH will Using luminescence differ from one compound to another. spectra the ionisation of acridine and five of its derivatives: 1-, 2-, 3-, 9-monoaminoacridines and The luminescence spectra were measured photoelectrically using a glass 3,6-diaminoacridine, was studied. Luminescence was excited with 436 and spectra were measured using a quartz spectrophotometer. spectrograph. Dependence of the luminescence spectra in the region

Card 1/2

CIA-RDP86-00513R0009297100 APPROVED FOR RELEASE: Monday, July 31, 2000

Study of the Ionisation of Molecules of Acridine and its Derivatives using Luminescence Spectra.

400 to 660 mm on pH for the substances studied is given in Figs.1, 2, 3, 4 and 6. Fig.5 shows the absorption spectra of double ions of 9-aminoacridine and acridinium in the region 220 to 460 mm. The degree of ionisation and its dependence on pH was found for each compound studied from its luminescence spectrum. The results obtained for 9-aminoacridine support acridonimine structure for that molecule. The authors thank Professor A. M. Grigorovskiy for his advice and a supply of the substances studied, and Z. A. Barmina for help in experiments. There are 6 figures, 21 structural formulae and 12 references, 7 of which are Slavic.

ASSOCIATION: Moscow State University. (Moskovskiy gosudarstvennyy universitet.)

SUBMITTED: November 19, 1956.

AVAILABLE: Library of Congress.

Card 2/2

AUTHORS:

Levshin, L.V. and Suvorov, V.S.

51-4-5-18/29

TITLE:

Association of Molecules of Rhodamine 6G and Grystal Violet Dyes in Concentrated Aqueous Solutions (Assotsiatziya molecul krasiteley podamina 6Zh i kristallicheskogo floletovogo v

kontsentrirovannykh vodnykh rastvorakh)

PERIODICAL:

Optika i Spektroskopiya, 1958, Vol IV, Nr 5, pp. 678-681 (USSR)

ABS TRACT:

The present paper deals with the possibility of formation of associates consisting of different dye molecules and with the study of their properties. Rhodamine 6G and crystal violet were the two dyes studied. The first of them luminesces strongly in solution while the second does not possess luminescent properties. Each of these two dyes readily forms associates in concentrated aqueous solutions. The absorption spectra were measured using a SF-4 spectrophotometer. The luminescence spectra were recorded by means of an ISP-51 spectrograph together with a photoelectric collimator PS-381. The results are shown in Fig 1 which gives the absorption spectra of the mixture (curves 1) and of the components (curves 2 and 3),

Card 1/3

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51-4-5-18/29

Association of Molecules of Rhodamine 6G and Crystal Violet Dyes in Concentrated Aqueous Solutions

whose concentrations are equal to the concentration of the mixture. The shaded portions in Fig 1 represent the difference between the absorption spectrum of the mixture obtained experimentally, and the sum of the absorption spectra of both components. Magnitude of this shaded region indicates the degree of non-additivity of the absorption spectra. Such non-additivity is the evidence for the existence of associates. Curves marked 4 in Fig 1, which are the boundaries of these shaded regions, are called non-additivity curves. Fig 2 shows the change with time of the non-additivity curve for the mixture of the two dyes. Fig 3 shows the effect of heating to 95°C on the non-additivity of the absorption spectra of the dye mixture. The effect of the solvent on the non-additivity of the absorption spectra of rhodamine 6G and crystal violet is shown in Fig 4. The existence of non-additivity of the absorption spectra of rhodamine 60 and crystal violet when they are mixed together, and the behaviour of the non-additivity curves are shown to be conclusive evidence for formation of associates consisting of rhodamine 6G and crystal violet molecules. There are 4 figures and 7 references 6 of which are Soviet and 1 German.

Card 2/3

Association of Molecules of Rhodamine 6G and Crystal Violet Dyes in Concentrated Aqueous Solutions

Moskovskiy gosudarstvennyy universitet, fizicheskiy fakul'tet ASSOCIATION:

(Physics Department, Moscow State University)

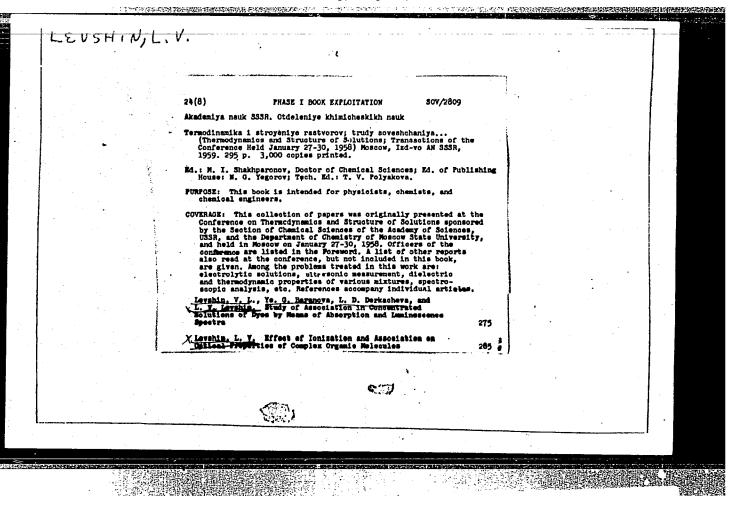
July 19, 1957 SUBMITTED:

Rhodamine - Luminescence 2. Crystal violet - Luminescence 3. Dyes - Analysis

Card 3/3

#### "APPROVED FOR RELEASE: Monday, July 31, 2000

CIA-RDP86-00513R000929710



0.4(7)	SOV/48-23-1-4/36
- WOINOW.	Levshin, L. V.
TITLE:	Influence of Ionization Upon the Luminescence Spectrum of 9-Amino Acridine and Its Monochloro Derivatives (Vliyaniye ionizatsii na spektry lyuminestsentsii 9-aminoakridina i ionizatsii na spektry lyuminestsentsii 9-aminoakridina i
	yego monokhlorproizvodnykh)
	yego monokholpio
PERIODICAL:	4 07 Nm 1 DD 17-64 \- '
ABSTRACT:	Due to the ionization of acridine and monoamino acriding to the ionization of acridine and absorption spectra vary in
•	a similar manner (net 1/)
•	by affiliating a late wave range. In the ionization of
	9-amino acridine the passassumed that 9-amino acriding
	possesses an additions to the amino group; second remains of the proton to the amino group; second remains and the proton to the proton to the amino group; second remains and the proton to the proton to the amino group; second remains and the proton to the proton to the amino group; second remains and the proton to the proton to the amino group; second remains and the proton to the amino group and the
	affiliation of the proton to the amino group; second affiliation of the proton to the ring(Refs 2,3). In this affiliation to the nitrogen of the ring(Refs 2,3). In this affiliation to the nitrogen of 1,2,3,4-chloro-9-amino acridines is paper the ionization of 1,2,3,4-chloro-9-amino acridines
Card 1/3	paper the ionization of the paper the ionization of the paper

CIA-RDP86-00513R0009297100 APPROVED FOR RELEASE: Monday, July 31, 2000

Influence of Ionization Upon the Luminescence SOV/48-23-1-4/36 Spectrum of 9-Amino Acridine and Its Monochloro Derivatives

studied according to the method of luminescence. The shift of the luminescence spectrum at single and double ionizations corresponds fully to the above-mentioned data on 9-amino acridine. Furthermore, the influence is investigated that is exercised by the substitution of chlorine in the various ring positions. Figures show the comparison of: 1) nonionized 9-amino acridine with its monochloro derivatives in 1,2,3,4-position, whereby the spectrum is not shifted considerably, however, the shape of the spectrum depends to a large extent on the position of chlorine; 2) the luminescence spectra of the mono-ions of 9-amino acridine with its chloro derivatives. Their shape varies only in a certain degree, whereas considerable shifts take place, especially in position 1. The minimum shift is to be found in position 3. The other two positions adopt a mid-position. The third comparison of the double-ionized 9-amino acridine with its chloro derivatives shows a particularly marked shift of the 1- and 4-chloro derivative toward the long-wave range. Both curves are almost equal, while for the positions 2 and 3 the spectrum of 9-amino acridine adopts a mid-position between them only

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Influence of Ionization Upon the Luminescence SOV/48-23-1-4/36 Spectrum of 9-Amino Acridine and Its Monochloro Derivatives

as far as the shape is concerned. The regularities resulting from these data are explained by induction effects. In the amino and imino structure, positive electric charges are arranged in the symmetrical position round the nitrogen of the ring or round the amino group. Chlorine enters into reaction with nitrogen and produces a deformation of the spectrum. There are 4 figures and 6 references, 4 of which are Soviet.

card 3/3

# PHASE I BOOK EXPLOITATION

sov/4698

# Levshin, Leonid Vadimovich

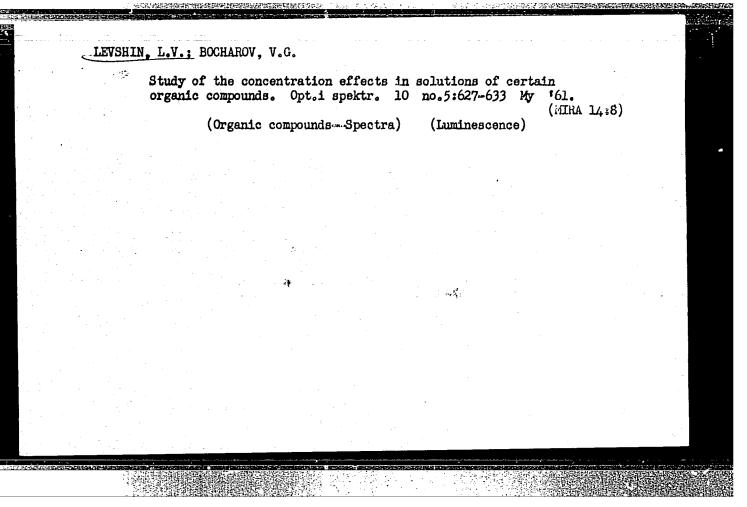
Sergey Ivanovich Vavilov. [Moscow] Izd-vo Mosk. univ., 1960. 101 p. (Series: Zamechatel'nyye uchenyye Moskovskogo universiteta, no. 24) Errata slip inserted. 2,000 copies printed.

