

KHION, Ya. V.

USSR/ Mathematics - Topology

Card 1/1 Pub. 22 - 7/47

Authors : Khion, Ya. V.

Title : ~~Regulated associative rings~~

Periodical : Dok. AN SSSR 101/6, 1005 - 1007, Apr. 21, 1955

Abstract : Regulated associative rings were studied and results presented. The study was accomplished with the help of defined concepts of the following: the embossed ideal, embossed sub-ring, and embossed sub-group. (Definition: A subset A of a regulated set B is called embossed in the B, if $b \in A$ when $a_1, a_2 \in A, b \in B$ and $a_1 \leq b \leq a_2$) • One USSR reference (1954).

Institution : M. V. Lomonosov State University, Moscow

Presented by: Academician P. S. Aleksandrov, December 21, 1954

Khion, Ya. V.

Call Nr: AF 1108825

Transactions of the Third All-union Mathematical Congress (Cont.) Moscow,
Jun-Jul '56, Trudy '56, V. 1, Sect. Rpts., Izdatel'stvo AN SSSR, Moscow, 1956, 237 pp.
Suprunenko, D. A. (Minsk). Linear Nilpotent Groups. 35

Turkin, V. K. (Moscow). Quasi-monomial Representations
of Groups. 35

Urazbayev, B. M. (Alma-Ata). On Some Asymptotic Formulas
in Algebra. 35-36

Mention is made of Delone, B. N. and Fadeyev, D. K.

Khion, Ya. V. (Tartu). Rings Normed by Means of Semi-groups. 35-36

Chernikov, S. N. (Molotov). Nilpotent Groups. 37-40

Mention is made of Kurosh, A. G., Shmidt, O. Yu., Ado, I. D.,
Mal'tsev, A. I., Myagkov, N. N., Mukhamedzhan, Kh. Kh.,
Glushkov, V. M., Golovin, O. N., Sesekin, N. F., Smirnov, D. M.,
Plotkin, B. I., Charin, V. S.

Card 12/80

Nhion Ya.V.
SUBJECT USSR/MATHEMATICS/Algebra CARD 1/3 PG - 908
AUTHOR KHION Ya.V.
TITLE Ordered semigroups.
PERIODICAL Izvestija Akad.Nauk 21, 209-222 (1957)
reviewed 7/1957

The set P is called an ordered semigroup if 1) P is closed with respect to the associative multiplication defined in it; 2) P is a (linear) ordered set; 3) for all $\alpha, \beta, \gamma \in P$ from $\alpha \geq \beta$ there follows: $\alpha\gamma \geq \beta\gamma$ and $\gamma\alpha \geq \gamma\beta$; 4) in P there exists an element 0 with the property $\alpha 0 = 0\alpha = 0$, $0 \leq \alpha$ for every $\alpha \in P$; 5) from $\alpha\gamma = \beta\gamma \neq 0$ it follows $\alpha = \beta$ and from $\gamma\alpha = \gamma\beta \neq 0$ it follows $\alpha = \beta$ for all $\alpha, \beta, \gamma \in P$.
A subset A of the semigroup P is called subsemigroup in P if A is closed with respect to the multiplication defined in P . A subset A of the semigroup P is called a right ideal in P if from $\alpha \in A$ and $\gamma \in P$ it follows $\alpha\gamma \in A$. A subset A of an ordered semigroup P is called convex in P if from $\alpha_1, \alpha_2 \in A$ and $\alpha_1 \leq \beta \leq \alpha_2$ it follows $\beta \in A$. If in an ordered semigroup P a convex two-sided ideal I is given, then with respect to it a new ordered semigroup can be formed which is denoted as a factor semigroup of P with respect to I : P/I . $\alpha \in P/I$ corresponds to the element $\alpha \in P$. From $\alpha \in I$ it follows $\alpha = 0$.

Izvestija Akad.Nauk 21, 209-222 (1957)

CARD 2/3

PG - 908

Let from $\alpha \geq \beta$ follow $\bar{\alpha} \geq \bar{\beta}$. Let $\bar{\alpha}\bar{\beta} = \bar{\alpha\beta}$. For an element α of an ordered semigroup we have either $\alpha^2 < \alpha$ or $\alpha^2 = \alpha$ or $\alpha^2 > \alpha$. The set of those α for which $\alpha^2 < \alpha$ is called K and these α are called integral.

Elements being different from zero, for which $\alpha^2 = \alpha$, are called unities. Let M be the set of integral elements which are no unities. If P contains no non-trivial convex two-sided ideals, then P is called simple.

Theorem 1: K and $P \setminus K$ are convex subsemigroups in P.

Theorem 2: Every convex subset in K which contains the element 0 is a two-sided ideal in K.

Theorem 3: Beside of the elements M, K contains at most one element $\bar{\varepsilon}$.

If $M \neq K$, then M is a maximal convex two-sided ideal in K; the ideal M is a prime ideal in K and the factor semigroup

K/M consists of the elements $\bar{0}$ and $\bar{\varepsilon}$, where $\bar{\varepsilon}^2 = \bar{\varepsilon}$.

Theorem 4: If I is a convex right (left) ideal in P and $I \neq P$, then $I \subset M$.

Theorem 5: The totality N of all nilpotent elements of an ordered semigroup P is a convex two-sided ideal in P. N is the union of all right nilpotent ideals of P. The factor semigroup P/N has no zero divisors. N is called the radical of P.

Theorem 6: If $P \neq M$, then in P it exists a non-trivial maximal convex

Izvestija Akad. Nauk 21, 209-222 (1957)

CARD 3/3

PG - 908

two-sided ideal Λ . The factor semigroup P/Λ contains no non-trivial convex two-sided ideals.

Theorem 7: A simple ordered semigroup contains no non-trivial convex one-sided ideals.

In an ordered semigroup archimedean classes are introduced as follows: let $\alpha \geq \beta$ and $\alpha, \beta \in K$, then α and β have to belong to a class if it exists an n such that $\alpha^n \leq \beta$. Let $\alpha \geq \beta$ and both α and β non-integral, then α and β have to belong to a class if it exists an n such that $\alpha \leq \beta^n$.

Theorem 8: The decomposition into archimedean classes is a decomposition into pairwise disjoint convex subsemigroups. Every decomposition of P into pairwise disjoint convex subsemigroups can be continued up to a decomposition into archimedean classes.

Some further theorems concern integral ordered groups, the notion of congruence being determined by the decomposition into archimedean classes and so-called archimedean semigroups.

AUTHOR

KHION YA.V.

TITLE

The Rings normalized with the help of half-groups.
(Kol'tsa, normirovannyye pri pomoshchi polugrupp, - Russian)
Izvestiia Akad. Nauk SSSR, Ser. Mat. 1957, Vol 21, Nr 3,
pp 311-328 (USSR).

PERIODICAL

ABSTRACT

The theory of these rings is here developed upon the basis
of the theory of ordered half-groups. The attempt can be
made to transfer the conception of norm from the fields
to the rings, so that the axiom $\|ab\| = \|a\| \|b\|$ is
conserved. If, however, it is intended to normalize suffi-
ciently voluminous classes of rings (e.g. rings with zero
divisors), also normalization with the help of ordered
half-groups (and not merely with half-groups) must be
admitted. The present work discusses the generalizations
in this direction.

The first chapter gives a definition of the conception
"normalized ring" and discusses the connection between
a ring and a normalizing half-group. One of the theorems

CARD 1/2

38-3-2/7

38-3-2/7

The Rings normalized with the help of half-groups.

mentioned here says: For any ordered half-group P there exists a ring R, which is normalized with the help of this half-group. The next chapter deals with the most simple properties of a normalized ring and the ideal. The maximum convex ideal and the simple normalized rings are then dealt with. Thus, a simple normalized ring contains no non-trivial unilateral convex ideals. The following chapter investigates whole normalized rings, i.e. such rings as have been normalized with the help of a whole ordered half-group. In this and in the preceding chapter a considerable number of theorems is mentioned. In conclusion the ordered rings are investigated.
(No Illustrations)

ASSOCIATION: not given.
PRESENTED BY: A.N. KOLMOGOROV, member of the Academy.
SUBMITTED: 23.12. 1955.
AVAILABLE: Library of Congress.

CARD 2/2

KHION, Ya.V.

Partially ordered semigroups in which the characteristic convex
subsemigroups do not intersect. Izv.AN SSSR.Ser.mat. 27 no.1
67-74 Ja-F '63. (MIRA 16:2)
(Groups, Theory of)

KHIPKOVA, G. A.

Dissertation: "The Effect of Some Physicochemical Factors on the Fluidity of Cement-Producing Slime." Cand Tech Sci, Novocherkassk Polytechnic Inst, Novocherkassk, 1953

SO: Referativnyy Zhurnal, No. 5, Dec 1953, Moscow, AN USSR, (W-30928)

KHIRA LAL; PETRIY, O.A.; PODLOVCHENKO, B.I.

Electrolytic oxidation of organic matters on platinized platinum at hydrogen adsorption potentials. Dokl. AN SSSR 158 no. 6:1416-1419 Ju '64. (MIRA 17:12)

1. Moskovskiy gosudarstvennyy universitet im. Lomonosova.
Predstavлено академиком A.N. Frumkinyem.

KHIRA LAL; PETRIY, O.A.; PODLOVCHENKO, B.I.

Role of adsorption of intermediate reaction products in the
electrooxidation of methanol in an acid solution. Elektrokhimia
1 no. 3:316-320 Mr '65. (MIRA 18:12)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova.

GRINSHPUN, L.Ya.; PYLAYKIN, P.A.; KHIRDZHIYEV, S.G.; PERTSOVSKAYA, Ye.V.

Tanks on powerful horizontal hydraulic presses for the extrusion of
aluminum alloys. Kuz.-shtam.proizv. 6 no.l:21-24 Ja '64.
(MIRA 17:3)

L 1359-66 EWT(m)/EMP(t)/EMP(k)/EMP(b)/EWA(h)/EWA(c) JD/HW
ACCESSION NR: AP5024359 UR/0286/65/000/015/0024/0024
621.984.2

36

B

AUTHOR: Yefimov, I. A.; Sysoev, P. M.; Pylaykin, P. A.; Shtin, L. M.
Khirdzhiev, S. G.

TITLE: A multilayer container for the extrusion process. Class 7, No. 173195

SOURCE: Byulleten' izobreteniij i tovarnykh znakov, no. 15, 1965, 24

TOPIC TAGS: metal extrusion, metallurgic process

ABSTRACT: This Author's Certificate introduces a multilayer container for the extrusion process. The device is built up from several hoops fitted concentrically one over the other. To economize on costly steels and lighten the container, spacers are placed between two or several pairs of hoops. These spacers are made in the form of hoops which are cut away in one or several places along the generatrix.

