

GUNDLACH, W.L.R.

Conference on measuring techniques in thermal and flow  
machines. Przegł mech 21 no.15:484 10 Ag '62.

L 17745-63  
 EPA/EWT(d)/EWT(l)/EWF(a)/EPF(n)-2/EDS/HH(r)  
 AEDC/AFPC/ASD/APCC/SSD ~~Pa-4/Pa-4/Pa-4~~ WW/JD/JG  
 P/0008/63/000/007/0166/0168 82-79

ACCESSION NR: AP3007184

AUTHOR: Gundlach, W. R. (Professor, Doctor, Engineer); Przybylski, R. (Mgr. Engineer)

TITLE: Research problem in the field of turbine engines 23

SOURCE: Technika lotnicza, no. 7, 1963, 166-168

TOPIC TAGS: turbine engine research, turbine engine, aircraft turbine engine, turbine engine test installation, aircraft institute, turbine research

ABSTRACT: Research programs on gas flow, combustion, heat transfer, and dynamic phenomena in small and medium turboengines have been planned by the Katedra Ciepłych Maszyn Przepływowych (Department of Thermal-Flow Machines), Lodz Polytechnic Institute. Test stands equipped with straight-flow tubular combustion chambers, reversible-flow combustion chambers, fuel injectors, axial and radial compression stages and expansion stages, and regenerative heat exchangers will be used. At the present time, satisfactory conditions exist only for research into aircraft engines.

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L 17745-63

ACCESSION NR: AP3007184

The facilities of the Instytut Lotnictwa (Aircraft Institute), industrial research units, and laboratories of institutions of higher education are inadequate for solving certain problems related to refractory materials, high-speed bearings, and instrumentation.

ASSOCIATION: Politechnika Lodzka (Lodz Polytechnic Institute)

SUBMITTED: 00

DATE ACQ: 030ct63

ENCL: 00

SUB CODE: PR

NO REF SOV: 000

OTHER: 000

Card 2/2

GUNDLACH, Wladyslaw R., prof. dr inż.

Course of developing the measuring unit systems. Ciepłota przepływu  
no.47/48:7-38 '63.

WIERZONKA, Benedykt, dr. hab.; KAMONOWSKI, J.; PRZYBYLSKI, R.  
PRZYBYLSKI, R.

Abstracts of publications. Siepl masz przepl/w no.51/52:93-  
104 '64.

GUNDLACH, Wladyslaw, prof. dr inz.

The State Council for Problems of Fuel and Power Management.  
Ciepl masz przeplyw no.53:45-47 '64.

GUNDLACH, W. R.; PRZYBYLSKI, R.

Correction to W. Kordzinski's article. Techn lotn 19  
no. 5:131 My '64.

1. GUNDOREH, A.A.

2. USSR (600)

4. Fishing Boats

7. Ways to increase the speed and lower the cost of fishing boat construction, *Itzh. khoz.* 29 no. 3, 1953.

9. Monthly List of Russian Accessions, Library of Congress, APRIL 1953, Uncl.



GUNDOBIN, A.A., inzhener.

Firsov's completed diagram. Sudostroenie 22 no.2:1-4 F 156.  
(Shipbuilding)(Stability of ships) (MLRA 9:7)

GUNDOBIN, A.A., inzh.

At the industrial exhibition of the Maritime Province. Sudostroenie 25  
no.2:57-58 F '59, (MIRA 12:4)  
(Ussuriisk--Ships--Maintenance and repair--Exhibitions)

GUNDOBIN, A.A.; PETRISHCHEV, K.F.

Standardization of corrugated ship bulkheads. Standartizatsia  
24 no.6:32-34 Je '60. (MIRA 13:7)  
(Bulkheads (Naval architecture)--Standards)

GUNDOBIN, A.A., inzh.

Some methods for determining the outline of a ships hull from the  
ship itself. Sudostroenie 26 no.9:55-56 S'60. (MIRA 13:10)  
(Ships--Maintenance and repair)

GUNDUBIN, A.A., inzh.; FETRISHCHEV, K.F., inzh.; KREYMER, I.D., inzh.

Testing of corrugated bulkheads. Sudostroenie 26 no.10:68-70  
0'60. (MIRA 13:10)  
(Bulkheads (Naval architecture))

GUNDOBIN, A.A., inzh.

One hundredth anniversary of Vladivostok. Sudostroenie 26  
no. 11:81-82 N '60. (MIRA 14:1)  
(Vladivostok)

GUNDOBIN, Anatoliy Andrianovich; CHASHKOV, Mikhail Timofeyevich; KHANOVICH, I.G., nauchnyy red.; KLITORINA, T.A., red.; TSAL, R.K., tekhn. red.

[Improving the stability of ships being re-equipped] Uluchshenie ostoichivosti pereoborudemykh sudov. Leningrad, Gos. soiuznoe izd-vo sudostroitel. promyshl., 1961. 69 p. (MIRA 14:6)  
(Stability of ships)

GUNDOBIN, A.A., inzh.

Fortieth anniversary of the use of electric welding at the "Dal'zavod."  
Svar. proizv. no.4:42-43 Ap '61. (MIRA 14:3)  
(Vladivostok--Ships--Welding)



GUNDOBIN, A.A., inzh.; EYDEL'KIND, L.Sh., inzh.

Plans for welded structures for the repair of riveted ships. Sudo-  
stroenie 27 no.12:52-55 D '61. (MIRA 15:1)  
(Ships--Maintenance and repair)

GUNDOBIN, A.A., inzh.; KREYMER, I.D., inzh.

Operation of pontoon dry dock gates. Sudostroenie 28 no.4:  
65-66 Ap '62. (MIRA 15:4)

(Sluice gates)

GUNDOBIN, A.A., inzh.

Icebreaker "Sibir'" in the Arctic. Sudostroenie 28 no.4:81  
Ap '62. (MIRA 15:4)  
(Arctic regions--Ice-breaking vessels)

GUNDOBIN, A.A., inzh.

Providing for the stability of ships being modernized and re-equipped.  
Sudostroenie 29 no.1:62-64 Ja '63. (MIRA 16-3)  
(Stability of ships) (Ships--Maintenance and repair)

GUNDOBIN, N.A.

Railroad transportation in the sixth five-year plan; discussion  
with N.A. Gundobin, Assistant Minister of Transportation of the  
U.S.S.R. Voen.znan. 31 no.7:3 J1 '56. (MLRA 10:8)

1. Ministr putey soobshcheniya SSSR.  
(Railroads)

GUNDOBIN, N.A.

Speeding up the turnover of railroad cars is an important condition for increases in haulage. Zhel.dor.transp. 37 no.5:9-15 My '56.  
(MLRA 9:8)

1. Zamestitel' Ministra putey soobshcheniya.  
(Railroads--Cars)

GUNDOBIN, N.A., red.; GUBAREVA, N.T., red.; VERINA, G.P., tekhn. red.