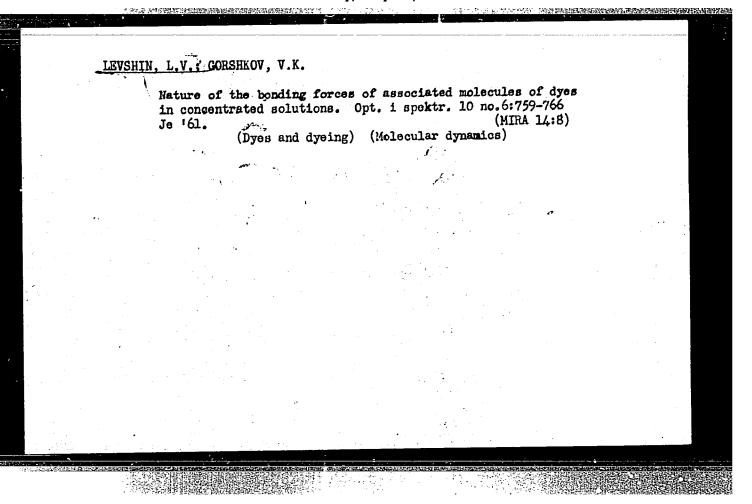
Editorial Board: S. B. Bernshteyn, Professor, D. G. Vilenskiy, Professor, G. D. Vovchenko (Chairman and Resp. Ed.) Professor, D. I. Gordeyev (Deputy Chairman) Professor, N. K. Gudziy, Professor, P. A. Zayonchkovskiy, Professor, S. F. Kechek'yan, Professor, K. P. Mel'nikova (Secretary), Candidate of Sciences, fessor, K. P. Mel'nikova (Secretary), Candidate of Sciences, F. Ya. Polyanskiy, Professor, K. A. Rybnikov, Professor, S. D. Skazkin, Academician, and A. N. Solov'yev, Docent; Ed.: M. G. Zaytseva: Tech. Ed.: G. I. Georgiveva. Zaytseva; Tech. Ed.: G. I. Georgiyeva.

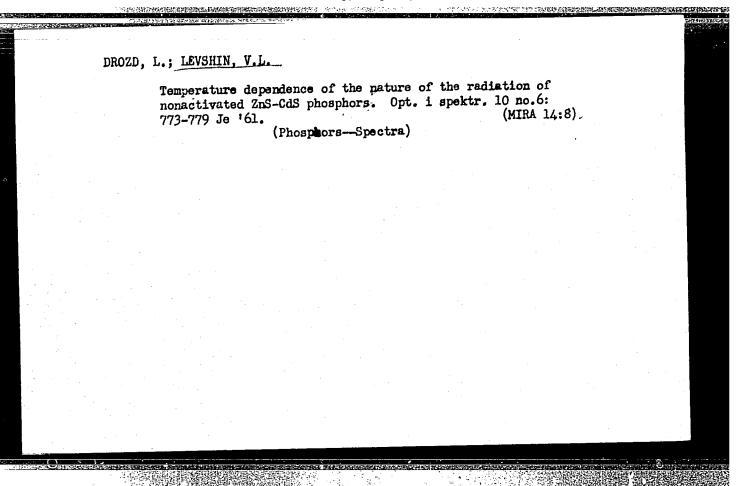
The booklet is intended for the general reader. be of particular interest to students of physics.

COVERAGE: This booklet outlines the life and scientific work of the physicist, S. I. Vavilov (1891-1951). The first half of the booklet includes biographical information, while the second half presents essays on Vavilov's contributions to the fields

Card 1/2







card 1/3

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CIA-RDP86-00513R000929710
                           LEVSHIN, L.V.; LONSKAYA, I.S.
                                                                                   Dependence of the association of rhodamines on their molecu-
                                                                                    lar structure and the nature of the solvent. Opt. i spektr.
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                 (MIRA 14:8)
                                                                                    11 no.2:278-282 Ag '61.
                                                                                                                                                                                                                                                             (Rhodamine)
                                                                                                                                                                                                                                                              (Molecular association)
                                                                                                                                                                                                                                                                                                                                                                                                            2000 GJA ABP 86-00513R000929
                                                                                                                                                                                                                                                        Tonday, July 31, 2000
                                                                                                                                                                                                                                                                                                                                                                                                                 B187/B102
                                                                                                                              Association of the molecules of rhodemine 6 X (6Zh) in mixtures of chloroform and CCl.
                                                                                                                                                                                                                                                                                                                                                                                                                                        Seriya III. Fizika,
                                                                                                                                  Association of the more cures of wixtures of chloroform and CC14
                                 TEXT: The binding energy U was determined quantitatively for associates of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which an indication of 62h rhodamine in mixtures of chloroform and CCl, for which are chloroform and CCl, for the chlorofor
                                    TEXT: The binding energy U was determined quantitatively for as indication of 6Zh rhodamine in mixtures of chloroform and in the infrared spectrum.

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                                               of the presence of hydrogen bonds was found in the infrared spectrum.

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found in the concentration of the seregy on the concentration of the means of the monomeric state at a spectra.

The dependence of the mixtures was determined from the absorption spectra.

The fraction of the solutions was determined from the spectra.

The fraction of the solutions was determined from the spectra.