ASSOCIATION: none

SUBMITTED: 27Mar64

NO REF Sov: 000

ENCL: 01

SUB CODE: IE, MM

OTHER: 000

Card 1/2

L 1359-66

ACCESSION NR: AP5024359

ENCLOSURE: 01

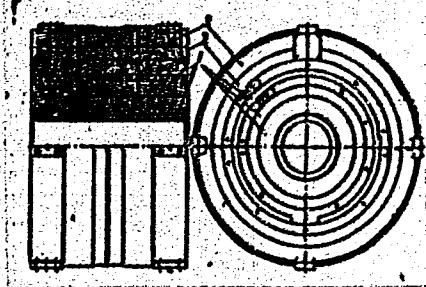


Fig. 1. 1--hoops; 2--spacer; 3--outside hoop

Card 2/2

L 15329-66 ENT(d)/ENT(m)/EWP(v)/T/EWP(k)/EWP(h)/EWP(l) DJ
ACC NRI AP6001006 (N)

SOURCE CODE: UR/0286/65/000/022/0077/0077

AUTHORS: Pylaykin, P. A.; Khirdzhiev, S. G.

ORG: none

TITLE: Hydraulic cylinder. Class 47, No. 176470 [announced by Scientific Research Construction Technological Institute of Heavy Machine Construction of Uralmashzavod (Nauchno-issledovatel'skiy konstruktorsko-tehnologicheskiy institut tyazheologa mashinostroyeniya Uralmashzavod)]

SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 22, 1965, 77

TOPIC TAGS: hydraulic device, hydraulic equipment

ABSTRACT: This Author Certificate presents a hydraulic cylinder including a case with a flange mounted on a fixed support by means of a ring. To decrease the stress at the support points of the flanges, to increase the reliability of the cylinder mounting, and to increase the wear resistance, an elastic element is placed between the outer surface of the case flange and the inner surface of the fixed support (see Fig. 1). To ensure the mechanical properties of the elastic element capable of withstanding the pressure of the working medium in the cylinder, the elastic element is reinforced.

Card 1/2

UDC: 621-222.1 621.888.6

I. 15329-66
ACC NR: AP6001006

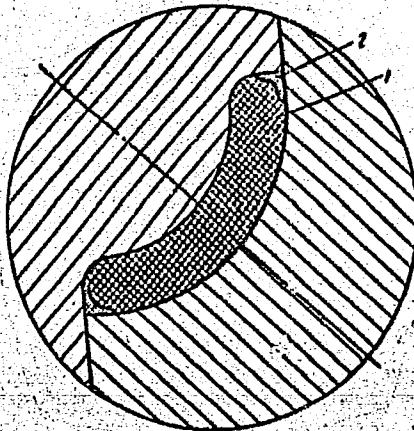


Fig. 1. 1 - Elastic
element; 2 - rein-
forcing.

Orig. art. has: 1 diagram.

SUB CODE: 13/

SUER DATE: 22Oct64

Card 2/2 Inv. No. _____

ACCESSION NR: APL011134

S/0182/64/000/001/0021/0024

AUTHORS: Grinshpun, L. Ya.; Pyatlaykin, P. A.; Khirdzhilyev, Ye. V.;
Pertsovskaya, Ye. V.

TITLE: Containers of high power horizontal hydraulic presses for pressing
aluminum alloys

SOURCE: Kuznechno-shtampovochnoye proizvodstvo, no. 1, 1964, 21-24

TOPIC TAGS: hydraulic press, press container, 5KhNV steel, 5KhNM steel, 5KhNM2
steel, 38Kh2N3M steel, 3Kh2N2MVF steel, 27Kh2N2MVF steel

ABSTRACT: The technological requirements of containers for pressing Al alloys
were limited by the temperatures up to 430°C, specific stresses up to 50 kg/mm²,
and the maximum press force 12 000 T. A commonly used container consisted of a
frame and a conical bushing. Both the frame and the bushing were made of high-
alloy steels 5KhNV or 5KhNM. They had a number of shortcomings associated with
the shape of the bushing and the metal used. For this reason, several research
projects leading to the design of more suitable containers were undertaken at

Card 1/3

ACCESSION NR: AP4011134

the Uralmashzavod (Ural Machine Plant). The new types had multilayer frames and cylindrical bushings (see Fig. 1 of the Enclosure). The problem of obtaining steels with high mechanical properties ($\sigma \geq 150$ kg/mm²) at 480C has not yet been solved. The steels studied so far were: 5KhNM2, 38Kh2N3M, 3Kh2N2MVF and 27Kh2N2MVF. A standard mathematical procedure for calculating the strength of a multilayered thick-wall cylinder subjected to internal pressure is presented. Orig. art. has: 1 table, 3 figures, and 2 formulas.

ASSOCIATION: none

SUBMITTED: 00

DATE ACQ: 14Feb64

ENCL: 01

SUB CODE: MM

NO REF Sov: 000

OTHER: 000

Card 2/3

L 35976-66 EWT(m)/EWP(t)/ETI IJP(c) JD

ACC NR: AP6016042 (N) SOURCE CODE: UR/0185/66/011/005/0502/0506 69

AUTHOR: Khirich, I.Ya.

66

B

ORG: Odessa Technological Institute of the Food and Refrigeration
Industry (Odes'kyy tekhnolohichnyy instytut kharchovoyi i kholodil'noy
promyslovosti)

TITLE: Thermoelectric and galvanomagnetic properties of bismuth -
antimony single crystals 27

SOURCE: Ukrayins'kyy fizychnyy zhurnal, v.11, no.5, 1966, 502-506

TOPIC TAGS: antimony crystal, bismuth crystal, thermal emf, magnetic
field, galvanomagnetic effect, thermoelectric property, electric resistance,
temperature dependence

ABSTRACT: Samples of Bi_{97%} Sb_{3%}, obtained by the zone leveling method,
have been investigated. The variation of the thermal emf in the
magnetic field (H=0-15 koe) was measured as a function of temperature.

Card 1/2

L-35976-66

ACC NR: AP6016042

The superposition of a magnetic field of 2 koe intensity increases the electromotive force. At a temperature T=200K and H=15 koe, the relative change in thermal emf was $\frac{\Delta\alpha}{\alpha} = 0.5 \alpha$. The effect of the

magnetic field intensity on the electric resistance was investigated. At a temperature of 100K and H=15 koe, the electric resistance amounted to $\frac{\Delta\rho_H}{\rho_0} = 14.6$. The temperature dependence of the resistivity and Hall

emf were used in determining the carrier concentrations, while the electron and hole mobilities were determined from formulas for isotropic metals. The author thanks V.A.Nayer, I.D.Konozenko, and V.L.Vinets'kiy for their valuable advice and help in this work. Orig. art. has: 4 figures and 7 formulas. [Based on author's abstract] [NT]

SUB CODE: 11,20/ SUBM DATE: 06Apr65/ ORIG REF: 003/ OTH REF: 007

ns
Card 2/2

CHIZHIKOV, D.M.; KHIRIK, A.S. [deceased]

Declassified

Determining thermophysical properties of oxygen and sulfur
compounds of zinc and copper. Trudy Inst. met. no. 12:79-84
'63. (MIRA 16:6)

(Zinc compounds--Thermal properties)
(Copper compounds--Thermal properties)

CHIZHIKOV, D.M.; KHIRIK, A.S.[deceased]

Determining thermophysical properties of oxygen and sulfur compounds of zinc and copper. Trudy Inst. met. no.12:79-84 '63. (MIRA 16:6)

(Zinc compounds--Thermal properties)
(Copper compounds--Thermal properties)

KHIRIK, A. S.

USSR/ Engineering - Metal machining

Card 1/1 Pub. 128 - 8/25

Authors : Khirik, A. Sh., Engineer

Title : New principles of heading with swinging punches on rotating automat

Periodical : Vest. mash. 35/4, 35-38, Apr 1955

Abstract : The testing of a new rotary action automatic heading machine is described. The basic kinematic feature of this header is that all its working organs and links execute a uniform rotary motion in one direction. In addition some of its links, e.g., the punches are displaced forward thus creating a rocking (swinging) motion. It is shown that the utilization of rocking punches in the heading automat warrants the possibility of heading even during the rotation of the billets and the working discs. The characteristics of the new automatic heading machine are described. Drawings; illustrations.

Institution :

Submitted :

ANDRUSHKO, A.F., prepodavatel'; VORONKOV, E.N., prepodavatel',
KUBETSKIY, G.A., prepodavatel', MALYSHEV, G.A., prepodava-
tel'; SETYUKOV, L.I., prepodavatel'; SOKOLOV, A.A., prepodavatel';
KHIRIN, A.A., prepodavatel'; SHALIMOVA, K.V., prof.; ENYUTIN, V.V.
rea., LARIONOV, G.Ye., tekhn. red.

[Specialized guide to semiconductors and semiconductor devices]
Spetsial'nyi praktikum po poluprovodnikam i poluprovodniko-
vym priborom. Moskva, Gos. energ. izd-vo, 1962. 303 p.
(MIRA 15:2)

(Semiconductors) (Transistors)

"APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722020007-1

1982. ADJUSTMENT AND OPERATING INSTRUCTIONS

APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722020007-1"

PESTRIY, N.V., inzhener; KHIRIN, N.D., inzhener.

Adjusting wetting-type ash removers of centrifugal scrubbers.
Elek.sta. 28 no.9:23-27 S '57. (MIRA 10:11)
(Scrubber (Chemical technology))

KHIRIN, N.D., inzh.; PESTRIY, N.V., inzh.

Performance of wet ash collectors operating with ashes from
milled peat. Teploenergetika 7 no.3:52-56 Mr '60.
(MIRA 13:5)

1. Yuzhnoye otdeleniye Gosudarstvennogo tresta po organizatsii
i rationalizatsii elektrostantsiy.
(Ash disposal)

VOLKOVA, Ye.I., inzh.; KHIRIN, N.D., inzh.; BARYSHNIKOV, A.P., inzh.;
KOZHEVNIKOV, G.A., inzh.; KHOKHRIN, K.G., inzh.; BABKOV, V.A.,
inzh.; VNUKOV, A.K., kand.tekhn.nauk

Starting clutch for draft and blowing machinery and pit mills.
Teploenergetika 8 no.6:31-32 Je '61. (MIRA 14:10)

1. Yuzhnoye otdeleniye Gosudarstvennogo tresta po organizatsii i
ratsionalizatsii elektrostantsiy.

(Clutches (Machinery))
(Electric power plants--Equipment and supplies)

PESTRIY, N.V., inzh.; KHIRIN, N.D., inzh.; LOZINSKIY, R.P., inzh.;
VESELOV, V.T., inzh.

