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transformations in eutectic alloys copper-aluminum.
 VI. P. L. Gruzin and G. Kurdyumov. *J. Tech. Phys. (U.S.S.R.)* 10, 1690-4 (1940).--The transformation $\beta' \rightarrow \beta_1$ in heated eutectic Cu-Al alloys was investigated by the method of Sykes (measurement of true specific heat). The thermal-capacity curves, obtained for the alloys with the concns. of Al from 11.85 to 12.9%, show a max. corresponding to the absorption of heat at the $\beta' \rightarrow \beta_1$ transformation point. The measured energy of transformation was 4.2-4.5 cal./g. The transformation $\beta' \rightarrow \beta_1$ covers the temp. interval analogous to that in the transformation of steels. (The width of the interval reaches 100°.) The energy of eutectic transformation $\alpha \rightarrow \beta$ was found to be 6 cal./g. VII. G. Kurdyumov and V. Miretskii. *Ibid.* 1685-90.--For the investigation of $\beta_1 \rightleftharpoons \beta'$ transformation of solid solns. in Cu-Al alloys, a photographic method was developed permitting making photographs in a few sec. The heating and cooling processes in the monocryst. alloys contg. 12.4, 12.7 and 13.00% Al were photographed. This permitted direct observation of structural changes corresponding to the $\beta \rightleftharpoons \beta_1 \rightleftharpoons \beta'$ transformation, and study of the phenomena of reversibility and temp. hysteresis for the $\beta_1 \rightleftharpoons \beta'$ transformation.
 Roksalana Gamow

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION
 REGION NUMBER

GROUP #1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	139	140	141	142	143	144	145	146	147	148	149	150	151	152	153	154	155	156	157	158	159	160	161	162	163	164	165	166	167	168	169	170	171	172	173	174	175	176	177	178	179	180	181	182	183	184	185	186	187	188	189	190	191	192	193	194	195	196	197	198	199	200

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Thermal effects in the high-temperature annealing of hardened steel. R. A. Abramovich, P. L. Gruzin, G. V. Kurdymov and R. I. Entin. *J. Tech. Phys. (U.S.S.R.)* 11, 1083 (1941). In order to collect expl. data on the copulation of Fe and carbides in troostite and sorbite, heat capacities were detd. and their changes on heating of steels which were annealed after hardening at 375°, and of cold-deformed steels. In the range 100-700° the heat capacity of cold-worked steels does not differ from that of the heat-treated steels. On heating hardened steels that were previously annealed at 375°, there was found a thermal effect in the range of 500 (600)°, the magnitude of which is about 1 cal/g. Several explanations for this effect are presented. G. M. Kosolapoff

ASB-31A METALLURGICAL LITERATURE CLASSIFICATION

GRUZIN, P. L.

USSR/Metals - Diffusion

1 Sep 51

"The Influence of Carbon on the Self-Diffusion of Iron," P. L. Gruzin, Yu. V. Korney, G. V. Kurdymov, Corr Mem, Acad Sci USSR, Inst of Metal Sci and Phys of Metals, Cen Sci Res Inst of Ferrous Metallurgy

"Dok Ak Nauk SSSR" Vol LXXX, No 1, pp 49-51

Preliminary data touching on the influence of carbon on the self-diffusion of gamma-iron was obtained during investigations of the self-diffusion of pure iron. It was shown that the addn of carbon strongly influences the parameters governing the self-diffusion of iron. Current article subjects this problem

221745

to a special study. Concludes that the addn of carbon in iron decreases the energy of the bond of the austenite lattice. Submitted 12 July 51.

221745

GRUZIN, P. I.

USSR/Engineering - Refractories, Testing, Radioactive Indicators

Oct 52

"Application of the Method of Radioactive Indicators for Investigation of Diffusion in Refractories," L. A. Shwartsman, G. A. Pechenev, P. I. Gruzin, Inst of Metallography and Phys of Metals, TsNIICM

"Ogneupory" No 10, pp 465-469

Studies character of motion in refractories of one of slag components, iron oxides, at temps when this substance is already solidified. Method of radioactive indicators, used in experiments, established that process of penetration of solid oxides in refractories at high temps is described by equation of diffusion. Coefficients of diffusion were determined in magnesite, chrome-magnesite, and dinas bricks at various temps. Used radioactive isotope Fe^{59} . Measured radiation intensity with counters for beta-particles and gamma-quanta.

PA 244172

GRUZIN, P. L.

"Using Artificial Radioactive Indicators for Studying Diffusion and Self-Diffusion Processes in Alloys: Self-Diffusion of Cobalt," Dokl. AN SSSR, 86, No.2, 1952

Abstract, R. L.

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J. of the Iron & Steel
Inst. Feb 1964
Corrosion

The Use of Artificial Radioactive Isotopes for the Study of Diffusion and Self-Diffusion in Alloys. Self-Diffusion in γ Iron. P. L. Gruzin / *Izvestiya Akademii Nauk S.S.S.R. Otdeleniye Tekhnicheskikh Nauk*, 1963, (3), 383-392. [In Russian]. A method of determining coefficients of diffusion and self-diffusion, based on the use of radioactive isotopes, which can be used for any solid suitable for mechanical working is described. More exact results were obtained when using γ -radiating isotopes. The temperature dependence of the coefficient of self-diffusion of γ iron in the 900-1250° C. range was determined.—v. a.

10-13-54
RMZ

P. L. GRUZIN

Journal of the Iron and Steel Inst.
June 1954
Metallography

The Influence of Carbon on the Self-Diffusion of Iron in the Iron-Nickel System. P. L. Gruzin and E. V. Kuznetsov. (*Doklady Akademii Nauk S.S.S.R.*, 1953, 93, (6), 809-812). [In Russian]. The influence of carbon on the self-diffusion of iron in Fe-Ni alloys was studied, using radioactive ^{59}Fe , in the temperature ranges 800-1300 °C and 1050-1330 °C, for alloys containing 20% and 25% of nickel respectively. The time of diffusion heating was such that the diffusion layer was 40-60 times thicker than the electroplated layer of radioactive iron. The dependence of the coefficients of self-diffusion of iron on the carbon content of the iron-nickel alloys are represented by the formulae: (1) for alloys containing 20% of nickel

$$D_{\text{Fe-Ni-C}} = 18 \cdot 10^{-0.02p} e^{-\frac{75000-6000p}{RT}};$$

and (2) for alloys containing 25% of nickel

$$D_{\text{Fe-Ni-C}} = 71 \cdot 10^{-0.03p} e^{-\frac{70000-5000p}{RT}};$$

where p is the carbon content in at.-%, R a gas constant, and T the temperature (° K.). It is concluded that carbon lowers the bond energy of atoms in solid solutions of iron and nickel.

GRUZIN, P.L.; KUZNETSOV, Ye.V.; KURDYUMOV, G.V., akademik.

Effect of intra-granular structure of austenite on the autodiffusion of iron. Dokl.AN SSSR 93 no.6:1021-1023 D '53. (MLRA 6:12)

1. Institut metallovedeniya i fiziki metallov TsNIIChM.
(Iron--Metallography) (Austenite)

GBUZIN, P. L.

~~Determination of diffusion coefficients in solid bodies by the method of radioactive isotopes. P. L. Gruzin and D. P. Na. Ilyin. Doklady Akad. Nauk S.S.S.R. 94, 41-4 (1964); cf. C.A. 48, 1223c.~~—The method was based on the fact that β -particles are more strongly absorbed than γ -rays, and therefore the relative intensities of the 2 radiations changed as a layer of radioactive isotopes diffused into the body of a specimen. The method was tested on tech. Fe by using Co^{60} as the diffusing substance. Fe specimens $3 \times 12 \times 25$ mm. were homogenized at $1050-1250^\circ$ and then covered with a $1-\mu$ layer of radioactive Co. Counts were made before and after diffusion with and without a 1-mm. Al filter. The diffusion anneal was done in a quartz tube evacuated to 10^{-4} mm. Hg and with temp. controlled to $\pm 5^\circ$. The back side was cleaned after diffusion to avoid surface-diffusion effects. Data were obtained for γ -Fe at 3 temps. in the range $1100-1200^\circ$ and for α -Fe at 5 temps. in the range $700-900^\circ$. The D values were: $D_\alpha = 0.2 \exp. (-54,000/RT)$; $D_\gamma = 300 \exp. (-87,000/RT)$. These agreed only moderately well with the results of a sectioning method. The present method had shortcomings, but it had such advantages as being independent of partial loss of the radioactive substance.

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*Diffusion of Cobalt, Chromium, and Tungsten in Iron and Steel, F. L. Gruzin (*Doklady Akad. Nauk S.S.S.R.*, 1964, 84, (1964) 2054-2057 (Russian). The temp-dependence of the diffusion coeff., D , of Co, Cr, and W in α and γ phases of tech. Fe (0.08% C) and steel of eutectoid compn. (0.83% C) was studied in the range 700°-1250° C. by the method of radioactive isotopes (using Co^{60} , Cr^{51} , W^{187}). The $\log D$ against $1/T$ curves for tech. Fe showed a break in the 1000°-1100° C. range, in the γ -phase field. The increase of the diffusion rate in austenite below 1050° C. was explained by the complex structure of the austenite grains (at that temp.), in which the grain boundaries of crystals of the transformed ferrite, pearlite, and ferrite-pearlite mixture are still discernible. The diffusion parameters D_0 and Q in the equation $D = D_0 e^{-Q/RT}$ were calculated for Co, Cr, and W in the ferrite of tech. Fe and in the austenite of tech. Fe and steel. The diffusion activation energy of Co, Cr, and W in austenite was much lower for steel than for tech. Fe, indicating that the presence of dissolved C loosened the bonds between the atoms of these elements in the austenite lattice. The differences between D for Co, Cr, and W in austenite were small; in ferrite the rate of diffusion of Co was 10-20 times that of Cr, with W occupying an intermediate position.—S. K. L.

Inst. Metal. & Physics of Metals, TsNIICM.

Evaluations: { B-77863 - 18 Aug 54
 B-78524 - 8 Sep 54
 B-81524 - 28 Dec 54

Gruzin, P. L.

USSR/Physics - Technical physics

Card 1/1 Pub. 22 - 16/40

Authors : Gruzin, P. L.; Noskov, B. M.; and Shirokov, V. I.

Title : Effect of Mn on the self-diffusion of Fe

Periodical : Dok. AN SSSR 99/2, 247-250, Nov 11, 1954

Abstract : Eight Fe-Mn alloys were investigated to determine the effect of Mn on the self-diffusion of the Fe in austenite. The thermal dependence of the self-diffusion coefficients of Fe in the gamma-phase of Fe-Mn alloys was investigated by the method of radioactive indicators through the utilization of the artificially-radioactive Fe⁵⁹ isotope. The self-diffusion coefficients were calculated on the basis of data obtained by measuring the integral radioactivity of the sample. It was found that the bond between the atoms of the basic alloy during the addition of the second element increases. The energy of activation of Fe self-diffusion at an Mn content of 8% was established as greater than the activation energy of pure iron self-diffusion. Seven references; 6-USSR and 1-USA (1938-1954). Tables; graphs.

