

GLUSHOV, A.A.

Principles of designing automatic quality control systems for
radiobroadcasting apparatus. Izv. vys. ucheb. zav.; radioelektr.
(MIRA 13:10)
no.4:463-370 Sl-ag '60.

1. Rekomendovani kafedroy radioveshchaniya i akustiki Moskovskogo
elektrotekhnicheskogo instituta svyazi.
(Information theory) (Radiobroadcasting)

GLUKHOV, A.A.

Study of the effect of phase distortions on the instantaneous value
of the signal in the radio broadcasting channel. *Elektronviaz'* 15
no.1:33-39 Ja '61. (MIRA 14:3)
(Information theory)

GLUKHOV, A. A.

Card Tech Sci - (diss) "Problems of the automatic control of the quality of performance of broadcasting equipment." Moscow, 1961. 14 pp; with illustrations; (Ministry of Communications USSR, Moscow Electrical Engineering Inst of Communications); 150 copies; price not given; (KL, 6-61 sup, 215)

3/106/60/000/009/000/XX
A055/A125

6,4000

AUTHOR: Glukhov, S. A.

TITLE: On the automatic monitoring of a broadcasting station

PERIODICAL: Elektrosvyaz', no. 9, 1960, 64 - 67

TEXT: The author presents the scheme of V. A. Nyurenberg [Ref. 1: "Ustroystvo dlya avtomaticheskogo kontrolya raboty veshchatel'nykh ustanovok", Vestnik Svyazi, no. 2, 1959] for an automatic monitor, in which one finds only one functional converter with synchronized commutation of input and output. The author thanks Professor I. E. Geron for his help. There are 3 figures, 3 photographs and 3 references: 3 Soviet-bloc and 2 non-Soviet-bloc. The references to the English-language publications read as follows: Dudley, "Remaining speech, 1. August" Rev. Am., 11, 1:39; Rantzen, Peachey and Gunn-Russell, "The Broad Principles in the design of Automatic Monitors", Electronic Engineering, v. 23, no. 275, 1951.

12

SUBMITTED: March 19, 1960

Card 1/1

GLEKHOV, A.A.

Measurement of nonlinear distortions in a wire broadcasting
channel. *Elektronika* 13 no.5:76-80 My '62 (MIRA 17:8)

L 174C7-66 EWT(m)/EWG(m)/EWP(t)/ETC(f) IJP(c) RDW/JD
ACC NR: AP6007247 SOURCE CODE: UR/0363/60/002/002/0245/0248

AUTHOR: Kharakhorin, F. F.; Glukhov, A. A.; Kuznetsova, Ye. S.; Potapov, V. I. 57

ORG: none B

TITLE: Some properties of tellurium doped indium and gallium arsenides 7

SOURCE: AN SSSR. Izvestiya. Neorganicheskiye materialy, v. 2, no. 2, 1966, 245-248

TOPIC TAGS: semiconducting material, gallium arsenide, indium compound, indium arsenide, single crystal, electric property, activated crystal, tellurium activator

ABSTRACT: Electron carrier concentration in relation to Te dopant content in the charge and Hall mobility of electrons in relation to the carrier concentration have been studied in indium arsenide and gallium arsenide single crystals/grown by the Czochralski-Gremmelmayr technique and, in the case of GaAs, by oriented crystallization. This latter technique was used to exclude interference of Si acceptor impurity (from the quartz container) with electrical characteristics of GaAs. In the Czochralski process, 99.999% Te was introduced directly into the melt. Hall coefficient and resistivity were measured at 300K. In both indium and gallium arsenides, carrier concentration increased with the increase in Te content of the charge up to a certain value ("saturation" point), then leveled off. However, the "saturation" point was reached with ten times higher Te content in InAs than in GaAs.

Card 1/2 UDC: 546.682'191+546.681'191+546.24 Z

L 17407-66

ACC NR: AP6007247

Consequently, the limit (maximum) carrier concentration was about an order of ² magnitude higher in InAs than in GaAs ($\sim 2 \times 10^{19}$ versus 3.1×10^{18} at/cc). These data were in satisfactory agreement with the literature. Presumably, the "saturation" in carrier concentration was reached at a point when Te atoms form electrically inactive Te-Te bonds. The Hall mobility in both arsenides studied displayed a similar pattern of gradual decrease with increased concentration. A wide dispersion of mobility data at a given carrier concentration for GaAs crystals prepared by Czochralski technique and by oriented crystallization was explained by the compensating effect of the uncontrollable acceptor impurity. Orig. art. has: 5 figures. [JK]

SUB CODE: 20 SUBM DATE: 12Jul65/ ORIG REF: 002/ OTH REF: 007/ ATD PRESS: 4206

Pure metal 44,18

Card ¹⁵ 2/2

L. P. 0010-0000 ENT(9) 0000(1) 1000(1) 00
ACC NR: AP6011317 SOURCE CODE: UR/0363/867002/003/0461/0453

AUTHOR: Kharakhorin, F. P.; Kuznetsova, Ye. S.; Petropov, V. I.;
Glukhov, A. A.

ORG: none

TITLE: Relation between mobility and concentration of carriers in
indium arsenide

SOURCE: AN SSSR. Izvestiya. Neorganicheskiye materialy, v. 2, no. 3,
1966, 461-463

TOPIC TAGS: indium compound, arsenide, indium arsenide, semiconductor
single crystal, electron mobility, carrier concentration

ABSTRACT: Variations of Hall mobility at different carrier (electron)
concentrations ($n = N_D + N_A$) in the $4 \cdot 10^{15} - 10^{17}/\text{cc}$ range have been
studied at 300K in indium arsenide, as one of the most promising
IIIIV compounds. The theoretical plot of mobility versus n was
calculated using the Brooks formula for uncompensated ($N_D = 0$) and
compensated materials which cover concentration regions with nondegen-
erated and weakly degenerated states, respectively. Comparison was
made of the calculated data with the experimental data from literature
and with the authors' own data. The latter were obtained with single

Card 1/2

UDC: 546.682'191:537.311.33

L 20610-66

ACC NR: AP6011317

crystals grown either by oriented crystallization or by Czochralski-Gremmelmayer technique. Most of the data for the samples grown by the first technique ($n = 3 \cdot 10^{16} - 8 \cdot 10^{16}/\text{cc}$ and mobility = 29,700--22,000 $\text{cm}^2/\text{v}/\text{sec}$) were in agreement with the calculated data. Data obtained with the samples grown by Czochralski technique ($n = 5 \cdot 10^{16} - 10^{17}/\text{cc}$ and mobility = 24,300--20,000 $\text{cm}^2/\text{v}/\text{sec}$) were somewhat lower and the literature data were considerably lower than theoretical. The discrepancy between theoretical and some of the experimental data was attributed to a variable degree of compensation by impurities. Orig. art. has: 2 figures and 3 formulas. [JK]

SUB CODE: 20/ SUBM DATE: 12Jul65/ OTH REF: 008/ ATD PRESS: 4225

Card 2/2. *sb*

L 32043-66 EAT(m)/SM(t)/ETI IJP(c) JD

ACC NR: AP6613335 SOURCE CODE: UR/0363/66/002/004/0582/0584

AUTHOR: Kharakhorin, F.F.; Kuznetsova, Ye. S.; Glukhov, A.A.; Potapov, V.I.