Studying the model of a wet ash collector with a gas overfeed system. Teploenergetika 9 no.1:11-14 Ja '62. (MIRA 14:12)

1. Yuzhnoye otdeleniye Gosudarstvennogo tresta po organizatsii
i ratsionalizatsii elektrostantsiy.

(Gases--Purification)
(Electric power plants--Equipment and supplies)

VESELOV, V.T., inzh.; DAROVSKIY, Ye.T., inzh.; LOZINSKIY, R.P., inzh.;
KHIRIN, N.D., inzh.

Adjustment and testing of type MP-VTL ash collectors with a
4,500 mm. diameter. Teploenergetika 9 no.11:41-45 N '62. (MIRA 15:10)

1. Yuzhnoye otdeleniye Gosudarstvennogo tresta po organizatsii i
ratsionalizatsii rayonnykh elektrostantsiy i setey.
(Fly ash)

+ 7 (SAC/EP)(SAC/EP(b)) (J) (SAC/EP)(SAC/EP(b)) (J)

REF ID: AP4047340

S/0133/64/MG/105/0008/0011

Shalimova, K. V.; Andrushko, A. F.; Khirin, V. S.; Moro-

TITLE: Optical properties of powders of cadmium sulfide of hexagonal modification at 77.3K

SOURCE: IVUZ. Fizika, no. 5, 1964, 3--11

TOPIC TAGS: cadmium sulfide, luminescence spectrum, luminescence analysis, polycrystal, reflection band, optical absorption

ABSTRACT: Inasmuch as earlier research on the fine structure in the absorption, reflection, emission and excitation of luminescence of hexagonal-modification cadmium sulfide was limited to single-crystal samples and thin films deposited on heated substrates, the authors have undertaken to determine the dependence of the optical properties of the hexagonal modification of cadmium sulfide on the condi-

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L-22-32-05

ACCESSION NR: AP4047340

C

the under which it is obtained. α -CdS powders with fine crystalline structure, obtained by different means at different temperatures and with different reagents, were used in the investigations. All the spectra were obtained in unpolarized light at liquid-nitrogen temperature. The reflection spectra were obtained with the ISP-51 spectrograph with UF-84 camera, while the radiation and excitation spectra were obtained with the same instrument equipped with an attachment. The wavelength range is 400-600 nm. The powders have fine reflections at room temperature, and the luminescence excitation spectra display a comparison of these spectra with the absorption maxima which corresponds to a maximum of intensity. The cadmium-sulfide powders give blue and green reflective spectrum. The blue band lies between with positions from sample to sample, while the green lies about four (4) $\times 10^{-4}$ nm, 5223, 5295, and 5394 Å. It depends on the spectrum on the preparation. The results will show the optimum

ACCESSION NR: AP4047340

The absorption in the cadmium sulfide is due to impurities and is connected with the violation of the stoichiometry of the crystals.
It has: 3 figures.

ASSOCIATION: Moskovskiy energeticheskiy institut (Moscow Power
Eng Institut)

SUBMITTED: 28Apr63

ENCL: 00

SUB CODE: OP, IC

NR REF Sov: 009

OTHER: 003

Card 3/3

ACC NR: AP6033438

SOURCE CODE: UR/0051/66/021/004/0456/0459

AUTHOR: Shalimova, K. V.; Khirin, V. N.; Korolev, O. I.

ORG: none

TITLE: Temperature dependence of the absorption, reflection, emission, and excitation spectra and of the quantum yield of luminescence in polycrystalline films of cadmium sulfide

SOURCE: Optika i spektroskopiya, v. 21, no. 4, 1966, 456-459

TOPIC TAGS: cadmium sulfide, absorption spectrum, emission spectrum, excitation spectrum, quantum yield, luminescence spectrum, semiconducting film, temperature dependence

ABSTRACT: This is a continuation of earlier work (DAN SSSR v. 139, 938, 1961) on the temperature dependence of the absorption and reflection of cadmium sulfide films, where a photographic procedure was used. In the present study, the tested polycrystalline films were of hexagonal modification, and were prepared by a thermal sublimation of the initial substance in vacuum, using a procedure described by the authors elsewhere (Kristallografiya v. 8, 774, 1963). The temperature dependence of the spectra and of the quantum yield were measured with an optical system based on a spectrometer (DFS-12), universal monochromator (UM-2), a double monochromator (DMR-4), and a photo-electronic continuous spectrum recorder (FEP-1). The absorption spectra in the temperature range 77 - 300K were plotted with a recording spectrophotometer SF-10. The

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UDC: 535.34 + 535.37.096

ACC NR:

APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722020000

use of the spectrophotometer in the present investigation made it possible to determine the absorption coefficient more accurately with allowance for reflection. The results show that all the absorption bands shift toward longer wavelengths with increasing temperature. The shift is linear. In addition, the bands become deformed with increasing temperature and their half-width increases. The quantum yield was also investigated by the authors earlier (Opt. i spektr. v. 20, 1063, 1966). The luminescence spectrum, on the other hand, shifts toward shorter wavelengths, with all bands of the spectrum experiencing an equal shift. The relative quantum yield of the luminescence decreases with increasing temperature. The temperature dependence of the photoluminescence is the same for different excitation wavelengths, agreeing well with the Gurney and Mott relation. Orig. art. has: 3 figures and 2 formulas.

SUB CODE: 20/ SUBM DATE: 07Apr65/ ORIG REF: 005/ OTH REF: 003

Card 2/2

L 13079-65 EWP(m)/EWP(t)/EWP(b) IJP(c)/AEDC(a)/AS(mp)-2/ESD(c)/ESD(gs)

ACCESSION NR: AP4047358

S/0139/64/000/005/0119/0124

AUTHORS: Shalimova, K. V.; Andrushko, A. F.; Khirin, V. N.; Morozova, N. K.

TITLE: Optical properties of powders of the cubic modification of cadmium sulfide and their changes in the $\beta \rightarrow \alpha$ phase transition

SOURCE: IVUZ. Fizika, no. 5, 1964, 119-124

TOPIC TAGS: cadmium sulfide, cubic crystal, powder, phase transition, light absorption, light emission

ABSTRACT: In order to obtain additional information on the mechanism of absorption and emission of light in CdS, the authors investigated the spectra of diffuse reflection, emission, and excitation of luminescence of CdS powders of cubic modification, and also investigated the spectra of β -CdS powders annealed at high temperatures in air and in vacuum atmosphere. All the spectra were investigated in unpo-

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L 13079-65

ACCESSION NR: AP4047358

larized light at liquid-nitrogen temperature. The reflection spectra were obtained with an ISP-51 spectrograph with UF-84 camera. The reflection and emission spectra were investigated with the same spectrograph and a photoelectric attachment. The cadmium sulfide powder was precipitated from solutions and heat treated at temperatures from 200°C. One band is observed in PCdS in reflection and in emission at 77K. After heating the β -CdS above 340°C, this band gradually attenuates and vanishes, and is replaced by reflection peaks characteristic of the hexagonal modification. The emission and reflection spectra show a similar behavior. art. has: 3 figures.

ASSOCIATION: Moskovskiy energeticheskiy institut (Moscow Power Engineering Institute)

DATE REC'D: 28Apr63

ENCL: 00

SUB CODE: OP, 88

NR REF Sov: 010

OTHER: 003

Page 2/2

$\text{EST}(1)/\text{EST}(\mathbf{a})/\text{EWP}(\mathbf{t})/\text{EWP}(\mathbf{b})$ $\{j=4\}$ (J^{P})

卷之三

Millora, A., V. I. Kirin, V. N. Korol'chuk, 2.

...decrease of polycrystalline films of cadmium sulfide at the

11-0158, DC, 1, 1965, 80-84

1. Luminescence, cadmium sulfide, polycrystalline, single crystal, spectral excitation, fine structure.

This was made of the spectra of absorption, reflection, extinction, and transmission of polycrystalline SiC films.

These investigations are in progress.

... of CDS files was never made before, and some were omitted.

elimination of powder in an atmosphere of argon or hydrogen sulfide in

subsequent crystallization on heated and unheated substrates. Samples

esters heated below 350°C., regardless of the substituents, which

rate absorption and emission processes in the system.

samples heated above 350°C show a significant increase in the relative amount of the 100% spinel phase.

AP5006056

APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722020007

reflection, luminescence excitation, and emission spectra, due to transitions between states of the excess gadolinium atoms and the host lattice. Transitions at 441, 4670, and 4740 Å were observed. The excitation spectra of the spectral distribution of the luminescence of the film was resulting wavelength in the 400-450 Å range. A fine structure reflection, luminescence excitation, and emission spectra was film having hexagonal-symmetry lattice.

1983) energeticheskiy institut (Moscow Power Engineering Institute).

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MR. COPE: CP 26

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L 42898-66 EWT(1)/EWT(m)/EWP(t)/ETI IJP(c) JD/GG
ACC NR: AP6018447 SOURCE CODE: UR/0051/66/020/006/1063/1065

AUTHOR: Shalimova, K. V.; Khirin, V. N.; Korolev, O. I.

ORG: none

74
B

TITLE: Spectral distribution of relative quantum yield for photoluminescence of poly-crystalline cadmium sulfide films at 77°K

SOURCE: Optika i spektroskopiya, v. 20, no. 6, 1966, 1063-1065

TOPIC TAGS: cadmium sulfide, photoluminescence, light excitation, excitation spectrum, absorption spectrum, spectrophotometry, quantum yield, monochromator, semiconducting film, polycrystalline film

ABSTRACT: The absorption, reflectance, excitation and quantum yield of luminescence from polycrystalline cadmium sulfide films at liquid nitrogen temperature (77°K) were investigated. In the corresponding spectra in the 210-360 m μ region, four broad bands with maxima at 330, 290, 240 and 215 m μ were observed while a fine structure consisting of 4-5 bands was observed in the 460-490 m μ region. The spectral distribution of green luminescence of polycrystalline CdS films is independent of the wavelength of the excitation light. The radiation intensity of this luminescence is directly proportional to the intensity of the excitation light over three orders of magnitude. These conclusions are based on a series of measurements carried out by the authors on hexa-

UDC: 535.37

Card 1/2

L 42898-66

ACC NR: AP6018447

gonal CdS polycrystalline films at 77°K. The experimental equipment included a double grating monochromator, a grating spectrometer, universal monochromator, recording spectrophotometer, and a photomultiplier detector. The quantum yield was calculated for given excitation and radiation conditions for samples of CdS films from 0.15 to 1.1 μ thick. All samples exhibited identical spectral distributions of their quantum yield. Spectral distribution of green fluorescence of the CdS film is shown in figure 1.