Institution : Central Scientific Research Institute of Ferrous Metals, Institute of Metallurgy and Physics and State University, Institute of Chemistry, Gorkiy

Presented by: Academician G. V. Kudryumov, June 5, 1954

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SELF DIFFUSION IN TANTALUM. P. L. Gruzin and V. I. Medvedev. Translated from *Voprasy Fiz. Metallov*. Akad. Nauk Ukrain. S.S.R. 370(1953). 6p.
The self-diffusion coefficients in Ta were determined at 1250, 1250, and 1300°C using Ta^{182} as tracer. Radioactive Ta was deposited on one of the surfaces of a neutralized Ta specimen which was in the form of a parallelepiped. Two samples, with their active faces toward each other, were sealed in quartz ampoules and given diffusion anneals of 170 to 320 hr, depending on the temperature. The self-diffusion coefficients were measured by cutting layers and measuring the remaining integral activity of the specimens. (L. T. W.)

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VINTAYKIN, Ye.Z.; GRUZIN, P.L., kand.fiz.-mat.nauk; KLYACHKO, Yu.A., prof., doktor
khim.nauk; SHOTOV, A.P.

Alloy phase analysis by the method of radioactive tracers. Probl.
metalloved. i fiz. met. no.4:269-276 '55. (MIRA 11:4)
(Alloys--Metallography) (Radioisotopes--Industrial applications)

GRUBIN, P.L., kand.fiz.-mat.nauk

Cobalt, chromium, and tungsten diffusion in iron and steel. Probl.
metaloved. i fiz. met. no.4:475-485 '55. (MIRA 11:4)
(Diffusion) (Iron alloys)

GRUZIN, P.L.

USSR/Crystals.

B-5

Abs Jour : Referat Zhur - Khimiya, No 6, 1957, 18298

Author : P.L. Gruzin, D.F. Litvin.

Title : To The Question of Diffusion Study by the Method of Radioactive Isotopes.

Orig Pub : Problemy metalloved. i fiz. metallov, sb. 4, 1955, 486-493.

Abstract : A variant of the absorption method of radiometric analysis was developed. At the measurement of the sample activity before and after the diffusion annealing, the ratio of the intensities of β - and γ - components of radiation in case of mixed radiators, or the ratio of the intensities of two β spectrum parts in case of pure β - radiators is determined by this method. The solution of the diffusion equations is modified in order to apply them to this method. The relative error of the determination of the diffusion factor is 25%. The experiments were successful

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USSR/Crystals.

B-5

Abs Jour : Referat Zhur - Khimiya, No 6, 1957, 18298

in reducing the error to from 10 to 15%.
The sensitivity of this method is greater than that of other modifications of the absorption method. The use of this method in the region of low temperatures is the most rational. The diffusion factor of Co in Fe determined by this method agrees well with data obtained with other methods.

Card 2/2

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GRUZIN, P.L., kand.fiz.-mat.nauk; KUZNETSOV, Ye.V.; KURDYUMOV, G.V., akademik

Effect of the intergranular structure of austenite on the self-
diffusion of iron. Probl. metalloved. i fiz. met. no.4:494-497 '55.
(Diffusion) (Iron alloys--Metallography) (MIRA 11:4)
(Austenite)

GRUZIN, P.L., kand.fiz.-mat.nauk; NOSKOV, B.M., kand.fiz.-mat.nauk; SHIROKOV,
V.I., kand.fiz.-mat.nauk.

Effect of manganese on the self-diffusion of iron. Probl. metalloved.
i fiz. met. no.4:503-508 '55. (MIRA 11:4)
(Diffusion) (Iron) (Manganese)

GRUZIN, P. L.

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Effect of alloying upon the self-diffusion of cobalt. *Izv. Akad. Nauk SSSR, Ser. Metallurg. i Fiz. Metal. Sbornik* No. 4, 300-16 (1955); *Referat. Zhur., Met.* 1955, No. 6644. — The coeffs. of self-diffusion D of Co in Co alloys contg.: (1) Cr 3.58, (2) Cr 7.08, (3) Ni 4.10, (4) Ni 7.9, (5) Ni 20.17, (6) Cr 0.21, Ni 26.38, (7) Ca 18.32, Ni 15.03, (8) Fe 20.8, and (9) Fe 15.10, Ti 3.76% were detd. at 1100-1150° by the method of integral residues (cf. *C.A.*, 47, 2407a) by means of the γ -activity of Co^{60} . The alloys after being cast were homogenized at 1050° for 20-26 hrs. in a H atm. and rolled into sheets. The curve of $\log D$ as a function of $1/T$ shows that at 1150° and above only vol. diffusion occurs; at temps. below 1150° mixed diffusion occurs. At 1150° for the alloys in the above order D is 0.57, 66.0, 839.3, 124.0, 0.155, 0.3, 0.4, 0.54, and 0.008 sq. cm./sec., energies of activation of self-diffusion are 65.8, 79.3, 97.4, 81.2, 69.3, 72.1, 64.2, 65.0, and 81.2 kcal./g-atom. This shows a considerable change in bond energy of the lattice on addn. of alloy components.

Alexis N. Pestun

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GRUZIN, P.L., kand.fiz.-mat.nauk; KOSTOGONOV, V.G.; PLATONOV, P.A.

Using an artificial C^{14} isotope in studying the diffusion of carbon
in steel. Probl. metalloved. i fiz. met. no.4:517-523 '55.

(Carbon--Isotopes) (Steel--Analysis)

(MIRA 11:4)

GRUZIN, P.L., kand.fiz.-mat.nauk; MESHKOV, V.I.

Self-diffusion of tantalum. Probl. metalloved. i fiz. met. no.4:570-573
'55. (MIRA 11:4)

(Diffusion) (Tantalum)

VINTAYKIN, Ye. Z.; GRUZIN, P. L.; FEDOROV, S. N.

Determining sublimation temperatures by means of the mass-spectrograph. Zav. lab. 21 no. 7: 835-837 '55. (MIRA 8:10)

1. Tsentral'nyy nauchno-issledovatel'skiy institut chernoy metallurgii.

(Sublimation (Chemistry)) (Mass spectrometry)

GRUZIN, P. L.

USSR/Physics - Metallurgy

Card 1/1 Pub. 22 - 17/50

Authors : Gruzín, P. L.

Title : Effect of chrome on the self-diffusion of iron

Periodical : DOK. AN SSSR 100/1, 65-67, Jan. 1, 1955

Abstract : Experiments were conducted to determine the effect of chrome on the self-diffusion of iron. The results indicate that the presence of chrome in iron increases the intermolecular couplings in the crystalline lattice and diminishes the velocities of the iron atoms. This, then increases the self-induction of iron. Four USSR references (1948-1953). Graphs; tables.

Institution : Institute of Metallography and Physics of Metals of TsNIIChM (Central Scientific Research Institute of Ferrous Metals)

Presented by: Academician G. V. Kurdyumov, May 26, 1954

GRUZIN, P. L.

USSR/ Physics - Diffusion

Card 1/2 Pub. 22 - 10/47

Authors : Gruzin, P. L.; Kostogonov, V. G.; and Platonov, P. A.

Title : ~~Application of the carbon isotope C¹⁴ for the study of carbon diffusion~~
 : Application of the carbon isotope C¹⁴ for the study of carbon diffusion
 : in steel

Periodical : Dok. AN SSSR, 100/6, 1069-1072, Feb 21, 1955

Abstract : The development of a method for measuring the diffusion coefficient in
 : metals by means of the radioactive C¹⁴ isotope is reported. The new
 : method has shown greater sensitivity than all other available methods
 : and requires no special assumptions regarding the mechanism of diffusion
 : or knowledge of the structural diagram.

Institution : Central Scientific Research Institute of Ferrous Metallurgy, Institute
 : of Metallurgy and Physics of Metals

Presented by : Academician G. V. Kurdyumov, June 22, 1954

Periodical : Dok. AN SSSR. 100/6, 1069-1072, Feb 21, 1955

Card 2/2 Pub. 22 - 10/47

Abstract : The method was found highly suitable for the study of the effect of alloying elements on the carbon diffusion in ferrite and austenite. Some practical experiments carried out by means of the radioactive isotope method are described. Nine references: 4 USA and 5 USSR (1937-1953). Graphs.

G RUZIN, P. L.

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Chromium diffusion in its nickel solid solutions. P. L. Gruzin and G. B. Fedorov (Eng. Phys. Inst., Moscow). *Doklady Akad. Nauk S.S.S.R.* 105, 207-210 (1957). Cr diffusion was studied in (1) 99.8% Ni, (2) Ni-Cr alloys, (3) Ni-Cr-Ti-Al, and (4) Ni-Cr-Ti-kov Al alloys. The isotope was plated on the sample surface; 2 samples were clamped together face to face and diffusively annealed for 5-3 hrs. at 875° and 1250°. The radioactivity was measured in thin layers taken off by machining the samples. In all the samples except No. 3, Cr diffused in the single-phase range; in No. 3, at 875°, 2 phases were present, whereas at higher ranges in No. 3, Cr also diffused in 1 phase. In all single phases the diffusion values fall on a straight line on the log D-T diagram. The diffusion coeffs. were evaluated for 700, 800, 900, and 1000°. In a 20% Cr alloy the diffusion coeff. is 0.1 to 0.2 that in pure Ni in the 700-1000° range, which corresponds to a rise in activity energy from 48 to 61 kcal./g. The Ti addn. further raises the activity coeff., i.e. increases the bond strength in the lattice. The calcd. m.p. of the alloys were very close to the expl., and were, in the 4 samples, resp.: 1455°, 1425°, 1370°, and 1330°, which explains the experimentally observed rise in the activation energy of Cr diffusion in the alloys of lower m.p.

W. M. Steynberg

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Gruzina, P.L.

3785. INVESTIGATION OF WEAR OF REFRACTORY LININGS OF METALLURGICAL FURNACES USING RADIOACTIVE ISOTOPES. Gruzina, P.L. and Zensky, B.B. (Zavod. Lab. (Fact. Lab., Moscow), 1956, (2), 169). To follow the wear of the shaft lining of a blast furnace, Co⁶⁰ was placed at intervals along the radius, and a counter outside the furnace registered the total radiation; the counter showed when the lining had worn back to such an extent that a radioactive pellet was removed. The source is placed in alumina ampoules, closed at one end; the other end is smeared with refractory paste, to reduce evaporation. At working temperatures of 1100° evaporation of cobalt from closed ampoules is negligible. To study wear on the lining of a mixer and of the roof, sides and hearth of an open hearth furnace, holes were made in chrome-magnesite bricks and radioactive iron powder was inserted. For measurements under production conditions, five radiation sources were placed 50 cm apart in the brick. This method cannot be used for the hearth of a blast furnace, the wear in which can be measured by impregnating the hearth with solutions of radioactive isotopes and watching for evidence of radioactivity in the pig-iron and slag. Experience shows that the hearth lining wears quickly at first - the top courses last from a day to a month - and short-lived isotopes such as

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p³², Ir¹⁹² were therefore used. In the lower courses, however, long life isotopes such as Co⁶⁰ are needed. The results are shown graphically. To avoid errors, isotopes with different spectra should be placed in the adjoining sections. Isotopes having a stable γ radiation with an average energy of 0.4-0.5 Mev. are most suitable. Radioactive isotopes are at present placed along the axis of the hearth. Isotopes are put into bricks during construction or repair of the furnace. The ampoules for certain brick-work are made from graphite, those for the firebrick lining from iron.

B. Gruzina, R.A.

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E-6

USSR / Diffusion. Sintering.

Abs Jour : Ref Zhir - Fizika, No 4, 1957, No 9339

Author : Arkharov, V.I., Gerasimov, A.F., Gruzin, P.L.

