

HAZAY, Istvan, dr., Kossuth-díjas, a muszaki tudományok doktora, egyetemi
tanár

Dimension of the auxiliary quantities of compensation. Geod kart 14
no.6:397-399 '62.

1. Építőipari és Kozlekedési Műszaki Egyetem; "Geodezia és Kartográfia"
szerkesztő bizottsági tagja.

HAZAY, Istvan, dr., Kossuth-dijas, a muszaki tudományok doktora, egyetemi tanár

Professional practice combined with correspondence courses for students of geodesy. Good kart 14 no.6:454-455 '62.

1. Építőipari és Közlekedési Műszaki Egyetem; "Geodesia es Kartografia" szerkeszto bizottsagi tagja.

HAZAY, Istvan, a muszaki tudományok doktora

Remark about the paper of Zoltan Heinemann entitled "Remarks about the principle of static equalization." Muzsaki kozl MTA 31 no.1/4:275-279 '62.

1. Epitoipari es Kozlekedesi Muszaki Egyetem II. Geodeziai Tanszeke.

HAZAY, Istvan, dr., egyetemi tanár, okleveles mérnök, a muszaki
tudományok doktora

General review of the reform curriculum proposal of the
Faculty of Engineering, Technical University of the Construction
Industry and Transportation. Melyepitestud szemle 13 no.2/3:
49-64 F-Mr '63.

1. Építőipari és Közlekedési Műszaki Egyetem Mérnöki Kara
dekanja, Budapest.

HAZAY, Istvan, dr.

Direct determination of function values from the normal equations of coordinate compensation. Geod kart 15 no.1: 14-15 '63.

1. "Geodezia es Kartografia" szerkeszto bizottsagi tagja.

HAZAY, Istvan, dr.

Calculation of mean errors in connection with fictive
measurements. Geol kart 15 no.2:89-99 '63.

1. "Geodezia es Kartografia" szerkeszto bizottsagi tagja.

HAZAY, Istvan, dr.

Loxodromes and orthodromes. Geod kart 16 no.2:110-119 '64

1. "Geodezia es Kartografia" szerkeszto bizottsagi tagja.

HAZAY, I., prof. Doctor of Technical Sciences

Dimension of the subsidiary quantities of adjustment. Acta techn
Hung 47 no. 1/2:123-129 '64.

1. Technical University of Building and Communication, Budapest.

HAZAY, Istvan, dr.

The image of the degree network in the Mercator projection with
oblique axis. Geod kart 17 no.1:1-4 '65.

I. Editorial Board Member, "Geodezia es Kartografia."

L 30185-66

ACC NR: AT6020305

SOURCE CODE: HU/2505/65/052/01-/0171/0201

AUTHOR: Hazay, I.--Khazai, I. (Doctor of technical sciences; Professor)

24
B+1

ORG: Department of Advanced Geodesy, Technical University for Construction and Transportation, Budapest

TITLE: Significance of the Tissot indicatrix

SOURCE: Academiae scientiarum hungaricae. Acta technica, v. 52, no. 1-2, 1965, 171-201

TOPIC TAGS: cartography, projective geometry

ABSTRACT: Studies were conducted to establish the suitability of the Tissot indicatrix as a property of the projection of a point with the aim of determining the usefulness of it in selecting the optimum type of projection for cartographic purposes. The indicatrix formation for various projection types, distortions caused in the process, and means of indicatrix determination were described and discussed. Orig. art. has: 11 figures, 83 formulas, and 1 table. [Orig. art. in German.] [JPRS]

SUB CODE: 08, 12 / SUBM DATE: 12Jun64 / ORIG REF: 001 / OTH REF: 003

Card 1/1 cc

HAZAY, S.

"Adjustment of national and continental triangulations." Acta Technica, Budapest, Vol. 6,
No. 3/4, 1953, p. 399.

SO: Eastern European Accessions List, Vol. 3, No. 11, Nov. 1954, I.C.

HAZDROVA, Milena, promovana geolozka

Calculation of the thermal water reserves in the basal Cretaceous sandstones in the Usti nad Labem area. Geol pruzkum 5 no.6: 178-180 Je '63.

1. Ustredni ustav geologicky, Praha.

KYSELA, J.; HAZE, K.

Some remarks on determination of the lowering of the freezing point according to Czechoslovakian pharmacopeid 1, supplement 1. Cesk. farm. 4 no.5:247 June 55.

1. Z krajske kontrolni laboratore Prazske Mediky n.p.

(PHARMACOPEIA

Czech. 1, supplement 1, determ. of lowering of freezing point.)

(DRUGS

freezing point, determ. of lowering in Czech. pharmacopeia)

GRUND, Mir.; HAZE, M., inz.; ZUZANEK, Jar., inz.

New design of a switch for a small point recorder. Automatizace
6 no.1:17-18 Ja '63.

1. Zavody prumyslove automatizace, n.p., zavod Nova Paka.

HAZEK, B

CZECHOSLOVAKIA/Inorganic Chemistry. Complex Compounds

C

Abs Jour : Ref Zhur - Khimiya, No 3, 1958, No 7363

Author : F. Petru., B. HazeK, J. Zavorka

Inst : Not Given

Title : On the Chemistry of the Rare Elements. II. On Scandium Pyrophosphate.

Orig Pub : Chem. listy, 1957, 51, No 1, 21-26, Sb. chekhosl. khim. rabor, 1957, 22, No 5, 1541-1546

Abstract : The deposit of scandium pyrophosphate, as a result of interaction of solutions of $ScCl_3$ and $Na_4P_2O_7$ or $Na_2H_2P_2O_7$ at pH 3.6 and 0.5 was studied. On the basis of potentiometric and conductometric studies of the course of the formation of the deposit and of the results of the analysis of obtained substances, the authors conclude that a formation of $Sc_4(P_2O_7)_3$ takes place in all cases. Part I see RZhKhim., 1957, 50969.

Card : 1/1

WRONSKI, Mieczyslaw; HAZEK, Lucyna

Kinetics of the hydrolysis of Phenyl isothiocyanate in solutions of sodium hydroxide. Nauki matem przyrod Lodz no.12:155-162 '62.

1. Katedra Technologii Chemicznej, Uniwersytet, Lodz.

HACHI, Jovan

Vale Vouk; obituary. Letopis SAZU 13: 51-53 '62
(publ. '63)

STRAUB, Gyula; HAZI, Endre

Structural analysis of substances by means of radioactive gamma rays with the aid of the Geiger-Muller counters.
Veszprem vegyip egy kozl 3 no.1/4:263-266 '59.

1. Veszpremi Vegyipari Egyetem Analitikai Kemia Tanszek.

HAZI, Endre

Application of contact potential measurement in studying the initial phase of oxidation on metal surfaces. Veszprem vegyip egy kozl 5 no.2:167-175 '61.

1. Veszpremi Vegyipari Egyetem Analitikai Kemia Tanszek.

BORLAI, Oszkar; HAZI, Endre; BADACSONYI, Tivadar

Dwelling period measurement of grains in fluidized layers
by means of isotopes. Veszprem vegyip egy kozl 7 no.1:
55-60 '63.

1. Magyar Tudomanyos Akademia Muszaki Kemiai Kutato
Intezete, Budapest-Veszprem; Veszpremi Vegyipari Egyetem
Radiokemiai Tanszek.

CSAPO, Zoltan; HAZI, Endre; KALDI, Pal

Examination of mixing and dwelling time distributions in
foam columns by means of radiometric method. Veszprem
vegyp egy kozl 7 no. 2:137-144 '63.

1. Chair of Radiochemistry, Chemical Industry University,
Veszprem, and Research Institute of Technical Chemistry,
Hungarian Academy of Sciences, Budapest-Veszprem.