The fraction of the solutions was determined from the spectra.
                                           of 62h rhodamine in mixtures of chloroform and ccl 4 for which an indice infrared spectrum. In the infrared spectrum the infrared sand on the of the presence of the binding energy on the concentration c and on the of the binding energy on the concentration.
AUTHORS:
         TITLE:
                                                               spectra. The fraction in the monomeric state at a given the absorption spectra.

The fraction in the monomeric state at a given the absorption spectra.

The fraction in the solutions was determined from the absorption of the solutions was determined from the absorption spectra.
                                                                    concentration of the solutions was determined from the absorption spectra.

The value of U

The tionality of linear function of proportionality of linear from the factor with increasing C of the solution of
                     PERIODICAL:
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IGNAT'YEVA, L.A.; LEVSHIN, L.V.; OSIPOVA, T.D.; POLUKHIN, Yu.M.

Study of the association of rhodamine 6G molecules based on electron and vibrational absorption spectra. Opt. i spektre 13 no.3:396-402 S '62. (MIRA 15:9) (Rhodamine—Spectra) (Molecular association)

5/188/63/800/601/604/01 B104/B102

L. V., Herdovskiy, V.

AUTHORS: The effect of the molecule ionization of some derivatives

of 9-aminoacridine on their spectra-

PERIODICAL: Moscow. Universitet. Vestnik. Seriya III. astronomiya, no. 1, 1963, 24 - 31

TEXT: In earlier papers (L. V. Levshin, A. P. Khovanskiy, Optika i spektroskopiya, 2, 747, 1957) a method was developed for investigating ionization of acridine molecules. This is based on a study of the change occurring in their luminescence spectra when the pH values of an alcoholic solution are varied. In another paper (L. V. Levshin, Izv. AN SSSR, ser. fizicheskaya, 23, 19, 1959) this method had already been used to investigate the ionization effect on the luminescence spectra. Here the changes in the luminescence spectra of the solutions of the following five 9-aminoacridine derivatives are investigated by a photoelectric spectrometer:  $9-(\delta-\text{diethylamino}-\alpha-\text{methylbutyl})$ -aminoacridine; 1-chlor-9-(δ-diethylamino-α-methylbutyl)-aminoacridine; 2-chlor-9-

Card 1/2

APPROVED FOR RELEASE: Monday, July 31, 2000

BABUSHKIN, Aleksandr Afanas'yevich, dots.; BAZHULIN, Pavel Alekseyevich, prof.; KOROLEV, Fedor Andreyevich, prof.; LEVSHIN, Leonid Vadimovich, prof.; PROKOF'YEV, Vladimir Konstantinovich, prof.; STRIGANOV, Arkadiy Romanovich, doktor fiziko-matem. nauk; GOL'DENEERG, G.S., red.; GEORGIYEVA, G.I., tekhn. red.

[Spectrum analysis methods]Metody spektral'nogo analiza. [By]
A.A.Babushkin i dr. Pod red. V.L.Levshina. Moskva, Izd-vo Mosk.
univ., 1962. 508 p. (MIRA 16:2)
(Spectrum analysis)

## LEVSHIN, L.V.; SLAYBOVA, T.D.

Association of rhodomine 6G molecules in mixtures of chloroform and carbon tetracholoride. Vest. Mosk.un. Ser. 3: Fiz., astron. 17 no. 6:24-27 N-D '62. (MIRA 15:12)

1. Kafedra optiki Moskovskogo universiteta.
(Rhodamine) (Carbon tetrachloride) (Molecular association)

LEVSHIN, L.V., BERDOVSKIY, V.

Effect of molecular ionization of some derivatives of 9-aminoacridine on their limit mescence spectra. Vest. Mosk.un.Ser3:Fiz., astron.18no.1: 24-31 Ja-F \*63.

(MIRA 16:5)

1. Kafedra optiki Moskovskogo universiteta.
(Aeridine—Spectra) (Ionization)

L 10165-63

EWT(m)/HDS-

ACCESSION NR:

AP3000307

s/0048/63/027/005/0590/0595

Bocharov, V. G.; Levshin, L. V.

TITLE: Association of rhodamine 60 molecules in binary solvents [Report; Eleventh Conference on Luminescence held in Minsk 10-15 Sept. 1962]

Izvestiya AN SSSR. Seriya fizicheskaya, v. 27, no. 5, 1963, 590-595

TOPIC TAGS: molecular association, rhodamine 6G, absorption in solvents

ABSTRACT: The purpose of the work was to investigate the influence of the polar component of a binary solvent on the association of rhodamine 6G molecules as evinced by the absorption spectra of the dye. The nonpolar component in all cases was carbon tetrachloride; the polar solvents were normal alcohols (methyl to amyl), iso-alcohols (propyl to amyl) and secondary and tertiary butyl alcohols. The binary mixtures were proportioned so that the molar ratio of the polar and nompolar components remained constant. There was observed the variation of the long wavelength electronic absorption band of rhodamine 6G as a function of the concentration (the microdensitometer traces are reproduced). It was found that the monomer absorption decreases in going from methyl to high alcohols, while the

Card 1/2

L 10165-63

ACCESSION NR: AP3000307

3

dimer absorption increases. The change in monomer absorption is attributed to changes in dipole-dipole interaction between the dye molecules and the solvent molecules. For the mixtures used the monomer absorption coefficient is proportional to the sum of the dipole moments in a unit volume of solvent. It is concluded that the different concentration induced alterations in the absorption spectra of rhodamine 6G are due to two factors: different dipole-dipole interactions between the dye and solvent molecules, and difference in the interaction between the dye molecules, resulting in different degrees of association in different solvents. "In conclusion the authors thank Ye. A. Byrsak and Z. A. Gorbunova for assistance in carrying out the work." Orig. art. has: 3

ASSOCIATION: Fizicheskiy fakul tet Moskovskogo gos. universiteta im. M. V. Lomonosova (Physics Dept, Moscow State University)

SUBMITTED: 00

DATE ACQ: 12Jun63

ENCL:

SUB CODE: PH /

NR REF SOV: 011

OTHER: 001

LEVSHIN, L. V.; AKBAROVA, D. M.