Inst : Ural'University USSR

Title : Investigation of the Phenomenon of Internal Adsorption on the Boundary of an Aluminum Silver Alloy with Oxide Film, Using the Radioactive Isotope Method.

Orig Pub : Fiz. metallov i metallovedeniye, 1956, 2, No 2, 294-302

Abstract : Plates of the alloy Al -- 0.29% Ag, containing a small amount of radioactive isotope Ag^{110} were subjected to a six-hour exposure at 550° in vacuum (10^{-2} -- 10^{-3} mm mercury) to obtain a thin (50 -- 100 A) oxide film, and to rapid cooling. After this, successive layers 500 -- 600 A thick were etched away from the specimen (the thickness of the removed layer was estimated from the decrease in weight of the specimen), and each layer was etched away in a dif-

Card : 1/2

USSR / Diffusion. Sintering.

E-6

Abs Jour : Ref Zhur - Fizika, No 4, 1957, No 9339

Abstract : ferent vessel. Such treatment was carried out cyclically 18 times until there was accumulated in all the vessels a sufficient amount of etching products, and after which the etching products were evaporated and their activity measured. It turned out that the first (surface) layer is somewhat poorer in silver, compared with the average contents of the silver in the alloy, which can be explained by the floating away of the silver atoms into the specimens as a result of the oxidation of the surface. The second and 3rd layers have considerably more silver compared with its average contents in the alloy. This enrichment cannot be attributed to selective oxidation of the alloy and must be ascribed to the internal adsorption of silver in the transition zone between the oxide and metal. The thickness of the transition zone, according to the described experiments, is estimated to be 1,600 -- 1,800 A.

Card : 2/2

P.L.

USSR / Diffusion. Sintering.

E-6

Abs Jour : Ref Zhur - Fizika, No 4, 1957, No 9328

Author : Gruzin, P.L., Tyutyunnik, A.D.

Inst : Institute of Metal Research and Metal Physics, Central Scientific Research Institute for Ferrous Metallurgy.

Title : Concerning the Diffusion Mobility of Atoms During the Melting and Crystallization of Metals.

Orig Pub : Fiz. metallov i metallovedeniye, 1956, 3, No 1, 70-75

Publ Ref :

Abstract : Certain problems concerning the role of diffusion in melting and recrystallization of metal are considered. An analysis of the available data on self-diffusion leads to the conclusion that the melting processes and the recrystallization of various metals are characterized by definite levels of diffusion mobility. The threshold levels of the diffusion mobility are determined by a coefficient of the order $10^{-22} \text{ cm}^2 \text{ sec}^{-1}$ in recrystallization and 10^{-8} cm^2

Card : 1/2

USSR / Diffusion. Sintering.

E-6

Abs Jour : Ref Zhur - Fizika, No 4, 1957, No 9328

Abstract : sec^{-1} in the melting of pure metals. The crystallization of solid solutions takes place at higher threshold levels of diffusion mobility ($D \sim 10^{-13} \text{ cm}^2 \text{ sec}^{-1}$). It can be assumed that other single-type processes that take place in solid bodies and that are connected with diffusion also have approximately equal threshold levels of diffusion mobility.

Card : 2/2

Gruzin, P. L.

~~Study of refractory lining wear in metallurgical furnaces~~
~~by the radioactive tracer technique. P. L. Gruzin and S. V.~~
~~Zemskil. Zavedskaya Lab. 22, 109-743557.~~ The detn.
 of blast-furnace, mixer, and open-hearth lining wear by
 the radioactive tracer technique, as practiced in Russia, is
 described. The techniques depend on the specific furnace
 construction, the lining thickness, etc., and the preferred
 radioactive isotopes are Co^{60} and Fe^{59} , the former in the
 blast-furnace shaft with a relatively thin lining (1350 mm.),
 by measuring the radiation through the shell and in the
 metal. A large increase in the latter invariably indicates
 a breaking off of the lining. In the blast furnace well lining,
 Fe^{59} and W^{187} are recommended. The selection of the
 proper isotope is dictated by the length of the service period
 of the furnace, by the lining thickness, by safety to the
 operating personnel, etc.

W. M. Sternberg

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WMS

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SAKHARUK, P.A.; KHAZANOVA, T.P.; GRUZIN, P.L.; TYUTYUNNIK, A.D.

Using radioisotopes to investigate the dephosphorization process
of pyrochloric concentrates. TSvet. met. 29 no.7:7-9 J1 '56.
(MLRA 9:10)

(Pyrochlore) (Radioisotopes--Industrial applications)

BABIKOVA, Yu.F.; GRUZIN, P.L., professor, doktor fiziko-matematicheskikh nauk, nauchnyy rukovoditel'.

[Using radioactive indicators in the study of the mobility of atoms in solid solutions; abstract of a dissertation for the degree of candidate of physical and mathematical sciences]
Izuchenie podvizhnosti atomov v tverdykh rastvorakh metodom radioaktivnykh indikatorov; avtoreferat dissertatsii na soiskanie uchenoi stepeni kandidata fiziko-matematicheskikh nauk.
Moskva, M-vo vysshego obrazovaniia SSSR, 1957. 15 p. (MIRA 11:12)
(Solutions, Solid) (Carbon--Isotopes) (Mass transfer)

GRUZIN, P. L.

137-58-5-10592

Translation from: Referativnyy zhurnal, Metallurgiya. 1958, Nr 5. p 246 (USSR)

AUTHORS: Gruzin, P. L., Kurdyumov, G. V., Tyutyunik, A. D., Entin, R. I.

TITLE: On the Role of Diffusive Displacements of Atoms in High-temperature Strength (O roli diffuzionnykh peremeshcheniy atomov v zharoprochnosti)

PERIODICAL: V sb.: Issled. po zharoprochn. splavam. Vol 2. Moscow, AN SSSR, 1957, pp 3-8

ABSTRACT: Some results of investigations of diffusion (D) in metals and alloys relative to the problem of high-temperature strength (H) are examined. It is noted that the special features of the behavior of metals at high temperatures under load are conditioned by the existence of rather frequent diffusive shifts of atoms (A) in the crystal lattice of the phases constituting the alloy. Therefore, along with the shear mechanism of plastic deformation, a diffusion mechanism becomes active. The number of atomic displacements, increasing with temperature, tends to limit the temperature zone in which hardened alloy phases may be employed, owing to the reduction in the resistance to plastic deformation due to the shear mechanism. The relatively higher A

Card 1/2

137-58-5-10592

On the Role of Diffusive (cont.)

mobility at the grain boundary or the intra-grain interface, as against that within the body, means that the grain boundaries constitute the weak spot in the resistance of a metal to deformation and failure at high temperatures. A reduction in the mobility of the A is required to increase the level of H. It is demonstrated that an identical level of mobility of the A can be attained at different temperatures with different metals. The temperature at which a given level of diffusive mobility of A is attained is determined primarily by the energy of activation. In some metals the level of mobility of the A is also significantly shifted by the change in the magnitude of the factor D_0 preceding the exponent in the expression for the relationship between the coefficient of diffusion (CD) and the temperature. Accumulated experimental data show that a variation in the CD may occur owing to changes in either parameter of the temperature dependence of the CD. At elevated energies of activation (due to alloying), there is usually an increase in the multiplier D_0 , with the result that at temperatures that are high for the given alloy base metal there is little change in the CD, while at low temperatures they may change by a full order of magnitude or even more. Addition to the alloy of elements that strengthen the bond in the solid solution causes a shift toward higher temperatures for the onset of the diffusive ductility mechanism.

1. Metals--Diffusion 2. Alloys--Diffusion 3. Metals--Temperature Properties
Card 2/2 4. Metals--Mechanical properties

137-58-4-6317

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 4, p 1 (USSR)

AUTHOR: Gruzin, P. L.

TITLE: Making Use of the Achievements of Nuclear Physics in Metallurgy
(Ispol'zovaniye dostizheniy yadrenoy fiziki v metallurgii)

PERIODICAL: V sb.: Primeneniye radioaktivn. izotopov v chernoy metallurgii.
Chelyabinsk, Knigoizdat, 1957, pp 19-31

ABSTRACT: The use of isotopes and nuclear radiations in metallurgy, instrument and machine manufacture, and the study and control of metallurgical processes.

1. Metallurgy--USSR
2. Isotopes--Applications
3. Instrumentation--Development

P. N

Card 1/1

000211 13.

137-1957-11-25467

Translation from: Referativny zhurnal, Metallurgiya, 1957, Nr 12, p 358 (USSR)

AUTHORS: Gruzin, P. L., Zemskiy, S. V.

TITLE: Inspection of Refractory Lining by Means of Radioactive Isotopes (Primeneniye radioaktivnykh izotopov dlya kontroiya sluzhby ognepornoy kladki)

PERIODICAL: V sb.: Primeneniye radioaktivn. izotopov v chernoy metallurgii. Chelyabinsk, Knigoizdat, 1957, pp 32-48

ABSTRACT: Methods of inspection for wear in the lining of open-hearth and blast furnaces are examined.

L V

1. Refractory materials-Inspection
2. Isotopes (Radioactive)-Applications
3. Furnaces-Maintenance

Card 1/1

Gruzin, P. L.

137-1958-1-280

Translation from Referativnyy zhurnal. Metallurgiya, 1958. Nr 1, p 43 (USSR)

AUTHORS: Gruzin, P. L., Zemskiy, S. V., Trekalo, S. K., Afanas'yev, V. N.

TITLE: A Study of the Motion of Charge Components in Blast Furnaces
(Izucheniye dvizheniya shikhtovykh materialov v domennykh pechakh)

PERIODICAL: V sb.: Primeneniye radioaktivn. izotopov v chernoy metallurgii.
Chelyabinsk. Knigoizdat. 1957, pp 59-66

ABSTRACT: Radioactive isotopes Fe⁵⁹ and Co⁶⁰ were used to determine the rate of motion of the charge at various distances from the wall of a blast furnace. Photon counters mounted either outside the furnace or in its lining (fixed counters) or introduced within it by A. A. Melikyan's method (mobile counters) were used to trace the progress of the radioactive isotopes through the various levels of the furnace. Data obtained by the investigation show that the time during which the materials remain within a 330 m³ furnace from the stock line to the tuyere belt is 3 hrs. 40 min. for the clinker in the middle of the furnace and 4 hrs. 45 min. at the periphery, while for coke it is 4 hrs. 50 min. both at the center and at the periphery.

Card 1/1

M. O.

1. Blast furnaces--Performance 2. Iron isotopes (Radioactive)
--Applications 3. Cobalt isotopes (Radioactive)--Applications

G. Gruzina, P. L.
USSR/Physical Chemistry - Crystals.

B-5

Abs Jour : Referat Zhur - Khimiya, No 1, 1958, 232

Author : P.L. Gruzina, Yu.A. Polikarpov, G.B. Fedorov.

Inst : ~~USSR~~

Title : Study of Diffusion of Carbon in Nickel and Its Alloys
Using Radioactive Isotope C^{14} .

Orig Pub : Fiz. metallov i metallovedeniye, 1957, 4, No 1, 94-102

Abstract : The method of the study of the diffusion of C in metals using C^{14} was developed. The tagged C was applied to the metal surface by short-time cementation. Parallel layers were machined off after the diffusion annealing. The total radioactivity I of the remaining portion of the specimen was measured after the removal of each layer, and graphs of the dependence of I on the thickness of the removed layers x, and of the dependence of $\ln I$ on x^2 were plotted. The diffusion factor D was determined by the equation $D = (\alpha/\beta - 1) / 4\alpha t$, where β is the slant of

Card 1/2

USSR/Physical Chemistry - Crystals.