HAZI, Endre; CSAPO, Foltan; JELINKO, Borbala

Continuous densimetry of streaming solutions and slurries.
Musz elet 19 no. 4: 10 13 F '64.

ACC NR: APT003591 (71) SOURCE CODE: HU/0033/66/003/012/0359/0015

AUTHOR: Hazkoto, Gizella (Chemical engineer); Szalontai, Imre (Chemical engineer); Szondy, Istvan (Building engineer)

ORG: [Hazkoto] Industrial Research Institute for Plastics (Muanyagipari Kutato Intezet); [Szalontai] BM [Ministry for Interior] National Headquarters for Fire Protection (BM Tuzrendeszet Orszagos Parancsnoksaga); [Szondy] Scientific Institute for Planning and Design (Epitestudomanyi Intezet)

TITLE: Behavior of plastics in fire

SOURCE: Muanyag es gumi, v. 3, no. 12, 1966, 359-365

TOPIC TAGS: polyethylene plastic, polypropylene plastic, polyester plastic, polyvinyl chloride plastic, fire resistant material, flammability, foam plastic, glass fiber reinforced plastic, plastic tubing

ABSTRACT: The authors describe two series of tests performed in 1965.

1. Several horizontal plastic and resin tubes (polyvinyl chloride, polyethylene, polypropylene) for carrying liquids, were tested for their behavior in fire, and the fire-resistance of glass-fiber-reinforced polyester and PVC polyester suction

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ACC NR: AP7003591

ducts was studied. PVC pipes were found to be least resistant to fire, while the glass-fiber-reinforced polyester showed good fire-resistant properties. 2. The flammability of sandwich structures made of several types of foam was investigated, but the results were considered inconclusive. The United Pharmaceutical and Food Plant participated in the tests. Orig. art. has: 15 figures and 2 tables.

[KS]

SUB CODE: 11/SUBM DATE: none/ORIG REF: 001/OTH REF: 002/

Card 2/2

HAZLER, T.

"Local Materials as a Basis for Rural Building." p. 11, (BUDOWNICTWO WIEJSKIE, Vol. 5, no. 1, Feb./Jan. 1953, Warszawa, Poland)

SO: Monthly Lists of East European Accession, LC, Vol. 3, no. 5, May 1954/Uncl.

HAZLER, T.

Local materials in the building investments of collective farms.

P. 9 (Budownictwo Wiejskie, Vol. 8, no. 3, Mar. 1956, Warsaw, Poland)

Monthly Index of East European Accessions (EEAI) LC. Vol. 7, no. 2,
February 1958

HAZLER, T

POLAND /Chemical Technology - Chemical Products and Their
Application. Ceramics. Glass. Binders. Concrete.

H-13

Abs Jour : Ref Zhur - Khimiya, No 8, 1958, 25974

Author : Hazler Tadeusz

Inst : -

Title : Slags and Their Use in Building.

Orig Pub : Budown. wiejskie, 1957, 8, No 8, 7-10

Abstract : Description of different kinds of blast-furnace and
fuel slag, and means of utilizing them in building.

Card 1/1

HAZLER, T.

Slags and their use in the building industry. II

p. 10 (Budownictwo Wiejskiej) Vol. 9, No. 9, Sept. 1957, Warszawa, Poland

SO: MONTHLY INDEX OF EAST EUROPEAN ACCESSIONS (EEAI) LC, VOL. 7, NO. 1, JAN. 1958

HAFLER, TADEUSZ

Budynki z zuzlobetonu. (Wyd. 1)

Warszawa, Poland. Arkady. 94 p.

Monthly List of East European Accessions (EMAI) LC, Vol. 8, no. 8
August 1959.

Uncl.

HAZSLINSZKY, Tamás

Stalactites. Elet tud 19 no.6:248-250 7 F'64.

HAZMAN, I.

Transistor preamplifier for electron-tube equipment. p.183.

RADIOTECHNIKA. Budapest, Hungary. Vol. 9, no. 6, June 1959.

Monthly List of East European Accessions (EEAI), LC. Vol. 8, No. 9, September 1959

Uncl.

HIDAS, Gyorgy; KEMENY, Adam; HAZMAN, Istvan; KISS, Erno; SOMOGYI, Janos

The use of transistors in radio receiving sets; also, remarks by
E.Kiss and I.Somogyi. Muszaki kozl MTA 26 no.1/4:101-104 '60.

(EEAI 9:10)

1. Hiradastechnikai Kutato Intezet (for Hidas, Kemany and Hazman)
(Radio) (Transistors)

S/274/63/000/002/012/019
A055/A126

AUTHOR: Házman, István

TITLE: Unified studio amplifier

PERIODICAL: Referativnyy zhurnal, Radiotekhnika i Elektrosvyaz', no. 2, 1963, 57, abstract 2B424 (Hiradástechn. ipari kutatás int. közl., 1961, v. 1, no. 1, 42 - 49, 60; Hungarian; summaries in German, English and Russian)

TEXT: The author describes the circuit, the parameters and the calculation of the elements of a unified studio amplifier with six semiconductor-triodes made in Hungary and in Western Europe. The pass-band of the amplifier is 30 - 15,000 cps with a ± 0.5 db irregularity; at a gain of 40 db and at a noise level of -54 db (noise of the amplifier itself), the sensitivity of the amplifier is 100 μ v across 30 ohm; the largest output signal is 6.2 v. The amplifier is intended for operation with a microphone, and also as a linear line-repeater in lines with 50 and 200 ohm impedances. The amplifier has a transformer-coupled input and output. It consists of five resistance-coupled

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Unified studio amplifier

S/274/63/000/002/012/019
A055/A126

stages. The output stage is a push-pull stage. A 24-v battery is used as power source.

D.Yu.

[Abstracter's note: Complete translation]

Card 2/2

HAZMAN, Istvan

Transistor pre-amplifiers used in the motion pictures industry.
Kep hang 7 no.3:88-93 Je '61.

HAZMAN, Istvan

Transistorized sound system in railway cars. Kép hang ?
no.3:91-94 Je '63.

HAZMAN, Istvan

On dynamics expansion. Radiotechnika 13 no.4:154-155 Ap '63. .

HAZMAN, Istvan

On dynamics expansion. Radiotekhnika 13 no.5:194-195 My '63.

HAZMAN, Istvan

Cutoff frequency measurements of junction transistors. Hir techn
15 no.3:75-76. Mr '64.

1. Research Institute of the Telecommunication Industry, budapest.

HAZMUKA, B.

"Prefabrication of products in the sideline building industries and in finishing work."

POZEMNI STAVBY, Praha, Czechoslovakia, Vol. 3, No. 10, October 1955.

Monthly List of East European Accessions (EEAI), LC, Vol. 8, No. 9, September 1959.

Unclassified.

1/1

HUNGARY

FERENCZY, Lajos, ZSOLT, Janos, HAZNAGY, Andras, TOTH, Laszlo, SZENDREI, Kalman; Jozsef Attila University, Institute of Plant Physiology (director: SZALAI, I.) (Jozsef Attila Egyetem, Novegyeletani Intezet), Szeged, and Medical University of Szeged, Institute of Pharmacognosy (director: NOVAK, I.) (Szegedi Orvostudomanyi Egyetem, Gyogyszerhatastani Intezet).

"The Antifungal Constituents of *Cynanchum Vincetoxicum* (L.) Pers I. The Quantitative Antifungal Spectrum of Substance C-1."

Budapest, Acta Microbiologica Academiae Scientiarum Hungaricae, Vol XII, No 4, 1965/66, pages 337-344.