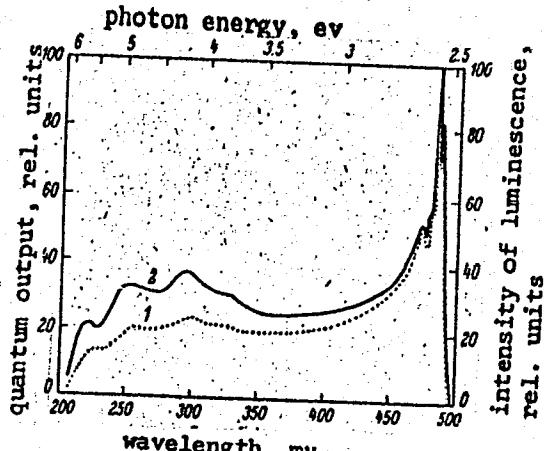


Fig. 1.

Curve 1 shows the spectral distribution of the excitation light for the green luminescence, curve 2 shows the relative quantum yield. Orig. art. has: 3 figures.

SUB CODE: 201

SUBM DATE: 07Apr65/

ORIG REF: 006/

OTH REF: 002

Card 2/2

GOLIYAT, Yu. S. (Veterinary Doctor, Peremyshl' Inter-District Veterinary Bacteriological Laboratory), KHILIVSKIY, V. A. (Head Veterinary Doctor, Bobr. District, L'vov Oblast). (Abstracted by NOSKOV, A. I.)

"Treating herpes tonsurans with a 50% phenothiazine emulsion".....
Veterinariya, vol. 39, no. 3, March 1952 pp. 27

"APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722020007-1

KHIRMUNSKIY, V. M.

"Sredneaziatskiye narodnye skaziteli (traditsiya i tvorcheskaya improvizatsiya)."
report submitted for 7th Intl Cong, Anthropological & Ethnological Sciences,
Moscow, 3-10 Aug 64.

APPROVED FOR RELEASE: 09/17/2001

CIA-RDP86-00513R000722020007-1"

KHIREVYKH, L.A., inzh.

Testing the remote control system in fields of the Abin Petroleum
Trust. Neftianik 5 no.6:11-12 Je '60. (MIRA 13:?)

1. Neftepromyslovoye upravleniye Abinneft' Krasnodarskogo sov-
narkhoza.
(Abin region (Krasnodar Territory)---Remote control)

GABRIYELOV, L.V.; SHEVTSOV, A.A.; LITVINOV, A.Ya.; KHIRNYKH, L.A.

Automation of group measuring installations. Neftianik ? no.4:11-13
Ap '62. (MIRA 15:11)

1. Nachal'nik tsekha avtomatiki neftepromyslovogo upravleniya
Abinneft' (for Gabriyelov). 2. Glavnyy inzh. neftepromyslqvogo
upravleniya Abinneft' (for Shevtsov). 3. Starshiy inzh. promyslovoy
gruppy tsekha avtomatiki neftepromyslovogo upravleniya Abinnert'
(for Litvinov). 4. Starshiy inzh. gruppy telemekhanizatsii tsekha
avtomatiki neftepromyslovogo upravleniya Abinneft' (for Khirnykh).
(Kuban-Petroleum-Measurement)
(Automatic control)

ABRUKIN, Abram L'vovich; KHIRNYKH, Leonid Andreyevich; PEREVERZEV,
V.V., red.; GOR'KOVA, A.A., ved. red.; YAKOVLEVA, Z.I.,
tekhn. red.

[Remote control in petroleum production] Telemekhanizatsiya
dobychi nefti. Mskva, Gostoptekhizdat, 1962. 302 p.
(MIRA 16:2)

(Remote control)
(Oil fields--Equipment and supplies)

KHIRNY, N.I., inzhener-podpolkonik; NIKOLAYEV, S.A., inzhener-mayor

Mechanization to replace manual labor. Vest.Vozd.Fl. no.3:61-68
Mr '61.

(Airplanes—Equipment and supplies) (MIRA 14:6)

KHIRNYY, V. P.

Sugar - Manufacutre and Refining

How we achieved rhythmic labor. Sakh. prom. 26. No. 6, 1952.

Monthly List of "ussian Accessions, Library of Congress, August 1952. Unclassified.

ALMAZOV, T. V.; KHIRNYY, Yu. M.

Sources of negative hydrogen ions. Prib. i tekhn. ekspl. no. 5:54-58
S-0 '57.

(Ionization of gases) (Hydrogen) (MIRA 10:12)

SOV/120-58-2-13/37

AUTHOR: Kbirnyy, Yu. M.

TITLE: A Source of Negative Hydrogen Ions (Istochnik otritsatel'nykh ionov vodoroda)

PERIODICAL: Pribory i Tekhnika Eksperimenta, 1958, Nr 2, pp 51-53
(USSR)

ABSTRACT: The device is based on the high frequency source of positive ions with a longitudinal magnetic field given in Ref.3 and is shown in Fig.1. The negative ions are produced by charge exchange between protons and hydrogen, or deuterons and deuterium, in the capillary of a high frequency ion source (Fig.1). The beam of ions thus produced is accelerated to 20 kev and analysed by a magnet. Analysis of the ion beams shows that the ions coming out from the source are practically monoenergetic. The working life of the source is more than 200 working hours and the amount of hydrogen supplied to obtain a current of 20 μ amp (H^-_1) is 28 cc/hour. A similar amount of deuterium gave a current of about 10 μ amp of D^-_1 . Fig.2 shows the dependence of the current on the diameter of

Card 1/2

A Source of Negative Hydrogen Ions.

SOV/120-58-2-13/37

the capillary. A. V. Almazov is thanked for his supervision.
There are 3 figures and 4 references, of which 1 is Soviet
and the rest English.

SUBMITTED: August 5, 1957.

Card 2/2

1. Hydrogen ions--Sources
2. Ion beams--Analysis

KHIRNYY, Yu. M., CAND TECH SCI, "SOURCES OF NEGATIVE
HYDROGEN IONS." Moscow, 1961. (Moscow, ENGINEERING-
PHYSICS INST). (KL, 3-61, 221).

279

21392

26.2312

S/120/61/000/002/002/042
E032/E114

AUTHORS: Khirnyy, Yu.M., and Kochemasova, L.N.

TITLE: An injector of negative hydrogen ions

PERIODICAL: Pribory i tekhnika eksperimenta, 1961, No. 2, pp. 14-19

TEXT: The electrostatic generator has retained its importance in nuclear reactor studies since it can be used to produce accelerated particle beams with a very small energy spread. In order to double the energy of particles obtained from electrostatic generators one can use the L.W. Alvarez scheme (Ref.1) whereby negative hydrogen ions are first accelerated by the field between the "earth" and the positive electrode and then, having given up two electrons in a stripping target, they are converted into positive ions which are accelerated again by the field but in the opposite direction. The energy thus obtained corresponds to twice the applied potential. The two ion injectors described in the present paper are designed for this type of application. The stripping target employed by these authors was in the form of a long thin tube with a gas circulated through its middle part. In order to obtain good vacuum and avoid charge exchange on Card 1/5

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S/120/61/000/002/002/042
E032/E114

An injector of negative hydrogen ions

residual gas atoms, the diameter in the tube had to be as small as possible. It was found that the most suitable electron-optical focussing scheme for the negative ions was a combination of a three-electrode lens with an immersion lens. The first of these lenses focusses the beam leaving the source in the immersion lens. The latter is used to match the beam energy to the accelerating voltage of the tube and to stabilize the position of the narrowest part of the beam in its object plane (C.H.Johnson, et al. Rev. Scient. Instrum., 1957, 28, 942. Ref.4). The three-electrode lens consists of electrodes of equal diameter ($D=30$ mm). The length of the middle electrode is 30 mm and the gap between the electrodes is 3 mm. The minimum focal length is about 50 mm. Fig.3 shows a schematic drawing of the H^- injector. In this figure, 1 is the source of the negative ions, 2 is the focussing system, 3 is the corrector which is used to adjust the position of the beam, 4 is an electron separator which consists of two permanent magnets and a moveable Faraday cup 5. The latter is used to measure the beam current. The electrodes 6 and

Card 2/5

S/120/61/000/002/002/042
E032/E114

An injector of negative hydrogen ions .

9 are used to suppress secondary electron emission, 7 is a valve, 8 is a trap which removes CO₂ which flows in from the stripping target, 11 is a slit and 12 is a second Faraday cup. Fig.5 shows another negative ion injector in which the H₁⁻ and H₁⁰ particles are separated by a magnetic field. In Fig.5, 1 is the negative ion source, 2 is a single lens, 3 is the magnetic analyser, 4 is a 4 mm diaphragm, 5 is an immersion lens, 6 is a corrector, 7 is a Faraday cup, 8 is a valve and 9 is a trap. With this arrangement only two lenses are necessary as compared with four in the apparatus described by L.E. Collins and A.C. Riviere (Ref.9: Nucl. Instrum. and Meth., 1959, 4, 121). Moreover, the length of the focussing system of the injector up to the object plane of the tube is smaller by a factor of 20 as compared with the length reported in Ref.9. The beam diameter obtained in the object plane is smaller than in Ref.9 and is practically independent of the input energy. The negative hydrogen-ion injector described in this paper was designed for an electrostatic generator with a working energy of 2 x 1.5 Mev.

Card 3/5

21392

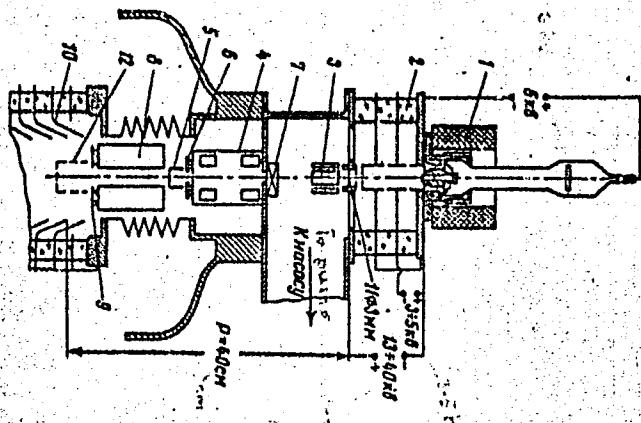
S/120/61/000/002/002/042

An injector of negative hydrogen... E032/E114

There are 5 figures, 3 tables and 9 references: 3 Soviet and
6 non-Soviet.

Acknowledgements are expressed to A.V. Almazov for his interest
in this work.

SUBMITTED: May 6 1960



Card 4/5

Fig.3

26. 2312

S/057/61/031/005/013/020
B104/B205

AUTHOR: Khirnyy, Yu. M.

TITLE: Formation of negative hydrogen ions by protons passing through gas targets

PERIODICAL: Zhurnal tekhnicheskoy fiziki, v. 31, no. 5, 1961, 597-605

TEXT: The results of an investigation of the content of H_1^- , H_1^+ , and H_1^0 in equilibrium beams, obtained by the passage of protons through H_2 , He, Ne, Ar, CO_2 , and C_3H_8 targets, are presented. Furthermore, data on the scattering of hydrogen beams by such targets are given. In the introduction, data from previous articles are discussed, reference being made to a paper of Ya. M. Fogel' and R. V. Mitin (ZhETF, 30, 450, 1956). The experimental arrangement is shown in Fig. 1. The neutral H_1^0 component in the beam was determined by measuring the current of secondary electrons produced by neutral atoms impinging on the beam catcher. The curve $H_1^-/H_1^+ = f(t_M)$ was drawn for the purpose of determining the thickness of

Card 1/9

22782

Formation of negative...