B-5

Abs Jour : Ref Zhur - Khimiya, No 1, 1958, 232

the line $\ln I = f(x^2)$, and α is the slant of a line of the same kind plotted for the thin surface layer of the specimen after cementation. It was found for the diffusion of C in Ni at 500 to 900° that the pre-exponential factor $D_0 = 0.1 \text{ cm}^2\text{sec}^{-2}$ and the activation energy $Q = 33 \text{ kcal per g-atom}$. The alloying of nickel with Cr, Co or Mo (1 to 5%) results in a small increase of D_0 and Q , D remaining practically constant, which agrees with the known peculiarities of the arrangement of C in metal lattices. D for C is 10^7 to 10^8 times greater than D for metal in metal. The importance of obtained results for metallurgy is discussed.

Card 2/2

Gruzina P.L.

AUTHORS: Babikova, Yu. F., and Gruzin, P. L. 126-2-10/35

TITLE: Study of the electrolytic transfer of carbon in steel by means of radioactive tracers. (Izucheniye elektroliticheskogo perenosa ugleroda v stali metodom radioaktivnykh indikatorov).

PERIODICAL: Fizika Metallov i Metallovedeniye, 1957, Vol.5, No.2, pp. 255-260 (USSR)

ABSTRACT: Literary data are available on investigations of electro-diffusion of carbon into austenite. Attempts by Guterma, V. M. (Ref.5) to establish electrolytic transfer of carbon into ferrite were unsuccessful. Diffusion data published by Gruzin, P. L. (Ref.6) indicate that the mobility of carbon atoms in ferrite is greater than in austenite. Therefore, it could be expected that electro-diffusion of carbon should take place in ferrite and that it could be detected if a proper technique is used. In this work the main object of investigation was α -iron. The specimens consisted of iron foils whereby the iron contained the following: 0.03% C; 0.04% Mn; 0.009% S; 0.13% Si; 0.005% P. The specimens consisted of 5 to 8 mm wide, 0.08 to 0.15 mm thick and 50 mm long strips which were copper-plated

Card 1/3

126-2-10/35

Study of the electrolytic transfer of carbon in steel by means of radioactive tracers.

throughout their entire length, except for a central section 1 to 5 mm long intended for case hardening with radioactive carbon. The case hardening was effected in special copper containers placed into evacuated quartz tubes using charcoal and BaCO_3 with radioactive C^{14} as the carburizing agent. Fig.2⁵ shows a photo of the general view of the set-up for electro-diffusion annealing; Fig.3 shows a photo of the vacuum part of the apparatus for electro-diffusion annealing, which was carried out in molten salts at the temperatures 550 to 800°C. The graph, Fig.1, shows the distribution of the radio-active carbon along the length of the specimen after case hardening; the graph, Fig.4 shows the distribution of the radio-activity on the cathode and the anode sections of the specimen, before and after heating, at a temperature of 600°C for ten minutes. Fig.5 shows the distribution of radio-activity at the ends of the specimen before and after diffusion annealing at 700°C for fifty minutes. Table 1 gives the number of carbon transfers for the ferrite for the temperatures 800, 700, 600 and 530°C, whilst Table 2 gives the values of carbon cation charges

Card 2/3

126-2-10/35

Study of the electrolytic transfer of carbon in steel by means of radioactive tracers.

for the ferrite for the same temperatures. The here described method of determining the carbon transfer number in alloys, based on utilising C^{14} , permits detection of the transfer of carbon in steels under the effect of the electrical field and to determine the degree of ionisation of its atoms. It was established that the carbon atoms in the ferrite are in the form of cathions and the cathion charge of the carbon is approximately equal to four elementary units of the electric charge.

There are 5 figures, 2 tables and 10 references, 7 of which are Slavic.

SUBMITTED: March 7, 1957.

ASSOCIATION: Institute of Metal Technology and Metal Physics
TsNIICHM. (Institut Metallovedeniya i Fiziki Metallov
TsNIICHM)
Moscow Physico-Engineering Institute. .
(Moskovskiy Inzhenerno-fizicheskiy Institut).

AVAILABLE: Library of Congress.
Card 3/3

GRUZIN, P. L.

Radioactive indicators for the study of processes of dephosphorizing pyrochlore concentrates. ~~E. A. Slabunsk, F. P. Khazanova, P. L. Gruzin, and A. D. Tyutyunuk. Tsvetnye Metally, 29, No. 1, 7-9 (1957).~~ Radioactive P dissolved in HNO₃ and neutralized with NaOH was added (a) to metallic Zn which had been treated with a soln. of apatite in 10% HCl, (b) to pyrochlore concentrate contg. 17% ZnO₂, and (c) to pyrochlore contg. 0.7% ZrO₂. In all cases about 68% of the activity was lost after the first washing with hot, 90°, water. On the other hand, 30-35% of the activity remained after 7 washings. Thus, while much of the P was adsorbed and could be removed by washing some remained as an insol. phosphate of Zr. B.

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- Ruz
- Gruz

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20-114-6-23/54

AUTHORS: Bardin, I. P., Member of the Academy, Gruzin, P. L., Zemskiy, S.V.

TITLE: An Investigation of the Blast Furnace Process by Means of Isotopes (Izucheniye domennogo protsesssa metodom izotopov)

PERIODICAL: Doklady Akademii Nauk SSSR, 1957, Vol. 114, Nr 6, pp. 1220-1223 (USSR)

ABSTRACT: The present paper gives the results of various experiments made by the Central Scientific Research Institute for Siderurgy and by metallurgical factories on the motion of the charge materials, the wear of the fire-proof lining and the elaboration of a method for studying the motion of gases in the blast furnace by means of isotopes. Under the influence of the air blast the pieces of coke perform a spherical circulation. With the aid of radioactive isotopes the motion of the parts of the charge in the blast furnaces can be studied by fundamentally new methods. A piece of the charge is labelled by an artificially radioactive isotope and its motion followed by radiation detectors. The simplest method for labelling the charge materials consists in the boring of holes in the pieces of coke and lime and the fixing of certain quantities of radioactive isotopes in these holes. Another very suitable

Card 1/2

20-114-6-23/54

An Investigation of the Blast Furnace Process by Means of Isotopes

method is based on the activation of the grains of the charge material in the nuclear reactors.

At present two methods are used for studying the motion of the charge. By the first method the total duration of stay of the charge in the furnace is determined; details on this are given. The second method requires a special equipment of the furnace with investigation holes which are arranged in a vertical. By these methods the authors performed a number of investigations of the motion of charges in furnaces with various volumes.

Several conclusions: The various components of the charge of the blast furnace have various speeds. In the axial part of the blast furnace the coke and the agglomerate are more rapidly displaced than on the periphery. Of great influence are the construction of the blast furnace, the cooling system and the forcing of the blast furnace process. There are 2 figures, 1 table, and 6 references, 5 of which are Slavic.

SUBMITTED: March 12, 1957

Card 2/2

81501

SOV/137-59-9-10629

187500

Translation from: Referativnyy zhurnal, Metallurgiya, 1959, Nr 5, p 166 (USSR)

AUTHORS: Gruzin, P.L., Polikarpov, Yu.A., Fedorov, G.B., Shumilov, M.A.

TITLE: Investigations Into Carbon Diffusion in Alloys With the Use of the C¹⁴ Radioactive Isotope

PERIODICAL: V sb.; Metallurgiya i metallovedeniye, Moscow, AS USSR, 1958, pp 246 - 252

ABSTRACT: The authors investigated the diffusion of C in ferrite (alloyed with Ni, Mo, Cr, Mn) Si in austenite (alloyed with Si) and in Ni (alloyed with Cr, Co, Mo). The diffusion coefficient D was determined by the method of removing the layers and measuring the integral radioactivity of the remaining section of the specimen. Introduction of Si into the ferrite increased the pre-exponential multiplier D₀ and also the activation energy Q. The same effect was exerted by alloying with Ni, while Mo and Cr produced the greatest effect. Thus if the Mo content was 2.5%, D₀ increased from 0.2 to 20.0 cm²/sec, and Q from 24.6 to 33.5 kcal/g-atom. The introduction of 0.93% Cr gave D = 16.4 cm²/sec and Q =

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81501

SOV/137-59-5-10629

Investigations Into Carbon Diffusion in Alloys With the Use of the C^{14} Radioactive Isotope

33.6 kcal/g-atom. Alloying of nickel with 0.74% and 4.65% Cr, 5.25% Co and 2.94% Mo, caused in any case an increase in Q due to the stronger bonds of C atoms was in the crystalline lattice of Ni. The same effect on the C diffusion in γ Fe was produced by an addition of Cr and Mo. The addition of Co reduced the value of Q for C diffusion in γ Fe. There was almost no difference in the rates of C diffusion in Ni and γ Fe. At all temperatures, D of C was $10^5 - 10^{10}$ times higher than D of the metal. The rates of C diffusion in the ferrite and austenite were very different. This explains some peculiarities in the process of stepped case-hardening of steel. The first case-hardening stage is conducted in the γ -phase range. At the second stage, case-hardening is conducted in the α phase range, since diffusion takes place at a higher rate.

I.D.

Card 2/2

RUSSIA, P. I. (auth. N. G. Bogdanova, G. I. Yermolayev, I. D. Nikulinskiy)

"APPLICATION OF RADIOACTIVE ISOTOPES FOR THE INVESTIGATION OF ENZYMATIC PROCESSES".

by N. G. Bogdanova, P. L. Gruzin, G. I. Yermolayev and I. D. Nikulinskiy.

Report presented at 2nd UN Atoms-for-Peace Conference, Geneva, 9-13 Sept. 1958.

GRUZIN, P. L. (Prof.)

"Employment of Isotopes and Nuclear Radiation in Research and Industry,"
report distributed at the International Seminar on Peaceful Uses of Atomic
Energy, and the Youth, Moscow, Aug 1958.

BABIKOVA, Yu.F.; GRUZIN, P.L.

Diffusion in nickel-base solid solutions. Issl. po zharopr. splyv.
3:109-112 '58. (MIRA 11:11)
(Diffusion) (Solutions, Solid) (Nickel alloys)

SOV/37-59-2-4054

Translation from: Referativnyy zhurnal. Metallurgiya, 1959, Nr 2, p 248 (USSR)

AUTHORS: Gruzin, P. L., Babikova, Yu. F., Borisov, Ye. V., Zemskii, S. V.,
Peregudov, N. P., Polikarpov, Yu. A., Tirkina, A. N., Fedorenko, G. B.,
Shumilov, M. A.

TITLE: Study of the Mobility of Carbon Atoms in Steel and Alloys Using C¹⁴
Isotope (Izucheniye podvizhnosti atomov ugleroda v stal'nykh spavakh
pri pomoshchi izotopa C¹⁴)

PERIODICAL: Sb. tr. In-t metaloved. i fiz. metallov Tsentr. n.-i. in-ta chernoy
metallurgii. 1958, Vol 5, pp 327-365

ABSTRACT: The authors examine methods for investigating the diffusion, elec-
trolytic transfer, and distribution of C in Fe, Ni, and some of their
alloys. Data were obtained by the direct (autoradiographic) method
on the effect of Cr, Ni, Mo, and Si on the diffusion of C in ferrite; Ni
and Si have appreciably less effect on the diffusion of C than the carbide-
forming Cr and Mo. It was established that the diffusion mobility
level changes very little when the Fe and Ni are highly alloyed; it is
displaced only when another base is used, as it happens in Fe-Cr, and
under these conditions the mobility level of C approaches that of the

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SOV/137-59-2-4054

Study of the Mobility of Carbon Atoms in Steel and Alloys Using C^{14} Isotope

alloying elements. It was experimentally verified that the C in Fe and Ni is in the cation state. It was established that the cation charge can change depending upon the character of the alloying element. Bibliography: 27 references.