[Transportation and economic councils] Transport i sovnarkhozy.  
Moskva, Gos. transp. zhgl-dor. idz-vo, 1958. 294 p. (MIRA 11:9)  
(Transportation)

*GUNDOBIN, N.A.*  
GUNDOBIN, N.A.

Advance work methods for all railroad dispatchers. Zhel. dor.  
transp. 40 no.1:8-11 Ja '58. (MIRA 11:1)

1. Zamestitel' ministra putey soobshcheniya.  
(Railroads--Train dispatching)



GUNDOBIN, N.A.

Improve the creative activity of engineers and technical workers.  
Zhel. dor. transp. 40 no.8:3-8 Ag '58. (MIRA 11:9)

1. Zamestitel' ministra putey soobshcheniya, predsedatel' Nauchno-  
tekhnicheskogo soveta Ministerstva putey soobshcheniya.  
(Railroad engineering)

GUNDOBIN, N. A.

Urgent program for speeding up technical progress in railroad transportation. Zhel.dor.transp. 42 no.8:3-9 Ag '60.  
(MIRA 13:8)

1. Zamestitel' ministra putey soobshcheniya.  
(Railroads)

GUNDOBIN, N. A.

Toward new heights of technological progress in transportation. Zhel. dor. transp. 45 no.1:3-9 JA '63.  
(MIRA 16:4)

1. Pervyy samestitel' ministra putey soobshcheniya.

(Railroads)

GUNDOBINA, T.K.

Antibiotic treatment for diseases in children caused by yeastlike fungi. *Pediatrics* 39 no.3:49-52 My-Je '56. (MLRA 9:9)

1. Iz Leningradskoy detskoy bol'nitsy imeni N.K.Krupskoy (glavnyy vrach A.I.Chezina, nauchnyy rukovoditel' - prof. N.A.Shalkov)  
(FUNGUS DISEASES, etiol. and pathogen, antibiotics, in child.)  
(ANTIBIOTICS, inj. eff. fungus dis. in child)

PETROV, A.P., doktor tekhn. nauk, prof.; TULUPOV, L.P., kand. tekhn. nauk; KRYUKOV, N.D., kand. tekhn.nauk; GUNDOBIN, V.N., inzh.; VASIL'YEV, G.S., kand. tekhn. nauk; ~~GR...~~, kand. tekhn. nauk; MORZOVA, K.N., inzh.; ROZE, V.A., inzh.; LEVSHIN, G.L., inzh.; BERNGARD, K.A., doktor tekhn. nauk, prof.; BIKHENTAY, M.A., inzh.; BUYANOV, V.A., inzh.; ILOVAYSKIY, N.D., inzh.; MUKHAMEDOV, G.A., kand. tekhn.nauk; MIROSHNICHENKO, A.P., inzh.; ANDRIANOV, V.P., inzh.; BUTS, V.D., inzh.; KAZIMOV, A.A., inzh.; KIREYEV, O.P., inzh.; DYUFUR, S.L., kand. tekhn. nauk; USTINSKIY, A.A., kand. tekhn. nauk; MIKHAYLOV, S.M., inzh.; NESTEROV, Ye.P., kand. tekhn. nauk, retsenzent; LIVSHITS, V.N., inzh., retsenzent; PREDE, V.Yu., inzh., red.; VOROTNIKOVA, L.F., tekhn. red.

[Control of transportation processes using electronic digital computers] Upravlenie perevozhnym protsessom s primeneniem elektronnykh tsifrovyykh vychislitel'nykh mashin. Pod obshchei red. A.P.Petrova. Moskva, Transzheldorizdat, 1963. 207 p.

(MIRA 16:8)

1. Chlen-korrespondent AN SSSR (for Petrov).  
(Railroads--Management) (Electronic digital computers)

L 53789-65

ACCESSION NR: AP5009874

UR/0115/65/000/002/0035/0036  
536.5:621.038

AUTHOR: Fomin, Yu. Ya.; Gundorin, B. K.

TITLE: Measuring the rapidly-changing temperature of fuel under high-pressure conditions

SOURCE: Izmeritel'naya tekhnika, no. 2, 1965, 35-36

TOPIC TAGS: diesel engine, diesel fuel

ABSTRACT: The measurement of the temperature of the fuel in a diesel-engine fuel injector is described. The constant temperature component was measured by a copper-constantan 0.03-0.06-mm thermocouple having a thermal inertia of 15-20 msec, and the variable component, by a 0.015-mm tungsten resistance thermometer with a thermal inertia of 4-6 msec. An oscillogram shows the entire injection process: lifting the injector needle, fuel pressure, and temperatures. The resistance thermometer reading was behind that of the pressure by only 1-2 msec. Orig. art. has: 3 figures.

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L 53789-65

ACCESSION NR: AP5009874

ASSOCIATION: none

SUBMITTED: 00

ENCL: 00

SUB CODE: PR, IE

NO REF SOV: 000

OTHER: 000

*Am*  
Card 2/2

GUNDOROVA, R. A.

"Novocaine in the treatment of glaucoma." Second Moscow State Medical  
Inst imeni I. V. Stalin. Moscow, 1959. (Dissertations for the Degree  
of Candidate in Medical Science)

So: Knizhaya letopis', No. 16, 1956



GUNDOROVA, R.A.

Injury of the eye caused by fuzz from a live caterpillar. Vest.oft.  
69 no.5:39-40 S-0 '56. (MIRA 9:12)

1. Iz Glaznoy kliniki (dir. - prof. N.A.Pletneva) II Moskovskogo  
meditsinskogo instituta  
(**SYE**, wounds and injuries  
caused by contact with fuzz of living caterpillar)

GUNDOROVA, R.A., kand.med.nauk

Method of photocoagulation in ophthalmological practice; survey of foreign literature. Vest.oft. 72 no.5:53-57 S-O '59. (MIRA 13:3)

1. Moskovskaya glasnaya klinicheskaya bol'nitsa (nauchnyy rukovoditel'-zasluzhennyy deyatel' nauki prof. M.L. Krasnov).  
(EYE DISEASES, surgery)  
(DIATHERMY)

GUNDOROVA, R.A.; SHCHEKINA, A.N.; KORTIKOVA, Ye.A.

Intermedin in the treatment of complicated myopia and pigmentary degeneration of the retina. Vest. oft. 73 no. 4:37-38 J1-Ag '60.  
(MIRA 14:1)

(PITUITARY BODY—SECRETIONS) (MYOPIA)  
(RETINA—DISEASES)

GUNDOROVA, R.A., kand.med.nauk

Gonioscopy in diseases of the vascular tract. Oft. zhur. 16 no.3:  
179-181 '61. (MIRA 14:5)

1. Iz Moskovskoy glaznoy klinicheskoy bol'nitsy (nauchnyy rukovoditel'  
zasluzhennyy deyatel' nauki prof. M.L.Krasnov).  
(GONIOSCOPY) (EYE—DISEASES AND DEFECTS)

SKEP'YAN, N.A., vrach; MELOKS, T.S., vrach; SIDEL'NIKOVA, T.Ya., kand.  
med.nauk; GUNDOROVA, R.A., kand.med.nauk; KRISTMAN, V.I., kand.  
med.nauk; GUSAROVA, A.S., kand.med.nauk; MARSHAK, M.S., prof.