ORG: none

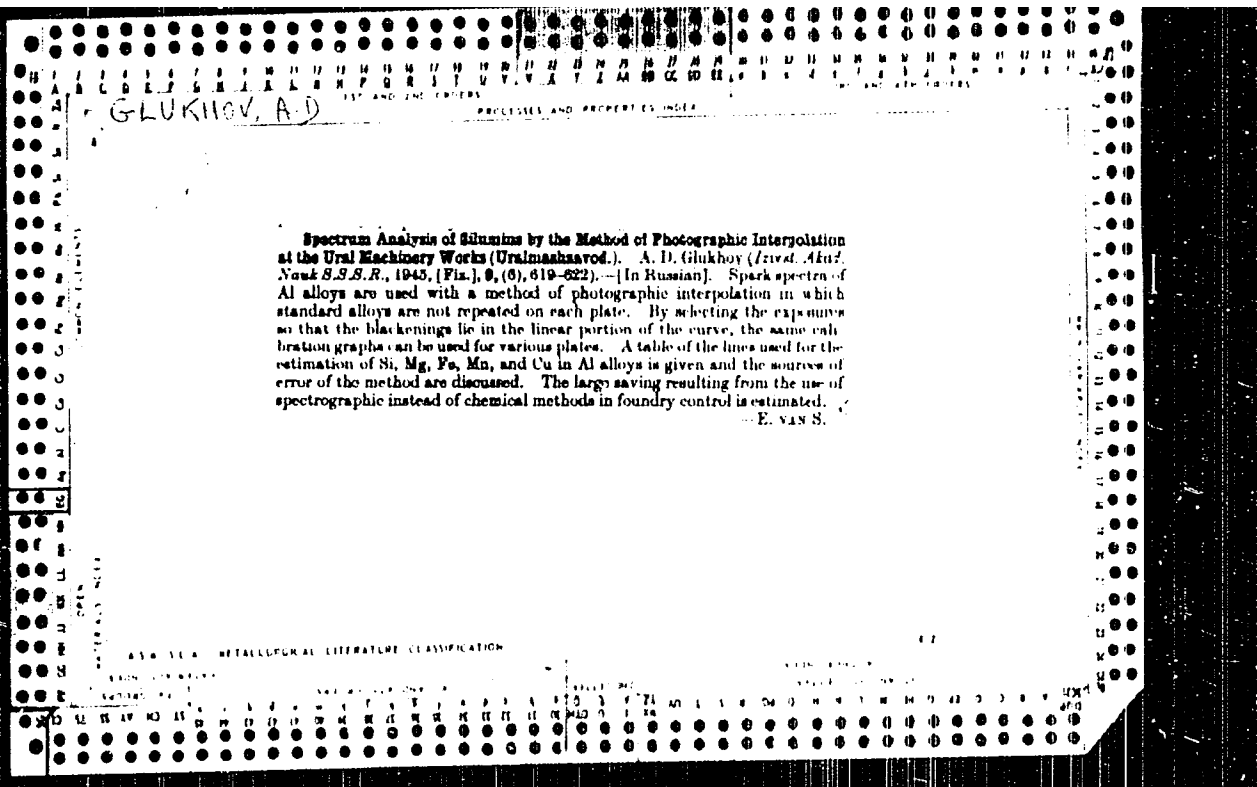
TITLE: Purification of arsenic by sublimation

SOURCE: AN SSSR. Izvestiya. Neorganicheskiye materialy, v. 2, no. 4, 1966, 582-584

TOPIC TAGS: arsenic, sublimation, metal purification

ABSTRACT: A process and the corresponding equipment have been developed for purifying arsenic by sublimation. Usually, one or two sublimations are performed, impurities of low vapor pressure such as copper, iron, and aluminum being thus removed. More sublimations are required to remove impurities having a substantial vapor pressure (zinc, cadmium, sulfur, selenium, tellurium). The process avoids contamination of the arsenic by eliminating its transfer from one ampoule to another. Radioactivation analysis has shown that after 4-5 sublimations, for a threefold decrease in the total impurity content, the amount of sulfur decreased by a factor of 6 - 10. Arsenic obtained after five sublimations was used to synthesize indium arsenide with a carrier concentration of $4 \times 10^{19} \text{cm}^{-3}$ and a mobility of $29,000 \text{ cm}^2/\text{V sec}$ at 300K, which also indicates that the
Card 1/2 UDC: 546.19

Card 2/2



ГЛУКHOV, A. I.

Glukhov, A. I. "The scientific victory of Soviet woman", (On the award of the Stalin Prize to K. Ya. Bakhitza for developing new types of tea), *Selektsiya i innovatsiya*, 1949, No. 5, p. 17-18.

SO: "4630, 10 Sept. 53, (Leningrad 'Zhurnal'nykh Statей, V. 11, 1953).

GLINCHOV, A. N.

"Fuel Elements for Light Water Cooled and Moderated Reactors of Atomic Power Stations", by S. S. Asbatsyan, A. N. Glukhov, D. I. Bondharov, A. I. Kovalyev, and G. A. Skvortsov.

Report presented at the IAEA Atom-for-Peace Conference, Geneva, 9-13 Sept 1958.

(LinkHoc, D.M.)

21(4) PHASE I BOOK EXPLOITATION SOV 2183

International Conference on the Peaceful Uses of Atomic Energy, 2nd, Geneva, 1958.

Doklady sovetzkikh uchenykh; yadernyye reaktory i yadernaya enerzhia. (Reports of Soviet Scientists on Nuclear Reactors and Nuclear Power) Moscow: Izdatel'stvo 1959, 707 p. (Series: Ita; Trudy, vol. 2) Errata slip inserted. 8,000 copies printed.

General Eds.: M.A. Dollezhal, Corresponding Member, USSR Academy of Sciences, A.K. Krasin, Doctor of Physical and Mathematical Sciences, A.I. Lepunskiy, Member, Ukrainian Academy of Sciences, I.I. Morikov, Corresponding Member, USSR Academy of Sciences, and V.S. Purov, Doctor of Physical and Mathematical Sciences. Ed.: A.V. Alyab'yev, tech. Ed.: Ye. I. Mazel'.

PURPOSE: This book is intended for scientists and engineers engaged in reactor designing, as well as for professional and students of higher technical schools where reactor design is taught.

COVERAGE: This is the second volume of a six-volume collection on the peaceful uses of atomic energy. The six volumes contain the reports presented by Soviet scientists at the Second International Conference on Peaceful Uses of Atomic Energy, held from September 1958 to 1958 in Geneva. Volume 2 consists of three parts. The first is devoted to atomic power plants and reactors in the Soviet Union; the second is devoted to atomic power plants and reactors abroad, which is predominantly theoretical, to problems of nuclear reactor physics and construction engineering; and the third is the science editor of this volume. See SOV 2181.

Dollezhal, M. A., A.K. Krasin, M.A. Mikolayev, A.M. Orlovskiy, and V.M. Purov. Experiences of Operating the First Atomic Power Plant in the USSR and the Plant's Work Under Boiling Conditions (Report No. 2183) 15

Dollezhal, M.A., A.K. Krasin, F.I. Alekshchenko, A.V. Gricorin, A.M. Orlovskiy, M.Ye. Morozov, Ye.Ye. Yezhlyayev, M.M. V. Zhurav, V.M. Zhurav, and I. Klyayev, and A.M. Zhukhin. A Graphite Uranium Reactor With Heat-Trajectory Steam Superheat (Report No. 2139) 36

Aleksandrov, A. P., I. I. Astashev, A. V. Brandaev, A. I. Brandaeva, G.M. Blazovskiy, G.M. Gerasimov, V. G. Gerasimov, and V.S. Zholobin. The Atomic Icebreaker "Lening" (Report No. 2140) 60

Shtrom, Ya. V. and P.G. Polozhik. Radiation Safety System of the Atomic Icebreaker (Report No. 2518) 87

Svortsov, S. A. Water-steam Power Reactors (VVER) in the USSR (Report No. 2184) 105

Aleksandrov, A. P., A.M. Orlovskiy, V. V. Orlovskiy, A. I. Kovalev, and S.A. Svortsov. Heat-producing Elements for Water-steam Reactors of Atomic Power Plants (Report No. 2190) 119

Kuzhikina, O.N. and V.I. Subbotin. Cooling Water-steam Reactors (Report No. 2144) 134

Yamakov, V.S. and I.V. Ivanov. A Study of Unsteady Heat Transfer in Heat-producing Elements of Nuclear Reactors (Report No. 2470) 153

Ivanovskiy, M. M., V. I. Subbotin, and E. A. Zhurav. High-speed Method for Measuring the Heat-transfer Coefficient in the Pipe (Report No. 2475) 165

Dutateladze, A.S., V. I. Subbotin, V.M. Borichenko, and P. L. Kirillov. Heat Exchange During the Flow of Liquid Metal in the Pipes (Report No. 2210) 176

Karabuzovskiy, O.D. Economics of Nuclear Fuel in Fast Power Reactors (Report No. 2028) 188

Belkin, V.M., E.A. Karabuzovskiy, Ye.S. Sijakov, and O.V. Shvedov. Thermal Neutron Density Distribution Along the Radius of Assemblies of Rod-shaped Heat Producing Elements (Report No. 2034) 199

L 24710-66 EWT(m)/ETC(f)/EPF(n)-2/ENG(m) WW

ACC NR: AT6008415

SOURCE CODE: UR/3136/65/000/993/0001/0017

AUTHOR: Ambartsumyan, R. S.; Goncharov, V. V.; Glukhov, A. M.; Yegorenkov, P. M.; Smirnova, R. F.; Shavrov, P. I.