M. G.

Card 2/2

SOV/137-58-8-16757

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 8, p 75 (USSR)

AUTHORS: Gruzin, P.L., Zemskiy, S.V., Tyutyunnik, A.D.

TITLE: Diffusion in Titanium and Alloys Based Thereon and the Sintering Process (Diffuziya v titane i splavakh na yego osnove i protsess spekaniya)

PERIODICAL: Sb. tr. In-t metalloved. i fiz. metallov Tsentr. n.-i. in-ta chernoy metallurgii, 1958, Vol 5, pp 366-382

ABSTRACT: Processes of diffusion in powder Ti (porosity <2%) and alloys thereof with 10% Cr or W (porosity <5%) are investigated by Cr, Fe, and W isotopes in the 700-1250°C interval. The temperature dependence of the value of D is determined on the basis of a layer-by-layer analysis of the integral radioactivity of the β (W isotope) and γ radiation. It is established that this dependence is of the same exponential character as that of compact metals. Ti differs from other refractory metals by a higher level of diffusive mobility. D and D_0 rise with increase in porosity. The D values may be employed to solve problems of rational duration of homogenation of powder alloys and to calculate the density of Ti in powder form sintered at various

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SOV/137-58-8-16757

Diffusion in Titanium and Alloys Based Thereon and the Sintering Process
temperatures in the course of a given period. The conclusion is arrived at
that the sintering of metallic Ti powder proceeds primarily by diffusion.

A.N.

1. Titanium--Diffusion
2. Titanium alloys--Diffusion
3. Radioisotopes--Applications
4. Diffusion--
Temperature factors

Card 2/2

GRUZIN, P.L.; KOHONYUK, I.F.; PAVLYUCHENKO, M.M.; POLIKARPOV, Yu.A.

Using the method of radioisotopes in studying the diffusion of sulfur in iron. Inzh.-fiz. zhur. no. 6:64-67 Je '58.

(MIRA 11:7)

1. Institut metallovedeniya i fiziki metallov Tsentral'nogo nauchno-issledovatel'skogo instituta chernoy metallurgii, Moskva i Belorusskiy gosudarstvennyy universitet im. V.I.Lenina, Minsk.
(Radioisotopes--Industrial applications)
(Sulfur--Isotopes)
(Iron--Metallography)

24(8)

PHASE I BOOK EXPLOITATION

SOV/2117

Soveshchaniye po eksperimental'noy tekhnike i metodam vysokotemperaturnykh issledovaniy, 1956

Ekspperimental'naya tekhnika i metody issledovaniy pri vysokikh temperaturakh; trudy soveshchaniya [Experimental Techniques and Methods of Investigation at High Temperatures; Transactions of the Conference on Experimental Techniques and Methods of Investigation at High Temperatures] Moscow, AN SSSR, 1959. 789 p. (Series: Akademiya nauk SSSR. Institut metallurgii. Komissiya po fiziko-khimiicheskim osnovam proizvodstva stali) 2,200 copies printed.

Resp. Ed.: A.M. Samarin, Corresponding Member, USSR Academy of Sciences; Ed. of Publishing House: A.L. Bankvits'er.

PURPOSE: This book is intended for metallurgists and metallurgical engineers.

COVERAGE: This collection of scientific papers is divided into six parts: 1) thermodynamic activity and kinetics of high-temperature processes 2) constitution diagram studies 3) physical properties of liquid metals and slags 4) new analytical methods and production of pure metals 5) pyrometry, and 6) general questions. For more specific coverage, see Table of Contents.

VI. GENERAL QUESTIONS

Kholodov, A.I., and G.V. Musorin. Instrument for Measuring the Rate of Tempering of Steel 675

Bogdanova, N.G., P.L. Grusin, G.I. Yermolayev, and I. D. Mikulinskiy. A Study of the Motion of Metal and the Distribution of Alloying Elements in Open-hearth Furnaces 682

Card 27/32

G Ruzin, P.L.

21(3,4); 17(10)^{P. 2}

PHASE I BOOK EXPLOITATION

SOV/3394

Neischerpayemy (The Inexhaustible) Moscow, Atomizdat, 1959. 149 p.
Errata slip inserted. 10,000 copies printed.

Compiler: V. P. Parkhit'ko; General Ed.: A. K. Krasin, Doctor of Physical and
Mathematical Sciences, Professor; Ed.: N. M. Pchelintseva; Tech. Ed.: N. A.
Vlasova.

PURPOSE: This book is intended for the layman interested in the peaceful use of
atomic energy.

COVERAGE: This book contains several reports by leading Soviet scientists,
specializing in the peaceful uses of atomic energy, at the international
seminar on "Youth and Peaceful Use of Atomic Energy," held in August, 1958,
under the auspices of the Committee on Youth Organizations of the USSR.

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Card 2/3

GRUTZIN, P.L.

RUSSIAN BOOK REPRODUCTION SOV/2113

International Conference on the Peaceful Uses of Atomic Energy. 2nd, Geneva, 1958

Резюме докладов ученых и промышленности и промышленности изотопов (Reports of Soviet Scientists, Production and Application of Isotopes) Moscow, Atomizdat, 1959. 368 p. (Series: Iza: Trudy, vol. 6) 5,000 copies printed.

Имя (Title page): G.V. Kurdyumov, Academician and I.I. Sorokov, Corresponding Member, USSR Academy of Sciences; Ed. (Inside book): Z.D. Andreyenko; Tech. Ed.: Z.D. Andreyenko.

PURPOSE: This book is intended for scientists, engineers, physicists, and biologists engaged in the production and application of atomic energy to peaceful uses; for post-graduate and undergraduate students of higher technical schools whose nuclear science is taught; and for the general public interested in atomic science and technology.

COVERAGE: This is volume 6 of a 6-volume set of reports submitted by Soviet scientists at the Second International Conference on the Peaceful Uses of Atomic Energy held in Geneva from September 1 to 13, 1958. Volume 6 contains 24 reports on: 1) modern methods for the production of stable and radioactive isotopes and their labeled compounds, 2) research results obtained with the aid of isotopes in the field of chemistry, analytical chemistry, biology, medicine, and agriculture, and 3) industry of isotopes.

6 was edited by: S.Y. Levinitskiy, Candidate of Physical-Mathematical Sciences, Institute of Chemistry, USSR Academy of Sciences; V.I. Prusakov, Candidate of Chemical Sciences; and V.I. Sobolev, Candidate of Medical Sciences. See SOV/2001 for titles of volumes of the set. References appear at the end of the articles.

5. Yakovlev, G.M. and V.B. Dolov. Means of Developing Remote Control Methods in the Radiochemical Laboratories of the A.S. Zinov'ev (Report No. 2025)

6. Mallov, M.P., A.G. Zaidovich, A.B. Prudkov, and I.B. Denilov. General Principles of Production of Deuterium by the Low-Temperature Distillation Method (Report No. 2027)

7. Gverditskii, I.G., R.Ya. Gubarev, and V.K. Tshakaya. Separation of Isotopes by Diffusion in a Steam Flow (Report No. 2028)

8. Zolotarev, V.S., A.I. Il'ina, and Ye.O. Kozlov. Separation of Isotopes on Electromagnetic Units in the Soviet Union (Report No. 2055)

9. Alekseyev, B.A., B.Y. Buligin, V.S. Zolotarev, B.Y. Papis, Ye.S. Chernomir, and G.Ya. Sherepin. Separation of Isotopes of Rare-earth Elements by the Electromagnetic Method (Report No. 2017)

10. Novosov, P.M., B.M. Mator, N.S. Ioffe, B.G. Bruchner, and G.M. Franklin. Ion Sources for the Separation of Stable Isotopes (Report No. 2053)

11. Kozlov, M.Y. and P.M. Novosov. Electric Field Effect in Ion Beams on Stable Isotope Separation by the Electromagnetic Method (Report No. 2054)

12. Shteynman, M.O., P.L. Grutzin, G.I. Yemelyayev, and I.D. Sivillinskiy. Use of Radioactive Tracers in Metallurgical Research (Report No. 2018)

13. Smalilovskiy, N.N., V.A. Yamshchikov, and I.M. Tatarskiy. The Theory and Practice of Pulse-type Instruments Based on Radioactive Isotopes (Report No. 2032)

14. Zaslavskiy, Yu.S., G.I. Ghor, and B.M. Shchepetov. Studying the Mechanism of Protection of Rubbing Surfaces Against Wear Due to Corrosion (Report No. 2028)

15. Baryshnikov, S.V. and L.S. Matsyuk. The T0170, Ra155, and Ce144 as Sources of Radiation for Checking Thin-valved Products (Report No. 2035)

16. Krak, B.I., A.S. Zavyalov, and G.I. Kaprin. Studying the Radioisotopic Titration of Elements in Metal Alloys and Their Compounds by Autoradiographic and Radiometric Methods (Report No. 2036)

17. Grutzin, P.L., A.I. Vostyubskiy, V.S. Yemelyayev, G.O. Ryabov, G.B. Fedorov. Studying the Diffusion and Distribution of Elements in Alloys of Zirconium and Titanium Based by the Radioactive Isotope Method (Report No. 2036)

G. RUZIN, P. L.

PHASE I BOOK EXPLOITATION 52/3559

Abstrakts nauk SSSR. Institut metallurgii. Nauchnyy sovet po probleme naruzhnykh sloyev
 Issledovaniya po sverkhochuvstviyu, z. 5 (Investigations of Heat-Resistant Alloys, Vol 5) Moscow, Izd-vo AN SSSR, 1979. 423 p. Errata slip inserted. 2,000 copies printed.

Ed. of Publishing House: V.A. Kuz'min; Tech. Ed.: I.P. Kur'shin; Editorial Board: M. A. Galitskiy, Academician, G.V. Kur'yakov, Academician, N.V. Agayev, Corresponding Member, USSR Academy of Sciences (Resp. Ed.), I.A. Oling, I.M. Perlov, and I.F. Zadko, Candidate of Technical Sciences.

PURPOSE: This book is intended for metallurgical engineers, research workers in metallurgy, and may also be of interest to students of advanced courses in metallurgy.

CONTENTS: This book, consisting of a number of papers, deals with the properties of heat-resisting metals and alloys. Each of the papers is devoted to the study of the factors which affect the properties and behavior of alloys. The effects of various elements such as Cr, Mo, and V on the heat-resisting properties of various alloys are studied. Deformability and workability of certain metals as related to the thermal conditions are the object of another study described. The effect of hydrogen embrittlement, diffusion and the deposition of oxides on metal surfaces by means of electrochemical methods. One paper describes the apparatus and methods used for metallographic studies of metals. Boron-base metals are critically examined and evaluated. Results are given of studies of interatomic bonds and the behavior of atoms in metal. Tests of turbine and compressor blades are described. No personalities are mentioned. References accompany most of the articles.