How to keep well. Zdorov'e 8 no.12:28-29 D '62. (MIRA 16:1)  
(HYGIENE)

GUNDOROVA, R.A.

Study of the crystalline lens in infrared light. Uzh.zap.  
GNIi glaz. bol. no.8:40-43'63. (MIRA 16:9)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut glaznykh  
bolezney imeni Gel'mgol'tsa.  
(INFRARED RAYS) (CRYSTALLINE LENS)

GUNDOROVA, R.A.; FRIDMAN, F.Ye.; MOROZOV, V.I.; POLYAKOVA, L.Ya.

Contemporary methods of treating traumatic cataracts. Uch.  
zap. GNII glaz.bol. no.8:113-118'63. (MIRA 16:9)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut glaz-  
nykh bolezney imeni Gel'mgol'tsa.  
(CATARACT) (EYE--WOUNDS AND INJURIES)

GUMDOROVA, R.A.; MOROZOV, V.I.; FRIDMAN, F.Ye.; POLYAKOVA, L.Ya.

Use of various means of removing the crystalline lens in  
the intracapsular extraction of a cataract. Uch. zap.  
GNII glaz.bol. no.8:119-127'63. (MIRA 16:9)

1. Gosudarstvennyy nauchno-issledovatel'skiy institut glaz-  
nykh bolezney imeni Gel'mgol'tsa.  
(CATARACT) (CRYSTALLINE LENS)



GUNDOROVA, R.A., kand.med. nauk (Moskva)

Surgical treatment of retinal detachment. Med. sestra 22 no.8:  
39-43 Ag'63. (MIRA 16:10)  
(RETINA--DISEASES) (EYE--SURGERY)

GUNDOROVA, R.A., kand. med. nauk

Surgical treatment of traumatic retinal detachment. Vest.  
oft. 76 no.3:11-17 My-Je '63. (MIRA 17:2)

1. Nauchno-issledovatel'skiy institut glaznykh bolezney  
imeni Gel'mgol'tsa (dir. A.V. Roslavytsev).

GUNDOROVA, R.A., kand. med. nauk; GASIMOV, V.G., aspirant

Analysis of the outcome of mechanical eye injuries due to  
foreign bodies for a period of 8 years. Azerb. med. zhur.  
42 no.8:55-62 Ag '65. (MIRA 18:11)

1. Iz gosudarstvennogo nauchno-issledovatel'skogo instituta  
glaznykh bolezney imeni Gel'mgol'tsa (dir. - A.V. Roslavytsev)  
(rukovoditel' - starshiy nauchnyy sotrudnik, kand. med. nauk  
Ye.S. Vaynshteyn).

GUNDORTSEV, I.V.

Organizations and principles of drilling operations in fields  
of the Al'met'yevsk Petroleum Industry Drilling Trust. Neft.  
khoz. 39 no.3:16-23 Mr '61. (MIRA 16:7)

(Al'met'yevsk region(Tatar A.S.S.R.)--Oil well drilling)

AND IZET, A.M.

Ioganzhen, I.G. and Guadrizer, A.M. "The NaCl concentration factor for certain  
fresh-water fish of Siberia," Ichen. zapiski (Izskiy oos. un-ta Ir. Kuyaysheva),  
No. 11, 1948 p. 27-39, - Bibliog: p. 31-39

SO: U3261, 10 April 53, (Letopis'zhurnal 'nykh Statey, No. 12, 1949)

GUNDRIZIER, A. N.

Session in Memory of Academician K. M. Ber.

The author reports on the session of the Tomsk Department (otdeleniye) of the Moscow Society of Naturalists and Tomsk Division (otdel) of the Geographical Society of the USSR, which was devoted to the 75th anniversary of the death of Karl Maksimovich Ber, who was an outstanding geographer, zoologist, and anthropologist. (RZhGeol, No. 4, 1955) Vopr. geografii Sibiri, No. 3, 1953, 324.

SO: Sum. No. 744, 8 Dec 55 - Supplementary Survey of Soviet Scientific Abstracts (17)

GUNDRIZER, A. N.

"The Ide of Western Siberia." Cand Biol Sci, Tomsk State U imeni V. V. Kuybyshev,  
Toms, 1955. (KL, No 17, Apr 55)

SO: Sum. No. 704, 2 Nov 55 - Survey of Scientific and Technical Dissertations Defended  
at USSR Higher Educational Institutions (16).

GUNDRIZER, A.N.

Hybrids of *Leuciscus idus* and the Siberian roach from Lake Chany.  
Zam. po faune i flore Sib. no.18:25-26 '55. (MIRA 11:1)

1. Laboratoriya ikhtiologii i gidrobiologii Tomskogo gosudarstvennogo  
universiteta imeni V.V. Kuybysheva.  
(Chany, Lake--Fishes)



IOGANZEN, B.G.; GUNDRIZER, A.N.

The technique of estimating stocks of migratory fishes and the degree of their exploitation under river conditions. Trudy sov. Ikht. kom. no.13:457-465 '61. (MIRA 14:8)

1. Tomskiy gosudarstvennyy universitet imeni V.V. Kuybysheva. (Ob' River--Fisheries)

GENERAL: 1200

"Problems of conservation in Western Siberia." Reviewed by  
S.N. Gurevizer. Okhr. priro. "Izv. 1 Dal'. Vest. no.1, 264-  
267, 1962. (MIRA 17:5)

GUNDREZER, A.N.

Biology of the whitefish *Coregonus pravdinellus* Dulkeit in  
Lake Teletskoye and the Biya River. Izv. Sib. otd. AN SSSR  
no.3:111-119 '62. (MIRA 17:7)

1. Tomskiy gosudarstvennyy universitet.

GUNDRIZER, A.N.

Possibilities of utilizing the lakes of the Gornyy Altai for  
fishing. Izv. Alt. otd. Geog. ob-va SSSR no.5:203-206 '65.  
(MIRA 18:12)

1. Tomskiy gosudarstvennyy universitet.

IOGANZEN, B.G.; GUNDRIZER, A.N.; KAFANOVA, V.V.; KRIVOSHCHENKOV, G.M.

Lake Teletskoye as a unique body of water of the Altai and  
an object deserving protection. Izv. Alt. otd. Geog. ob-va  
SSSR no.5:216-217 '65. (MIRA 18:12)

1. Tomskiy gosudarstvennyy universitet.

SKIDANOVA, N.I.; GUNDYREV, A.A.; CHERNOZHUKOV, N.I.