S/057/61/031/005/013/020
B104/B205

the target corresponding to a beam of equilibrium composition. The thickness at which this curve begins to form a plateau was assumed to suffice for generating an equilibrium beam. The H_1 and H_2^+ currents were measured simultaneously in order to exclude the effect of fluctuations in the beam intensity upon the results of measurement which are diagrammatically represented. Hydrogen target: The composition attained equilibrium at a target thickness of $8.6 \cdot 10^{15}$ atoms/cm² and maintained it up to $2.5 \cdot 10^{16}$. All types of particles in this target were equally scattered. The good agreement between the charge-exchange coefficients and the content of H_1 ions in the equilibrium beam indicates that protons are slightly scattered while passing through the target. The charge exchange between protons and gas is negligible when charge exchange takes place also between protons and target. The charge-exchange coefficient of H_2^+ ions is much lower than that of protons. Equilibrium composition of a hydrogen beam inciding upon a helium target was attained at a thickness of $1 \cdot 10^{16}$ atoms/cm². With a neon target, this state was reached at

Card 2/9

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B104/B205

Formation of negative...

$0.9 \cdot 10^{16}$ atoms/cm² and remained constant up to a thickness of $2.5 \cdot 10^{16}$ atoms/cm²; with an argon target, it was reached at $0.6 \cdot 10^{16}$ atoms/cm²; with a carbon dioxide target, at $3 \cdot 10^{15}$ atoms/cm²; and with a propane gas target, at $5.6 \cdot 10^{15}$ atoms/cm². A. V. Almazov is thanked for his interest in the work. There are 12 figures, 1 table, and 13 references: 8 Soviet-bloc and 5 non-Soviet-bloc. The two references to English-language publications read as follows: Stier et al., Phys. Rev., 96, 973, 1954; Collins et al. Nucl. Instr. and Methods, 4, 121, 1959.

SUBMITTED: June 18, 1960

Legend to Fig. 1: 1) Ion source; 2) focusing lenses; 3) accelerator tube; 4) electrostatic hyperbolic lens; 5) diaphragm; 6) channel of 6 mm diameter and 26 mm length; 7) target chamber; 8) beam catcher; 9) palladium capillary; 10) diaphragm; 11) analyzer; 12) beam catcher; 13) to diffusion pump.

Card 3/9

L 10305-63 ENT(1)/ENT(m)/BDS/ES(s)-2/ES(w)-2--
AFFTC/ASD/ESD-3/SSD-Pt-4/Pab-4--IJP(C)
ACCESSION NR: AP3002714

8/0120/63/000/003/0025/0029

76

69

21

AUTHOR: Khirnyy, Yu. M.; Kochemasova, L. M.

TITLE: Study of a model of a charge-transfer electrostatic generator

SOURCE: Pribory i tekhnika eksperimenta, no. 3, 1963, 25-29

TOPIC TAGS: charge transfer, electrostatic generator, ion accelerator

ABSTRACT: Construction and performance of a 3-Mev charge-transfer electrostatic generator are described. The design, shown in Fig. 1 of Enclosure, included accelerating porcelain-ring tubes, each 1.5-m long and separated by a section containing a stripping tube 340 mm long by 8 mm in diameter into which carbon dioxide was injected as the stripping agent. Hydrogen was used as the active source and was injected into the accelerating tube in neutral atomic and positive and negative ion form. The beam charging source was 500 w, alternating at 500 cps. The vacuum system consisted of two diffusion pumps with nitrogen traps. The method used for reducing the loading effect of secondary particles on the accelerating tubes was to add grids and to stop the beam down to a 3-mm diameter with a diaphragm, rather than to actually separate out the unwanted particles. Tests on the carbon dioxide injection rate showed that optimum positive ion

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ACCESSION NR: AP3002714

7

formation occurred at an injection rate of 35 cm^{sup} 3 per hr, or a chamber pressure of 6.5×10^{-5} mm Hg. However, it appeared that the accelerator operation is not very sensitive to this pressure and that one pump capable of 500 l/sec would be adequate. Results show that approximately 70% of the injected ion beam emerges in ion form and 10% as neutral hydrogen atoms. It is concluded that a construction of this type would also meet the demands of a 12-Mev accelerator, but that the question of separating out high-speed neutral atoms from the beam needs further investigation. "The authors thank A. V. Almazov, F. F. Myntsov, B. F. Ometov, V. A. Tabachkovskiy, I. G. Sugurov, and L. H. Budnikov for their help." Orig. art. has: 5 figures.

ASSOCIATION: none

SUBMITTED: 08May62 DATE ACQ: 12Jul63 ENCL: 01

SUB CODE: 00 NO REF SOV: 003 OTHER: 000

Card 2/3

KHIRNYY, Yu.M.; KOCHEMASOVA, L.N.

Analysis of a model for an ion recharge electrostatic generator.
Prib. i tekhn. eksp. 8 no.3:25-29 My-Je '63. (MIRA 16:9)
(Electric generators)

ACC NR: AP7001934

SOURCE CODE: UR/0120/66/000/006/0032/0036

AUTHOR: Khirnyy, Yu. M.; Kochemasova, L. N.

ORG: none

TITLE: Universal injector of negatively charged ions of hydrogen isotopes for a charge exchange generator

SOURCE: Pribory i tekhnika eksperimenta, no. 6, 1966, 32-36

TOPIC TAGS: ion source, ion beam, electrostatic generator, charge exchange

ABSTRACT:

A description of an injector is given in which the negatively charged ions are obtained through the charge exchange of the positively charged ions in a supersonic flow of mercury vapor. The injector, shown in Fig. 1, consists of: 1 - a high-frequency source of positively charged ions with a water-cooled gas discharge bulb; 2 - a single lens located 25 mm from the channel of the source probe; 3 - an immersion lens; 4 - a charge-exchange chamber; 5 - a nozzle which creates a supersonic jet from the mercury vapor used as the charge-exchange target; 6 - a trap with liquid nitrogen; 7 - a second single lens; 8 - a vacuum valve which separates the chamber of the deflecting magnet from the charge-exchange chamber; 9 - a deflecting magnet; 10 - plates for the electrostatic corrector; 11 - Faraday cylinder for controlling the beam;

Card 1/3

UDC: 537.534.2

ACC NR: AP7001934

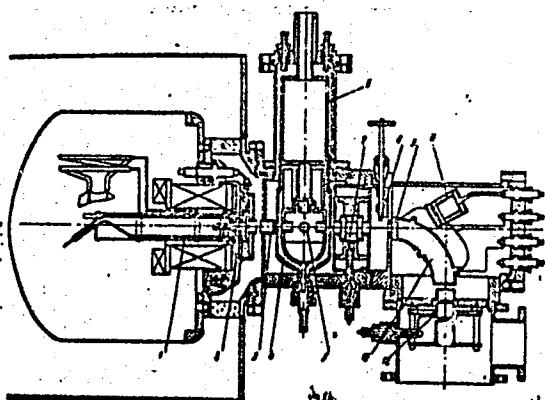


Fig. 1. Schematic of the Injector

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ACC NR: AP7001934

and t2 - a third single lens for focusing the beam at the exit from the injector. The injector was tested for 60 hr on a stand while operating on a D + He mixture. The D_1^- ion current was 14—17 μ A, and the consumption of mixture was 40—65 cm^3/hr . All of the units of the injector were in a normal state after the test and no mercury traces could be found in the chamber of the deflecting magnet or at the exit from the injector. The injector makes it possible to obtain bunches of H_1^- ions of 20—27 μ A, D_1^- ions of approximately 16 μ A, and T_1^- ions of 12—15 μ A. It is also simple to operate. Orig. art. has: 6 figures and 2 tables.

SUB CODE: 20/ SUBM DATE: 19Nov65/ ORIG REF: 004/ OTH REF: 002/

ATD PRESS: 5112

Card 3/3

ACC NR: AP7001935

SOURCE CODE: CIA-RDP86-00513R000722020007-1
 APPROVED FOR RELEASE: 09/17/2001 AUTHOR: Almazov, A. V.; Khrnyy, Yu. M.; Kochemasova, L. N.

ORG: none

TITLE: Compact duoplasmatron source of negatively charged ions for a charge exchange generator

SOURCE: Pribory i tekhnika eksperimenta, no. 6, 1966, 36-39

TOPIC TAGS: ion source, ion beam, electrostatic generator, charge exchange

ABSTRACT:

The description is given of a duoplasmatron (see Fig. 1) with a constant magnet. At a 0.95-mm diameter of the anode aperture and a voltage of approximately 80 kv, the source current reached about 140 μ amp. The injector and source passed the stand test before being installed in the generator. At an aperture of 0.5 mm the source current was 40 μ amp. After 120 hr of constant operation the parameters of the source had not changed. Generally speaking, the source does not require the attention of an operator. Orig. art. has: 6 figures.

Card 1/2

UDC: 537.534.2

ACC NR: AP7001935

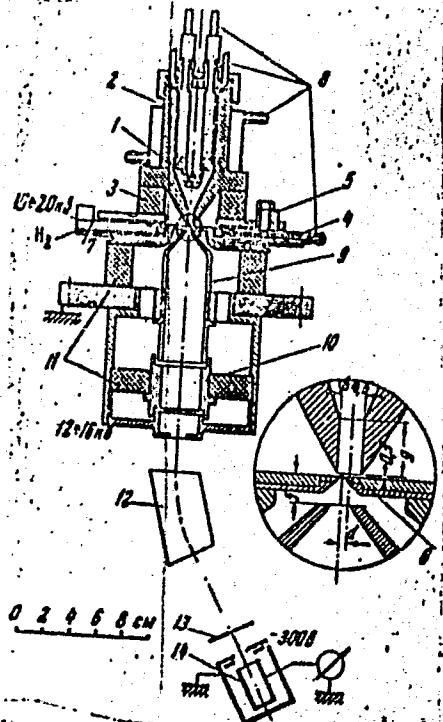


Fig. 1. Schematic diagram of duoplasmatron

1 - Cathode; 2 - intermediate electrode (steel); 3 - magnet;
4 - anode (steel); 5 - cam;
6 - copper insertion; 7 - gap, measured with clearance gage;
8 - water cooling pipes;
9 - extraction electrode;
10 - single lens; 11 - insulators;
12 - deflecting magnet;
13 - slit; 14 - Faraday cylinder.