Effect of the nonpolar components of a binary solvent on the development of molecular association in rhodamine 6G. Vest. Mosk.un Ser.3:Fiz., astron.19 no. 2:16-24 Mr-Ap '64. (MIRA 17:5)

1. Kafedra optiki Moskvoskogo universiteta.

ACCESSION NR: AP4040667

8/0075/64/019/006/0693/0696

AUTHOR: Glovadskiy, Ya.; Golovina, A. P.; Levshin, L. V.; Mitteel', Yu. A.

TITLE: Rhodamine 3B as a fluorescent reagent for indium

SOURCE: Zhurnal analiticheskoy khimii, v. 19, no. 6, 1964, 693-696

TOPIC TAGS: indium determination fluorescence analysis, rhodamine dye, rhodamine 3B, fluorescent reagent, germanium tetrachloride analysis

ABSTRACT: Rhodamine 3B was selected as the least known of the rhodamine dyes and the most promising fluorescent reagent for indium. It has been shown that the fluorescence of benzene extracts of rhodamine bromoindates can be excited by a visible light with a wave length near the maximum absorption of their solutions. The optimum concentrations of acid, potassium bromide, and reagent were determined. Indium can be determined in the presence of 2000—3000 times its amount of germanium, i.e., in germanium tetrachloride or in the presence of Cord 1/2

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ASSOCIATION Lomonosov (	: Moskovskiy g Moscow State Un	30sudarstvenny*y unive niversity)	rsitet im. M. V.
SUBMITTED:	04Ju163	DATE ACQ: 06Jul64	
SUB CODE:	'GC	NO REP SOV: 009	OTHER:- 000
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LEVSHIN, V.L., doktor fiz.-matem. nauk

Successes of molecular luminescence; conference in Kharkov. Vest. AN SSSR 34 no.10:108-111 0 '64.

(MIRA 17:11)

LEVSHIN, L.V.; AKBAROVA, D.M.

Spectroscopic study of the association of 6G-rhodamine in mixtures of carbon tetrachloride with various polar solvents. Zhur. prikl. spektr. 3 no.5:441-448 N '65.

(MIRA 18:11)

# BOCHAROV, V.G., LEVSHIN, L.V.

Effect of molecular interactions on the optical properties of molecules of the rhodamine 6G dye in binary solvents. Vest. Mosk.un. Ser. 3: Fiz., astron. 20 no.4:78-82 J1-Ag '65.

(MIRA 18:12)

1. Kafedra optiki Moskovskogo gosudarstvennogo universiteta. Submitted May 24, 1964.

L 01275-66 ETT(1)/ETT(m)/ETP(j)/T IJP(c) RM

ACCESSION NR: AP5020785

UR/0048/65/029/008/1295/1301

AUTHOR: Levshin, L. V.

TITLE: Association of rhodamine series dye molecules in binary solvents Report, 13th Conference on Luminescence held in Khar'kov 25 June to 1 July 1964

SOURCE: AN SSSR. Izvestiya. Seriya fizicheskaya, v. 29, No. 8, 1965, 1295-1301

TOPIC TAGS: luminescence quenching, absorption spectrum, solution property, dye chemical, nonaqueous solution, hydrogen bonding

ABSTRACT: The author discusses the experimental data on the absorption spectra of rhodamine dyes in binary mixtures of polar and nonpolar solvents, which he and collaborators have published in a series of papers beginning in 1961. The investigations were undertaken to study association of the dye molecules. It was found, however, that the absorption spectra of rhodamine 60 varied considerably in different solvents in which its degree of association was practically the same. Both the polar and nonpolar components of the solvent affected the spectrum and also influenced the concentration damping of the <a href="Luminescence">Luminescence</a>. Some of the observed phenomena are explained in terms of the conjugated bonds and shared

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ACCESSION NR: AP5020785

7-electron clouds of the aromatic solvents, which facilitate the formation of certain types of hydrogen bonds. Rhodamine 3B was not associated in a mixture of carbon tetrachloride and propyl alcohol, whereas rhodamine 6G was associated in this solvent. This is ascribed to the formation of hydrogen bonds with the N H groups that occur in rhodamine 6G but not in rhodamine 3B. It is concluded that the solvent molecules exert a very great influence on the association of type molecules, and that in binary solvents each solvent molecular species has its own very specific effect, which is determined by the nature and structure of the molecules. Orig. art. has: 5 figures.