ORG: none

TITLE: Increasing the power of VVR-S reactors

SOURCE: Moscow. Institut atomnoy energii. Doklady, IAE-993, 1965. O povyshenii moshchnosti reaktorov VVR-S, 1-17

TOPIC TAGS: water cooled nuclear reactor, water moderated reactor, reactor fuel element, nuclear reactor power / VVR-S water cooled nuclear reactor

ABSTRACT: The authors consider the possibilities for using slightly modified MR fuel assemblies for increasing the power of VVR-S water-cooled water-moderated reactors. A figure is given showing the construction and dimensions of the MR fuel assembly. The assembly consists of five tubular fuel elements of circular cross section. The heat-transfer area of the MR fuel assembly is 2.35 times as great as assemblies using EK-10 elements. The elements are interchangeable, i.e. they may be

Card 1/2

L 24710-66

ACC NR: AT6008415

placed in any cell of the reactor core. The efficient design of the MR elements assures that 90% of the water passing through the core flows through the fuel assembly. The assembly contains 173 grams of U-235, i.e. 35% more than an assembly with EK-10 elements. The use of these elements makes it possible to irradiate specimens in experimental channels or ampules with an outside diameter of 14 mm. Larger specimens may be irradiated by using fuel assemblies with fewer tubular fuel elements. However, use of the MR fuel assembly cuts down the volumetric fraction of water in the reactor core to 0.65 as against 0.7 when assemblies with EK-10 elements are used. The volumetric water fraction is cut still further to 0.52 by the use of beryllium moderators to reduce nonuniformity in heat release due to localized increases in neutron density in the water spaces between adjacent MR fuel assemblies. The use of these fuel assemblies increases the power of the reactor to 8-11 Mw and the maximum neutron intensity (U-235) to $\sim 9 \cdot 10^{13}$ neutrons/cm² sec. The authors discuss the experimental possibilities of the VVR-S reactor with MR fuel assemblies. Orig. art. has: 6 figures, 1 table.

SUB CODE: 18/ SUBM DATE: 00/ ORIG REF: 001/ OTH REF: 003

Card 2/2 *fv*

GLUKHOV, A.S.

[Accounting of the means of economic organizations and
communal economy enterprises] Uchet denezhnykh sredstv
v khozorganakh i predpriyatiyakh kommunal'nogo khoziaistva.
Izd.2., perer. i dop. Moskva, Izd-vo Ministerstva kommunal'-
nogo khozyaystva RSFSR, 1951. 94 p. (MIRA 12:6)
(Housing--Accounting)

Glukhov, A. S.

N/E
756
.05
1954

Operativnoye planirovaniye, uchet i planirovaniye, uchet i khoraschet v tramvaynykh i trolleybusnykh khozyaystvakh (Operative planning, accounting and cost-accounting in the tramway and trolley bus economy) Izd. 2, Perer. I dop. Moskva, Izd-vo Ministerstvo Kommunal'nogo Khozyaystva RSFSR, 1954.
274 p. ^Diagrs., Tables.

GLUKHCV, A. S.

Glukhov, A. S. -- "On the Brucellocidal and Brucellostatic Influence of Saliva, Bile, the Contents of the Duodenum, and Extracts of Certain Organs of Healthy Sheep." Min Higher Education USSR, Novocherkasak Zootechnical-Veterinary Inst imeni First Cavalry Army, Novocherkassk, 1954 (Dissertation for the Degree of Candidate in Veterinary Sciences)

SO: Knizhnaya Letopis', No. 23, Moscow, Jun 55, pp 87-104

GERALD, JIMMY CARROLL

175
011
.85

United States Dept. of Housing and Urban Development (Department of Housing
in Housing Management) 120-76 0000, 1971.
128 p. Tables.
Photostatic copy.

128

GLUKHOV, Arkadiy Semenovich; ISAKOV, V.I., red.; BODANOVA, A.P.,
tekh. red.

[Mechanization of accounting in motorbus transportation units]
Mekhanizatsiia ucheta v avtomobnykh khoziaistvakh. Moskva,
Avtotransizdat, 1962. 141 p. (MIRA 15:12)
(Motorbus lines--Accounting)

ГУКНОВ, А.Я.

Pick hammer point of drill steel. Shakti. steel. 9 no.3:27
№ 165. (MIRA 18:7)

1. NIS-6 kombinata Karagandashakhtostroy.

NO. 1000
Card 1/2

OTHER: 000

NO. 1000

L 64380-65

ACCESSION NR: AP5021635

ENCLOSURE: 01

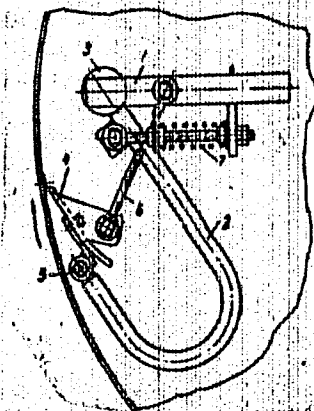


Fig. 1.

1- guide bar; 2- bracket; 3- bracket hinge; 4- scraper blade;
5- blade hinge; 6- rigid connecting rod; 7- returning spring

llc

Card 2/2

BEKISHEV, I.S., inzh.; GLUKHOV, B.A., inzh.; IVACHENSKAYA, M.M., inzh.

New working components of blade paddle concrete mixers of the rotary (turbine) type. Stroit. i dor.mash. 10 no.12:80-71 D 165. (MIRA 1951)

GLUKHOV, S.F., kand. tekhn. nauk, dotsent

Improving furnaces with shaft-type impact mills. Izv. vya. ucheb.
zav.; energ. no. 2:66-71 F '58. (MIRA 11:7)

1. Frunzenskiy politekhnicheskiy institut.
(Furnaces)

GLUKHOV, B.F., kand.tekhn.nauk, dotsent

Qualitative analysis of the aerodynamics of furnaces with shaft-
type impact mills. Izv. vys.ucheb.zav.; energ. no.5:73-76 My '58.
(MIRA 11:8)

(Furnaces---Aerodynamics)

GLUKHOV, B.F., kand. tekhn. nauk, dots.

About another "new" method of calculating heat transfer in
furnaces. Izv. vys. ucheb. zav.; energ. 2 no.7:129-133 J1 '59.
(MIRA 13:1)

1. Belorusskiy politekhnicheskiy institut.
(Heat--Transmission) (Furnaces)

GLUKHOV, B.F., kand. tekhn. nauk

Analysis of the efficiency of steam-gas and steam turbine
industrial thermal electric power plants. Energ. i elektrotekh.
prom. no.2:22-27 Ap-Je '63. (MIRA 16:7)

1. Belorusskiy politekhnicheskii institut.
(Electric power plants)

NEBRNSH, A. I. (AKA) MELNIK, M. I. (AKA) ...
... GLUKOV, B. P., kand.
tekh. nauk, dokladovatel' ...
... doklady ...

