

FEDOROV, G.F., inzh.

Studying the kinematics of the wave resulting from the release of water  
from the reservoir. Trudy TSNIEVT no. 19:83-95 '60.

(MIRA 14:5)

(Hydraulics)

SOV/124-58-7-7630

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 7, p 43 (USSR)

AUTHOR: Fedorov, G.F.

TITLE: Local Deformations of the Translatory Wave Heading a Supplementary Water Flow Released From an Elevated Reservoir and the Effect of Said Wave on a River Bed (Mestnyye deformatsii volny popuska i yeye vozdeystviye na ruslo)

PERIODICAL: Tr. Tsent. n.-i. in-ta ekon. i ekspluat. vodn. transp.,  
1957, Nr 12, pp 87-96

ABSTRACT: Bibliographic entry

1. Inland waterways--Physical properties    2. Water waves--Geophysical effects

Card 1/1

AUTHOR: Fedorov, G.F.

SOV/140-58-3-28/34

TITLE: Properties of the Matrices of Second Order and Their Application to the Quadrature of Some Systems of Differential Equations (Svoystva matrits vtorogo poryadka i ikh primeneniye k resheniyu v kvadraturakh nekotorykh sistem differentsial'nykh uravneniy)

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy : Matematika, 1958, Nr 3, pp 217-224 (USSR)

ABSTRACT: The paper is a generalization of former results of the author (Vestnik Leningradskogo universiteta Nr 11, 1953). In § 1 he explicitly considers the properties of the matrices of second order with vanishing determinant. In § 2 he gives with the aid of these properties three conditions under which the system

$$\frac{dy}{dx} = y (\bar{u}_0 \varphi_0 + \bar{u}_1 \varphi_1 + \bar{u}_2 \varphi_2)$$

is integrable by quadratures. Here the  $u_j$  are constant matrices of second order and the  $\varphi_j$  are scalar functions of  $x$ .

There is 1 Soviet reference.

Card 1/2

Properties of the Matrices of Second Order and Their      SOV/140-58-3-28/34  
Application to the Quadrature of Some Systems of Differential Equations

ASSOCIATION: Leningradskiy institut kinoinzhenerov (Leningrad Institute  
for Cinema Engineers)

SUBMITTED:    November 29, 1957

Card 2/2

FEDOROV G.G.

FEDERAU, G.G.; GULIN, S.V.

Polarographic method of determining ferric oxide in cement  
slurry. TSement 28 no.2:22 Mr-Ap '62. (MIRA 15:8)

1. Semipalatinskiy tsementnyy zavod.  
(Cement) (Iron oxides)

FEDOROV, G. G. and GOLOVINA, E. S.

"Influence of Physical and Chemical Factors in the Velocity of Flame Propagation," a paper presented at the 6th International Conference on Combustion, New Haven, 19-24 August 1956

Candidates of Tech. Sci., Inst. of Energetics AS USSR, Moscow

A-52806, 9 July 56

FEDOROV, G. G., and GOLOVINA, Ye. S.,

"Effect of Physicochemical Factors on the Velocity of Flame Propagation." (Study of Combustion Processes; Collection of Articles on Work Done by the Power Institute imeni G. M. Krzhizhanovskogo AS USSR) Moscow Izd-vo AN SSSR, 1958, 123 p.

(Laboratory of Combustion Physics)

for Abstract see Khitrin, L. N.

ZARIF'YANTS, Yu.A.; KISELEV, V.F.; FEDOROV, G.G.

Chemical adsorption of oxygen on a freshly cracked graphite surface. Zhur.fiz.khim. 35 no.8:1885-1886 Ag '61.

(MIRA 14:8)

1. Moskovskiy gosudarstvennyy universitet, fizicheskiy fakul'tet.  
(Oxygen) (Adsorption) (Graphite)



27267

S/020/61/139/005/021/021  
B103/B208

5.4400

AUTHORS: Fedorov, G. G., Zarif'yants, Yu. A., and Kiselev, V. F.

TITLE: Chemical and physical adsorption of oxygen on the surface of freshly split graphite

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 139. no. 5. 1961, 1166-1169

TEXT: The authors studied the adsorption mentioned in the title by using finely ground natural graphite which was pulverized in an argon atmosphere. Studies of this kind are regarded as being important in the search for new ways of surface modification, in order to obtain efficient adsorbents and polymer fillers. One of the applied samples was specially purified. The pulverized samples had a specific surface of 275 and 400 m<sup>2</sup>/g. They were put into the volumeter under exclusion of air. The specific surface was determined after a cycle of absorption measurements with oxygen from the low-temperature nitrogen adsorption by the BET (BET) method. The specific adsorption values of both samples were found to fit well single isotherm. This indicates the good reproducibility of the surface properties for the grinding method applied. Control experiments showed that

Card 1/4

Chemical and physical...