ASSOCIATION: Fizicheskiy fakul tet Moskovskogo gosudarstvenmogo universiteta im. M.V.Lomonosova (Physics Department, Moscow State University)

SULMITTED: 00

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CIA-RDP86-00513R000929710

6988-66

ACC NR: AP5020243

SOURCE CODE: UR/0188/65/000/004/0078/0082

AUTHOR: Rocharov, V. G.; Levshin, L. V.

ORG: Department of Optics (Kafedra optiki)

TITLE: The influence of molecular interaction on optical properties of the Rhoda-

SOURCE: Moscow. Universitet. Vestnik. Seriya 3. Fizika, astronomiya, no. 4, 1965,

TOPIC TAGS: molecular interaction, dye chemical, organic solvent, alcohol, carbontetrachloride, band spectrum, solution concentration

ABSTRACT: In concentrated solution many organic substances, especially dyes, associate with molecules of the solvent. This is usually accompanied by a change in the absorption spectra and it initiates a diminishing of the luminescence of the dye. The latter depends on the nature of the solution used. To study the intermolecular interaction effects on Rhodamine 6-G (a brilliant yellowish red basic dye used in coloring paper) measurements were carried out with both polarized and unpolarized

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UDC: \_\_ 535.34.

Card 2/2

EWT(m)/EWP(j)T. 33243-66

ACC NR: AR6016196 SOURCE CODE: UR/0058/65/000/011/1027/1028

38 GREAT B

AUTHOR: Bocharov, V. G.; Levshin, L. V.

TITIE: Influence of intermolecular interaction on the optical properties of molecules

of rhodamine 6Zh dye in binary solvents

SOURCE: Ref. zh. Fizika, Abs. 11D213 REF SOURCE: Tr. Komis. po spektroskopii. AN SSSR, t. 3, vyp. 1, 1964, 313-318

TOPIC TAGS: molecular interaction, optic property, dye chemical, absorption band, solution concentration, monomer, organic solvent

ABSTRACT: The authors investigated the concentration effects in solutions of rhodamine 6Zh dye dissolved in binary mixtures of 10 different alcohols with CCL, and water. It is shown that the concentration changes of their electronic absorption bands, and also the concentration luminescence quenching, depend strongly on the nature of the solvent. It is established that the concentration changes in the absorption spectra are due both to the formation of associated molecules of dye, and to unequal influences exerted on the absorption band of its monomers and associates on the part of the molecules of the different binary solvents. Taking these effects into account, the degrees of association are calculated for all the investigated solutions and the absorption spectra of the molecules of rhodamine 6Zh, which are in a monomer and associated states, are determined. [Translation of abstract]

SUB CODE: 20, 07/

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LEVSHIN, S.A., inzh.; GOL'TSMAN, V.Kh.

Precast reinforced concrete in the construction of the Kaunes

Gidr. stroi. 30 no.4:15-19 Ap '60.

Hydroelectric Power Station.

(Kaunas Hydroelectric Power Station)

(Precast concrete construction)

LITVINOV, L.N., kand.tekhn.nauk; MORIGEROVSKIY, V.M., kand.tekhn.nauk; LEVSHIN, S.V., inzh.; SHKLYAYEV, A.V., inzh.

Driving piles with diesel hammers not of the drop hammer type.

Transp. stroi. 11 no.7:13-14 J1 \*61. (MIRA 14:7)

(Piling (Civil engineering))

LEVSHIN, S.A., inzh.

Precast reinforced concrete in the construction of the Kaunas
Hydroelectric Power Station. Energ. stroi. no.20:30-36 '61.

(MIR. 15:1)

1. Kaunasgesstroy. (Kaunas Hydroelectric Power Station---Precast concrete construction)

SOV/124-57-4-4561

Translation from: Referativnyy zhurnal. Mekhanika, 1957, Nr 4, p 97 (USSR)

AUTHOR:

Levshin, V.A.

TITLE:

On One Application of Toroidal Coordinates in the Theory of Elasticity (Ob odnom primenenii toroidal'nykh koordinat v teorii uprugosti)

PERIODICAL:

Nauch. tr. Mosk. tekhnol. in-ta legkoy prom-sti, 1956, Nr 7,

pp 260-283

ABSTRACT: The author discusses the application of toroidal coordinates to the solution of a problem on the equilibrium of an elastic body. The solution is given in terms of Papkovich-Neuber harmonic functions. Under conditions of axisymmetrical loading of the tore, the solution retains one harmonic function, the general expression of which contains toroidal functions as introduced by Neumann. No specific examples are included in the article. It should be noted that the basic results of this work appear in an article by A. F. Zakharevich (Zap. Leningr. gorn. in-ta, 1952, Vol 26, Nr 1) in which the approach described above is applied to the solution of a problem on the stresses in a V. K. Prokopov revolving tore.

Card 1/1

CIA-RDP86-00513R000929710( APPROVED FOR RELEASE: Monday, July 31, 2000

40511

s/044/62/000/008/006/073 C111/C333

AUTHOR:

Levshin, V. A.

On the solution of homogeneous one-dimensional linear

differential equations TITLE:

PERIODICAL: Referativnyy zhurnal, Matematika, no. 8, 1962, 33, abstract 8B148. ("Nauchn. tr. Mosk. tekhnol. in-t legkoy

prom-sti", 1961, no. 20, 216-219) A simple integration method is suggested for the homogeneous

one-dimensional linear differential equation

where  $a_{l}$  are constants (k=1,...,n). It is shown that, using the Euler-substitution r=e (where t is the new parameter) for all k=1,...