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GLUKHOV, B.F., kand. tekhn. nauk

Block-type high-pressure steam generator with natural
circulation for and industrial thermal electric power plant.
Energ. i elektrotekh. prom. no.1:11-14 Ja-Mr'64.

(MIRA 17:6)

GLUKHOV, D.N.

Lightweight device for testing long-distance call stations. Vest. svyazi
20 no.9:11-12 S'60. (MIRA 13:10)

1. Starshiy inzhener laboratorii Tsentral'noy meshdugerozdnoy telefonnoy
stantsii.
(Telephone, Automatic--Testing) (Electric meters)

USSR/Metals - Cast Iron

Jan 51

"Increasing the Wear Resistance of Cylinder Blocks by Application of Titanium-Copper Cast Irons," D. P. Glukhov, Engr, P. G. Petrov, Engr, Laureate of Stalin Prize, ZIS

"Litsey Proiz" No 1, pp 4-7

Since 1949 nearly 1,000 automobiles were produced with blocks made of cast iron with added titanium and copper. This cast iron demonstrated improved mech properties. Expts established decrease in wear of cylinders by 30-50%. In addn wear was more uniformly distributed. Suggests chem compn:

185TP88

USSR/Metals - Cast Iron (Cont'd)

Jan 51

3.2-3.4% C, 2.0-2.2% Si, 0.6-0.8% Mn, 0.16-0.22% P, max 0.12% Cu, 0.25-0.35% Cr, 0.25-0.35% Ni, 0.35-0.50% Cu, 0.08-0.15% Ti. Despite higher hardness of cylinders, wear of piston rings is same or somewhat lower than usual.

185TP88

SECRET, U.S.S.R.

185257

USSR/Engineering - Foundry

Feb 51

"Casting Permanent Magnets," D. P. Glukhov,
Engr, ZIS

"Litey Proiz" No 2, pp 34, 35

Permanent magnets, simpler in operation and less expensive than electromagnets, may be used for magnetic chucks of surface grinders and cutting-off lathes. Presents chem compn and magnetic properties of 3 alloys used for making permanent magnets, and describes melting and pouring procedure and heat-treatment process.

185257

GLUKHOV, D. P.

PA 196T62

USSR/Engineering - Automobiles, Parts, Jul 51
Friction

"Increasing Durability of Friction Surfaces,"
D. P. Glukhov, Engr, Moscow Plant Imeni I. V.
Stalin

"Itley Protzvod" No 7, p 28

Suggests preventing cast-iron valve tappets
from crumbling-out by coating friction surfaces
with magnetic iron oxide after preliminary
pickling of tappets in 30% soln of hydrochloric
acid. Coating of this kind increases life of
piston rings by 25-30% and may also be used for
196T62

USSR/Engineering - Automobiles, Parts Jul 51
Friction (Contd)

other cast-iron or steel parts which are
not subjected to heat treatment.

196T62

ГЛУКОВ, Д. П. (Engr.)

"Investigation of Factors Having an Influence on the Structure and Wear Resistance of Piston Rings of Individual Cast and the Selection of Cast Iron for a Ring-Cylinder Pair of Auto - mobile Engines." Cand Tech Sci, Moscow Automotive Mechanics Inst. 5 Mar 54. Dissertation (Vechernyaya Moskva Moscow, 24 Feb 54)

SO: SUM 186, 19 Aug 1954

GLJKHOV, D.P.

Wear-resistant cast iron build up of automobile engine steel
push-rod disks. Lit.proizv.no.1:28-29 Ja '55. (MIRA 8:3)
(Die casting) (Automobiles-Engines)

GLUKHOV, D.P., kandidat tekhnicheskikh nauk.

Examination of foreign automobile cast iron parts. Lit.proizv. no.12;
9-12 D '55. (MIBA 9:3)
(Automobiles--Apparatus and supplies)(Cast iron--Metallography)

Distr: 4E2b/4E2c

~~Naturally Alloyed Khallors Cast Iron Machine Components~~
~~Wm. D. P. Glukhov, G. M. Kozlov and B. M. Popyev~~

"APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515420004-0

Glatkov, D.P.

APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515420004-0"

BLURNEY, J. T.

Cast cam shafts for automobile engines. E. J. Blukhly
Metals Progress 1937, No. 7, 4-7. Cast steel cam
 in their performance to forged steel cam shafts made by casting
 iron cont. C.S. 3.5, Si 1.2, Mn 0.8, P 0.01, S 0.005, and
 0.4-0.6, and P 0.25, has been used in their manufacture.

15(2), 15(7)

128-111-107

AUTHOR: Glukhov, D.F., Candidate of Technical Sciences and
TETURNOV, T.A., Engineer

TITLE: Choice of Cast Iron of Automobile Cylinders

PERIODICAL: *Viteynoye Proizvodstvo*, 1959, No 7, pp 16-17. (USSR)

ABSTRACT: For the production of internal combustion engines the Automobile Plant at Gor'kiy has started to produce those areas of the cylinder block, which are the most exposed to damage, from corrosion and heat resistant steel in the shape of special liners. In case these liners are irregularly formed they damage the pistons. The expensive nickel (17,5%) and copper (1,5%) admixtures increase the price of the cast iron. Therefore the Automobile Plant at Yaroslavl has launched the liners made from special steel over the active layer of the cylinder. But this can mean faster wear of the cylinder. At the Automobile Plant "Imeni Lenina" no liners are produced but the cylinder blocks are cast from a chrome-nickel-alloy (since 1951). The

Cont. 1/2

1971-11-11-027-027

Choice of Cast Iron of Automobile Cylinders

production method ensures a better coverage of the cylinder wall of the piston. The bearing capacity for the compression rate for cylinders lined with conventional Flucton for those of the Arana 1100 cc. III (made from titanium-copper alloy). The authors analyze the influence of the particular cast alloy on the composition and heat resistance of the cast iron: chrome, nickel, copper and titanium. The results of the experiments are published in a table. Micro-photographs of diameter and 1 micro-photograph.

Card 2/2

GLUKHOV, D.P.

Casting piston rings from iron prepared in cupola furnaces and by
the duplex process. Lit. proizv. no.11:33-34 N '60. (MIRA 13:12)
(Iron foundation)

GLADILIN, A.A.; GLUKHOV, D.S.; YEREMIN, V.I.; ZVEREVA, N.P.; LAPIN, K.N.;
MAMONOVA, A.S.; MARTYNOV, M.K.; GHIKOV, N.Ye.; MIKHAL'CHIKOV,
P.I.; POLYACHKIN, M.A.,red.; ANTOHOV, V.P., tekhn. red.

[Economy of Penza Province; a statistical collection] Narodnoe
khoziaistvo Penzenskoi oblasti; statisticheski sbornik. Penza,
1958. 190 p. (MIRA 11:11)

1. Penzenskaya oblast'. Statisticheskoye upravleniye.(for all except
Mikal'chikov and Antonev).

(Penza Province--Statistics)

MEMO, 1. 1961.

MEMO, 1. 1961. - [Faint, illegible text]

TO: Richard L. Hall, [Faint, illegible text]

8(9)

SOV/112-59-2-2934

Translation from: Referativnyy zhurnal. Elektrotehnika, 1959, Nr 2, p 96 (URSS.)

AUTHOR: Glukhov, D. Ya.