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BABIKOVA, Yu.F.; GRUZIN, P.L.

Use of the radioactive isotope C^{14} to study the electrolytic transfer of carbon in metals and alloys. Met. i metalloved. chist. (MIRA 12:10)
met. no.1:200-212 '59.
(Electrolysis) (Carbon--Isotopes)

25551

S/137/61/000/009/026/087
A060/A101

18 7500

AUTHORS: Borisov, Ye.V., Gruzin, P.L., Pavlinov, L.V., Fedorov, G.B.
TITLE: Self-diffusion of molybdenum and diffusion of tungsten in molybdenum
PERIODICAL: Referativnyy zhurnal. Metallurgiya, no. 9, 1961, 3, abstract 9Zh11
("Metallurgiya i metallovedenie chist. metallov", no. 1, Moscow, 1959, 213 - 218)

TEXT: The self-diffusion of Mo and the diffusion of W in Mo were studied in the interval 1,800-2,175°C by the use of radioactive isotopes Mo⁶⁹ and W¹⁸⁵. The annealing time was from a few hours up to tens of hours. Samples of Mo obtained by arc-smelting in vacuum were subjected to preliminary annealing at 1,500°C for a period of 20 hrs in a H₂ atmosphere. To determine the diffusion coefficient the method of measuring the total activity of the sample remainder was used. The following temperature dependence was obtained for the diffusion coefficient for Mo self-diffusion: $D = 4 \exp(-115,000/RT) \text{ cm}^2/\text{sec}$. For diffusion of W in Mo it was found that $D = 5 \cdot 10^{-4} \exp(-78,000/RT) \text{ cm}^2/\text{sec}$.

[Abstracter's note: Complete translation]

A. Rusakov

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Card 1/1

28552

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AC 60/A101

18.7500

AUTHORS: Gruzin, P.L., Fedorov, G.B.

TITLE: Diffusion in heat-resistant cobalt-base alloys

PERIODICAL: Referativnyy zhurnal. Metallurgiya, no.9, 1961, 3-4, abstract 9Zh17 (V sb. "Metallurgiya i metalloved. chist. metallov", no. 1, Moscow, 1959, 219 - 223)

TEXT: The diffusion of Co-Cu-W-Mo alloy BK 36 (VK36) under deformation was studied. The diffusion of both the elements entering in the base of the alloy (Co) and of those used for heat-proofing (C and Mo) was investigated. To investigate the influence of C and Mo upon the diffusion constant, an alloy with a minimum C content (0.025%) was prepared, and also a similar alloy without Mo. After casting the alloys were subjected to a preliminary compression to break down the dendritic structure. Diffusion annealing in H₂ at 1,100°C for 50 hrs was then carried out. Thereupon the ingots were forged into bars from which the specimens were prepared. Before the diffusion annealing a thin layer of the radioactive element (Co⁶⁰, Cr⁵¹, W¹⁸⁵) was applied to one of the sides of the specimen. Diffusion annealing proceeded at 850-1,250°C in a tubular furnace. The diffusion co-

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Diffusion in heat-resistant cobalt-base alloys

efficients were measured by determining the total activity of the remaining portion of the sample after parallel layers were removed from it. The study of Co self-diffusion in the alloys investigated has shown that an addition of C reduces somewhat the activation energy and increases the self-diffusion coefficient. The admixture of Mo raises the activation energy and reduces the self-diffusion coefficient of Co. The increased tendency towards carbide formation determines the highest activation energy of Cr in an alloy with the maximum Cr content. Thus, the presence of Cr in these alloys raises the activation energy somewhat and reduces the diffusion coefficient of Cr. The retardation of Cr diffusion is the essential factor improving the heat-resistance properties of the VK₆ alloy.

A. Rusakov

X

[Abstracter's note: Complete translation]

Card 2/2

807/133-59-4-2/32

AUTHORS: ~~Grazin, P.L.~~, Doctor of Phys.-Mathematical Sciences,
Professor, Afanas'yev, V.N., Engineer, and Zemskiy, S.V.

TITLE: Investigation and Control of the Blast Furnace Process
Using radioactive Isotopes and Radiations
(Issledovaniye i kontrol' domennogo protsessa s
pomoshch'yu radioaktivnykh izotopov i izlucheniya)

PERIODICAL: Stal', 1959, Nr 4, pp 291-297 (USSR)

ABSTRACT: A brief review of the application of radioactive sources
in the USSR for the investigation and control of the
blast furnace process is given. The following
applications are briefly discussed: the determination of
the rate of descent and residence time of burden
materials in a blast furnace; control of the rate of
wear of the blast furnace lining; studies of the
movement of gases in a blast furnace (in this respect a
method of utilising inert gases with subsequent
application of a mass-spectroscope for the analysis of
the top gas instead of radon, developed by TsNIICHM is
mentioned); control of stock level and of the quality
of coke and sinter charged to skips (the latter relates

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SOV/133-59-4-2/32

Investigation and Control of the Blast Furnace Process Using
Radioactive Isotopes and Radiations

to an indirect indication of the size distribution of sinter and coke based on the absorption of γ radiation of the layer of the above materials in the weighing hopper). It is concluded that a further effort should be made in applying radiometric methods for research and control purposes with special attention to utilising isotopes with a short radioactive life. There are 6 figures and 27 references of which 22 are Soviet, 4 English and 1 German.

ASSOCIATION: TsNIICHM

Card 2/2

ALIYAROVA, Z.A.; GRUZIN, P.L.; ZEMSKIY, S.V.

Studying the distribution of an admixture of sulfur in selenium
by autoradiography. Dokl. AN Azerb. SSR 15 no. 6:467-471 '59.
(MIRA 12:9)

1. Predstavleno akademikom AN Azerbaydzhanskoy SSR Z.I. Khalilovym.
(Sulfur--Analysis) (Selenium--Analysis)
(Autoradiography)

GRUZIN, P.L., doktor fiz.-mat. nauk, otv. red.; ERYANTSEVA, V.P., inzh., ved. red.; SHKOVSKAYA, I.Yu., inzh., ved. red.; SINITSYN, V.I., inzh., nauchnyy red.; LADONINA, L.V., tekhn. red.

[Use of radioactive isotopes and nuclear radiations in hydraulic engineering and construction] Primenenie radioaktivnykh izotopov i iadernykh izlucheni' v gidrotekhnike i stroitel'stve. Moskva, (Peredovoi nauchno-tekhnicheskii i proizvodstvennyi opyt. Tema 19) No.14. 1960. 35 p. (MIRA 15:3)

1. Moscow. Institut tekhniko-ekonomicheskoy informatsii.
(Construction industry) (Hydraulic engineering)
(Radioactive substances--Industrial applications)

13.7500

001, 13.7500-1774

AUTHORS: Grunin, P. L., Shvachov, A. M., Martinson, V. G.,
Polikarpov, Yu. A.

TITLE: Study of the Dependence of Self-Diffusion Coefficient
of Iron in Steel on Temperature

PERIODICAL: Investiya vysshikh izhobnykh nauchnykh issledovaniy. (Zhurnal)
metallurgiya, 1960, No 1, pp 107-110 (USSR)

ABSTRACT: Previous experiments of P. L. Grunin proved that the
diffusion coefficients for Fe, Cr, and Mn in Armco iron
and steel with 0.8% C, obey the rule of exponential
dependence at 1,100 to 1,250°C, but exhibit deviations
from the rule below 1,100°C. The authors checked and
specified more precisely the coefficient deviations
in the case of self-diffusion of Fe and tried to deter-
mine the extent to which the self-diffusion coefficient
for Fe depends on the size of austenite grains. Steel
U7 containing 0.75% C, i.e., close to that at eutectic
point, was selected for this purpose, since it is the
best suitable steel for the study of self-diffusion at

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Study of the Dependence of Self-Diffusion Coefficient of Iron in Steel on Temperature

11100
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the lowest possible temperatures. The self-diffusion coefficients were determined according to the radio-activity of different layers caused by the diffusion of radioactive Fe-54 isotope. This had been deposited electrolytically prior to a high-temperature treatment for grain growth (or without such treatment) and subsequent annealing at 300 to 1,000°C for 10 to 300 hr. The grain size of austenite was measured in parallel specimens having exposed their grain boundaries by evaporation (thermal etching) in vacuum for 5 to 12 hr. The experiments proved that self-diffusion coefficient D for Fe depends on temperature as illustrated in Fig. 2 and is defined by

$$D = 2 \cdot 10^{-2} \exp\left(-\frac{52000}{RT}\right) \text{ cm}^2 \text{ s}^{-1}$$

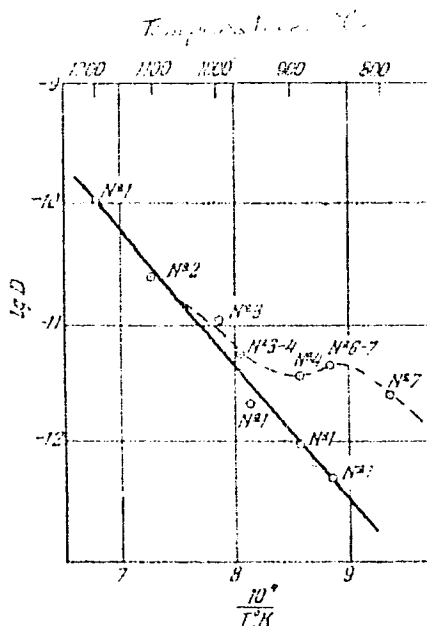
If the treatment fails to make austenite equiaxial, for instance at low temperatures, the equation does not hold. Figure 1 illustrates this case on the right side, where the grain size of austenite shows anomalously high diffusion coefficients at low temperatures. This is the combined effect of: (1) diffusion of grain boundaries where grain growth is observed with the decreasing grain size; (2) change in austenite grain

Cont 2/4

Study of the Temperature of Self-Diffusion
Coefficient of Iron in Steel as a Function of Temperature

NOV/148-60-1-23/34

Fig. 2. Self-diffusion coefficient as a function of temperature. (Nos 1 to 7 denote relative grain sizes, 7 being the smallest). specimens which have undergone a high-temperature treatment for grain growth; ---- specimens without such treatment.



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Study of the Dependence of Self-Diffusion
Coefficient of Iron in Steel on Temperature

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seem to be incomplete or instable at lower temperatures of annealing. In conclusion, further experiments are recommended in which steels: (1) of identical composition but with different tendency to grain growth; and (2) undergoing no phase transformations, should be used. Such experiments would permit one to establish diffusion coefficients depending on grain size and phase transformations separately, instead of both factors combined. There are 2 figures; 1 table; and 4 references, 3 Soviet, 1 U.S. The U.S. reference is: H. W. Mead, C. T. Birchenal, J. of Metals, 1956, 8, Nr 10, 1336.

ASSOCIATION: Moscow Machine Construction Evening Institute and the Central Scientific Research Institute of Ferrous Metallurgy (Moskovskiy vecherniy mashinostroitel'nyy institut i TsNIIChM)

SUBMITTED: November 4, 1958

Card 4/4

RYABOVA, G.G.; BABIKOVA, Yu.F.; GRUZIN, P.L.

Distribution and electrodiffusion of tin in zirconium alloys.
Met. i metalloved. chist. met. no. 2:115-127 '60. (MIRA 13:12)
(Zirconium alloys--Metallography)
(Diffusion)

23061 3/137/61/000/004/22/000
ADSE/AD01

188100

1160 1418 1138

AUTHORS: Babikova, Yu. F., Gruzina, P. L.