.., n, the relation

card 
$$1/2^{dr^{K}}$$
 =  $\prod_{i=0}^{k-1} \left( \frac{d}{dt} - i \right)$ 

CIA-RDP86-00513R00092971 APPROVED FOR RELEASE: Monday, July 31, 2000

IEVSHIN, Vladimir Arturovich; FILONENKO-BORODICH, M.M., doktor tekhn.nauk, prof., retsenzent; VOSTROKNUTOV, S.P., doktor tekhn.nauk, prof., retsenzent; SINDEYEV, V.A., prof., retsenzent; SOKOLOV, V.I., doktor tekhn.nauk, prof., retsenzent; MINAYEVA, T.M., red.; SHAPENKOVA, T.A., tekhn.red.

[Strength of materials] Soprotivlenie materialov. Moskva, Izd-vo nauchno-tekhn.lit-ry RSFSR, 1961. 475 p. (MIRA 14:6)

(Strength of materials)

\$\124\63\000\001\035\080 D234/D308

AUTHOR:

Lëvshin, V.A.

TITLE:

A generalization of the solutions of some problem

of the theory of elasticity

PERIODICAL:

Referativnyy zhurnal, Mekhanika, no. 1, 1963, 3, abstract 1V13 (Nauchn. tr. Mosk. tekhnol. in-t legkoy prom-sti. 1961, no. 23, 198-239)

TEXT: The author describes general equations of the theory of elasticity and thermoelasticity, the equations of the plane prob-lem in Cartesian and polar coordinates. He considers known solutions of the following problem. Lamé's problem for a cylinder in this field, the problem of rotating cylinder and sphere, the problem of temperature stresses in a hollow cylinder and a hollow sphere. Abstracter's note: Complete translation

Card 1/1



S/124/63/000/003/023/065 D234/D308

AUTHOR:

Levshin, V. A.

TITLE:

Choice of stress functions for an axisymmetrically

loaded tore

PERIODICAL: Referativnyy zhurnal, Mekhanika, no. 3, 1963, 4 abstract 3V21 (Nauchn. tr. Mosk. tekhnol. in-t legkoy prom-sti. 1962, no. 24, 197-201)

TEXT: The author proposes a determined choice of harmonic functions  $\tilde{\Phi}_0$  and  $\tilde{\Phi}$  in the expression for the stress function

 $\mathbf{F} = \overline{\mathbf{Q}}_0 + \frac{\mathbf{c} \sin \theta}{\mathbf{ch} \, \eta - \cos \theta}$ 

in the case of axisymmetrically loaded tore (see Nauchn. tr. Mosk. tekhnol. in-t legkoy prom-sti, 1956, sb. 7, 260-283 - RZhMekh, 1957, no. 4, 4561). The case of a hollow tore loaded by external

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APPROVED FOR RELEASE: Monday, July 31, 2000

CIA-RDP86-00513R000929710

Choice of stress ...

and internal pressure is studied. Abstracter's note: Complete

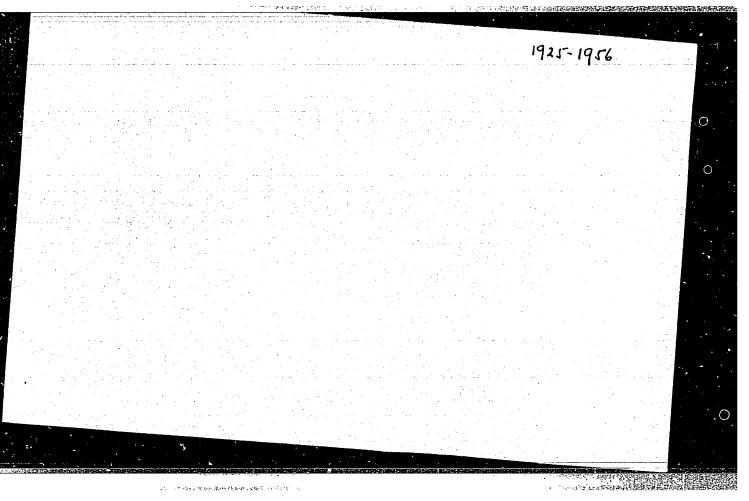
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UR/0368/66/004/004/0316/0322 IJP(c) EWT(1)/EWT(m)/T/EWP(t) SOURCE CODE: 23570-66 ACC NR. AP6012854 AUTHORS: Goryunov, V. A.; Levshin, V. L. TITLE: The influence of repeated electron localization on the photostimulated ORG: none luminescence and conductivity of ZnS-Cu monocrystals SOURCE: Zhurnal prikladnoy spektroskopii, v. 4, no. 4, 1966, 316-322 TOPIC TAGS: zinc sulfide copper, monocrystal, conductivity, electron radiation, ABSTRACT: The relation between thermoluminescence and thrmostimulated conductivity curves has been determined by optical and electrical measurements. Electron migration from deep traps to shallow ones has been studied on thermoluminescence curves under exposure to IR radiation of 1.2  $\mu$ . An estimation of the influence of repeated electron localization on photostimulated radiation at specific temperatures has been carried out. Electrons transferred by the IR radiation from the deep traps into the carried out. Discussors transferred by the in radiation from the deep traps into the conductivity band are repeatedly captured by the shallow adhesion levels due to the high probability of repeated localizations. With temperature decrease, more and more high probability of repeated localizations. With temperature decrease, more and more electrons are delegated on shallow traps. electrons are delayed on shallow traps. This leads to "freezing" of photostimulated radiation. Orig. art. has: 4 figures and 2 formulas. [Based on author's abstract][AM] SUEM DATE: 10Jul65/ 20/ SUB CODE: Card 1/1 