TITLE: Determination of Synchronous-Motor Reactances in Case of Two Motors
(Opredeleniye reaktivnostey sinkhronnogo dvigatelya v sluchaye dvukh dvigateley)

PERIODICAL: Izv. Kiyevsk. politekhn. in-ta, 1957, Vol 22, pp 453-461

ABSTRACT: Determination of synchronous and subtransient reactances of a synchronous motor under operating conditions is considered. Presence of a second synchronous motor operating from the same bus is assumed. To determine reactances x_d and x_q , this second motor is brought to under-excitation conditions; after that, both motors being connected with each other are disconnected from the power system. Retardation of the first motor with its field circuit open is considered. Under these conditions, the second motor operates as a generator substituting a controlled-voltage source. From

Card 1/2

803/114-59-2-2934

Determination of Synchronous-Motor Reactances in Case of Two Motors

current and voltage oscillograms taken under retardation conditions, reactances x_d and x_q of the first motor can be determined (from the maximum and minimum values of current). The above method differs from the conventional method of reactance determination in that the stator of the first motor is fed by a variable-frequency voltage and that a slip due to different retardation speeds of both machines is present. To determine subtransient reactances x_d'' and x_q'' , retardation of the second underexcited motor is used, with the locked rotor of the first motor (actually with a slow rotor rotation of the first motor or, if this is difficult, during the terminal part of its retardation). On the basis of Goetz-Park's general equations for a synchronous machine, the errors are analyzed which are possible in determining synchronous and subtransient reactances by the above methods. Bibliography: 4 items.

F. M. S.

Card 2/2

CHIZHENKO, I.M.; NEMIROVSKIY, A.Sh.; GLUKHOV, D.Ya.; IVANOV, Yu.M.

The first compensated mercury rectifying converter of the aluminum
plant and results of its testing. Izv. KPI 26:139-169 '57.
(MIRA 11:6)

1. Kafedra teoreticheskikh osnov elektrotekhniki Kiyevskogo politekh-
nicheskogo instituta.

(Mercury-arc rectifiers--Testing)

GLUKHOV, D.Ya.

Experimental determination of synchronous and supertransient reactances of synchronous motors under operational conditions. Izv. EPI 26:363-370 '57. (MIRA 11:6)

1. Kafedra teoreticheskikh osnov elektrotehniki Kiyevskogo politekhnicheskogo instituta.
(Reactance (Electricity)) (Electric motors, Synchronous)

8(0)

SOV/112-59-4-7094

Translation from: Referativnyy zhurnal. Elektrotekhnika, 1959, Nr 4, p 82 (USSR)

AUTHOR: Glukhov, D. Ya.

TITLE: Determining the Principal Characteristics of a Synchronous Machine by Retardation Test

PERIODICAL: Izv. Kiyevsk. politekhn. in-ta, 1957, Vol 26, pp 371-385

ABSTRACT: It is suggested that the retardation test of a synchronous motor be used for determining its no-load and short-circuit characteristics under operating conditions. To plot a no-load characteristic, it is suggested that the retardation test be made with the excitation on and the stator winding open. The oscillograms $i_f = f(t)$, $n = f(t)$, and $U = f(t)$ can be taken; the voltage U and the speed n can be determined from the oscillograms for various values of the exciting current i_f . Then, the voltage corresponding to each point should be translated to the synchronous speed using the formula $U_0 = U/n$, where n is the relative speed. An analysis of the principal equations of the synchronous

Card 1/2

SOV/112-59-4-7004

Determining the Principal Characteristics of a Synchronous Machine by

machine showed that the error in determining the no-load characteristic will not exceed 2% if the time constant of an open-stator retardation is about 25-30 sec. To plot a short-circuit characteristic, a closed-stator retardation test should be staged. Here, the machine working as a motor without load is cut off from the supply network, and its excitation is opened. Then, the stator winding is short-circuited, and the excitation is applied (to make the stator current 1.1-1.3 of the rated current), and the oscillograms $I = f(t)$, and $i_f = f(t)$ are taken. The oscillogram processing includes determining the scales for the stator current I and for the exciting current i_f . The error will not exceed 2%, if the time constant is about 5-6 sec. Oscillograms are presented, and sample processing is shown.

Ya.B.D.

Card 2/2

11339

S/194/62/000/007/091/160
D295/D308

AUTHORS: Chiznenko, I.M., and Glukhov, D.Ya.
TITLE: Converter for feeding an electric arc load
PERIODICAL: Referativnyy zhurnal. Avtomatika i radioelektronika,
no. 7, 1962, abstract 7-5-59 y (Tr. Kiyevsk. politekhn.
in-ta Sb. statey elektrotekhn. fak., Kiev, 1961,
90 - 102)

TEXT: A description and an approximate calculation are given of new converter circuits intended for feeding an electric arc load. Capacitive and inductive converter differ from the usual ones in that they comprise 3 capacitors or three chokes in series with the three feeding buses of a three-phase bridge rectifier, which ensures a decreasing load characteristic. By operating in parallel, a capacitive and an inductive set, one obtains a single compensating converter that has a phase-shift angle equal to 1, or a converter with leading phase angle. The paper gives a calculation of the currents and voltages of a capacitive and an inductive converter, and oscillograms of various modes of operation determined by means
Card 1/2

Converter for feeding an ...

S/194/62/C00/007/091/160
D295/D308

of a laboratory model. The rectifier suggested has a number of technical advantages in comparison with standard bridge converters with wide-angle quick-acting grid control. [Abstracter's note: Complete translation.]

Card 2/2

GLUKHOV, G. K. Cand Agr Sci -- (diss) " Sheep milk and ~~the~~ experiments of processing
it into cheese and other dairy products." Saratov, 1956. 15 pp 20 cm. (Min of
Agriculture USSR. Saratov Agr Inst), 150 copies
(KL, 7-57, 108)

50

L 45304-66 EWT(1)

ACC NR: AR6015988

SOURCE CODE: UR/0044/65/000/011/B070/B070

AUTHOR: Glukhov, G. M.

TITLE: Constructing a solution for the boundary value problem for a linear heat equation. 1.

SOURCE: Ref. zh. Matematika, Abs. 11B320

REF SOURCE: Tr. Saratovsk. in-ta mekhaniz. s. kh., vyp. 33, ch. 2, 1964, 24-31

TOPIC TAGS: heat equation, successive approximation, approximation convergence

ABSTRACT: The author constructs a solution for the one-dimensional nonstationary heat equation with interior sources of heat, depending on temperature according to the law

$$H_c(x, t) \sum_{k=0}^N q_k e^{\lambda_k T^k}$$

where T is temperature, $g_0(x, t)$ is a known function of the coordinate and time, c_k are known constants, ϵ is a parameter. The dependence of the coefficient of heat conductivity on temperature is taken in the form $\lambda(T) = \sum_{k=0}^n \lambda_k e^{\lambda_k T^k}$, where λ_k are known constants. The solution is sought in the form of a series in powers of ϵ

$$T = \sum_{i=0}^{\infty} \epsilon^i T_i(x, t).$$

Card 1/2

UDC: 517.9:536.2

L 45304-66

ACC NR: AR6015988

For determining the functions T_i the author obtains an infinite system of linear equations of the heat type, from which T_i are successively determined. He investigates convergence of the series determining the temperature, and obtains bounds for its region of convergence. V. Vilenskiy [Translation of abstract]

SUB CODE: 12

Card 2/2 mjs

YEVLEV, V.I., kapitan 2-go ranga; GARKHON, G.P., kapitan 2-go ranga; ZARUBIN, L.K., kapitan 2-go ranga; TIMOSHIN, V.D., kapitan 3-go ranga; KARTSEV, R.P., kapitan 1-go ranga; MICHURIN, V.I., kapitan 1-go ranga.

Matured problems. Mor. sbor. 49 no. 1244-45. 1945.

CHAGIN, M.; REZAK.

"The Nature of Titrating Curves" part III.
"The titration of toluene in the presence
of sulfuric and acetic acid," *Kur.
Khim.*, 18, No 5, 1971. Laboratory
of Physical Chemistry, Central Asiatic State
University. Received 21 July 1970.

Report 8-1520, 20 Oct 62.

GLUKHOV, I.A.