Abstract: [English article, authors' English summary modified] The antifungal activity of a compound isolated from *Cynanchum vincetoxicum* (L.) Pers and designated as Substance C-1 has been studied against 40 yeasts, 40 moulds and 20 dermatophytes. With the exception of a few moulds, most of the examined species and strains, including all of the human pathogenic fungi tested, were found to be highly sensitive to the compound. The minimal inhibitory concentration on glucose-broth-peptone culture medium at pH 7 was found to be between 0.015 and 1.0 µg/ml. 2 Eastern European, 6 Western references. [Manuscript received 31 Jul. 65.]

1/1

HAZSLINSKY, B.

MD
The toxic effect of honey from belladonna (*Atropa bella-*
donna). B. Hazslinsky (Chem. Inst., Budapest, Hung.).
Z. Bienenforsch. 3, 93-6 (1956) (English summary).—Inges-
tion of honey and pollen collected by bees from *A. belladonna*
caused dizziness in 2 persons and serious poisoning in
another. The honey, contg. *atropa alkaloids*, was not in-
jurious to the bees.
Martin Jacobson

Hazzlinsky, E.

102
Poisonous honey from deadly nightshade. B. Hazzlinsky
(Chem. Inst. Hauptstadt, Budapest). *Z. Bienenforsch.* 3,
93-6(1950); *Bee World* 37, 164(1958).—The poisonous effect
of deadly nightshade honey, due to the presence of bella-
donna alkaloids, is described. The method by which the
slightly H₂O-sol. alkaloids get into the nectar has not yet
been established. P. H. Wells

HAZSLINSZKY, Tamas

"The world of caves" by Jakucs-Kessler. Reviewed by
Tamas Hazslinszky. Elet tud 18 no.7:212 17 F '63.

HAZSLINSZKY, Tamas

"Adventures in the depth of the sea" by Odon Radai. Reviewed
by Tamas Hazslinszky. Elet tud 18 no.19:582 12 My '63.

HAZUKA, Miroslav, MUDr

Effect of trauma on etiology and course of tuberculosis. Prakt.
lek., Praha 34 no.22:509-511 20 Nov 54.

1. Z chirurg. klin. sakladny pro doskoleni lekaru pri obv. nem.
v Praze 8-Bulovka; predn. prof. MUDr Jan Knobloch
(TUBERCULOSIS, etiol. and pathogen.
trauma)
(WOUNDS AND INJURIES
trauma in etiol. of tuberc.)

HAZUKA, M., Dr.; HERZ, J., Dr.; MACH, F., Dr.

Fracture of hand bones. Acta chir. orthop. traum. cech.
23 no.2:72-75 Feb 56.

1. Z Chirurgické Klinické Zakladny UDL v Praze 8, Bulovka,
prednosta prof. MUDr. Jan Knobloch.

(HAND, fract.
management. (Cz))

(FRACTURES,
hand, management. (Cz))

HAZUKA, M.; MACH, F.

Surgical technics in simultaneous injuries of the brain and other parts of the body. Rozhl. chir. 41 no.4:289-291 Ap '62.

1. Chirurgická klinická základna UDL v nemocnici na Bulovce v Praze 8, přednosta prof. MUDr. J. Knobloch, DrSc.
(BRAIN wds & inj) (SHOCK ther)

ORSZAGH, J.; KAS, S.; HAZUKA, V.

The autonomic nervous system in infectious hepatitis. Cesk.
gastroent. vyz. 17 no.3:180-184 Ap '63.

1. Neurologické oddelení nemocnice v Praze-Motole, vedoucí doc.
dr. K. Mathon Oddelení infekčních zloutenek nemocnice v Praze-
Motole, vedoucí MUDr. O. Soušek.

(HEPATITIS, INFECTIOUS)
(AUTONOMIC NERVOUS SYSTEM)
(PULSE) (BLOOD PRESSURE)
(ELECTROCARDIOGRAPHY)

HAZUKOVA, J.

STICH, Z., MAREK, A., PERNICKY, J., HAZUKOVA, J., WALLENFELDS, V.,
ZOFKA, J., STRITESKY, J., MALEK, I. —————

Socialisation of medicine in Czechoslovakia. Zdravot. rev.
25:6, June 50. p. 155-6

CLML 19, 5, Nov., 1950

HEBDA, Michal, mjr. mgr. inz. (Warszawa)

The present state and development of research methods on the
properties of the subsurface layer. Pt. 1. Przegl mech 21
no.13:398-400 10 J1 '62.

BROS, Jan, dr inż.; HEBDA, Michał, dr inż.

Problems of research centers of friction and wear of the Institute of Basic Technical Problems, Polish Academy of Sciences, during the years 1957-1964. Przegł mech 23 no.23: 688-689 10 D '64.

1. Technical University, Krakow (for Bros). 2. Military Engineering School, Warsaw (for Hebda).

HEBDA, Michal, mjr., mgr inz. (Warszawa)

State and development of the research methods on the properties
of the subsurface layer. Pt.2. Przegł mech 21 no.14: ~~441-444~~
25 J1 '62.

HEBANOWSKI, Marek

Contribution to acute renal failure due to castor bean poisoning. Pol. tyg. lek. 19 no.31:1204-1205 3 Ag'64

1. Z II Kliniki Chorob Wewnętrznych Akademii medycznej w Gdansku; kierownik : prof. dr. Jakub Penson.

HEWDELSKA, E.

Analysis of Fulfilled Tasks in the Six-Year Plan. p.3.
PRZEGLĄD GOSPODARSTWA (Polskie Wydawnictwa Gospodarcze) Warszawa
Vol. 11, no. 4, Apr. 1956

So. East European Accessions List

Vol. 5, No. 9

September 1956

HEBDZYNSKA, Z.; Chojecki, M.

Some observations on the preparations for the summer season in summer resorts.
p. 8.
(PRZEMYSŁ GASTRONOMICZNY. Vol. 11, no. 6, June 1956, Warszawa, Poland)

SO: Monthly List of East European Accessions (EEAL) LC. Vol. 6, No. 12, Dec. 1957.
Uncl.

HEBEANU, Gh.

Concrete, dynamic, and combative visual agitation.
Munca sindic 7 no.12:45-47 D '63.

1. Presedintele comitetului sindicatului de la uzinele
"Industria Sirmei" Cimpia Turzii.

HEBERT, J.

Program for specialization in roentgenography. Med. glasn. 11 no.5:
194-196 May 57.

(SPECIALISM
in roentgenography (Ser))

CZECHOSLOVAKIA/Nuclear Physics - Installations and Instruments. C-
Methods of Measurement and Research.

Abs Jour : Ref Zhur Fizika, No 3, 1960, 5149
Author : Hebek Antonin
Inst : -
Title : Effect of Liquid Vapor on Pointed Geiger-Muller Counters
Orig Pub : Chekhosl. fiz. zh., 1959, 9, No 1, 127-128
Abstract : No abstract.