17267

S/020/61/139/005/021/021  
B103/B208

the fine iron powder which is always present (some fractions of one per cent) in graphite thus ground adsorbs not more than 0.5% of the adsorbed amount of graphite. The interaction of oxygen with the fresh cleavage plane of graphite at room temperature is a typical case of activated chemical adsorption. The physical adsorption is extremely low at these temperatures. Adsorption equilibrium is quickly attained in the initial stage, which indicates the low activation energy of the process. With further pressure increase the adsorption increases very slowly. The graphite surface is thought to be inhomogeneous. It consists of sites with different adsorption activity. The surface is saturated with chemisorbed oxygen already at a pressure close to the atmospheric one. Desorption measurements disclosed that all the oxygen adsorbed was irreversibly bound to the surface. No further adsorption takes place on an oxidized sample. The adsorption activity of samples kept at room temperature in an argon atmosphere for 1, 10, and 30 days did not change. Specific adsorption was reduced by about 20% by heating an ampul with non-oxidized graphite at 300°C for 50 hr. This may be due to partial saturation of the surface by gas molecules which escaped from the glass during heating. Graphite roasted at 800°C for 24 hr in an ampul which had

Card 2/4

27267

S/020/61/139/005/021/021  
B103/B208

Chemical and physical...

previously been degassed, lost only 10% of its adsorption. Such a high stability of free radicals on the surface makes graphite a suitable test object of the elementary interactions of various molecules with the surface radicals. Pulverization of graphite in dry air destroys the crystals in all crystallographic directions. The large surface thus formed has a high content of ruptured  $\pi$  or  $\delta$  bonds which form centers of chemical  $O_2$  adsorption on the surface. On the other hand, the activity of  $\pi$ -electrons may be increased by lattice deformations in the basal plane which also contributes to an increase in adsorption activity. By measuring the adsorption isotherms at  $-196^\circ C$ , the authors proved that the chemisorption of oxygen at these temperatures is insignificant. The isotherms have hysteresis loops both on the initial and on the oxidized graphite sample. The authors believe that this loop which extends to the range of low relative pressures, is due to a certain swelling of the sample. The  $O_2$  molecules may penetrate into the interplanar spaces of the graphite crystals which were largely destroyed during grinding. In the high pressure range, the hysteresis is obviously due to capillary condensation in the pores and gaps between the particles. The desorption isotherm shows a

Card 3/4

17267

S/O20/61/139/005/021/02  
B103/B208

Chemical and physical ...

break at  $P/P_S = 0.25$ . This break may be related to the clearing of some pores which were filled at high vapor tensions. This break did not appear at low pressures. The adsorption isotherm of nitrogen shows no hysteresis phenomena in this pressure range. This is explained by a higher penetrability of  $O_2$  at low temperatures. For this reason, the specific surface of fine-disperse graphite which was calculated from the low-temperature  $N_2$  adsorption, may be a little too low, as compared with the surface accessible to  $O_2$  molecules. This fact is however negligible for the authors' conclusions. There are 3 figures and 6 references: 4 Soviet-bloc and 2 non-Soviet-bloc. The reference to English language publications reads as follows: T. Kuczeký, H. Tomášek, Phys. Rev., 120, 1212 (1960).

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im M. V. Lomonosova  
(Moscow State University imeni M. V. Lomonosov)

PRESENTED: March 15, 1961, by M. M. Dubinin, Academician

SUBMITTED: March 13, 1961  
Card 4/4

ZARIF'YANTS, Yu.A.; KISELEV, V.F.; LEZHNEV, N.N.; NOVIKOVA, I.S.; FEDOROV,  
G.G.

Synthesis and functional analysis of oxygen complexes on a sur-  
face of freshly cleft graphite. Dokl. AN SSSR 143 no.6:1358-  
1361 Ap '62. (MIRA 15:4)

1. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova i  
Nauchno-issledovatel'skiy institut shinnoy promyshlennosti.  
Predstavleno akademikom M.M.Dubininym.  
(Graphite) (Surface chemistry) (Oxygen compounds)

37521

S/020/62/144/001/021/024

B124, B101

5.4400

AUTHORS: Zarif'yants, Yu. A., Kiselev, V. F., and Fedorov, G. G.

TITLE: Differential adsorption heats of oxygen and water vapor on the surface of graphite

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 144, no. 1, 1962, 151-154

TEXT: Fine-disperse graphite with a specific surface  $S = 350 \text{ m}^2/\text{g}$ , which had been obtained from natural graphite crushed in a ball mill in an argon atmosphere, was used in the experiments. Adsorption heats were measured using an automatic calorimeter with constant heat exchange; oxygen adsorption was determined by a volumetric method, and by water vapor adsorption with magnesium perchlorate. These methods were described in Ref. 7 (ZhFKh, No. 6 (1962)). The initial adsorption heats of oxygen were very high (110 kcal/mole). With a surface coverage of about  $0.5 \mu\text{M}/\text{m}^2$  of oxygen, the differential adsorption heat begins to decrease, and with  $1 \mu\text{M}/\text{m}^2$ , its value is only  $\sim 50$  kcal/mole. The results obtained show a high degree of reproducibility on fresh cleavage areas of graphite. The high initial adsorption heats are due to the formation of carbonyl groups,