Solubility in furfurole of aromatic hydrocarbons isolated  
from the oil fractions of the Kotur-Tepe petroleum. Izv.vys.  
ucheb.zav.; neft' i gaz 5 no.2:59-65 '62. (MIRA 15:7)

1. Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti  
imeni akademika I.M. Gubkina. (Furaldehyde)  
(Hydrocarbons)  
(Kotur-Tepe region--Petroleum--Refining)

AUTHORS: ~~Gundvray, A.A.~~ Nametkin, N.S., SOV/20-121-6-22/45  
Topchiyev, A.V., Member, Academy of Sciences, USSR

TITLE: Dipole Moments of the Hexalkyl Derivatives of Disiloxane  
(Dipol'nyye momenty geksalkilproizvodnykh disiloksana)

PERIODICAL: Doklady Akademii nauk SSSR, 1958, Vol 121, Nr 6, pp 1031 - 1033  
(USSR)

ABSTRACT: In the present paper the values of the dipole moments of the 4 derivatives referred to in the title: a) hexa-methyl-disiloxane b) hexa-ethyl-disiloxane c) hexa-propyl-disiloxane and d) hexa-butyl-disiloxane are quoted. They were produced by hydrolysis of the corresponding tri-alkyl-haloid-(ethoxy)-silanes (Ref 1). Only the dipole moment for a) was known (Refs 2,3). The others have been determined here for the first time. Results and references are given in table 1. A comparison of these results shows a very high precision of the used apparatus. This equipment worked according to the pulsation method. Its stabilization consisted of piezoquartz with a frequency of  $7,95 \cdot 10^5$  cycles. The dipole moments were computed by the formula:

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$$\mu = 0,012813 \cdot 10^{-18} \sqrt{P_{or} T} \quad (1)$$

Dipole Moments of the Hexalkyl Derivatives of  
Disiloxane

SOV/20-121-6-22/45

according to the method by Hederstrand (Gederstrand, Ref 7).  
 $\mu$  means the dipole moment,  $P_{or}$  an orientational polarization,  
and  $T$  the absolute temperature. The orientational polarization is  
associated with the general polarization of the dissolved com-  
pound  $P_{\infty}$  and with the deformation polarization  $P_{D\infty}$  (supposed be-  
ing identical with the molar refraction) by the ratio

$$P_{or} = P_{\infty} - P_{D\infty} \quad (2)$$

Table 2 shows the constants of the pure solvent for the deter-  
mination of  $P_{\infty}$  and  $P_{D\infty}$ . Table 3 contains the determination of  
dielectric permeability ( $\epsilon$ ), density ( $d$ ), and refractive index  
( $n_D$ ) of the substances in question, dissolved in n-hexane, for  
various concentrations of the dissolved compounds. They are ex-  
pressed in mol-ports (molarity) ( $C_2$ ) (molyarnyye doli Pl.).

From the data of table 2 and 3, the constants for the determina-  
tion of the dipole moments according to the Hederstrand method  
are calculated (Table 4). This evaluation of the dipole moments  
proves the Si-O-bond being a distinctly polar one. This is due

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Dipole Moments of the Hexalkyl Derivatives of  
Disiloxane

SOV/20-121-6-22/45

to the significant difference of the electronegativity of Si  
and O. There are 4 tables and 7 references.

SUBMITTED: May 13, 1958

Card 3/3

GUNDYREV, A.A.

A.V. Topchiyev, G.M. Fanchenkov, N.S. Mametkin, A.A. Gundyrev and Ku Ch'ang-li,  
"Temperature Dependence of the Viscosity and Density of Certain Silicon-  
Organic Compounds.

Report presented at the Second All-Union Conference on the Chemistry and  
Practical Application of Silicon-Organic Compounds held in Leningrad from  
25-27 September 1958.  
Zhurnal prikladnoy khimii, 1959, pp 238-240 (USSR)

68168

5(4) 5.3700(B)

SOV/20-129-6-35/69

AUTHORS: Gundyrev, A. A., Nametkin, N. S., Panchenkov, G. M.,  
Topchiyev, A. V., Academician

TITLE: The Dielectric Constants<sup>1</sup> and the Dipole Moments of Some Organo-  
silicon Compounds

PERIODICAL: Doklady Akademii nauk SSSR, 1959, Vol 129, Nr 6, pp 1325-1327  
(USSR)

ABSTRACT: The authors determined the dielectric constants of 15 organo-  
silicon compounds at a frequency of  $7.95 \cdot 10^5$  cps: hexaalkyl  
derivatives of disilane methane, disilane ethane, disiloxane  
and linear polyethyl siloxanes. The values determined at  
 $25 \pm 0.05^\circ$  are given in table 1. (Abstracter's note: This table  
has erroneously been printed under the heading "dipole moments"  
instead of "dielectric constants"). Within the series of hexa-  
alkyl derivatives of disilane methane and disilane ethane the  
dielectric constant in each series increases from the hexamethyl-  
to the hexabutylene derivatives, the greatest change occurring  
in transition from the hexamethyl- to the hexaethyl derivative.  
The dielectric constants of the hexaalkyl derivatives of di-  
siloxane vary anomalously with an increase in the molecular  
weight of the compounds. From the densities mentioned in table  
2, refraction indices and dielectric constants of the compounds  
with different concentrations dissolved in n-hexane, the dipole

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SOV/20-129-6-35/69

The Dielectric Constants and the Dipole Moments of Some Organosilicon Compounds

moments were calculated according to G. Hedestrand (Ref 4) (Table 3). Hitherto, only the data of hexamethyl- and hexaethyl disiloxane had been known. The values found by the authors for these two compounds agree with the data published in literature. The nearly equal values of the dipole moments for the derivatives of disilane methane and disilane ethane show that they are not influenced by the length of the chain and the nature of the alkyl molecules. In disiloxane derivatives the dipole moment decreases with an increase in the molecular weight of the alkyl radicals, whereas in the linear polyethyl siloxanes  $(C_2H_5)_3SiO[(C_2H_5)_2SiO]_nSi(C_2H_5)_3$  the dipole moments increase with an increase in molecular weight and follow the empirical equation:  $\mu = 0.63 \sqrt{n + 1}$  ( $\mu$  = dipole moment,  $n$  = number of oxygen atoms). There are 3 tables and 7 references, 2 of which are Soviet.