Card 1/1

S/194/62/000/002/006/096
D230/D301

AUTHORS: Przybylski, Tadeusz, Zuk, Mieszysław and Hebel,
Zygmunt

TITLE: Constant-current analyzer

PERIODICAL: Referativnyy zhurnal, Avtomatika i radioelektronika,
no. 2, 1962, abstract 2-1-33zh (Zesz. nauk. Politechn.
gdąnsk., 1960, no. 21, 63-70)

TEXT: The Department of Electrical Engineering of the Gdąnsk Poly-
technic Institute has developed, for own use, a new constant-cur-
rent analyzer. Five types of the variable potentiometer were used
as standard resistance boxes: 0.2; 2.5; 5; 50 and 100 kohms; the
resistance values for a given position were determined using an
ohmmeter. The potentiometers are divided into two groups of 60, in
each of the vertical analyzer fields; they are connected by means
of two single-cord jack and sockets in the "simulation field" cir-
cuit containing 48 switching stations with sockets. 200 and 500
ohm-potentiometers are used as auxiliary junction points for the

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D230/D301

Constant-current analyzer

simulation field; these points are connected with the corresponding socket junction points of the measured field which are used for measuring resistances, currents and voltages. The measuring instruments are mounted on the board of the analyzer. In the upper part of the analyzer there is the power supply switching field, distributing the voltage to 5 junction points and to sockets (each junction point consists of 5 sockets). The measuring ammeter is supplied with a range switch, current polarity switch and an auxiliary resistance used for calculating errors introduced by the internal resistance of the ammeter. In the analyzer circuit there are two class 0.5 ammeters for currents 1.5; 3 and 7.5 and 15, 30 and 75 ma. Voltage and resistance measurements are made with a valve voltmeter having ranges 3, 10, 30 and 100 V. The accuracy of voltage and resistance measurements is 3 and 5 percent, respectively (for the measuring range 0.1 to 100 kohms). A. c. or d.c. mains at 220 V provides the power supply; the circuit has five stabilized rails, 0 to 80 V (max. load of each rail 30 ma). The analyzer is inexpensive and small in size; it is thus suitable for use at the

Card 2/3

Constant-current analyzer

S/194/62/000/002/006/096
D230/D301

Polytechnic laboratories, technical colleges and also for design
(and similar) organizations. 4 figures. [Abstracter's note: Com-
plete translation.]

Card 3/3

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CZECHOSLOVAKIA

JEZEK, Z; HEBELKA, M; SVANOVA, E.

Institute of Epidemiology and Microbiology (Ústav epidemiologie a mikrobiologie), Prague (for all)

Prague, Rozhledy v tuberkulose, No 6-7, 1963, pp 454-460

"The Significance of *Mycobacterium bovis* in the Spread of Tuberculosis Infection in the Rural Population."

SERY, V.; JEZEK, Z.; SVANDOVA, E.; FUCHSOVA, M.; HEBELKA, M.

Use of tuberculin tests in the study of *Mycobacterium bovis*.
II. Analysis of allergy to tuberculin in children and adolescents
in relation to *Mycobacterium bovis* infection. *Cesk. epidem.* 12
no.5:262-267 S '63.

1. Ustav epidemiologie a mikrobiologie v Praze - Tuberkulozni
oddeleni OUNZ v Litomericich.
(TUBERCULIN REACTION) (TUBERCULOSIS, BOVINE)
(TUBERCULOSIS IN CHILDHOOD) (MYCOBACTERIUM BOVIS)

JEZEK, Z.; SERY, V.; HEBELKA, M.; SVANDOVA, E.

The value of simultaneous application of human and bovine
tuberculin. Cesk. epidem. 14 no.3:143-148 My '65

1. Ustav epidemiologie a mikrobiologie, Praha, Katedra nemoci
tropu a subtropu, UDL, Praha.

CZECHOSLOVAKIA

POPLUHAR, L.; ROSIVAL, F.; JEZEK, Z.; HEBELKA, M.; Veterinary Faculty, Chair of Infectious Diseases, College of Agriculture (VSP, Veter. Fakulta, Katedra Infekcnych Chorob), Kosice; Okresny Epizootologist (Epizootolog), Kosice; Institute of Epidemiology and Microbiology (Ustav Epidemiologie a Mikrobiologie), Prague.

"On the Problems of Tuberculosis in Pigs."

Prague, Veterinarni Medicina, Vol 11, No 8, Aug 66, pp 485-496

Abstract [Authors' English summary modified]: Mammalian PPD tuberculin was used for allergic diagnosis of tuberculosis in pigs, using a dose of 5000 Tu, and a dose of 2500 Tu of avian tuberculin. The swellings were excessive; when only 500 Tu of either tuberculin were used reliable results were obtained. Swellings which had a diameter of 8 mm and over were considered to be positive reactions. 4 Figures, 4 Tables, 5 Western, 5 Czech, 3 Russian, 1 Hungarian reference. (Manuscript received 10 May 65).
1/1

L 31845-66 T JK

ACC NR: AP6021322 (A) SOURCE CODE: PO/0081/65/019/003/0309/0313 48
 AUTHOR: Jeliazowicz, J.; Hawigor, J.; Czacka, J.; Cygankiewicza-Siennicka, M.;
Gorska, A.; Gulinski, J.; Hebenstreit, C.; Klimek, H.; Klapowska, K.; Krol, J.;
Lenartowicz, G.; Luft, A.; Moskwa, Z.; Nocen, I.; Pawlowska, J.; Padrycz, H.; Pernal, C.
C.; Pogorzelska, A.; Rodzinski, L.; Siennicki, W.; Sikora, G.; Szymanczyk, I.; Terech,
I.; Hawrzynska, M.; Nencel, Z.; Znis, A.
 ORG: Institute of Bacteriology, PZH, Warsaw (Zaklad Bakteriologii); Regional and
City Sanitary Epidemiological Centers, Bydgoszcz, Katowice, Kielca, Krakow, Lodz, Opole,
Rzeszow, Warsaw, Wroclaw (Wojewodska i Miejska Stacj Sanitarno-Epidemiologiczna);
Bacteriologic Laboratory, No. 3, PSK, Wroclaw (Laboratorium Bakteriologiczny)
 TITLE: Antibiotic-resistant strains of *Streptococcus viridans*, *Streptococcus Fecalis*,
Escherichia coli, *Pseudomonas aeruginosa*, *Proteus* species and *Klebsiella* species,
 isolated in Poland in 1960-1963
 SOURCE: Przegląd epidemiologiczny, v. 19, no. 3, 1965, 309-313
 TOPIC TAGS: bacteriology, penicillin, streptomycin, tetracycline, erythromycin,
 neomycin
 ABSTRACT: Sensitivity tests of the above strains were carried out in respect to peni-
 cillin, ~~streptomycin~~, tetracyclines, chloramphenicol, erythromycin and neomycin. It
 was found that resistance to antibiotics in *Streptococci* differed from that in Gram-
 negative bacilli. *Streptococcus fecalis* was found highly resistant to penicillin and
 erythromycin. Appreciable resistance to all antibiotics was noted in strains identified
 as *Streptococcus viridans*. Resistance varied according to samples and territorial dis-
 tribution. Experiments were conducted in 11 centers throughout the country simultane-
 ously; results were compared with those obtained in an identical experimental series in
 a single hospital environment. Orig. art. has: 2 tables. ^(SPRS)
 SUB CODE: 06/ SUBM DATE: none/ ORIG REF: 001/ OTH REF: 001
 Card 1/1 JS

~~HEBER, G.~~

HUNGARY/Theoretical Physics - Quantum Field Theory

B-6

Abs Jour : Ref Zhur - Fizika, No 2, 1958, No 2733

Author : Heber, G.

Inst : Not Given

Title : Latest Results of the Meson Theory of Nucleons

Orig Pub : Magyar fiz. folyoirat, 1957, 5, No 1, 73-78

Abstract : No abstract

Card : 1/1

HUNGARY/Theoretical Physics - Quantum Theory of Fields

Abs Jour : Ref Zhur - Fizika, No 9, 1958, No 19690

Author : Heber G

Inst : Not Given

Title : Remarks on the Measurement of the Space-Time Continuum.

Orig Pub : Fiz. szonle, 1957, 7, No 5, 166-167

Abstract : In connection with the difficulties of modern local quantum theory with nonlinear coupling, the author indicates that a study of the problem of the measurement of space-time continuum gives information on the possibility of constructing a single generalized field theory, which resolves the existing difficulties. The author reaches the conclusion that the coordinates of the world point cannot be considered to be c-numbers, since the commutators formed out of the field operators do not always vanish. This result is discussed from the point of view of the field theory, and the results obtained are compared with those of Bohr and Rosenfeld on the measurement of quantum magnitudes.