SUB.CODE: 20/ SUBM DATE: 19Nov65/
ORIG REF: 002/ OTH REF: 007/
ATD PRESS: 5112

Card 2/2

KHIROV, A. A., Cand Agric Sci (diss) --- "The development of a method of investigating first-class pine cultures". Leningrad, 1959. 18 pp (Min Higher and Inter Spec Educ RSFSR, Leningrad Order of Lenin Forestry Engineering Acad im S. M. Kirov), 200 copies (KL, No 10, 1960, 135)

KHIROV nauchnyy sotrudnik; DAVIDENKO, L.K., nauchnyy sotrudnik

Pests of pine grafts and their control. Zaishch. rast. ot vred.
i bol. 7 no.9:50 S '62. (MIRA 16:8)

1. Borovaya lesnaya optytnaya stantsiya Vsesoyuznogo nauchno-
issledovatel'skogo instituta lesovedstva i mekhanizatsii
lesnogo khozyaystva.

(Buzuluk region—Pine—Diseases and pests)
(Buzuluk region—Insects, Injurious and beneficial—Control)

OGIYEVSKIY, Vasilii Vasil'yevich, doktor sel'khoz. nauk; KHIROV,
Aleksandr Aleksandrovich, kand. sel'khoz. nauk

[Inspection and study of forest plantations; methodological
manual for foresters] Obsledovanie i issledovanie lesnykh
kul'tur; metodicheskoe posobie dlia lesovedov. Moskva, Les-
naia promyshlennost', 1964. 47 p. (MIRA 17:9)

ACC NR: AP6022074

SOURCE CODE: UR/0141/66/009/003/0489/0496

AUTHOR: Giorgadze, N. P.; Khirseli, Ye. M.; Tsintsadze, N. L.

ORG: Institute of Physics, AN GruzSSR (Institut fiziki AN GruzSSR)

TITLE: Disintegrating interaction of waves in magnetoactive plasma

SOURCE: IVUZ. Radiofizika, v. 9, no. 3, 1966, 489-496

TOPIC TAGS: magnetoactive plasma, electromagnetic wave

ABSTRACT: A theoretical study is presented of the disintegrating interaction of waves in a magnetoactive plasma in the simplest case when all wave processes take place along the magnetic field (single-variate problem). The ordinary and extraordinary transverse waves and a longitudinal wave are considered; frequencies of linear transverse waves are assumed to lie far from cyclotron frequencies. The mean power of disintegrating generation, the disintegrating dispersion of transverse waves by longitudinal waves, and the disintegrating transformation of transverse waves into longitudinal waves are considered. Finally, the generation of Langmuir oscillations due to a disintegrating interaction of a helicon with a h-f wave is analyzed. Orig. art. has: 60 formulas.

SUB CODE: 20 / SUBM DATE: 22May65 / ORIG REF: 008

Card 1/1

UDC:621.371.182

KHIRVONEN, Kh. P.

AID P - 447

Subject : USSR/Electricity

Card 1/1 Pub. 27 - 10/34

Authors : Akodis, M. M., Dr. of Tech. Sci., Bril', M. V. Eng.,
Rudnyy, V. M., Eng., and Khirvonen, Kh. P., Eng.

Title : Study of Reliable Ionic Valve Action of Gas-Filled Tubes
in an Experimental Circuit

Periodical : Elektrichestvo, 7, 52-56, Jl 1954

Abstract : Experiments were made with cathode-ray oscilloscope in
order to determine the moment of back-fire, the value of
inverse voltage, and the moment of rupture. The depend-
ence of valve action of the I-50/5000 ignitron from the
rate of growth of inverse voltage is presented. 8 dia-
grams, 3 tables and 6 Russian references (1940-53).

Institution : Ural Polytechnical Institute im. Kirov

Submitted : Mr 6, 1954

(VNITOE).

Submitted : No date

KHIEVONI, Kh. F., Cand.Tech.Sci—(disc) "Study of ~~the~~ ^a synthetic ~~system~~ system
for the testing of ion~~e~~ valves." Sverdlovsk, 1958. 17 pp with drawings
(Min of Higher Education USSR. Ural Polytech Inst in S.M. Kirov. Chair of
High Voltage ~~Technology~~ ^{Engineering}), 150 copies. Bibliography at end of text (12 titles)
(KL, 31-58, 104)

-71-

SOV/112-59-4-7854

Translation from: Referativnyy zhurnal. Elektrotehnika, 1959, Nr 4,
pp 204-205 (USSR)

AUTHOR: Khirvonen, Kh. P.

TITLE: Accuracy of Reproducing Real Conditions of Operation of Ionic Rectifying
Devices When the Synthetic Testing Method is Used

PERIODICAL: Izv. vyssh. uchebn. zavedeniy. Elektromekhanika, 1958, Nr 2,
pp 114-121

ABSTRACT: An artificial circuit for testing ionic rectifying devices permits
obtaining current and voltage wave-shapes identical with real wave-shapes in
the forward and reverse half-cycles; the circuit ensures absence of the interval
between the moment of terminating the forward current and the moment of
applying the reverse voltage; the circuit secures similarity of switching-angle
conditions. The inductance of the oscillatory circuit is estimated, and the
current error of a low-voltage circuit is presented. It is assumed that if the

Card 1/2

SOV/112-59-4-7854

Accuracy of Reproducing Real Conditions of Operation of Ionic Rectifying

current error does not exceed 10-15% at the beginning of commutation, it is sufficient for a practical equivalence, all the more because the end of commutation is reproduced exactly. In testing under inverter conditions, a complete reproduction of the anode-cathode voltage curve, the required extinction angle, and the rate of voltage change near zero is possible. With some complications, the artificial testing scheme can be made universal; it would be suitable for complete testing of ionic rectifying devices with various parameters in both rectifier and inverter operations. Bibliography: 6 items. (Ural Polytechnic Institute, Sverdlovsk.)

I.L.R.

Card 2/2

KHIRVONEN, Kh.P.

Consumption of electric power in the synthetic method of testing
ionic rectifiers. Trudy Ural. politekh. inst. no.79:93-104 '59.
(Electric current rectifiers--Testing)

9.4120

S/194/61/000/008/045/092
D201/D304

AUTHOR: Khirvonen, Kh.P.

TITLE: The exactness of reproducing real operating conditions of gas-filled rectifiers in an artificial method of analysis

PERIODICAL: Referativnyy zhurnal. Avtomatika i radioelektronika, no. 8, 1961, 28-29, abstract 8 G225 (Tr. Ural'skogo politekhn. in-ta, 1960, v. 77, 71-86)

TEXT: The theoretical analysis is given of an installation for testing gas-filled rectifiers, as developed by the Ural Polytechnic Institute. Equations are derived which permit choosing the inductances and capacitances of the oscillating circuit taking into account current evaluation errors in the circuit operation. It is shown that the most exactly reproduced parameters are the shapes of voltages and currents at small angles of flow. See also RZhE, 1959, 7854. [Abstracter's note: Complete translation] ✓C

Card 1/1

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S/194/61/000/008/048/092
D201/D304

AUTHORS: Khirvonen, Kh.P. and Rudnyy, V.M.

TITLE: Experimental study of an artificial arrangement for testing gas-filled rectifiers

PERIODICAL: Referativnyy zhurnal. Avtomatika i radioelektronika, no. 8, 1961, 29, abstract 8 G232 (Tr. Ural'skogo politekhn. in-ta, 1960, v. 77, 87-98)

TEXT: The experimental installation developed by the Ural Polytechnic Institute for testing gas-filled rectifiers utilizes thyratrons and consists of 2 circuits - of a current circuit of 50 kVA and of an oscillating circuit fed by a 10 kVA transformer. The experiments were carried out with rectifiers working in a 3-phase bridge circuit, max. rectified current 120 A, max. reverse voltage 7-8 kV. The rectifier under test was a type ТГ-120/15000 (TG-120/15000) thyratron. The experimental tests of the installation provided some material on the rectifier reliability. The power consumed

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Experimental study of an artificial...

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was also measured. It was found to be 1.5% of the rectified current power with the rectifier working in a 3-phase bridge connection.

[Abstracter's note: Complete translation]

Card 2/2

diagram of the experimental arrangement is given, together with rectifier current and voltage shapes. The main difference between the experimental set-up and practical circuit is the absence of the

Card 1/2

APPROVED FOR RELEASE: 09/17/2001

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Artificial method...

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period with reversed voltage. In the experimental circuit the voltage across the rectifier is zero after extinction for the whole of the de-ionization time, after which the direct voltage sharply increases. Methods are considered of more accurate reproduction of the voltage waveform at the rectifier during the non-conducting part of the period; methods are also considered of reproducing the direct current shape by means of additional circuits. The experiments were carried out with a thyratron model. 3 references.

[Abstracter's note: Complete translation] *Vc*

Card 2/2

AKODIS, M.M., prof., doktor tekhn. nauk, red.; KHIVONEN, Kh.P.,
dots., kand. tekhn. nauk, red.; KONSTANTINOV, A.G., inzh.,
red.

[Transactions of the Interuniversity Scientific and Technical
Conference on Overvoltages] Trudy Mezhvuzovskogo nauchno-
tekhnicheskogo soveshchaniia po perenapriazheniam. Sverdlovsk,
Izd.UPI, 1963. 2 v. (MIRA 17:4)

1. Mezhvuzovskoye nauchno-teknicheskoye soveshchaniye po pere-
napryazheniyam, Sverdlovsk, 1961.

KRICHENOVА, I.A., kанд. tekhn. nauk, dotsent; KHIRVONEN, Kh.P., kанд. tekhn. nauk, dotsent.

Review of A.I. Rutschii's book "Electric power plants and substations." Izv. vya. ucheb. zav.; energ. 6 no.11:107-110 N'63.

(MIRA 17:2)

1. Ural'skiy politekhnicheskiy institut imeni S.M. Kirova.

KHIRTINA, G.V.

Postoperative hemorrhages following extrapleural pneumolysis
[with summary in French]. Probl. tub. 36 no.4:63-68 '58
(MIRA 11:7)

1. Iz Barnaul'skogo gorodskogo protivotuberkuleznogo dispensera
(glavnnyy vrach T.I. Turosova); nauchnyye rukovoditeli raboty -
prof. I.V. Toroptsev, i prof. S.P. Khodkevich (iz Tomskogo meditsinskogo
instituta.)

(COLLAPSE THERAPY, compl.
postop.hemorrh. after extrapleural pneumolysis (Rus))

1. KHIRYA, YE. A.
2. USSR (600)
4. Swine--Feeding and Feeding Stuffs
7. How we fatten swine, Sots. zhiv., 15, No. 1, 1953.

9. Monthly List of Russian Accessions, Library of Congress, April, 1953, Uncl.

XHISAMOV, A.V.

Violent atmospheric phenomena in 1954. Izv. Uzb. fil. Geog. ob-vn
2:155-162 '56.

(MIRA 11:4)

(Meteorology)

KHISAMOV, A.V.