ASSOCIATION: Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti  
I. M. Gubkina (Moscow Institute for the Petroleum-chemical  
and Gas Industry I. M. Gubkin)

SUBMITTED: August 10, 1959  
Card 2/2

S/661/61/000/006/057/031  
D267/DJ02

AUTHORS: Gundyrev, A. A., Topchiyev, A. V., Panchenkov, G. H.,  
Kametkin, N. S. and Ku Ch'ang-ling

TITLE: Dependence of the viscosity and density of some classes  
of organosilicon compounds on temperature, and the relat-  
ion between the interaction energies of molecules of  
these compounds and their structure

SOURCE: Khimiya i prakticheskoye primeneniye kremneorganicheskikh  
soyedineniy; trudy konferentsii, no. 6: Doklady, diskus-  
sii, resheniye. II Vses. konfer. po khimii i prakt. prim.  
kremneorg. soyed., Len. 1958. Leningrad, Izd-vo, AN SSSR,  
1961, 239-240

TEXT: A discussion relating to a minor detail of the above paper  
(this publication, no. 3, p. 80), in which Ya. I. Vabel' (Moscow)  
took part. One of the authors stated that viscosity of mixtures of  
polysiloxane liquids had not been calculated, and that viscosities  
at temperatures below 0°C had not been measured.

Card 1/2



Dependence of the viscosity ...

S/661/61/000/006/057/081  
D267/D302

ASSOCIATION: Institut neftekhimicheskogo sinteza Akademii nauk  
SSSR, Moskva (Institute of Petrochemical Synthesis,  
Academy of Sciences, USSR, Moscow)

✓

Card 2/2

S/832/62/000/000/012/015  
D424/D307

AUTHORS:

Alexandrova, Z.A., Gundyrev, A.A., Nametkin,  
N.S., Panchenkov, G.M. and Topchiyev, A.V.

TITLE:

Surface tension of organosilicon compounds  
of a number of classes

SOURCE:

Issledovaniya v oblasti krmeniyorganiches-  
kikh soyedineniy; sintez i fiziko-khimiches-  
kiye svoystva. Sbornik statey. Inst. neftekhim.  
sint. AN SSSR. Moscow, Izd-vo AN SSSR, 1962,  
219 - 226

TEXT:

The investigation was designed to supply  
information in the little-studied field of the dependence of  
the surface tension of organosilicon compounds on the tempera-  
ture. The surface tension at an air boundary of 39 organosili-  
con compounds of 12 different classes was measured over the  
range from 20 to 50°C, at 10° intervals. The measurements were  
carried out in an air-bubble viscometer as proposed by Sugden

✓

Card 1/3

Surface tension ...

S/832/62/000/000/012/015  
D424/D307

(J.Chem.Soc., 121, 857 (1922) ) [Abstractor's note: reference corrected, improved by Quale and Smart (J.Amer.Chem.Soc., 66, 935 (1944) ), and further improved by the present authors. The constant of the viscometer was determined with benzene and checked with n-heptane and water. The following classes of compounds were investigated: hexaalkyldisilylmethanes, -ethanes, and propanes, hexaalkyldisiloxanes, phenyl-, 4-methylphenyl- and 3,4-dimethylphenyltrialkylsilanes, diphenyl-, bis-4-methylphenyl-, and bis-3,4-dimethylphenyldialkylsilanes, and linear and cyclic polyethoxysiloxanes. In all cases the surface tensions were found to obey the linear relation

$$\sigma_t = a - bt$$

where  $\sigma_t$  is the surface tension in dynes/cm at  $t^\circ\text{C}$ , and  $a$  and  $b$  are constants for a given compound. The surface tensions of hexaalkyldisiloxanes are 2-3 dyne/cm less than those of the corresponding hexaalkyldisilylmethanes or -ethanes. The transition

Card 2/3

Surface tension ...

S/832/62/000/000/012/015  
D424/D307

from hexamethyldisilylmethane to the corresponding disilylethane and disilylpropane derivatives is accompanied by a regular increase in  $\sigma_t$ . Lengthening the polyethylsiloxane chain by successive diethylsiloxo units is also accompanied by a regular increase in  $\sigma_t$ . In addition, the following pairs of analogous compounds were investigated: diethylsilane-n-pentane, dipropylsilane-n-heptane, dibutylsilane-n-nonane, phenylmethylsilane-ethylbenzene, and phenyldimethylsilane-iso-propylbenzene. The surface tensions of the alkylsilanes were close to those of the corresponding hydrocarbons, while the surface tensions of the alkylarylsilanes were 1.5-2 dyne/cm higher than those of the corresponding hydrocarbons. There are 5 tables.

Card 3/3



S/832/62/000/000/013/015  
D424/D307

AUTHORS: Gar, P.P., Gundyrev, A.A., Nametkin, N.S.,  
Panchenkov, G.M. and Topchiyev, A.V.

TITLE: Refractometric investigations of some  
organosilicon compounds

SOURCE: Issledovaniya v oblasti kremniyorganiches-  
kikh soyedineniy; sintez i fiziko-khimiches-  
kiye svoystva. Sbornik statey. Inst. neftek-  
him. sint. AN SSSR. Moscow. Izd-vo AN SSSR,  
1962, 228 - 234

TEXT: The refractive indices at 20°C for the C and F  
lines of hydrogen (for the first time) and for the D line of  
sodium have been measured for 46 compounds of the following classes:  
the hexaalkyl derivatives of disilylmethane, disilylethane, and  
disilylpropane, and of disiloxane, aryltrialkylsilanes, diaryl-  
dialkylsilanes, and linear and cyclic polyethoxysilanes. From the  
results, the individual, mean and specific dispersions, and the

Card 1/3

Refractometric investigations ... S/832/62/000/000/013/015  
D424/D307

molecular refraction for the D line, have been calculated. As a rule, in any one class of alkylarylsilanes those in which the alkyl group is ethyl have the highest refractive index. The refractive indices of dialkyldiarylsilanes are about 0.05 units higher than those of the aryltrialkylsilanes with the same alkyl and aryl groups. In the case of the phenyltrialkylsilanes, the introduction of a methyl group into the 4- position of the phenyl ring and then a second methyl group into the 3- position increases the refractive index, while the situation is reversed in the case of the aryltrialkylsilanes. The specific dispersions of aryltrialkylsilanes are 15 - 20 % lower than those of the corresponding diaryldialkylsilanes. The refractive indices of the hexaalkyl derivatives of disilylmethane, disilylethane, and disiloxane increase with a rise in the molecular weight, those of the disiloxanes being less than those of the corresponding members of the other classes. On passing from hexamethyldisilylmethane to the corresponding ethane and propane derivatives, the refractive index increases by 0.0025 units per CH<sub>2</sub> group added, while the specific dispersion is decreased. The change in the specific

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Refractometric investigations ...

S/832/62/000/000/013/015  
D424/D307

dispersion of the hexaalkyldisiloxanes is anomalous, the ethyl compound having the lowest value. The refractive indices of polyethylsiloxanes increase by 0.007 per diethylsiloxy group and are much higher than those of the corresponding polymethyl compounds. Dialkylsilanes have much higher refractive indices and specific dispersions than the corresponding hydrocarbons. The molecular refractions of all the compounds investigated were calculated by Sauer's method (J.Amer.Chem.Soc., 68, 691 (1946) ) from established group and bond values and the results were found to be in good agreement with those calculated from the experimental data. There are 3 tables.