Card : 1/1

HEMER, G.

The present state of the research work on the theory of space. p. 248.

FIZIKAI SZEMLE. (Eotvos Lorand Fizikai Tarsulat) Budapest, Hungary. Vol. 9, no. 8, 1959.

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HEBER, Gerhard (Jena)

Present state of research in the field of space theory. Fiz szemle
9 no.8:248-249 Ag '59.

HEBER, G. (Jena); SZABO, Janos [translator]

Remarks about the mensuration of the space-time continuum. Fiz.
szemle 7 no.5:166-167 0 '57.

CZECHOSLOVAKIA/Nuclear Physics - Installations and Instruments. C-
Methods of Research and Measurement.

Abs Jour : Ref Zhur Fizika, No 3, 1960, 5206

Author : Heber Milos

Inst :

Title : Radiochemical Laboratories for Work with High Activities

Orig Pub : Jaderna energie, 1959, 5, No 6, 184-189

Abstract : A survey is given of materials on the equipment of radio-
chemical laboratories, as presented at the Second Interna-
tional Conference on Peaceful Use of Atomic Energy.

Card 1/1

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"Workshop Tricks; Glass Cutting." . . 635 (HUNGARIAN TECHNICA. Vol. 9, No. 11,
Nov. 1954; Budapest, Hungary.)

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Methods for determination of the size of series in foundries. p.408

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Budapest, Hungary
Vol. 11, no.10, Oct. 1959

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Uncl.

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The tasks of assistant-professors at the Polytechnical University.
Periodica polytechn eng 4 no.3:277-299 '60. (EEAI 10:6)

1. Department of Mechanical Engineering Technology, Polytechnical
University, Budapest.
(Hungary--Technical education)

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"Ten years of the Chair of Technology of Machine Building."
Reviewed by K.Heberger. Periodica polytechn eng 6 no.2:173-174
'62.

HEBERGER, Karoly, adjunktus

Typification. Gép 14 no.9:331-336 S '62.

1. Budapesti Muszaki Egyetem Gepgyartastechnologiai Tanszek.

HEBERGER, Karoly, dr.

Results and problems in the training of engineers. Muzs elet
18 no.14:4 4 JI '63.

1. Muvelodesugyi Miniszterium.

HE51B, D

Problem which must be solved speedily. p. 256. (BEOGRAD, Vol 10, No. 7, July 1954.)

EG: Monthly Lists of East European Accessions. (EEAL, LC, Vol 4, No. 6, June 1955, Uncl.

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Air brakes and safety of railroad traffic, p. 257. (SECSTATE, Vol 10, No 7, July 1954.)

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HEBKY , A; PETRANEK, J.

Hydraulic transmission for motorcycles.

p. 200 (Automobil) Vol. 1, no. 6, June 1957 Praha, Czechoslovakia

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HEBKY, A.

New methods for regulating pressure water turbines. p. 11. (Strojnoelektrotechnicky Casopis. Bratislava. Vol. 3, no. 3, 1952)

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 4, No. 6, June 1955, Uncl.

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SOURCE: East European Accessions List (EEAL) Library of Congress, Vol. 4, No. 12, December 1954

HEBKY, A.

✓ Practical consequences of compressibility of liquids on pumping. A. Hebký. *Strojrenitel* 8, 837-40 (1958).—
The effect of compressibility of liquids in stationary and transient flow is analyzed. Results are applied to 2 extreme cases of stationary and transient flow with rapid velocity changes where hydraulic shocks develop. The role of compressibility is shown. J. G. Tschinkel

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Hebky, A.

Hydrodynamic similarity and the equations of Navier-Stokes. p. 483.

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"Hydrodynamic couplings and converters" by E. Kickbusch. Reviewed by Alois Hebký. Stroj vyr 12 no.8:612 '64.

"Pumpy" by Fuchslocher, Schulz. Reviewed by Alois Hebký. Ibid.: 613

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"Design of hydraulic control systems" by Ernest E. Lewis and Hansjoerg Stern. Reviewed by Alois Hebky. Stroj vyr 10 no.12:637 '62.

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Discussion on teaching chemistry in secondary and higher schools.
Chem listy 58 no.9:1129-1130 S '64.

HEBKY, J.

Preparation of substances of the *o*-aminoazotoluene type and their derivatives substituted in the amino group. I. V. Etki and J. Hebky. *Collection Czechoslov. Chem. Commun.* 13, 461 (1948) (in English).—A no. of compds. of the *o*-aminoazotoluene type are modified by the addn. of salicylate, 3-hydroxy-2-naphthoate, or sulfanilamide groups, and tested for epithelization and bactericidal activities. *o*-Aminoazotoluene, 2',3'-dimethyl-4'-aminoazobenzene (I), 2',3'-dimethyl-4'-sulfamyl-4'-aminoazobenzene (II), and 3-methyl-4'-sulfamyl-4'-aminoazobenzene (III) are used as intermediates. I is prepd. by diazotizing *o*-toluidine. 3,4-Me(AcNH)C₆H₃SO₂NH₂ prepd. according to Child and Smiles (C.A. 21, 234) is hydrolyzed with 1:1 HCl by heating to 80° until dissolved, the soln. decolorized with C, and the compd. pptd. by addn. of base. After crystn. from H₂O, 4-amino-*m*-toluenesulfonamide [3-methyl-4-*o*-inobenzenesulfonamide] (IV), m. 160°, is obtained in 28% yield. To IV (37.2 g.) in 400 cc. H₂O and 56 cc. concd. HCl is added with stirring at 7° 15.1 g. 92% NaNO₂ in 80 cc. H₂O, then, after 15 min., 22.4 g. *o*-toluidine in 60 cc. H₂O and 22 cc. concd. HCl, the temp. raised to 22°, 55 g. NaOAc added, and after 5 hrs. the ppt. filtered, washed, and crystd. from alc., giving II, m. 215° or 220°. To 172 g. sulfanilamide in 2 l. H₂O and 280 cc. concd. HCl is added at 8° with stirring 74.5 g. 92% NaNO₂ in 400 cc. H₂O, then, after testing with starch-I paper, 112 g. *o*-toluidine in 300 cc. H₂O and 110 cc. concd. HCl, the temp. raised to 22°, 280 g. NaOAc added, and, after 4 hrs. stirring, the ppt. filtered, washed, and crystd. from alc. to give 67% III, m. 203°. Ac₂O (102 g.) and 2 drops concd. H₂SO₄ are added to 22.5 g. *o*-aminoazotoluene, the mixt. refluxed 4 hrs. and ground in a mortar until solid, decolorized with

COMMON ELEMENTS

MATERIALS INDEX

PROPERTIES INDEX

C, and crystd. from MeOH, giving 25 g. (84% yield) 2,3'-dimethyl-4'-(diacetylsulfamyl)azobenzene (V), m. 65°. To 11.3 g. II in 100 cc. dry C₆H₆ is added 81.3 g. *o*-HO-C₆H₄COCl in 80 cc. C₆H₆, the mixt. heated 2 hrs. at 50-60°, the ppt. supd., dissolved in cold 10% NaOH, acidified with HOAc, and crystd. from alc., giving 49.2 g. 2,3'-dimethyl-4'-(salicyloylsulfamyl)azobenzene, m. 208°. To I (7.5 g.) in 30 cc. dry Me₂CO is added 30 g. fresh 3,2-HOC₆H₄COCl in 60 cc. Me₂CO, the mixt. refluxed 1 hr., poured hot, with stirring, into 10% NaOH, filtered, 10% NaOH poured over the filter, and the combined filtrates made acid with HOAc, giving, after crystn. from alc., 7.6 g. 2,3'-bis(4'-3-hydroxy-2-naphthyl)aminoazobenzene, m. 245°. To pulverized I (14.7 g.) in 75 cc. anhyd. pyridine bases (b. 120-40°) at 40-60° is added over 20 min. 80.3 g. powd. *p*-MeCO₂NC₆H₄SO₂NH₂, the mixt. boiled 10 hrs., the pyridine bases distd. off under reduced pressure, and the residue, after boiling with 45 cc. MeOH and 24 cc. concd. HCl 1 hr., filtered and ground with concd. NaOH soln., giving, after crystn. from alc.,

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1/2 V. E. H. ...