The volatile oil of archa (*Juniperus sibirica*).
 I. A. Glukhov, *Sovetskaya Tadzhik. Filial Akad. Nauk*
 No. 20, 11-18(1950) (in Russian). *J. sibirica*
 contains 0.5-0.8% volatile oil in the leaves.
 Compon. of 3 oil samples was established by a high-efficiency
 distn. under vacuum. The hydrocarbon part of the oil con-
 tained the following: α -pinene 40-50% (I), myrcene 18-
 20% (II), limonene 2-4% (III), and cadinene 3-4% (IV).
 I was identified by n and d. values; II was demonstrated by
 condensation with maleic anhydride. Tetrabromide deriv.
 of III and nitroso chloride and dihydrochloride deriv. of IV
 were prepd. Values of n and d. of various fractions were
 compared with initial values. M.p.s. of deriva. were given.
 An unidentified hydrocarbon fraction between II and III
 was suggested as the source of the antiseptic properties of the
 oil.
 A. W. Duly

CH

MA
7/15

SOV 137-59 3 3483

Translation from: Referativnyy zhurnal. Metallurgiya, 1959, Nr 3, p 78 (USSR)

AUTHOR: Glukhov, I. A.

TITLE: Investigations on the Chemistry of Rare and Dispersed Elements
(Issledovaniya v oblasti khimii redkikh i rassevannykh elementov)

PERIODICAL: Izv. Otd. yestestv. nauk. AN TadzhSSR, 1957, Nr 24, pp 21-24

ABSTRACT: The author renders a brief account of the results of investigations of the Institute of Chemistry, Academy of Sciences, Tadzhik SSR on the chemistry of rare and dispersed elements: 1) On the interaction of Ca and Fe wolframates with a gaseous mixture of Cl_2 and S_2Cl_2 ; 2) on the kinetics and chemism of the reaction of passage of MoS_2 into volatile compounds in the process of chlorination with pure Cl_2 in the presence of atmospheric O_2 ; 3) on the reaction of chlorination of ReS_2 .

Ye. Z.

Card 1 of 1

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30V/81-59-5-14687

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Translation from: Referativnyy zhurnal, Khimiya, 1959, Nr 5, p 69 (USSR)

AUTHORS: Glukhov, I A., Bekhta, G A.

TITLE: Oxidizing Chlorination of Molybdenite

PERIODICAL: Izv. AS TadzhSSR, 1958, Vol 84, pp 35 - 46

ABSTRACT: The chlorination of molybdenite in the presence of O_2 takes place more rapidly, than in an atmosphere of pure Cl_2 , whereby the products of sublimation are the highly volatile oxychlorides, mainly molybdenum di-oxydichloride, MoO_2Cl_2 . With a reduced supply of O_2 during oxidizing chlorination, highly volatile molybdenum oxychlorides are formed, having the lowest degree of oxidity, apparently mainly oxy-tetrachloride $MoOCl_4$ and molybdenum oxytrichloride, $MoOCl_3$. The presence of moisture, within the limits of atmospheric humidity, is not detrimental to the main process. An elevated humidity promotes the formation of low-volatile MoO_3 , as a result of thermal dissociation of the dioxychloride monohydrate formed $MoO_2Cl_2 \cdot H_2O$. Upon the formation of oxychlorides from molybdenite, during the oxidizing chlorination,

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Oxidizing Chlorination of Molybdenite

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SCV/81-59-5-14687

chlorination generally takes place first and then oxidation. However, at temperatures of $> 400^{\circ}\text{C}$ the reverse order of the reaction is partially possible, i.e., first the oxidation of the molybdenite and then chlorination. The optimum conditions of the oxidizing chlorination process of molybdenite with simultaneous sublimation of the reaction products, is a temperature of 450 to 480°C with a chlorine and oxygen supply at a ratio of 1:3 by volume. The highly-volatile molybdenum oxychlorides formed have both a higher thermal and chemical stability than molybdenum pentachloride, obtained by simple chlorination. 44

Authors' résumé

Card 2/2

GLUKHOV, I.A.; YELISEYEV, S.S.

A new oxchloride of pentavalent molybdenum - MoOCl_3 . Izv. Otd.
geol.-khim. i tekhn. nauk AN Tadzh. SSR no.1:9-82 159. (MIRA 14:8)

Institut khimii AN Tadzhikskoy SSR.
(Molybdenum chlorides)

GLUKHOV, I.A.; BEKTELE, G.A.

Chlorination reaction of molybdenite. Trudy AN SSSR 20:12-13
199. (MIRA 18:1)

(Molybdenite) (Chlorination)

BEKHEPLE, G.A.; GLEKHOV, I.A.; KALASHNIKOVA, G.M.

Sulfatization of copper chloride. Trudy AN Tadzh. SSR Ser. 13-3
'59. (MIRA 13:3)

(Copper chloride) (Copper sulfate)

BEKHTLE, G.A.; GLOKHOV, I.A.; KALASHNIKOVA, G.M.

Sulfatization of copper sulfide by intermediate chlorination.
Trudy AN Taish. SSR 84:53-55 '59. (MIRA 13:3)
(Copper sulfide) (Chlorination) (Copper sulfate)

GLUKHOV, I.A.; RODIONOVA, R.A.

Reaction of molybdenum dihydroxychloride in an atmosphere of chlorine and sulfur chloride. Dokl. An Tadh. SSR 2 no. 5:15-17 '59. (MIRA 13:12)

1. Institut khimii An Tadhikskoy SSR. Predstavleno akademikom AN Tadhikskoy SSR S. Yusupovoy.
(Molybdenum chloride) (Chlorine) (Sulfur chloride)

GLUKHOV, I.A.; SHALUKHINA, L.M.

Reductive chlorination of calcium molybdate. Dokl. AN Tadz. SSR
3 no.1:23-26 '60. (MIRA 13:12)

1. Institut khimii AN Tadzhijskoy SSR. Predstavleno akademikom
AN Tadzhijskoy SSR S.Yusupovoy. (Chlorination)
(Calcium molybdate)

GLUKHOV, I.A.; TIKHOMIROV, L.A.

Method for obtaining molybdenoxytetrachloride MoOCl_4 . Dokl. AN
Tadzh. SSR 3 no. 2:15-18 '60. (MIRA 14:4)

1. Institut khimii AN Tadzhikskoy SSR. Predstavleno chlenom-
korrespondentom AN Tadzhikskoy SSR R.B. Baratovym.
(Molybdenum chlorides)

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5/076/67/005/076/002/013
R110/8205

AUTHORS: Glukhov, I. A., Davidiyants, S. B., Yunusov, M. A.,
Yel'manova, N. A.

TITLE: Chlorination mechanism of rhenium heptasulfide Re_2S_7

PERIODICAL: Zhurnal neorganicheskoy khimii, v. 6, no. 6, 1967, 1264-1266

TEXT: The authors wanted to determine some intermediate stages of the
rhenium heptasulfide chlorination: $ReS_2 \rightarrow \dots \rightarrow ReS_2Cl_2 \rightarrow ReCl_4 \rightarrow ReCl_5$.

It was obvious to suppose (Ref. 1: S. B. Davidiyants et al: Tr. Akademi
nauk Tadzh. SSR, 1959, v. 31, no. 2, p. 105) that besides these known
stages between ReS_2 and ReS_2Cl_2 , the intermediate product ReS_2Cl_2 was
formed. Saturated sulfides (e.g. that of rhenium) react readily with
free chlorine, while saturated oxides react only at red heat.