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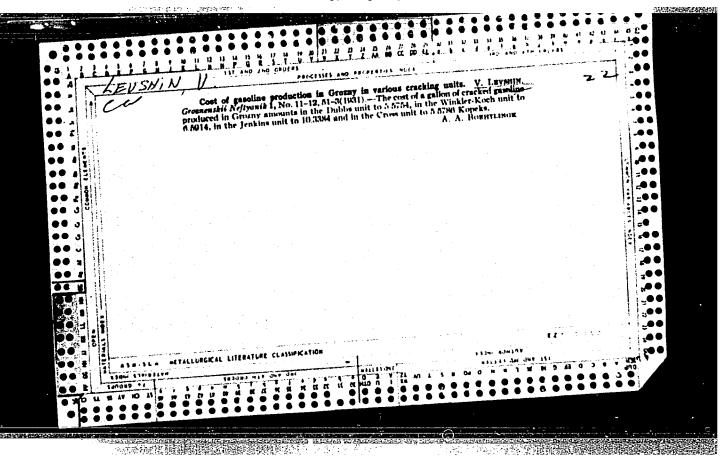
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LEVSHIN, Vadim Leonidovich

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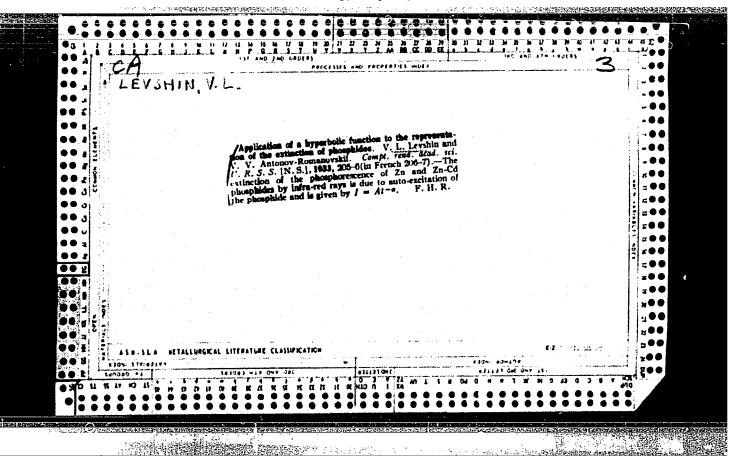
"The Polarized Fluorescence and Phosphorescence of Dye Sokutions," Zhur. Russkogo fis-khimich. ob-va. Ch. fizicheskaya, Nos. 3-4, 1925

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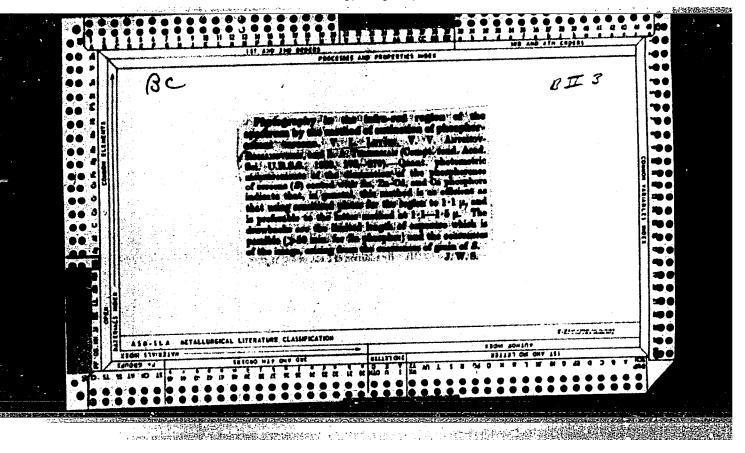


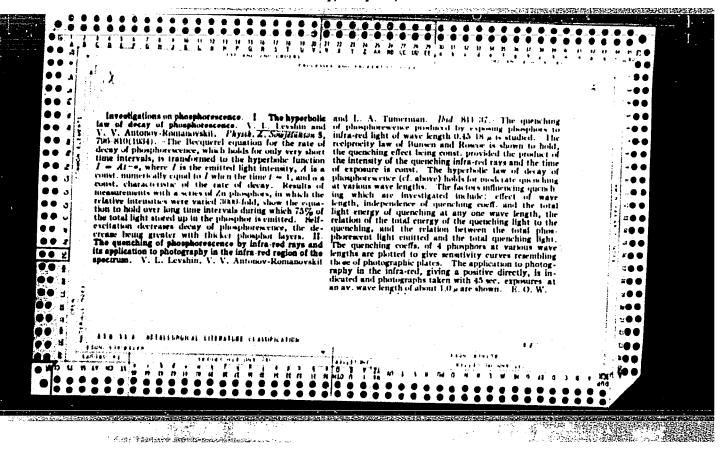
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