22 K. (80%) 2,3-dimethyl-4-sulfamylaminobenzene, m. 190°. Ac₂O (10 g.) and 15.2 g. II are heated 4 hrs. on a water bath, ground with H₂O, filtered, and crystd. from alc., giving 3,2'-dimethyl-4-acetamidobenzene-4'-sulfonamide, m. 244°. II (13 g.) in 600 cc. Me₂CO and 13 g. o-HOC₆H₄COCl are allowed to stand 0.5 hr., then refluxed 1 hr., filtered, and the residue dissolved in 10% NaOH, filtered, and made acid with HOAc, giving after filtration and crystn. from 80% alc. 11 g. (82%) 2,3-dimethyl-4-sulfamylamino-4'-azobenzene-sulfonamide (VI), m. 233°. II (13.6 g.) in 75 cc. dry Me₂CO and 2.7 g. 3,2-HOC₆H₄COCl in 15 cc. Me₂CO are refluxed 4 hrs. and the ppt. filtered, ground with NaOH soln., washed, and crystd. from alc. to yield 3.5 g. (63%) 2,3-dimethyl-4-(3-hydroxy-2-naphthylamino)azobenzene-4'-sulfonamide, m. 203° (decolor.). To 4 g. II in 15 cc. dry pyridine bases (b. 120-40°) at 40° is added, with stirring, 4 g. fresh p-McCONH₂SO₂NH₂, the mixt. heated at 100-10° 4 hrs., the pyridine bases distd. off at reduced pressure, the residue refluxed with 12 cc. MeOH and 6.0 cc. concd. HCl 1.5 hrs., the mixt. poured into 150 cc. H₂O, filtered, and the solid treated with warm dil. Na₂CO₃, washed with H₂O, and crystd. from 80% alc., giving 5.5 g. (92%) 2,3-dimethyl-4-sulfamylamidoazobenzene-4'-sulfonamide, m. 217°. 3-Methyl-4-acetamidobenzene-4'-sulfonamide, m. 217°. 3-Methyl-4-acetamidobenzene, m. 238° (from 80% HOAc, 44% yield); 2,4-dimethyl-4-sulfamylamino-4'-sulfamylamidoazobenzene-4'-sulfonamide, m. 205° (67% yield); 3-methyl-4-(3-hydroxy-2-naphthylamino)azobenzene-4'-sulfonamide, m. 233-4° (57% yield); 2-methyl-4-sulfamylamidoazobenzene-4'-sulfonamide, m. 233° (80% yield) are prepd. similarly. The *in vitro* bactericidal activities of II, III, V, and VI are about equal, but the bactericidal activity is low. Clarence T. Mason

Demethylation of carcinogenic aminoazo dyes by autoxidizing linoleic acid. J. P. Busch and J. A. Miller (Univ. of Wisconsin, Madison). *Proc. Soc. Exptl. Biol. Med.* 68, 1-40-3(1948); 67, C.A. 42, 2515c. p-PhN₂NC₆H₄NCMe₂ (I) and p-PhN₂NC₆H₄NMe (II) increased the latent period of autooxidation of linoleic acid. The effect was proportional to the concn. of I and II. There was a little more effective than II. p-PhN₂NC₆H₄NMe and the m⁺ and p⁺-Me derivs. of I had approx. the same inhibiting power as I. p-PhN₂NC₆H₄NMe was only slightly inhibitory and p-PhN₂NC₆H₄NMe was fully inhibitory. During the latent period, formation of Es (probably CO₂) was observed when any of the methylated dyes were mixed with the linoleic acid; Es formation was also observed with autoxidizing linoleic acid alone at the end of the oxidation period. As autooxidation of the dye-acid mixts. proceeded, demethylation of I and II occurred. After 30 hrs. 90% of the I initially added had disappeared and as much as 85% of the II initially added had disappeared after the first 24 hrs. Small amounts of p-PhN₂NC₆H₄NMe were detected. No demethylation occurred when the oxidation was inhibited by adding 2% of troloxol.

J. E. Gilson

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HECK, J.

Synthesis of compounds of the *o*-aminoazotoluene type and their amino-substituted derivatives. H. V. Ritel and J. Hebký (United Chem. and Met. Works, Prague). *Collection Czechoslov. Chem. Commun.* 13, 65-72 (1930) (in English); cf. *Chem. Abstr.* 42, 7291d. *o*-Aminoazotoluene (I) heated with aq. sol. $\text{HClO}_4 \cdot \text{NaHSO}_4$ for 4 hours at 90° with stirring gave 70% 4,5- $\text{NaSO}_3\text{CH}_2\text{NH}(\text{Me})\text{C}_6\text{H}_4\text{N}:\text{NC}_6\text{H}_4\text{Me}-2$, yellow, m. 207° . Refluxing I with succinic anhydride in abs. EtOH with a trace of H_2SO_4 for 4 hrs. gave, after recryst. of the crude solid from EtOH, 60% brick-red 2-(2-dimethyl-4-(succinylamino)azobenzene (II), m. 187° . The Na salt of II is readily sol. in water, and seps. as brick red crystals from EtOH-H₂O. The citraconyl deriv. of I prepd. in Me_2CO soln., m. $155-6^\circ$ (from alk.). D-Glucose in Me_2CO , refluxed for 3 hrs. with a slight excess of I and a catalytic amt. of NH_4Cl , gave 70% of the glucoside, m. 135° (decompn.), sparingly sol. in cold H₂O, readily sol. in EtOH or Me_2CO . *o*- $\text{MeC}_6\text{H}_4\text{NHCH}_2\text{SO}_3\text{Na}$ coupled with the diazonium salt from 3,4- $\text{Me}_2\text{H}_2\text{N}(\text{C}_6\text{H}_4)_2\text{SO}_3\text{NH}_2$ gave 4,5- $\text{NaSO}_3\text{CH}_2\text{NH}(\text{Me})$ -

$\text{C}_6\text{H}_4\text{N}:\text{NC}_6\text{H}_4(\text{Me})\text{SO}_3\text{NH}_2-2,4$, vermilion red, m. 212° , sparingly sol. in H₂O. Boiling 1,1,1- $\text{Me}_3\text{H}_3\text{N}(\text{C}_6\text{H}_4)_2\text{Na}^+$ with $\text{NaHSO}_4\text{NH}_2$ in Me_2CO with succinic anhydride and a trace of H_2SO_4 for 4 hrs. distg. off the Me_2CO , extg. with NaHCO_3 soln., and adding gave the succinyl deriv., red, m. 210° . *o*- $\text{MeC}_6\text{H}_4\text{NHCH}_2\text{SO}_3\text{Na}$ coupled with PhNH_2 gave 4,5- $\text{NaSO}_3\text{CH}_2\text{NH}(\text{Me})\text{C}_6\text{H}_4\text{N}:\text{NC}_6\text{H}_4$, yellow, m. 212° , which on hydrolysis with NaOH and treatment with succinic anhydride gave 3-methyl-4-(succinylamino)azobenzene, m. 192° . *p*- $\text{H}_2\text{NC}_6\text{H}_4\text{NaPh}$ (III) heated with $\text{HClO}_4 \cdot \text{NaHSO}_4$ gave Na *p*-phenyl-azobenzene-*o*-methanesulfonate (orange plate), m. 251° . *o*-Azobenzene-*o*-methanesulfonate (orange plate), III with toluidine gave Na *o*-toluidine-*o*-methanesulfonate, yellow succinic anhydride gave *p*-succinylamino-*o*-toluidine, yellow, m. 210° . *o*- $\text{NaSO}_3\text{CH}_2\text{NHCH}_2\text{Me}$ and *p*- $\text{H}_2\text{NC}_6\text{H}_4\text{NHCH}_2\text{SO}_3\text{Na}$, prepd. from $\text{HClO}_4 \cdot \text{NaHSO}_4$ and the corresponding amines, are unstable in warm aq. soln. The water-sol. derivs. of I possess epithelization properties and are bactericidal, resembling the derivs. of 3,4- Me_2 and are bactericidal, resembling the derivs. of 3,4- Me_2 (III) has only a very slight epithelization effect and is about half as bactericidal as I. The derivs. of *o*-toluidine have almost as much epithelization activity as I, but are not bactericidal. Benzidine has weak epithelization and no bactericidal activity. The water-sol. derivs. of *o*-toluidine show neither activity. F. E. Löffler