$S=Me=S + Cl_2 \rightarrow S=Me-\overset{Cl}{\underset{|}{S}}-\overset{Cl}{\underset{|}{S}}$ forms probably in this connection under opening
of the first double bond followed by the opening of the second one. Only

Card 1/4

3/078/6/306/006/002/0'3
B110/B206

Chlorination mechanism of rhenium

substitution is possible for saturated Re_2S_7 . As the valence of Re drops from 7 (Re_2S_7) to 5 (ReCl_5), the reaction must take its course over a number of intermediates. The synthetic Re_2S_7 reacts with chlorine already at low temperatures. It should therefore be possible to observe a number of unstable intermediates under mild reaction conditions. Re_2S_7 was produced by precipitation of a potassium perrhenate solution with ammonium sulfide (8% sulfide sulfur). After washing out by decanting with hot hydrochloric acid (70-80 ml concentrated HCl to 1 l H_2O), drying took place at 160°C in a CO_2 current. In order to prevent exothermic heating, a dry chlorine-carbon dioxide mixture ($\text{Cl}_2:\text{CO}_2 = 1:5$) was conveyed through 3-5 g Re_2S_7 in an electric glass furnace. The optimum temperature was established to be around 120°C during experiments at temperatures between 25 and 180°C . At lower temperatures, chlorination did not proceed quantitatively, and at higher ones, the intermediates were chlorinated further. In the CO_2 current, the water was first totally removed, then

Card 2/4

6/0/1974/7/1/002/01/011
3/10/1974

Chlorination mechanism of vanadium

the Cl_2-ClO_2 mixture was introduced at a rate of 0.2 liter per hour at $100^\circ C$, and for 2 hours at $20^\circ C$ under development of sulfur chlorides. The intermediate obtained was well soluble in water and alcohol in contrast to the final product, thus making it possible to control the completeness of chlorination. The elementary analysis provided as the average of three investigations: $Re_{0.4}S_{0.4}Cl_{0.4}$. This result is well agreed with the calculated values for $Re_2S_2Cl_2$. The product in chloride form is probably according to $Re_2S_2Cl_2 \cdot 2H_2O$. $Re_2S_2Cl_2$ is an amorphous (established roentgenographically), dark brown powder, well soluble in water and ethyl alcohol, insoluble in gasoline, chloroform and ether. After its aqueous solution is acidified, alkalinized and treated, hydrolysis takes place under formation of a flaky dark brown precipitate and liberation of hydrochloric acid. It is oxidized in alkaline solution by bromine, chlorine and perhydroly to alkali perchlorate. In order to investigate its further reactions, dry chlorine gas was introduced at $100-200^\circ C$. Re_2Cl_6 and sulfur chloride were formed thereby. To verify the end of the reaction, the furnace was kept for one hour at $100^\circ C$. A black or grey powder residue was then formed.

Card 3/1

S/137/63/000/002/011/034
A006/A101

AUTHORS: Glukhov, I. A., Shalukhina, L. M.

TITLE: On the reaction of reduction chlorination of lead molybdate and molybdenum trioxide

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 2, 1963, 28, abstract 2G154
("Dokl. Akad. Fankhoi RSS Tochikiston, Dokl. AN TadzhSSH", 1961, v. 4, no. 4, 19 - 23, Tadzhik summary)

TEXT: An investigation was made of $PbMoO_4$ and MoO_3 behavior under conditions of reduction chlorination. The experiments show that the mechanism of chlorinating $PbMoO_4$ and MoO_3 consists in the following process: $2PbMoO_4 + S_2Cl_2 + 3Cl_2 = 2MoO_2Cl_2 + PbCl_2 + 2SO_2$. $4MoO_3 + S_2Cl_2 + 3Cl_2 = 4MoO_2Cl_2 + 2SO_2$. The interaction of $MoPbO_4$ with a gaseous Cl_2 and S_2Cl_2 mixture begins below $160^\circ C$ and is fully completed with 40 - 50 min at $200^\circ C$. MoO_3 is able to react to 45 - 60% under the same conditions. Full chlorination of MoO_3 begins within 1 hour only at $300^\circ C$. Thermodynamical calculations show that under the

Card 1/2

On the reaction of reduction chlorination of...

S/137/63/C00/002/011/034
A006/A101

selected conditions $PbMoO_4$ chlorination is more advantageous from the energy point of view: its enthalpy exceeds 6.5 times and free energy about 3 times the enthalpy and free energy of the MoO_3 chlorination reaction.

G. Svodtseva

[Abstracter's note: Complete translation]

Card 2/2

S/078/63/008/001/010/026
B101/B186

AUTHORS: Glukhov, I. A., Davidyants, S. B., Yel'manova, N. A.,
Yunusov, M. A.

TITLE: Synthesis of rhenium sulfides and oxysulfides from rhenium
thiochlorides

PERIODICAL: Zhurnal neorganicheskoy khimii, v. 8, no. 1, 1963, 94-95

TEXT: The synthesis of the hitherto unknown compounds ReS , Re_2S_3 , ReOS
and $\text{Re}_2\text{S}_3\text{O}_2$ is described. ReS was obtained by heating ReSCl_2 in a current
of hydrogen. The liberation of HCl begins at 350°C . After 1.5 to 2 hr the
substance is heated at 500°C until no HCl can be traced in the H_2 . In the
same way, Re_2S_3 is obtained from $\text{Re}_2\text{S}_3\text{Cl}_4$. Both substances are steel gray
powders which do not change in air and are more stable towards perhydrol
and bromine water than Re_2S_7 and ReS_2 . From the blurred Debye patterns it
is concluded that the synthesized sulfides are cryptocrystalline. ReOS and
 $\text{Re}_2\text{S}_3\text{O}_2$ were obtained from ReSCl_2 and $\text{Re}_2\text{S}_3\text{Cl}_4$, respectively, by heating at
Card 1/2

Synthesis of rhenium sulfides...

S/078/63/008/001/010/026
B101/B186

350 to 500°C in water-vapor-containing CO₂. The reaction is terminated in 2 hr. The oxysulfides are black, amorphous powders.

ASSOCIATION: Institut khimii Akademii nauk Tadzhikskoy SSR (Institute of Chemistry of the Academy of Sciences Tadzhikskaya SSR)

SUBMITTED: April 5, 1962

Card 2/2

S/078/63/008/001/011/026
B101/B186

AUTHORS: Glukhov, I. A., Yelisseyev, S. S.

TITLE: Vapor pressure and thermal dissociation of molybdenum
oxy-chloride MoOCl_3

PERIODICAL: Zhurnal neorganicheskoy khimii, v. 8, no. 1, 1963, 100-104

TEXT: During the sublimation of MoOCl_3 , a disproportionation sets in at about 240°C , according to the equation $3 \text{MoOCl}_3 \rightleftharpoons \text{MoCl}_3 + \text{MoOCl}_4 + \text{MoO}_2\text{Cl}_2$. This process was investigated. The amount of nonvolatile MoCl_3 developed was determined by removing the residual MoOCl_3 through dissolution in H_2O . Analysis of the sublimate led to the empirical formula $\text{Mo}_2\text{O}_3\text{Cl}_6$. There exists, however, a mixture of MoOCl_4 and MoO_2Cl_2 which cannot be separated by fractionated sublimation, as proved with a 1:1 mixture of these compounds. The thermogram of the sublimate, too, showed endothermal effects at 102°C corresponding to the m.p. of MoOCl_4 , and at 152°C corresponding to the m.p. of MoO_2Cl_2 . Separation and identification of the

Card 1/2

Vapor pressure and thermal dissociation...