Marked weather anomalies in 1955. Izv.Uzb.fil.geog.ob-va
no.3:143-150 '57. (MIRA 11:4)
(Meteorology)

KHISAMOV, A. V.

Improvement of the microclimate of southern cities. Uch.zap.
Tashk. gos. ped. inst. no.18:51-57 '59. (MIRA 13:9)
(Russia, Southern--Microclimatology)

AKRAMOV, Z.M., kand. geogr. nauk; RAKITNIKOV, A.N., kand. geograf. nauk; ZAMKOV, O.K., kand. geograf. nauk; SHERMUKHAMEDOV, A.M. [deceased]; SAUSHKIN, Yu.G., doktor geograf. nauk, prof, otv. red.; DEGTYAR', V.I., red.; KHISAMOV, A.V., kand. geograf. nauk, red.; ASTAKHOV, A., red.; GOR'KOVAYA, Z.P., tekhn. red.

[Agricultural geography of Samarkand and Bukhara Provinces]
Geografiia sel'skogo khozaiystva Samarkandskoi i Bukharskoi oblasti. [By] Z.M. Akramov i dr. Tashkent, Izd-vo Akad. nauk UzSSR. Pt.2. 1961. 323 p. (Materialy Zeravshanskoi ekspeditsii SOAPS AN UzSSR, no.1) (MIRA 16:4)

1. Akademiya nauk Uzbekskoy SSR. Tashkent. Otdel geografii.
2. Nachal'nik Otdela sel'skogo khozyaystva Gosplana Uzbekskoy SSR (for Degtyar').

(Bukhara Province--Agricultural geography)
(Samarkand Province--Agricultural geography)

RYABCHIKOV, F.D., inzh.; KUSTOBAYEV, G.G., inzh.; SOKOLOV, V.A., inzh.;
KHISAMOV, F.N., inzh.

Accelerating the cooling of sheet steel in bell furnaces.
Stal' 22 no.8:748-749 Ag '62. (MIRA 15:7)

1. Magnitogorskiy metallurgicheskiy kombinat.
(Furnaces, Heat-treating)

ASTAF'YEVA, M.S.; KLIMUSHIN, I.M.; KHISAMOV, R.B.

Using the specific resistance of rocks in the testing of methods
for determining the permeability of terrigenous layers. Geol.
nefti 1 gaza 5 no. 5:42-44 My '61. (MIRA 14:4)

1. Tatarskiy nauchno-issledovatel'skiy neftyanoy institut.
(Rocks—Permeability)

S/169/63/000/002/120/127
D263/D307

AUTHORS: Sultanov, S. A., Astaf'yeva, M. S., Kilimushin, I. M.
and Khisamov, R. B.

TITLE: Use of industrial geophysical methods of determining
rock properties of terrigenous ores at Romashkinskoye
deposit

PERIODICAL: Referativnyy zhurnal, Geofizika, no. 2, 1963, 35-36,
abstract 2D211 (Tr. Tatarsk. neft. n.-i. in-t, 1961,
no. 3, 49-59)

TEXT: At Romashkinskoye deposit different methods of determining
the porosity of ores (K_p) from natural potential (NP) diagrams,
the permeability (K_{pr}) from the data of the resistance method, and
N. V. Vilkov's methods were checked. A comparative analysis of me-
thods of determining K_p from NP was made for strata having K_p lar-
ger than 16% in boreholes of the central part of the deposit, cha-
racterized by three and more cores. The research group method of

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S/169/63/000/002/120/127
D263/D307

Use of industrial ...

'Tatneftegeofizika' trust, and methods of A. I. Krinari and L. P. Dolina were checked. The minimum mean relative error in determining K_{pr} was obtained using the research group method, and the maximum error using Dolina's method. Maximum relative error in all methods is observed for strata less than 3 m thick. Methods of determining K_{pr} from the resistivity ρ_p of L. P. Dolina, S. A. Sul'tanov, V. M. Dobrynin and 'Tatneftegeofizika' trust were checked. G. S. Morozov's method was not checked as it gives high errors. Best results were obtained by L. P. Dolina's method, worst by the trust's method. All methods give small errors for strata with $\rho_p < 100 \text{ ohm.m}$, all methods give a low value of K_{pr} . The error in determining K_{pr} by all methods increases in strata less than 2 m thick. N. V. Vilkov's method of determining K_{pr} from NP is unsatisfactory, as it takes no account of the lack of connection between the NP amplitude, A_{NP} , and the permeability for $K_{pr} < 100 \text{ millidarcies}$, and of the very weak connection between A_{NP} and K_{pr} for

Card 2/3

S/169/63/000/002/120/127
D263/D307

Use of industrial ...

K_{pr} 100 millidarcies; no corrections are made in the values of A_{NP} for the effect of thickness and resistivity of the stratum. The general character of the connection between A_{NP} and K_{pr} indicated by Vilkov differs from the actual one. (Abstracter's note: Complete translation.)

Card 3/3

KAMAY, GILLIM; KHISAMOVA, Z.L.

Trichlormethylparatolylphosphinic acid and its derivatives.
Izv.Kazan.fil.AN SSSR Ser.khim.nauk no.1:59-61 '50.

(MIRA 10:5)

(Phosphinic acids)

CA

Action of carbon tetrachloride on esters of α -toluenephosphorous acid. Z. I. Khisanova and Gilmur Kamal (A. R. Arbuzov Chem. Inst., Kazan), Zash. Obshch. Akad. Nauk. (J. Gen. Chem.) 20, 1149-70 (1950); cf. C.A. 41, 16135g; 42, 7723d. — Esters of the type $R(C_6H_5)P(O)R'$ ($R = \text{p}-\text{MeC}_6H_4$) were converted to $R(C_6H_5)P(O)OH$ by the action of CCl_4 . RPC_6H_5 , b.p. 120°, d₄²⁰ 1.2001, m.p. 1.0600, d₄²⁰ 1.0421, added slowly to 13.5 g. MeOH and 60.2 g. Me_2NPh in 100 ml. Et_2O , refluxed 30 min., let stand overnight, and filtered, gave 52% $R(Et)_2P(O)OEt$, b.p. 107-9°, d₄²⁰ 1.0700, d₄²⁰ 1.0421, m.p. 1.5823, which reacts with CuX_2 , addn. of 5 g. MeI to 9 g. ester, gave 90% $RMeP(O)OEt$, b.p. 151-2°, d₄²⁰ 1.1315, d₄²⁰ 1.1201, m.p. 1.5280; this boiled with conc'd. HCl 3 hrs. and evapd. gave $RMeP(O)OH$, m.p. 119-20° (cf. Michaelis, Ber. 31, 1046 (1898)). Addn. of 12 g. phosphonite to 9.5 g. CCl_4 caused a vigorous reaction yielding 100% MeCl and $R(C_6H_5)P(O)OMe$, b.p. 128-30°, d₄²⁰ 1.2328, d₄²⁰ 1.2140, m.p. 1.5312. Similarly, ROH gave 50% $R(P(O)OR)$, b.p. 123-5°, d₄²⁰ 1.0680, d₄²⁰ 1.0210, m.p. 1.5138, which boiled 2 hrs. with EtOH yielded $REt_2P(O)OEt$, b.p. 101°, d₄²⁰ 1.0630, d₄²⁰ 1.0097, m.p. 1.5105, while addn. of 8 g. CCl_4 to 11 g. phosphonite, followed by refluxing 2 hrs., gave $R-(C_6H_5)P(O)(OEt)_2$, b.p. 152°, d₄²⁰ 1.3103, m.p. 1.6424.

This (4 g.) and 12 ml. conc'd. HCl refluxed 2 hrs., then repeatedly evapd. with Et_2O , yielded $R(C_6H_5)P(O)OH$, m.p. 161.5-60° (from Et_2O), sol. in EtOH and Et_2O . A reaction sequence as above with PrOH gave 81.0% $R(P(O)Pr)$, b.p. 129-30°, d₄²⁰ 1.0091, d₄²⁰ 0.9637, m.p. 1.5650, which boiled 3 hrs. with MeI gave 92% $RMeP(O)OPr$, b.p. 107°, d₄²⁰ 1.0899, d₄²⁰ 1.0050, m.p. 1.6185 (its hydrolysis gave $RMeP(O)OH$, as described above), while 10 g. CCl_4 with 15 g. phosphonite gave 12% $R(C_6H_5)P(O)OPr$, b.p. 160-70°, d₄²⁰ 1.3001, d₄²⁰ 1.2014, m.p. 1.5350 (hydrolysis with conc'd. HCl gave the free acid, m.p. 181°, as above). Bu_2N^+ , as above, yielded 80.3% $R(P(O)Bu_2)$, b.p. 170-172°, d₄²⁰ 0.9899, d₄²⁰ 0.9770, m.p. 1.5024, which heated with MeI gave 90% $RMeP(O)OBu_2$, b.p. 178°, d₄²⁰ 1.0722, d₄²⁰ 1.0680, m.p. 1.5002, while CCl_4 gave $R(C_6H_5)P(O)OBu_2$, b.p. 180-182°, d₄²⁰ 1.2230, d₄²⁰ 1.2143, m.p. 1.5207. Similarly, iso-BuCl yielded 77% $RPO(Bu-isobutyl)$, b.p. 155-6°, d₄²⁰ 0.9807, d₄²⁰ 0.9637, m.p. 1.5057; this (10 g.) and 9.2 g. CCl_4 reacted with heat evolution, yielding almost 100% iso-BuCl and 58% $R(C_6H_5)P(O)OBu-isobutyl$, b.p. 178-80°, d₄²⁰ 1.2601, d₄²⁰ 1.2230, m.p. 1.5291. Continued hydrolysis of $R(C_6H_5)P(O)H$ eventually leads to the formation of $RPO(OH)$. It may be noted that the consts. of the initial ArPCl_3 differ from the properties indicated by Michaelis and Pech, Ann. 212, 213 (1892), although the M. and P. procedure was used for the prepn.

G. M. Kosolapoff

KHISAMOVA, Z. I.

The action of carbon tetrachloride on the esters of p-tolylphosphorous acid.
Z.L. Khisamova and G.Iw. Kamai (Kazan Branch Acad. Sci. U.S.S.R.). J. Gen. Chem.
U.S.S.R. 20, 1207-14 (1950) (Engl. translation).

See C.A. 45, 1531d.

R.M.S.

immediate source clipping

KHISANOVA, Z. L.