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HELSKY, W.

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Pyridone derivatives of kojic acid. V. Ettel and J. Heblík. *Collection Czechoslov. Chem. Commun.* 15, 324 (1950) (in French). A cooled soln of 60 g. kojic acid and 75 g. cryst. NaOH in 100 cc. H₂O was treated over a 10 min. period with a soln of diazotized PhNH₂ (from 35 g. PhNH₂) and the colored mixt. sepd. after 2 hrs., washed with H₂O, and dried in air, giving 147 g. (92.8%) 2-hydroxy-methyl-5-hydroxy-6-phenylazo-4-pyridone (I). I (8.5 g.) in 70 cc. abs. EtOH, boiled 2 min., treated with CH₃N₃, allowed to stand overnight, and neutralized with 5% Na₂CO₃ gave 3.75 g. (44%) 2-hydroxymethyl-5-methoxy analog (II), orange-red needles, m. 170° (2-acetate, red needles, m. 134°; 2-benzoate, wine-red crystals, m. 128°; Me ether, red-orange crystals, m. 124°). 2-Methoxymethyl-5-methoxy-4-pyridone (4 g.) and 5 g. NH₃ heated 3 hrs. at 100° in an autoclave gave upon treatment with picric acid 2-methoxymethyl-5-methoxy-4(III)-pyridone picrate (III), m. 178°. 2-Chloromethyl-5-methoxy-4(III)-pyridone (4 g.) heated 6 hrs. at 85° with 10 cc. C₆H₆ and 5 cc. Et₃NH in a sealed tube gave on removal of the Et₃NH and chromatographic purification on Al₂O₃ 2-diethylaminomethyl-5-methoxy-4(III)-pyridone (IV), m. 41°. IV autoclaved 6 hrs. at 100° with 6 cc. of a satd. aq. NH₃ and purified chromatographically (Al₂O₃) gave 2-diethylaminomethyl-5-methoxy-4(III)-pyridone, isolated as the dipicrate, m. 188°; the

monopicrate could not be obtained pure. 2-Diethylamino-methyl-5-methoxy-4-pyridone (1 g.) in 8 cc. MeOH was allowed to stand 2 hrs. with 1 g. MeNH₂ in 5.5 cc. MeOH, heated 10 min. at 100°, the solvent removed, and the residue dried and pptd. as the 2-diethylaminomethyl-5-methoxy-4-pyridone picrate, m. 177°. 2-Chloromethyl-5-hydroxy-4-pyridone (3.2 g.) heated 8 hrs. at 80° in a sealed tube with 20 cc. Me₂S gave the trimethylammonium chloride, C₇H₁₀O₂NCl₂, m. 220° (decompn.); perchlorate, C₇H₁₀O₂NCl₄, m. 221°. 5-Me ether trimethylammonium chloride (V); perchlorate C₇H₁₀O₂NCl₄, m. 102° (effervescent). The trimethylammonium chloride of 2-chloromethyl-5-methoxy-4-pyridone (2 g.) heated 2 hrs. in a closed tube with 3 cc. satd. aq. NH₃ gave on treatment with picric acid, the picrate of (1,4-dihydro-5-methoxy-4-oxo-2-pyridylmethyl)trimethylammonium chloride (VI), m. 204°. From 2 g. V and 3 g. MeNH₂ is obtained the 1-Me deriv. of VI, obtained as the chloroplatinate, m. 221°. 1,4-Dihydro-5-methoxy-4-oxo-2-pyridylmethylacetic acid (5 g.) in 100 cc. EtOH treated with dry HCl 5 hrs. at 15-20°, then 5 hrs. at 50°, and refluxed 2 hrs. gave 2.5% Et ester, m. 145.6°. Fully pure 1-chloro-5-methoxy-2-pyridylmethylcarboxamide (2.0 g.) treated with 2.5 g. Br and 13.5 g. KOH in 240 cc. H₂O gave 1.8 g. 4-oxo-5-methoxy-2-aminopyridine, m. 99-100° B. K.

1951

CA HECCY, J.

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Synthesis of 2-diethylaminoethyl α -phenylvalerate V
 Eitel and J. Heblky. *Collection Czechoslov. Chem. Commun.* 15, 367-70 (1950) (in English).—Two syntheses of 2-diethylaminoethyl α -phenylvalerate (I), which has spasmolytic properties, are described. In the first, the K salt of $\text{PhCH}_2\text{CO}_2\text{CH}_2\text{CH}_2\text{NEt}_2$ is treated with PrBr at room temp. In the alternative procedure $\text{PhCH}_2\text{CO}_2\text{Et}$ is propylated and the $\text{PrCH}_2\text{PhCO}_2\text{Et}$ transesterified with $\text{Et}_2\text{NCH}_2\text{CH}_2\text{OH}$ (II). The over-all yields in either case are practically identical. To EtONa from 5.8 g. Na and 75 cc. abs. EtOH is added 17.3 g. $\text{PhCH}_2\text{CO}_2\text{Et}$ (from PhCH_2CN and CO_2Et), the mixt. allowed to stand overnight, heated 3 hrs. at 40° , 6 hrs. at 60° , 6 hrs. at 80° , and 6 hrs. at 100° , filtered, neutralized with 5 drops AcOH, the EtOH distd., and the residue distd., giving 46.3 g. *Et α -cyano- α -phenylvalerate* (III), bp $151-7^\circ$. To 75 cc. EtOH, 10 g. Na, and 8 cc. H_2O is added 22.5 g. III, the soln. refluxed 15 hrs., the EtOH distd., the residue taken up in 50 cc. H_2O , charcoaled, acidified with HCl, extd. with C_6H_6 , the solvent removed, and the residue distd., giving 16.5 g. *α -phenylvaleric acid* (IV), bp $148-0^\circ$, m. 53.5° . IV (31.2 g.) refluxed with 82 g. abs. EtOH and 17.5 g. H_2SO_4 2.5 hrs. gives 32.5 g. *Et ester* (V), bp 107° . V (41.2 g.) and 47 g. II contg. 0.06 g. Na are heated in N under a Widmer column, a mixt. of EtOH and II collected at 25-30 mm., then, with the column removed, 20.2 g. (excess) II is collected (2 hrs.);

the principal fraction, 53.4 g. I, bp $163-4^\circ$; HCl salt, m. $147-8^\circ$; picolonate, m. 146° . The transesterification of 90.3 g. $\text{PhCH}_2\text{CO}_2\text{Et}$ with 130 g. II and 0.125 g. Na is carried out identically, giving 91.3 g. 2-diethylaminoethyl *phenylacetate* (VI), bp $138-42^\circ$; picolonate, m. $103-4^\circ$. A suspension of 7.9 g. powd. K in 300 cc. boiling EtO is treated in 1.5 hrs. with 47 g. VI in 50 cc. EtO, 24.9 g. PrBr added, the mixt. allowed to stand 20 hrs., H_2O added, the Et_2O layer washed, dried, the solvent removed, and the residue fractionally distd., giving 23.2 g. I, bp $150-0^\circ$.
 Bernard Klein