Card 3/3

S/832/62/000/000/014/015  
D424/D307

AUTHORS: Gundyrev, A.A., Nametkin, N.S., Panchenkov,  
G.M. and Topchiyev, A.V.

TITLE: Dielectric constants and dipole moments of  
some organosilicon compounds

SOURCE: Issledovaniya v oblasti kremniyorgani-  
cheskikh soyedineniy; sintez i fiziko-  
khimicheskiye svoystva. Sbornik statey.  
Inst. neftekhim. sint. AN SSSR. Moscow.  
Izd-vo AN SSSR, 1962, 235 - 242

TEXT: The dielectric constants of 16 organosilicon  
compounds of the following types have been investigated: hexa-  
alkyl derivatives of disilylmethane, disilylethane, and disilox-  
ane, and polyethylsiloxanes, most of them for the first time.  
The dielectric constants were measured at 25° in n-hexane at  
a frequency of  $7.95 \times 10^5$  cps by the method of beats, with an  
accuracy of 0.02 %. The apparatus was calibrated with benzene

Card 1/2

Dielectric constants ...

S/832/62/000/000/014/015  
D424/D307

and checked on toluene and *n*-hexane. The value of the dielectric constants given in the literature for a few of the compounds (measured in other solvents) agreed well with those obtained in the present work. In the hexaalkyldisilylmethanes and -ethanes, the dielectric constant increases with the molecular weight, while in the hexaalkylsiloxanes the ethyl compound has the highest constant. From the figures for the dielectric constant, the density, and the refractive index in dilute *n*-hexane solution, the dipole moments of all the above compounds were calculated by Hedestrand's method (Z.phys. Chem., 2, 429 (1929)). All the hexaalkyl derivatives of disilylmethane and disilylethane have practically the same dipole moment of about 0.57 D. In the case of the hexaalkyldisiloxanes, the methyl compound has the highest dipole moment of those investigated. The dipole moments of the polyethylsiloxanes increase regularly as the molecular weight increases, in accordance with the relation

$$\mu = 0.63 \sqrt{n + 1},$$

where  $n+1$  is the number of oxygen atoms in the molecule of the polyethylsiloxane. There are 5 tables.  
Card 2/2

S/832/62/000/000/011/015  
D424/D307

AUTHORS:

Gundyrev, A.A., Nametkin, N.S., Panchenkov,  
G.M. and Topchiyev, A.V.

TITLE:

Viscosity of some liquid organosilicon  
compounds and the energy of interaction  
of the molecules of these compounds

SOURCE:

Issledovaniya v oblasti kremniyorganiches-  
kikh soyedineniy; sintez i fiziko-khimiches-  
kiye svoystva. Sbornik statey. Inst. neftekhim.  
sint. AN SSSR. Moscow. Izd-vo AN SSSR, 1962,  
205 - 218

TEXT:  
With the object of establishing general laws relating their physical properties to their structure, the densities and viscosities of 43 organosilicon compounds have been measured over the temperature range from 10 to 85°C, at 15° intervals, measurements for three of them being carried out also between -60° and 0°C (at 20° intervals). The compounds are:

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Viscosity of some ...

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hexalkyl derivatives of disilylmethane, -ethane, and propane, and of disiloxane, trialkyl derivatives of phenyl-, 4-methylphenyl-, and 3,4-dimethylphenylsilanes, and dialkyl derivatives of diphenyl-, bis-4-methylphenyl-, and bis-3,4-dimethylphenylsilanes. The densities, measured in a weight dilatometer to an accuracy of 0.0001 g/ml, were found to obey the linear relation

$$d^T = a + bT,$$

where  $d^T$  is the density at absolute temperature  $T$ , and  $a$  and  $b$  are constants. The viscosities,  $\eta$ , measured in a modified Ostwald viscometer to an accuracy of 0.00001 poise, were found to agree well with Panchenkov's formula

$$\eta = Ad^T \frac{4/3}{T} e^{\epsilon_0/RT} (1 - e^{-\epsilon_0/RT})$$

where  $\epsilon_0$  is the energy of bond formation between molecules, per mole,

Card 2/3

Viscosity of some ...

S/832/62/000/000/011/015  
D424/D307

R the universal gas constant, and A a magnitude which can be considered constant over a relatively narrow range of temperature. The formula was used to calculate  $\epsilon_0$ . Replacing an alkyl radical in aryltrialkylsilanes by an aryl radical leads to a considerable increase in  $\eta$  and  $\epsilon_0$ . T - dependence of  $\eta$  of the hexaalkyl derivatives of disiloxane is lower than that of the corresponding derivatives of disilylmethane and disilylethane. In the hexaalkyl derivatives of disilylmethane and disilylethane, the increase in  $\epsilon_0$  as the main chain is increased by one CH<sub>2</sub> group averages 295 cal/mole, as compared with 209 cal/mole for the n-alkanes. The intermolecular bond energies for the polymethylsiloxanes and the polyethylsiloxanes increase with a rise in the molecular weight, those for the former being higher than those for the latter. There are 10 figures and 4 tables.

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13557

S/832/62/000/000/015/015  
D424/D307

E 2700

AUTHORS: Kolesnikov, I.M., Gundyrev, A.A., Nametkin,  
N.S., Panchenkov, G.M. and Topchiyev, A.V.

TITLE: Behavior of some organosilicon compounds in  
the region of the solidification temperature

SOURCE: Issledovaniya v oblasti kremniyorganiches-  
kikh soyedineniy; sintez i fiziko-khimicheskiye  
svoystva. Sbornik statey. Inst. neftekhim. sint.  
AN SSSR. Moscow, Izd-vo AN SSSR, 1962, 243-253

TEXT: The solidification behavior of 31 organosilicon  
compounds has been investigated in order to determine their true  
melting points and purity, and the conditions under which glasses  
are formed. The compounds were of the following types: hexaalkyl  
derivatives of disilylmethane, -ethane, and -propane, and of  
disiloxane, phenyl-, 4-methylphenyl-, and 3, 4-dimethylphenyl-  
trialkylsilanes, diphenyl-, bis-4-methylphenyl-, and bis-3,4-  
-dimethylphenyldialkylsilanes, linear and cyclic polyethylsiloxanes,

Card 1/3



behavior of some ...