USSR/Chemistry - Organic Arsenic Compounds

1 Feb 51

178T17
Synthesis and Properties of Some Cyclic Esters of Ethyleneglycolsarsenous Acid," Gil'm Ramay,
Z. L. Khisanova, Chem Inst tmeni A. Ye. Arbusov,
DAN, Affiliate, Acad Sci USSR

Dok Akad Nauk SSSR" Vol LXXXI, No 4, pp 535-538

By reacting ethylene glycol with arsenic trichloride in presence of pyridine, synthesized chloride of ethyleneglycolsarsenous acid (bp 71-72° at 11 mm Hg, mp 44-45°). Prep'd number of triesters of arsenous acid from this cyclic diester monochloride. The

USSR/Chemistry - Organic Arsenic Compounds (Contd)

1 Feb 51

bp of alkyl esters of ethyleneglycolsarsenous acid lie very close to those of the analogous p compd. In the case of Me and Et esters of either ethyleneglycolsarsenous acid or ethyleneglycolphosphorous acid, they are in the neighborhood of 50-60° at 11-23 mm. Prep'd triester As₂(-OCH₂CH₂O-)₃. By oxidizing latter with bromine, obtained pentester of arsenic acid As₂(-OCH₂CH₂O-)₅.

178T17

esters of ethyleneglycol arachidic acid. Oil (m. Kamm and Z. L. Khisanova, A. E. Arbusov Chem. Inst., Kazan, Zhur. Osnichel. Khim., 23, 1323-30 (1953).—Add. of 103 g. NaCl with cooling to 37.6 g. $(\text{CH}_3\text{OH})_2$, to 87.2 g. pyridine and 400 ml. dry Et_2O , followed by standing overnight at room temp., and 1 hr. at reflux gave on distil. of the titrant

some 81 g. $O_2CH_2CH_3O_2AsCl$ (I), b.p. 71-8°, m. 44-5°, along with higher-boiling products, which included 10 g. $(CH_3O)_2AsOCH_2CH_3O_2As(OCH_3)_2$ (II), b.p. 150-7°, d₄²⁰ 1.801, n_D²⁰ 1.6433. I is instantly hydrolyzed by H₂O yielding As₂O₃. Addn. of I to $(CH_3O)_2As$, in pyridine and Et₂O, as above, gave an unstated yield of II. Heating 80 g. As₂O₃ with 36.4 g. $(CH_3O)_2As$ 30 min. at 140-50°, followed by continued heating under water-pump vacuum for removal to H₂O, gave 92.4% II, b.p. 160-1°. Addn. of 6 g. AsCl₃ to 16 g. II, followed by 0.6 hr. heating to unstated temp., gave largely I, b.p. 60-7°. Refluxing 2.5 g. II with 50 ml. H₂O gave As₂O₃. To 51 g. II in CCl₄ was added with cooling 23 g. Br₂, the resulting ppt. of $(CH_3O)_2As(OH)(OCH_3)_2$, m. 127° (Englund, C.A. 23, 505), was s^{er}d. and the filtrate dried, yielding 32.2 g. greenish lacquerary liquid, identified as $(CH_3O)AsBr_3$, b.p. 83°, d₄²⁰ 2.0770, d₂₅²⁰ 2.5038, n_D²⁰ 1.8022. Thus, the previously proposed structure of the substance, m. 127° (C.A. 43, 101664) is in error; probably the initially formed $[(CH_3O)_2AsOCH_2CH_3O_2As(OCH_3)_2$ is hydrolyzed in handling to the acidic product given above. Addn. of 20 g. I to 3.8 g. MeOH in 9.2 g. pyridine and 250 ml. Et₂O with cooling, followed by 1 hr. reflux, gave 8.6 g.

O.CHI., *CH₂OAsOMe*, *b.p.* 60–1°, *d*₄²⁰ 1.6163, *d*₄²⁵ 1.6310,
*n*_D²⁰ 1.4935. Similarly were obtained the following esters.
(R shown followed by *d*₄, *d*₄²⁰ and *n*_D²⁰): *E*, *b.p.* 61–3°,
1.5031, 1.4734, 1.4788; *P*, *b.p.* 74–5°, 1.4188, 1.3923,
1.4702; *Bu*, *b.p.* 93°, 1.4024, 1.3764, 1.4785; *n-C₄H₉*,
120–1°, 1.3035, 1.2813, 1.4742; *n-C₅H₁₁*, *b.p.* 130–1°,
1.2071, 1.2405, 1.4730; *n-C₆H₁₃*, *b.p.* 110°, 1.2440, 1.2250,
1.4737; *n-C₇H₁₅*, *b.p.* 117–18°, 1.2210, 1.2041, 1.4748; (cyclic
olefins); *n-C₈H₁₇*, *b.p.* 118°, 1.2720, 1.3044, 1.5070 (from cyclo-
hexanol); *Pn*, *b.p.* 143–4°, 1.5087, 1.5440, 1.5716. All
these are readily hydrolyzed by *H₂O* yielding *As₂O₃*. Heat-
(over)

Kamai, G. 1(2)

ing $\text{O}(\text{CH}_2\text{CH}_2)_2\text{OAsOMe}$ with MeI 0 hrs. at reflux resulted in no reaction. Similarly II failed to react with MeI , S,

or Cu_2Br_2 . Akin. of 4.4 g. Br to 6 g. $\text{O}(\text{CH}_2\text{CH}_2)_2\text{OAsOMe}$ in CCl_4 gave an unstated yield of $\text{C}_6\text{H}_5\text{O}_2\text{A}_2\text{Br}_2$, b.p. 72-3°, n_{D}^{20} 2.2152, d $_{4}^{20}$ 1.1707, n_{D}^{25} 1.6830. When 10.3 g. I in Et_2O was slowly added to cooled 16 g. $(\text{Bu}_4\text{O})_2\text{P}$ in Et_2O vigorous reaction took place with deposition of red P. The filtrate gave some BuOH and a fraction, b.p. 140°, d_4^{20} 1.1185, n_{D}^{25} 1.4333, which contained P and As but decompd. rapidly to a black solid; no analyses were made, presumably the substance was crude $(\text{C}_6\text{H}_5\text{O}_2)_2\text{P}(\text{O})(\text{OEt})_2$. To a soln. of $(\text{EtO})_2\text{PONa}$ (from 24.6 g. $(\text{RO})_2\text{POH}$) in Et_2O was added 30 g. I with cooling; the mixt. acquired a red color and deposited NaCl (9.0 g.); on distn. the filtrate decompd. and no individual substances were isolated. G. M. Kosolapoff

(3)

2/2

KHISAMOVA, Z. L.

(3) 5

Preparation of alkyl and glycolic esters of arsenious and arylarsonous acids. Gil'm'Kani, Z. L., Khisamova, and N. A. Chudovs'ka. Doklady Akad. Nauk S.S.R. 89, 1015-10 (1953); cf. C.A. 47, 10470c. —Cyclic esters of

general type ArAs.O.R'.O ($\text{R}' = \text{alkylene radical}$) were obtained by 2 procedures: reaction of ArAsCl_3 with the glycol in the presence of 2 moles pyridine or by heating ArAsO with the glycol without any water-abstrating agents (method not described further). The products obtained are listed below; the 2nd method generally gives somewhat better yields (R, R' properties of the products and yields are listed in the same order as the procedures cited above; all d , n values are for d_{20}^2 and n_D^2). Ph, $-(\text{CH}_2)-$: 56.6%, b_1 121-2°, d. 1.5305, n 1.6100; 64.5%, b_1 123-2.5°, d. 1.5379, n 1.6061. $p\text{-MeC}_6\text{H}_4$, $-(\text{CH}_2)-$: 45.6%, b_1 138-9°, m. 134-5°; 60.4%, b_1 137-1°, m. 134-5°. Ph, $-\text{CHMeCH}_2-$: 65.3%, b_1 150-8°, d. 1.3584, n 1.5540; 74.3%, b_1 158-9°, d. 1.3577, n 1.5512. Ph, $-(\text{CH}_2)-$: 62.1%, b_1 135-0°, d. 1.4618, n 1.5920; 69%, b_1 133°, d. 1.4042, n 1.4642. Ph, $-(\text{CH}_2)\text{O}(\text{CH}_2)-$: 63.3%, b_1 171-2°, d. 1.4591, n 1.5841; 59.8%, b_1 155°, d. 1.4504, n 1.5805. Ph, $o\text{-C}_6\text{H}_4$: 20.1%, b_1 155°, m. 85-1°; 86%, b_1 178-0°, m. 85-8°. Heating 60 g. As_2O_3 and 80.4 g. $\text{HOCH}_2\text{CH}_2\text{OH}$ in a distg. app. at 150° 15-20 min., removal of the H_2O *in vacuo*, and distn. of the mixt. gave

92.4% [$\text{O}(\text{CH}_2\text{CH}_2\text{OAsOCH}_2\text{CH}_2\text{O})$]₂, b_1 100-1°. G. M. K.

Evaluation of paper in B-76836, 19 Jul 54

KHISOMAVA, Z. L.

USSR/Chemistry

Card 1/1

Authors : Kamay, Gil'm; and Khisomava, Z. L.

Title : Acid chlorides and mixed esters of tetramethylethyleneglycolarsenous and pyrocatechinarsenous acids

Periodical : Zhur. Ob. Khim. 24, Ed. 5, 816 - 820, May 1954

Abstract : First obtained and investigated were acid chlorides of tetramethyl-ethyleneglycolarsenous and pyrocatechinarsenous acids. The acid chlorides obtained are crystalline substances soluble in water and some organic solvents. The reaction of these acid chlorides with various alcohols in a benzene medium in the presence of anhydrous pyridine results in the formation of mixed esters of tetramethylene-glycolarsenous and pyrocatechniarsenous acids the properties of which are described. Two USSR references. Table.

Institution : Acad. of Scs. USSR, Kazan Branch, The A. E. Arbuzov Chemical Institute

Submitted : December 26, 1953

KHISAMOVA, Z. L.
USSR/Chemistry

Card 1/1

Authors : Kamay, Gil'm; and Khisamova, Z. L.

Title : Reaction of cyclic arsenyl chlorides with dialkylamines

Periodical : Zhur. Ob. Khim. 24, Ed. 5, 821 - 824, May 1954

Abstract : The reaction of cyclic arsenyl chlorides with dialkylamines was investigated. The first products were dialkyl substituted amides of cyclic esters of arsenous acid (cyclic arsenates). The bond $>\text{As-N}<$ in these compounds is very unstable and breaks under the effect of alkyl halide, atmospheric oxygen and water. The products of oxidation of dialkylamido esters of arsenous acid are cyclic esters of pyroarsenous acid. One USSR reference. Tables.

Institution : Acad. Scs. USSR, Kazan Branch, The A. E. Arbuzov Chemical Institute

Submitted : December 26, 1953

KAMAY, Gil'm.; KHISAMOVA, Z.L.

Synthesis of new arsenic-nitrogen organic compounds with the As-N bond. Dokl.AN SSSR 105 no.3:489-491 N '55. (MLRA 9:3)

1. Khimicheskiy institut imeni A.Ye. Arbuzova Kavanskogo filiala
Akademii nauk SSSR.
(Arsenic organic compounds)