1957

HEBKY, J

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Derivatives of pyrons, 4(1H)-pyridones, and pyridines. I. Preparation of some 1-phenyl-4(1H)-pyridones. V. Eitel and J. Hebký (Czech. Chem. works, Prague-Vysocany). *Collection Czech. Chem. Commun.* 15, 639-62 (1950) (in English). — PhNH₂ (I) (25 g.) in 80 ml. glacial AcOH was added to 21.5 g. di-Et 2,6-dimethyl-4-oxo-4H-pyran-3,5-dicarboxylate (II) [Conrad and Gutzzeit, *Ber.* 19, 19 (1886)], and the mixt. refluxed 15 min., dil. with 270 ml. water, and cooled; the crystals were suction filtered and recrystd. from 100 ml. 50% EtOH with charcoal to yield 26.1 g. di-Et 2,6-dianilino-4-oxo-2,5-heptadecano-3,5-dicarboxylate (III), colorless, m. 137° (dried 6 hrs. in vacuo at 80°), and 1 g. di-Et 1,4-dihydro-2,6-dimethyl-4-oxo-3,5-pyridinedicarboxylate (IIIa) (from the mother liquor), colorless, m. 170°; free acid (IV) of IIIa, colorless, m. 225-8° (from 50% EtOH). II (21.1 g.) in 40 ml. hot 50% AcOH and 16 g. p-H₂CN₂·OH (V) in 70 ml. 50% AcOH were heated 25 min. (water bath, shaking), and the crystals which sepd. on cooling suction filtered, combined with other crystals from the mother liquor, and recrystd. from 500 ml. EtOH; evapn. of the mother liquor gave 28 g. di-Et 1,4-dihydro-1-(p-hydroxyphenyl)-2,6-dimethyl-4-oxo-3,5-pyridinedicarboxylate (VI), colorless, m. 259-60° (decompn.); free acid (VII), colorless, m. 246° (from 40% EtOH, foaming). Di-Et chelidonate (VIII) [Willstätter and Pummerer, *Ber.* 37, 3737 (1904)] (40.8 g.) in 150 ml. warm 50% AcOH and 27.8 g. V in 250 ml.

50% AcOH were heated 25 min. (water bath), 400 ml. water added to the hot mixt., and the crude cryst. product which sepd. on cooling filtered with suction; recrystn. from 500 ml. 50% EtOH gave di-Et 1,4-dihydro-1-(p-hydroxyphenyl)-4-oxo-2,6-pyridinedicarboxylate (IX), colorless, m. 185°; free acid (X), colorless, m. 198° (decompn.). II (13.3 g.) in 60 ml. 50% AcOH and 12 g. p-H₂CN₂·OH (XI) in 105 ml. 50% AcOH were heated 20 min. on a water bath; after cooling, the crystals were suction-filtered (addnl. crystals were obtained by addn. of 100 ml. water to the mother liquor), and recrystd. from 200 ml. 90% EtOH with charcoal to yield 23 g. di-Et 1,4-dihydro-1-(m-hydroxyphenyl)-2,6-dimethyl-4-oxo-3,5-pyridinedicarboxylate, colorless, m. 241°; free acid, colorless, m. 249-50° (decompn. from water). VIII (18 g.) in 25 ml. 50% AcOH was heated 20 min. (water bath) with 8 g. XI in 40 ml. 50% AcOH, 200 ml. water added to the hot soln., and the oil which sepd. by scratching with a glass rod; an addnl. crop from the mother liquor was obtained by further diln. with water; and suction filtration of both crops, and repeated recrystns. from 50% EtOH gave 6 g. di-Et 1,4-dihydro-1-(m-hydroxyphenyl)-4-oxo-2,6-pyridinedicarboxylate (XII), colorless, m. 175-6°; free acid (XIII), colorless, m. 191° (decompn., rate of heating = 1°/min.). 2,6-(MeO)₂C₆H₃OH (C4 g.) was dissolved in a hot soln. of 60 g. calcined Na₂CO₃ in 1400 ml. water, and the soln. cooled with ice water and treated with Ph₂NCI (from 37.2 g. I) (stirring); suction filtration and

CA [unclear]

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2,2'-Dinitro-4,4'-diformamido-3,5'-dimethoxybiphenyl.
J. Heblý and V. Letovský (Biochem. Research Inst.,
Prague, Czech.). *Chem. Listy* 49, 401(1961).—In an at-
tempt to reduce [2,4,6-O₂N(H₂N)(MeO)C₆H₃]₂ electro-
lytically in HCONH₂, 2,2'-dinitro-4,4'-diformamido-3,5'-
dimethoxybiphenyl was isolated. M. Hudlický

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CA: 47:11167

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"Derivatives of pyrone, pyridone, and pyridine. IV. Preparation of di-arylidene-2,6-dimethyl-4H-pyran-4-ones and 4(1H)-pyridones."

Chem. List; 46, 736-9(1952); cf. CA 47, 8070c.

MOHYL, J.; STANEK J.; ZVERINA, V.

Reactivity of the methyl group on the heterocyclic ring. III Methiodide of
2-(phenyl-hydroxymethyl)pyridine and its reactions; synthesis of
d, l-sedamine. p.735 (Chemické Listy, Praha. Vol. 46, No. 12, Dec. 1952)

SC: Monthly List of East European Associations, (ESAL), 13, Vol. 1, No. 6,
June 1955, Encl.

MINY, I.; RADEK, C.; URBAN, J.

Derivatives of pyrene, pyridone, and pyridine. IV. Preparation of some N-ary-
lideno-2,6-dimethyl- pyrenes and -pyridones. p.736 (Chemické Listy, Praha,
Vol. 46, No. 12, Dec. 1952)
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June 1955, Uncl.

STANEK, J.; HEBKY, J.; ZVERINA, V.

Reactivity of methyl groups on heterocyclic nuclei. Part 3. The methiodide of 2- (β -phenyl- β -hydroxyethyl) pyridine and its reactions; a synthesis of dl-sedamine [with summary in English]. Sbor.Chekh.khim.rab. 18 no.5: 679-683 0 '53. (MLRA 7:6)

1. Department of Organic Chemistry, Charles University, and Pharmaceutical and Biochemical Research Institute, Prague. (Heterocyclic compounds)
(Pyridine) (Piperidine)

Hebky, J.

4 Contribution to the synthesis of lobeline. J. Hebky and
J. Kuba (Farm. biochem. ústav, Praha, Czech. Chem. Listy 47, 1952, 1117-1121). A mixture of
lobeline (I), lobelanine (II), and lobelanidine (III) obtained
by catalytic hydrogenation of 2,6-diphenylpyridine p-
toluenesulfonate with 5 moles H₂ over PtO₂ was sepd.
by chromatography on Al₂O₃ (activity I). Similarly, the
mixture obtained from 22 g. II with 1.5 g. LiAlH₄ in tetra-
hydrofuran at 0° was chromatographed to yield II, m. 98°,
HCl salt, m. 193° (decomp.), and III. I, m. 100°, gave
with (+)-dibromosuccinic acid a salt, m. 142°, free base,
m. 148°; HCl salt, m. 150° (decomp.), [α]_D²⁰ -57.2°.
III, m. 146°; HCl salt, m. 138°; sulfate, m. 101°.
M. Baudry