TITLE: Electrodiffusion of carbon in zirconium

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 4, 1961, 4-6, pp. 128-133
(V sb. "Metallurgiya i metallovedeniye mistykh metallov", P. M. Atomizdat, 1960, 128-133)

TEXT: The author gives the data of the study of carbon in zirconium under the action of an electric field. The samples were manufactured of Zr-foils. In the central part of the samples was produced a zone of 0.3-3 mm length, containing a radioactive isotope of the element the transfer of which was studied. This zone is, for high temperatures, a source of dissolved carbon transfer. For the study of the C-transfer, the zone was formed by means of pin point welding of a Zr-wire containing the radioactive isotope. After the formation of the radioactive zone on the sample the distribution of radioactive on both sides of the zone, constituting the origin of the reading, was determined. The electrodiffusive annealing of the samples was conducted in a special cell under a vacuum of $\sim 10^{-5}$ mm/Hg, at 900, 1,000, 1,100 and 1,200 C during 2, 4, 8.

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A09/1101

Electrodiffusion of carbon in zirconium

and 4 hours respectively. The samples were heated by a d.c. of corresponding density. After annealing a redistribution of radioactivity was determined in the cathodic and anodic sections of the samples. In each sample a quantity of C to the cathode was displayed. This points to the fact that C is present in the Zr-crystal lattice in ionized state in the form of carbon. Comparison of the temperature-dependence obtained of the C-transfer number with the data for unalloyed austenite shows that the electroactivity of C in Zr is about 10² times greater (possibly < 1). The transfer numbers for 900 and 1,000 °C are of a high value in comparison with that which could be expected from the data for 1,100 and 1,200 °C. Such anomaly of dependence of the temperature coefficient is explained by the influence of the structure factor - temperature dependence at high temperature. The results obtained prove that the distribution of C is possibly its transfer are chiefly of a boundary nature. This fact is confirmed by the data of autoradiography.

A. B.

[Abstractor's note: Complete translation]

Card 2/2

18 7500

29032
S/081/61/000/018/002/027
B104/B101

AUTHORS: Gruzin, P. L., Ryabova, G. G.

TITLE: Study of the influence of the structure factor on diffusion in zirconium and its alloys

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 18, 1961, 31, abstract 18B196 (Sb. "Metallurgiya i metallovedeniye chistykh metallov". M., Atomizdat, no. 2, 1960, 134-140)

TEXT: The influence of the structure factor on self-diffusion of Zr and diffusion of Sn in zirconium alloys was studied. The alloys were smelted down in an arc furnace and subjected to a homogenizing annealing for 10 hr. The self-diffusion coefficient and the diffusion coefficient were determined by using Zr⁹⁵ and Sn^{113,123} isotopes, and the method of removing layers, and by measuring the radioactivity of the rest. In order to study the influence of the structure factor on self-diffusion in β -Zr, some of the specimens were annealed by the conventional method. The other specimens were heated to 1200°C, and kept at this temperature for

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Study of the influence of the...

20 minutes. Subsequently, the specimens were cooled to the temperature of diffusion annealing and kept at this temperature. The self-diffusion parameters were found to depend on the structure state. This dependence is the greater, the lower the temperature. The influence of the structure (hardening and annealing) on Sn diffusion in Zr-Sn alloys in the range of α -solutions (0.5; 1, and 2% Sn) was studied. In all cases, the diffusion coefficient of Sn in specimens with martensite structure was greater than in annealed specimens. With an increase of the Sn content in the alloys and with a temperature drop, the difference between the diffusion coefficients increased for both structure states.

[Abstracter's note: Complete translation.]

W

Card 2/2

GRUZIN, P.L.: POLIKARPOV, Yu.A.

Studying the effect of structural factors on diffusion in
steels and alloys by means of radioactive isotopes. Met.
i metalloid. chist. met. no. 2:259-275 '60. (MIRA 13:12)
(Steel alloys--Metallography)
(Radioisotopes--Industrial applications)

GRUZIN, P.L.; YEVSTYUKHIN, A.I.; ZEMSKIY, S.V.; NIKISHANOV, V.V.

Investigating the redistribution of sulfur during the zonal
melting of chromium in arc furnaces. Met. i metalloved.
chist. met. no. 2:276-279 '60. (MIRA 13:12)
(Chromium--Metallography) (Sulfur--Isotopes)

S/129/60/000/010/002/009
E193/E483

AUTHOR: Gruzin, P.L., Doctor of Physico-Mathematical Sciences,
Professor

TITLE: Certain Laws Relating to Diffusion¹⁶ and Distribution of
Elements in Alloys

PERIODICAL: Metallovedeniye i termicheskaya obrabotka metallov,
1960, No.10, pp.5-13 + 3 plates

TEXT: In a general review of the present state of knowledge of
diffusion processes in metallic materials, the following subjects
are discussed: (1) Application of the auto-radiographic¹⁹ method
in studies of diffusion phenomena with particular reference to
work carried out by V.M.Golikov, V.T.Borisov and B.Ya.Lyubov,
who have derived an expression for the kinetic curve of integral
radio-activity of diffusion test pieces. (2) The exponential
character of the temperature-dependence of the self-diffusion
coefficients of practically all metals above their recrystallization¹⁸
temperature, as illustrated by graphs reproduced in Fig.1 and 2.
(3) The relationship between the diffusion characteristics and
other properties of metals such as the melting point and
recrystallization temperature, and the effect of alloying
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E193/E483

Certain Laws Relating to Diffusion and Distribution of Elements in Alloys

additions on this relationship. (4) The anomalous temperature-dependence of the self-diffusion coefficients of iron, titanium and zirconium, with particular reference to the effect of the structural factor on diffusion processes. (5) The findings of Yu.F. Babikova, A.D. Tyutyunnik, G.G. Ryabova and the present author, who have studied localized concentration of the alloying additions in titanium and zirconium. (6) The effect of alloying additions on diffusion processes in iron, nickel, chromium and various other metals, studied by L.M. Mirski, S.Z. Bokshteyn and S.T. Kishkin. (7) The effect of phase transformations on the diffusion process, and the anomalous variation of the coefficient of diffusion of cobalt and iron in steel U8 at temperatures below 1100°C (Fig. 7 and 9), as studied by A.P. Gulyayev, Yu.A. Polikarpov, Yu.D. Zharov, V.G. Martinson and the present author. (8) The effect of other alloying elements (titanium, boron) on diffusion of cobalt in steel U8. (9) The effect of the structural factor (grain size and the degree of fragmentation of grains) on self-diffusion in austenite, as studied by Ye.Z. Vitaykin, V.D. Sadovskiy, V.M. Golikov,
Card 2/4

S/129/60/000/010/002/009
E193/E483

Certain Laws Relating to Diffusion and Distribution of Elements in Alloys

V.T.Borisov and G.V.Shcherbedinskiy. (10) The effect of various alloying additions (silicon, molybdenum, nickel, chromium, cobalt) on diffusion of carbon in steels; (compare Table 2, giving the magnitude of the pre-exponential factor D_0 and activation energy Q and Fig.10 and 11, showing the temperature-dependence of the coefficient of diffusion of carbon in various steels, as studied by M.Ye.Blanter, S.Z.Bokshteyn, S.V.Zemskiy, F.R.Florensova, Ye.M.Morozova and the present author).

(11) The findings of Yu.A.Polikarpov and the present author on the effect of the structural factor on diffusion of carbon in the Fe-Ni-Mn alloys. (12) The results of an investigation, carried out by Yu.F.Babikova and the present author, of the state of the carbon atoms in iron, nickel, zirconium and some other alloys, according to which carbon atoms are present in ferrite and nickel in the form of cations. (13) The effect of alloying additions (silicon, chromium) on the degree of ionization of carbon in nickel and iron. (14) Electron transfer from

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E193/E483

Certain Laws Relating to Diffusion and Distribution of Elements in Alloys

carbon in titanium and zirconium. There are 12 figures and 2 tables. ✓

ASSOCIATION: TsNIChM

Card 4/4

FEDOROV, G.B.; BABIKOVA, Yu.F.; GRUZIN, P.L.; ZHOMOV, F.I.; RYABOVA, G.G.

Radioactive-tracer techniques in the study of the mobility, interatomic interaction, and distribution of elements in zirconium and its alloys. Izv.vys.ucheb.zav.;khim. i khim.tekh. 3 no.3: 395-401 '60. (MIRA 14:9)

1. Moskovskiy inzhenerno-fizicheskiy institut, kafedra metallurgii i metallovedeniya.
(Zirconium alloys) (Radioactive tracers)

18.7000,21.1200

77220
SOV/89-8-1-14/29

AUTHORS: Gruzin, P. L., Ryabova, G. G., Fedorov, G. B.

TITLE: Iron Distribution in Microvolumes of Zirconium Alloys.
Letter to the Editor

PERIODICAL: Atomnaya energiya, 1960, Vol 8, Nr 1, pp 56-59 (USSR)

ABSTRACT: The use of zirconium in nuclear power reactors is very much reduced because of its poor strength and stability against corrosion. Although it is a well established fact that small impurities of different elements can decrease or increase its stability, little is known about the mechanism of these influences. Investigation of element distributions in alloys could, therefore, be very helpful, and the authors undertook to study the distribution of iron, which causes an extreme reduction of stability against corrosion especially in iodine containing zirconium. They used zirconium alloy with 0.15 weight

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Iron Distribution in Microvolumes of
Zirconium Alloys. Letter to the Editor

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% of iron and zircalloy type alloys (on the iodine zirconium basis) containing (in weight %): 1.1 lead, 0.1 iron, 0.1 chromium, and 0.05 nickel, and investigated the distribution of iron by means of contact autoradiography. Powdered radioactive isotope Fe^{59} was introduced into the alloy melted by means of an arc in the atmosphere of argon. Selfradiograms were taken on the MR type NIKFI plates by means of the 1.295 mev γ -ray and the 0.46 mev γ -ray of Fe^{59} . To insure sufficient resolution, sample thickness was of the order of a few tenths of a micron, which supplied a 5 to 10,000 imp/min \cdot cm² intensity of radiation. Exposure time depended on the particular setup. The authors found that the iron distribution in the cast zirconium alloy was nonhomogeneous even after various thermomechanical treatments. Largest part of the iron remains concentrated on the boundaries between blocks obtained during the $\beta \rightarrow \alpha$ phase transition, and another part remains in the solid solution having not enough

Card 2/5

Iron Distribution in Microvolumes of
Zirconium Alloys. Letter to the Editor

77-55
507

time to separate out of the alloy during its fast cooling. Similar iron distribution was recorded in cast alloy-2. A cold 5-10% deformation produces almost the same picture as in cast alloys. Annealing of cold deformed alloys achieves a transfer of iron from solid solution to the α -phase boundaries. Forging hot alloys in air at temperatures between 350 and 750° C results in a strong granulation of their structure but the iron inhomogeneity remains. Thermal treatments of cast and hot-forged alloys occurred inside quartz tubing evacuated to approximately 10^{-4} mm Hg. Tempering the alloys from various temperatures from β -region, the iron distribution stayed similar to those in cast alloys. This follows from the fact, pointed out by Hayes and others, that at high temperatures (in the β -region) iron is in solid solution and fast cooling leads only to its partial separation on the boundaries developed during the $\beta \rightarrow \alpha$ transition. Slow cooling in oven from the

Card 3/5

Iron Distribution in Microvolumes of
Zirconium Alloys. Letter to the Editor

7722
SOV 86-1-1-12

β -region temperatures leads to a more complete separation of iron along the boundaries and sub-boundaries of the α -phase. A sample of cast, unforged zirconium-iron alloy tempered at its eutectoid temperature of 800° C showed almost complete separation of iron from the solution in the form of the intermetallide $ZrFe_2$ distributed along the boundaries and inside the grains of the α -phase of zirconium. Annealing zirconium-iron alloy in the α -region at 600° C (20 h) and 500° C (40 h) after tempering at 1200 and 900° C preserved the inhomogeneity of the iron distribution. In zircalloy-2 the redistribution of iron proceeds at a slower rate than in pure zirconium. This is probably due to the presence of various elements with a distribution pattern similar to that of iron according to preliminary data. One can assume that the higher resistance to corrosion of the zirconium-lead compared with zirconium-iron alloys is due to the fact that lead distributes uniformly in zirconium, increasing

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Iron Distribution in Microvolumes of
Zirconium Alloys. Letter to the Editor

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thereby the resistivity of the solid solution (representing the basis of the alloy), while Fe, Cr, and Ni concentrate on the boundaries of grains and blocks, slowing down the corrosion at the boundaries. There are 8 figures; and 7 references, 5 Soviet, 2 U.S. The U.S. references are: B. Lustman, F. Kerze, The Metallurgy of Zirconium, London, McGraw-Hill Book Co., 1955, p 608; E. Hayes, A. Roberson, W. O'Brien, Trans. Amer. Soc. Metals, 43, 888 (1951).