S/832/62/000/000/015/015  
D424/D307

dialkylsilanes, and an arylalkylsilane. The main feature of the apparatus used is that the sample was contained in a double-walled glass vessel immersed in liquid nitrogen, the rate of cooling being controlled by the degree of evacuation of the space between the walls. Many of the compounds underwent pronounced supercooling and others did not crystallize but formed glasses. Where a glass was not formed, the m.p. of the material as such and the true m.p. of its main constituent were determined graphically by Rossini's method (J. Res. Nat. Bureau Standards, 32, 197 (1944)). In the series of phenyl-trialkylsilanes, the methyl and propyl compounds supercool only slightly and then crystallize spontaneously, while the ethyl compound crystallizes only after the deeply supercooled liquid is seeded, and the butyl compound forms a glass. Somewhat similar behavior is found with other series of compounds, the following also forming glasses on cooling: 3,4-dimethylphenyltripropyl- and -tributylsilanes, diphenyldiethyl- and -dibutylsilanes, bis-4-methylphenyldimethyl- and -dipropylsilanes, bis-3,4-dimethyldiethylsilane, hexapropyldisilylmethane and -ethane, and the polyethylsiloxanes containing 5, 6, 7 and 8 oxygen atoms. The viscosities

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Behavior of some ...

S/832/62/000/000/015/015  
D424/D307

of the compounds are also given, together with the intermolecular bond energies calculated from the temperature dependence of the viscosities by Pachenkov's formula (Zh. fiz. khim. 24, 1930 (1950)). These show that, in general, compounds with intermolecular bond energies less than about 4000 cal/mole crystallize on cooling, while compounds with a greater intermolecular bond energy form glasses. There are 8 figures and 1 table.

Card 3/3

L 10589-63

EPF(c)/EWT(m)/BDS Pr-l RM/WW

ACCESSION NR: AP3001471

S/0152/63/000/004/0043/0048

58  
59

AUTHOR: Skidanova, N. I.; Gundyrev, A. A.; Chernozhukov, N. I.

TITLE: Solubility of aromatic hydrocarbons (found in oil fractions of Koturtepin petroleum) in furfural and its dependence on the structure of their compounds

SOURCE: IVUZ. Neft' i gaz, no. 4, 1963, 43-48

TOPIC TAGS: Intermolecular bonds, bond energy, solubility of aromatic hydrocarbons, furfural, aromatic hydrocarbons

ABSTRACT: The viscosities and densities of closely-distilled aromatic hydrocarbons and their mixtures were measured in the temperature interval between 50 and 100C, and the intermolecular bond energy was measured on the basis of these values. It was shown that, with an increase of molecular weight of the fraction, the bond energy decreases on account of the increased number of hydrogen atoms in the side chains, and the number of cycles in the aromatic hydrocarbon molecules are simultaneously decreased. The bond energies were calculated for the solutions of various concentrations of one of the closely-distilled fractions in furfural. It was found that the lowest bond energy is present in compounds which are closely related to the compounds with the worst mutual solubility. It was also shown

Card 1/2

7. 10589-53

ACCESSION NR: AP3001471

that the value  $b$ , which is a constant for a given fraction, depends on the structure of mixed hydrocarbons and it is always higher than the value of  $b$  for the original fractions of binary mixtures of aromatic hydrocarbons. The study of solubility of close aromatic hydrocarbon fractions in furfural can be useful in establishing the structure of hydrocarbons present in these fractions. Orig. art. has: 3 tables and 2 graphs.

ASSOCIATION: Moskovskiy institut neftekhimicheskoy i gasovoy promyshlennosti im. akad. I. M. Gubkina (Moscow Institute for the Petrochemical and Gas Industry)

SUBMITTED: 22Jun62

DATE ACQD: 10Jun63

ENCL: 00

SUB CODE: 00

NO REF SOV: 003

OTHER: 001

*rh/gz*  
Card 2/2

SKIDANOVA, N.I.; GUNDYREV, A.A.; CHERNOZHUKOV, N.I.

Solubility in furfural of aromatic hydrocarbons separated from  
oily fractions from the petroleum of the Koturtepe field. Trudy  
MINKHIGP no.44:235-241 '63. (MIRA 18:5)

GUNDYRIN, Petr Afanas'yevich; KLEYMAN, M.Ya., red.; KALLISTOVA, G.A.,  
tekh.red.

[Our province in 40 years] Nasha oblast' za 40 let. Stalingrad,  
Stalingradskoe knizhnoe izd-vo, 1957. 65 p. (MIRA 13:8)  
(Stalingrad Province--Economic conditions)

GUNDYRIN, Petr Afanas'yevich

[Stalingrad economic region and prospects for its development]  
Stalingradskii ekonomicheskii raion i perspektivy ego razvitiia.  
Stalingrad, Stalingradskoe knizhnoe izd-vo, 1958. 107 p.

(MIRA 13:3)

(Stalingrad Province--Industries)

GUNDZHUVA, T. CR.

180 46  
783  
49-30  
1941.  
1941.  
А.  
Казань  
1941.

76  
А.  
Казань  
1941

78  
р.  
1941.  
1941.  
1941.  
1941.

78  
1941.  
1941.  
1941.  
1941.

772. Чигиневая Давида Ме-  
1955, 221 с. (Инст. ф-на им П.Н. Ле-  
блева АН СССР в Инст. ф-на АН  
Груз. ССР).  
Защ. 1956, 25 л.  
775. Чумбадзе Илья Салава-  
нович. Об основных энергетических  
соотношениях в лазерах.  
Защ. 1941, 30 с.  
776. Шалаур Ростом Сесео-  
вич. Об атомной и военной доктрине  
топика 8-О в лазере. 1954, 78 с.  
Защ. 1955, 16 л.

777. Березина Георгий Павло-  
вич. Исследование гомополимера в  
молочной и полиуретановой системах  
1951, 98, 111 с., 12 илл.  
Защ. 1953, 30 с.  
778. Березина Георгий Павло-  
вич. Исследование гомополимера в  
молочной и полиуретановой системах  
1951, 98, 111 с., 12 илл.  
Защ. 1953, 30 с.

779. Березина Георгий Павло-  
вич. Исследование гомополимера в  
молочной и полиуретановой системах  
1951, 98, 111 с., 12 илл.  
Защ. 1953, 30 с.

780. Березина Георгий Павло-  
вич. Исследование гомополимера в  
молочной и полиуретановой системах  
1951, 98, 111 с., 12 илл.  
Защ. 1953, 30 с.

781. Березина Георгий Павло-  
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молочной и полиуретановой системах  
1951, 98, 111 с., 12 илл.  
Защ. 1953, 30 с.

782. Березина Георгий Павло-  
вич. Исследование гомополимера в  
молочной и полиуретановой системах  
1951, 98, 111 с., 12 илл.  
Защ. 1953, 30 с.

783. Березина Георгий Павло-  
вич. Исследование гомополимера в  
молочной и полиуретановой системах  
1951, 98, 111 с., 12 илл.  
Защ. 1953, 30 с.

784. Березина Георгий Павло-  
вич. Исследование гомополимера в  
молочной и полиуретановой системах  
1951, 98, 111 с., 12 илл.  
Защ. 1953, 30 с.

619  
Dissertation for Degree of  
Candidate Physical-Mathematical Sciences

Def. at  
Tbilisi State U.