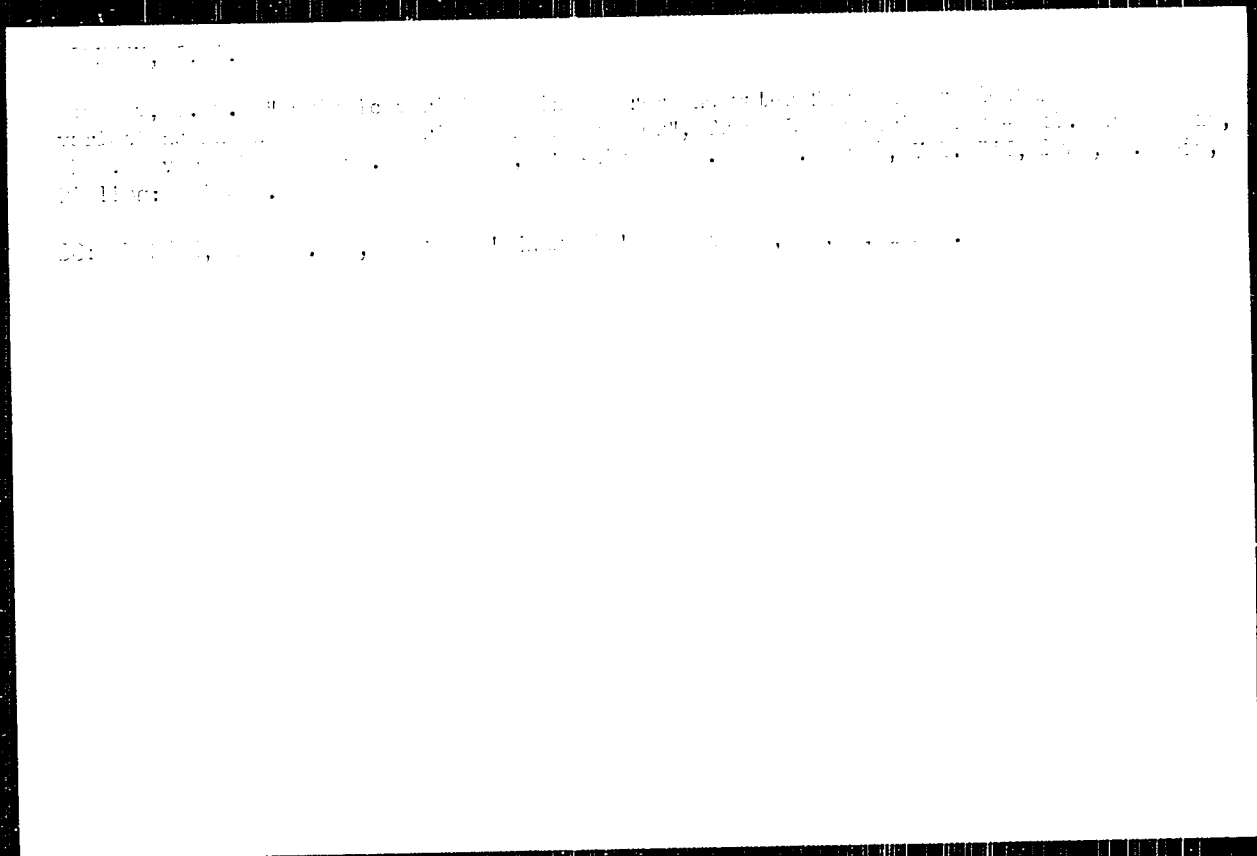
S/078/63/008/001/011/026
B101/B186

two components was carried out by extraction with CHCl_3 or CCl_4 , in which MoOCl_4 is better soluble. For the vapor pressure of MoOCl_3 , the following equation was found: $\log p_{\text{atm}} = 8.764 - 5484/T$, from which $\Delta H = 25$ kcal/mole, $\Delta S = 40$ entropy units was calculated for the sublimation. By extrapolation it was calculated that the vapor pressure of MoOCl_3 amounts to 1 atm at 352°C , and that disproportionation sets in at 215°C . There are 3 figures and 2 tables.

ASSOCIATION: Institut khimii Akademii nauk Tadzhikskoy SSR (Institute of Chemistry of the Academy of Sciences Tadzhikskaya SSR)

SUBMITTED: March 22, 1962

Card 2/2



GLUKHOV, I.G., kandidat geologo-mineralogicheskikh nauk.

Irrigation by underground water. Gidr. i mel. B no.9:
20-30 S '56.

(MLRA 9:10)

(Water, Underground) (Irrigation)

GLUKHOV, I.G., kandidat geologo-mineralogicheskikh nauk

Seepage filtration in loess rocks in canals and subsidence phenomena on irrigated fields. Gidr.i mel.8 no.10:9-18 0 '56. (MLRA 9:10)
(Loess) (Soil percolation) (Soil mechanics)

GLUKHOV, I.G.

~~SECRET~~
Some data on Quaternary deposits in the Sivash region of the
Crimean steppe. *Biul.MOIP.Otd.geol.* 31 no.3:81-88 My-Je '56.
(Sivash region--Geology, Stratigraphic) (MLRA 9:12)

G I. G

5-3-24/37

AUTHOR: Glukhov, I.G.

TITLE: Loesses of Water Origin in Some Regions of Central Asia (Lässy vodnogo proiskhozhdeniya nekotorykh rayonov Sredney Azii)

PERIODICAL: Byulleten' Moskovskogo Obshchestva Ispytateley Prirody, Otdel Geologicheskiiy, 1957, # 3, p 170 (USSR)

ABSTRACT: Five river terraces were found in the valleys of most of the rivers in Central Asia. Loesses were developed in the 5th, 4th and 3rd terraces, and their thickness amounts to 170 m. Their formation occurred probably during the early phase of the Quaternary period. The wetting of loesses is accompanied with sagging phenomena: arising of fissures, terrace-shaped benches and funnels. The loessial rocks can be divided into 3 groups according to the values of a relative sagging factor and the content of loess fraction. The loessial rocks possess sagging properties because of their high porosity and the presence of macropores. The strength of loessial rocks depends on structural couplings between the particles of rocks and salt cement. The salt cement is dissolved by wetting a rock, and this leads to sagging.

AVAILABLE: Library of Congress
Card 1/1

Library of Congress, Department of Geology, Smithsonian Institution, 1957

GLUKHOV, I.S.

Water balance and hydrogeological regions in the northwestern part
of the main Crimean mountain chain. Vestn. Mosk. univ. Ser. Biol.,
geogr., zool., geol. 12 no. 4: 143-150 (1971). (BIRA 11:5)

1. Institute of Hydrogeology, Moscow Institute of Geology, University.
(Ovchinnikov, M. A., Underprint?)

GLUKHOV, I.G.

Interrelation between loessal rocks of aqueous origin and the
genesis and age of river terraces in some regions of Central
Asia. Vest. Mosk. un. Ser. biol., pochv., geol., geog. 13
no.2:177-194 '58. (MIRA 11:9)

1. Moskovskiy gos. universitet, Kafedra gidrogeologii.
(Asia, Central--Loess) (Valleys)

GLUKHOV, I.G.

Earthquakes as one of the factors activating landslides in the
Crimean Mountains. Vest.Mosk.un.Ser.biol., pochv., geol., geog.
14 no.4:143-147 '59. (MIRA 13:6)

1. Kafedra gidrogeologii Moskovskogo universiteta.
(Beketovo (Crimea)--Landslides)

GLUKHOV, I.G.

Sagging phenomena in the irrigated part of the Zamsangir Plateau
in the Vakhsh Valley. Vop. gidrogeol. i inzh. geol. no.17:82-85
'59. (MIRA 14:1)
(Vakhsh Valley--Geology, Structural)

GLUKHOV, I.G.

"Genetic types of loess in the central and southern parts of Central Asia" by G.A. Maslianov. Reviewed by I.G. Glukhov. Uzb. geol. zhur. no.4:61-62 '60. (MIRA 13:10)

(Soviet Central Asia--Loess)
(Maslianov, G.A.)

GLUKHOV, I.G.

Formation of cavern waters in the Crimean Mountains as illustrated by the Skel'skiy, Karasubashi, and Subashi springs. Vest.Mosk.un. Ser.4: Geol. 15 no.1:43-50 '60. (MIRA 14:4)

1. Kafedra gidrogeologii Moskovskogo universiteta..

(Crimean Mountains--Water, Underground)

GLUKHOV, I.G.

Hydrogeological indications of the presence of various types
of karst in the Crimean Mountains. Nov.kar.i spel. no.2:
30-35 '61. (MIRA 15:9)

(Crimean Mountains--Karst)

GLUKHOV, I.G.

Conditions of springs as a basis for determining the infiltration of sediments and the change in karst water resources in the Crimean Mountains. Vest.Mosk.un.Ser.4:1961. 14 no.6:12-57 N-D '61. (MIRA 14:12)

1. Kafedra gidrogeologii Moskovskogo universiteta.
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