SUBMITTED: August 5, 1959

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S/089/60/009/003/011/014
B006/B063

AUTHORS: Gruzin, P. L., Babikova, Yu. F.

TITLE: Application of Radioactive Isotopes¹⁹ and Nuclear Radiations
in Metallurgy

PERIODICAL: Atomnaya energiya, 1960, Vol. 9, No. 3, pp. 223 - 225

TEXT: The present article gives a survey of the Soviet plants at which radioisotopes and nuclear radiation sources are successfully used as well as of the various fields of application. Radioisotopes and nuclear radiation were first used in metallurgy in 1948-49. Among the various fields of application are to be mentioned the use of isotopes as indicators (tracer technique), nuclear radiation as a power source, and the use of isotopes for the control and automation of technological processes. The Novo-Tul'skiy metallurgicheskiy zavod (Novo-Tula Metallurgical Plant) and the Kuznetskiy metallurgicheskiy kombinat (Kuznetsk Metallurgical Plant) were the first to use radioisotopes. One of the principal problems in the production of pig-iron was to find new methods for an increase in the efficiency of blast furnaces and the automation of

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individual units. In this connection, radioisotopes were used for control purposes, especially for the determination of the most favorable conditions. The plant imeni Dzerzhinskiy achieved great success with the use of a radiometric instrument that checks the charge level in blast furnaces. This plant in conjunction with the Tsentral'nyy nauchno-issledovatel'skiy institut chernoy metallurgii (Central Scientific Research Institute of Ferrous Metallurgy) developed and tested a method of checking the charge material. Analogous work was carried out at the Kuznetsk Metallurgical Kombinat. Further work has been carried out for the control of the enrichment of iron ore and the sintering of agglomerates. The Kuznetsk Metallurgical Kombinat developed a radiometer that is suited for determining the density of agglomerates. The Kuznetsk Metallurgical Kombinat, the plants imeni Dzerzhinskiy, "Azovstal", Makeyevskiy metallurgicheskii zavod (Makeyevka Metallurgical Plant), and others employ radiometric methods to check the wear and tear of the fireclay lining of blast-furnaces. The annual amount saved by radiometric control at the Plant imeni Dzerzhinskiy and at the Institut ekonomiki AN SSSR (Institute of Economics of the AS USSR) was calculated to be some million rubles. The Magnitogorskiy metallurgicheskii kombinat im. Il'icha (Magnitogorsk

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Metallurgical Kombinat imeni Il'ich) is also mentioned in this connection. Radioisotopes are also much used in steel production. Slag formation in open-hearth furnaces was investigated at the Magnitogorsk Metallurgical Kombinat by a radiometric method. Similar investigations were performed at the "Azovstal'" Plant and the Stalinskiy metallurgicheskiy zavod (Stalino Metallurgical Plant). The two last-mentioned plants and the "Zapcrozhtal'" Plant use radioisotopes to determine the melting rate and to check the weight of liquid steel during the melting process in open-hearth furnaces. The nature of non-metallic inclusions in steel was examined with the help of radioisotopes at the plants of the Kuznetsk and Magnitogorsk Metallurgical Kombinats as well as at the kombinat im. Serova (Kombinat imeni Serov), the Stalino and Chelyabinsk Plants, the plants "Serp i molot", "Dneprospetsstal", "Elektrostal", and at various institutes. The Kuznetsk and Makeyevka Metallurgical Kombinats and the "Azovstal'" Plant devised a method of checking the wear and tear of the fireclay lining of open-hearth furnaces. Radiometric examinations of the flow of metal during the rolling process were carried out at the Kuznetsk Metallurgical Kombinat and the Stalino Metallurgical Plant. Iron and tin radioisotopes were used to develop a new tinning technique which was

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tested by the Novo-Moskovskiy zavod (Novo-Moskovsk Plant) and the "Zaporozhstal" Plant. The Central Scientific Research Institute of Ferrous Metallurgy designed a radiometric level-meter for liquid metal. Such level-meters and level-regulators are in use at the Khartsyzskiy trubnyy zavod (Khartsyzsk Tube Mill), the Kalinin Plant imeni May 1, the Sinarskiy trubnyy zavod (Sinara Tube Mill), and the Mogilevskiy metallurgicheskii zavod (Mogilev Metallurgical Plant). The Leningradskiy staleprokatnyy zavod (Leningrad Steel Rolling Mill) and the Izhorskiy zavod (Izhora Plant), for example, made considerable savings by using radiometric thickness gauges and controllers for rolled stock. The non-ferrous metallurgical industry also makes use of radioisotopes for control purposes, such as, e.g., the Yuzhno-Ural'skiy nikelovyy kombinat (South Ural Nickel Kombinat). The Volkhovskiy alyuminiyevyy zavod (Volkhov Aluminum Plant) and the Dneprovskiy titano-magniyevyy zavod (Dnepr Titanium - Magnesium Plant) use radioactive densimeters for automation purposes. The use of radioisotopes for the production of high-purity metals was tested by the plants "Ukrtsink" and Novosibirskiy olovozavod (Novosibirsk Tin Plant).

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S/137/62/000/007/058/072
A160/A101

AUTHORS: Babikova, Yu. F., Ryabova, G. G., Gruzin, P. L.

TITLE: The distribution of carbon admixture in zirconium and titanium

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 7, 1962, 72, abstract 7I477
(In collection: "Metallurgiya i metalloved. chist. metallov". no. 3,
Moscow, Gosatomizdat, 1961, 28 - 33)

TEXT: The method of contact autoradiography was used in the work. Radio-
active $BaCO_3$ salt was introduced in a hole drilled in a Ti or Zr ingot. The hole
was plugged with a stopper from a corresponding metal and the ingot remelted in
an arc furnace. The content of C^{14} in Zr was $\leq 0.005\%$, and the total content of
C in commercial Ti ($C^{12} - C^{14}$) was $\leq 0.2\%$. In the martensitic structure of Zr and
Ti, C distributes along the boundaries of the β -phase grains and along the interior
boundaries developed as a result of martensitic conversion. In the slowly-cooled
 β -zone of Zr, C distributes along the boundaries and subboundaries. In Ti, a si-
milar thermal treatment leads to the formation of a δ dispersion phase rich with
C. After the annealing, the distribution of C in Zr and Ti in the α -zone ($800^\circ C$,
 ~ 20 hours) becomes uniform, which is apparently caused by a high mobility of C
atoms. The considerable non-uniformity in the distribution of C in Zr and Ti,
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cooled from the β -zone, is connected with the martensitic nature of the $\alpha \rightarrow \beta$ -conversion. During the conversion, a large number of blocks arise inside the initial grain, due to the shear processes. Along their boundaries, an accumulation of admixtures, and especially of C, takes place. The non-uniform distribution of the admixtures in the solid solution leads to the fact that their effect on the properties proves to be non-proportional to their mean concentration in the metal.

P. Novik

[Abstracter's note: Complete translation]

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S/755/61/000/003/0107027

AUTHORS: Ryabova, G. G., Gruzin, P. L.

TITLE: Study of the distribution of various elements in zirconium and its alloys by means of radioautography.

SOURCE: Moscow. Inzhenerno-fizicheskiy institut. Metallurgiya i metallovedeniye chistyykh metallov. no.3. 1961, 96-119.

TEXT: The paper describes a series of tests in which contact radioautography (RA) was employed to investigate the distribution (D) of Sn, W, Fe, Ni, Cr, Nb, and C in Zr and some of its alloys. It was found that the character of the D of Fe, Ni, Cr, Nb, and C is substantially affected by the existence of a polymorphic transformation (PT) in Zr, which leads to the formation of subgranular concentration non-uniformities in the D of these elements, but not in that of Sn and W. It is shown that certain other factors, such as deformation and heat treatment, affect the D of the various admixtures. A brief state-of-the-art survey on the effect of various impurities on the properties of Zr and its alloys is based primarily on 3 references: (1) "The metallurgy of Zr," (B. Lustman, F. Kerze, Jr., eds. McGraw-Hill, 1955; For. Lit. Publ. House, Moscow, 1959); (2) "Materials of the U. S. AEC," v. III - Nuclear Reactors, For. Lit. Publ. House, Moscow, 1956); (3) Ambartsumyan, R. S.,

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et al. (In Trans. 2d Intern'l Conf. on the Peaceful Uses of Atomic Energy, Geneva, 1958, v.3, Atomizdat, 1959). The test alloys were smelted in the MIFI-9-3 arc furnace in an atmosphere of purified Ar. Radioactive isotopes were introduced into the melt. The D of the various elements was studied after casting and following various work and heat treatments performed with all necessary precautions against the admittance of gases and other impurities to the alloys. RA was performed either with ordinary specimens or with thin (10-100 μ) specimens, depending on the radiation intensity of the isotope, which ranged from 2,000-30,000 pulses/cm²·min. NIKFI MK and MR film was used. Exposures ranged from 2 to 20 days. Sn: Some of the test results with Zr alloys and Zircalloy-type alloys containing 0.3-2.0 wt-% Sn were previously published by the senior authoress et al., in no.2 of the present sbornik, 1960, 128, and by G. B. Fedorov, et al. in Izv. vyssh. uchebn. zav. SSSR, "Khimiya i khim. tekhnologiya," no.3, 1960, 295. The dendritic liquation of Sn in cast specimens, increasing with Sn concentration and more pronounced in the Zircalloy-type alloy, is noted. Heating and holding at β -phase T's eliminated this nonuniform D and yielded a homogeneous solid solution. Hot forging at 800-850°C intensified the dendritic liquation of Sn. A 19-hr 620°C anneal of cold-worked alloys did likewise, despite recrystallization. W: The W concentrated in the interdendritic regions. Hot forging as well as cold-working with subsequent 12-hr 700°C anneal preserved the intracrystalline liquation of the W despite complete

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