TIKKOYEV, V.A.; SERDYUK, N.F.; SAPUTO, M.P.; GORISHNIY, Ya.I.; VOROB'YEV,  
V.F.; GUNDZILOVICH, A.A.; PRIVALOV, V.G.; MARIN, V.I.;  
LEVCHENKO, R.S.

The best in the profession. Put' i put.khoz. 6 no.12:4-9, 11,  
16-17 '62. (MIRA 16:1)

1. Zamestitel' nachal'nika Petrozavodskoy distantzii puti Oktyabr'skoy dorogi (for Tikkoyev).
2. Nachal'nik Solvychevodskoy distantzii Severnoy dorogi (for Serdyuk).
3. Nachal'nik Shchorsskoy distantzii puti Yugo-Zapadnoy dorogi (for Saputo).
4. Nachal'nik Kotovskoy distantzii puti, Odesskoy dorogi (for Gorishniy).
5. Nachal'nik Sverdlovsk-Passazhirskey distantzii puti Sverdlovskoy dorogi (for Vorob'yev).
6. Nachal'nik L'govskoy distantzii puti Moskovskoy dorogi (for Marin).
7. Zamestitel' nachal'nika Shar'inskoy distantzii Severnoy dorogi (for Levchenko).

(Railroads--Employees)

GUNENKO, V.I.; SLIVKOV, V.I.; SHEPTUNOV, V.I.

Efficiency of field geophysical and geochemical studies in  
the productive sediments of the Zapadnyy Tebuk field.  
Razved. i okhr. nedr 28 no.10:20-26 0 '62. (MIRA 15:11)

1. Ukhtinskoye geologicheskoye upravleniye.  
(Ukhta region—Petroleum geology)

ACCESSION NR: AP4018368

S/0120/64/000/001/0076/0081

AUTHOR: Boos, E. G.; Pavlova, N. P.; Volkova, O. I.; Gunenкова, O. V.;  
Zaytsev, K. G.; Kholmetskaya, A. V.

TITLE: Methods of measuring ionization losses of relativistic particles in a nuclear emulsion

SOURCE: Pribory\* i tekhnika eksperimenta, no. 1, 1964, 75-81

TOPIC TAGS: ionization loss, relativistic particle, relativistic particle  
ionization loss, nuclear emulsion, Ilford G-5 emulsion, emulsion development,  
emulsion development irregularity

ABSTRACT: Irregularities of development of Ilford G-5 nuclear emulsion were studied; methods of eliminating them are suggested. A stack of 40 G-5 films, 600-micron thick, 12x20 cm was irradiated (in CERN) by a 91.8-Gev/s-mean-impulse proton beam. To find the irregularity of development of the emulsion films, the density of blobs on the relativistic-particle tracks was investigated both in the plane parallel to the emulsion and in depth. The effects of the micro-

Card 1/2

ACCESSION NR: AP4018368

scope field-of-view illumination, experimenters' characteristics, and the track immersion angle upon the accuracy of measurements were studied. It was proven that a desirable accuracy (2% or better) in determining ionization losses with immersion angles up to  $10^{\circ}$  is attainable. The technique of "joining" tracks in adjacent emulsion layers is discussed. "The authors wish to thank Zh. S. Takibayev and I. Ya. Chasnikov for a useful discussion of this project, and the workers of the High-Energy-Particle Laboratory, A. A. Alpy\*sbayeva, Ts. Ya. Kagasova, D. I. Yermilova, F. N. Trushlyakov, T. T. Temiraliyev and G. A. Grigor'yeva, for their help in carrying out this project." Orig. art. has: 3 figures and 3 tables.

ASSOCIATION: Institut yadernoy fiziki AN KazSSR (Institute of Nuclear Physics, AN KazSSR)

SUBMITTED: 11Jan63

DATE ACQ: 18Mar64

ENCL: 00

SUB CODE: NS

NO REF SOV: 000

OTHER: 007

Card 2/2

MAZUROVA, T.M.; POPOVA, T.I.; SHMUSHKOVICH, A.Ya.; SHEVELEVA, A.A.;  
GURER, I.I.; LAVRENOVA, V.A.

Letter to the editors. Stomatologiya 38 no.3:72 My-Je '59.  
(MIRA 12:8)

(PLASTICS)

CONFIDENTIAL

Under the control of the Ministry of Oil and Gas, the Ministry of Oil and Gas, the Ministry of Oil and Gas.  
4-19-11-10 10:10. (REF: 14:10)

1. Sekretar' informatsionnoy byuroy  
natsionalnoy ekonomiki i planirovaniya.

**(BASHKIRIA--PETROLEUM INDUSTRIES--ENGINEERING ASPECTS)**

RUMANIA / Chemical Technology. Processing of Natural Gases and Petroleum.. Motor and Rocket Fuel, Lubricants. H-23

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 79125.

Author : ~~Gunesch, H.~~, Stadtmuller, R.

Inst : Not given.

Title : Chromatographic Separation and Determination of Acetylene Derivative Obtained During the Cracking of Methane in an Electric Arc.

Orig Pub: Rev. chim., 1958, 9, No 1, 35-38.

Abstract: A method is described for the qualitative and quantitative analysis of the acetylene derivatives based on the chromatographic principles of separation. As an example an analysis of a mixture consisting of methyl-, vinyl-, and ethyl-acetylene as well as that of diacetylene is cited.

Card 1/1

GUNESCH, H.; BRANDSCH, J.; HEITZ, Jutta; BOTEIU, Aurelia; LOFFLER, Ana

Determination of the crotonic aldehyde from monomer vinyl acetate and its effect on the emulsion polymerization process. Rev chimie Min petr 14 no.1:36-39 Ja '63.

1. Laboratorul central al Uzinelor chimice-Risnov.



GUNESCH, H.; DRAGHICI, R.; FLOREA, E.

Plasticization of polyvinyl acetate dispersions with dibutylphthalate. Rev chimie Min petr 15 no.6:342-344 de '64.

NIAC, G.; GUNESCH, H. WEISS, Gh.; LITERAT, L.

Contributions to the practical application of the kinetic method for determining desorption isotherms. Rev chimie Min petr 15 no. 4:198-203 Ap '64.

L 29773-66 EWI(j)/T DS/WW/IM

ACC NR: AP6020885

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AUTHOR: Gunesch, H.; Brandsch, J.

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TITLE: Concerning the protective-colloid power of polyvinyl alcohol

SOURCE: Revista de chimie, v. 16, no. 9, 1965, 424-427

TOPIC TAGS: polyvinyl alcohol, colloid chemistry

ABSTRACT: The authors suggest two methods to determine the protecting-colloid ability of polyvinyl alcohol, and present experimental results showing that this protecting ability in the polymerization of vinyl acetate can only be defined in terms of the specific composition and polymerization conditions of a particular reaction. Orig. art. has: 2 figures and 9 tables. [Based on authors' Eng. abstract] [JPRS]

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SOV REF: 001

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