Card 1/3

S/O20/62/144/001/021/024  
B124/B101

Differential adsorption heats of ...

radicals, and groups of the types ...  $\text{C}=\text{O}$  (157 kcal/M),

$\text{-C-O-O-C-}$  (111 kcal/M) or  $\text{-C-O-O-C-}$  (105 kcal/M) which probably appear prior to the formation of the peroxide radicals  $\text{R-O-O}\cdot$  on the surface. Some of them are successively converted to yield saturated oxygen-containing complexes. Chemisorption phenomena could not be observed. The adsorption heats of water vapor on oxidized graphite surfaces were also measured. The graphite sample previously used for the adsorption of oxygen was evacuated down to  $10^{-5}$  mm Hg and contacted with water vapor. The high initial adsorption heats (40 kcal/M), which exceed the physical adsorption heat of water considerably, are due to the formation of hydroxyl groups on the graphite surface. The adsorption heats decrease with increasing coverage, and approach condensation heats when the coverage is 0.5 to  $0.6 \mu\text{M}/\text{m}^2$ . The adsorption heats on oxidized and hydrated graphite were very close to the condensation heats. On an average, water molecules form two hydrogen bonds with oxygen-containing groups and hydroxyl groups on the surface. The probability of adsorption on single hydroxyl groups is small. There are 2 figures.

Card 2/3

L 17720-3

EPR/EFF(c)/ENP(q)/EWT(m)/BDS AFTC/ASD FS-4/Pr-4 WW/WH

ACCESSION NR: AP3004073

S/0076/63/037/007/1619/1622

AUTHORS: Fedcrov, G. G.; Zarif'yants, Yu. A.; Kiselev, V. F. 67  
66TITLE: ~~XXXXXXXXXXXXXXXXXXXX~~ Analysis of the surface properties of a fresh fracture in graphite. 15

SOURCE: Zhurnal fizicheskoy khimii, v. 37, no. 7, 1963, 1619-1622

TOPIC TAGS: graphite, oxygen, oxygen adsorption, heat of adsorption.

ABSTRACT: Authors analyzed the reaction capability of the surface of a fresh graphite fracture with respect to oxygen. A specially-purified, natural graphite was used in the experiments. A thin grist was obtained in a laboratory-type steel vibratory mill with an overpressure of purified argon. The powder was then poured into a glass ampule which was then soldered to the adsorption apparatus. After prolonged evacuation and deaeration of the apparatus to a vacuum of  $1 \times 10^{-5}$  mm mercury column, the diaphragm was broken, the ampule was evacuated and the oxygen adsorption was measured. Differential heats of oxygen adsorption were determined. The inhibiting effect of the reaction products on the oxidation process of a fresh fracture was demonstrated. Assumptions concerning a possible mechanism of oxidation of the graphite surface were pointed out. Orig. art. has 2 figures.

ASSOCIATION: Moscow state university.

Card 1/2



L 18307-63

EPR/EPF(c)/ENP(q)/EWT(m)/BDS

AFFTC/ASD

Ps-4/Pr-4

ACCESSION NR: AP3004981

WH/K

S/0076/63/037/008/1846/1849

68  
67  
5

AUTHORS: Zarif'yants, Yu. A.; Kiselev, V. F.; Fedorov, G. G.

TITLE: Investigation of the surface properties of the freshly splitted graphite

SOURCE: Zhurnal fiz. khimii, v. 37, no. 8, 1963, 1846-1849

TOPIC TAGS: heat adsorption by graphite, water vapor heat adsorption

ABSTRACT: Authors studied the adsorption of water vapors on a fresh surface of graphite. Measurement of the water vapor adsorption shows that the physical adsorption isotherm is complicated by a chemical reaction. The value of the irreversible adsorption measured after the adsorption cycle on the surface of fresh graphite, kept for a period of 48 hrs. in saturated water vapors, consists of 0.53  $\mu$  mole/ $m^2$ . The differential of heat adsorption in the region of small surface area is almost four times greater than the heat of physical adsorption for the usual oxidizing hydrocarbon adsorptions. Apparently, the first traces of water adsorb chemically on the valence-saturated carbon atoms. The graphite was left for 12 hrs. to be exposed to atmospheric oxygen before the desorption was measured. Almost all adsorbed oxygen was irreversibly bound with the graphite surface and its value was 1  $\mu$  mole/ $m^2$ . Thus, the retention of fresh graphite in

Card 1/2

L 18307-63

ACCESSION NR: AP3004981

saturated water vapors does not result in full passivation of the surface. The adsorption of oxygen on its surface is accompanied by the release of a large amount of heat which indicates chemical adsorption. Orig. art. has: 4 figures.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet fizicheskiy facul'tet (Moscow state university, physics faculty)

SUBMITTED: 19Sep62

DATE ACQ: 06Sep63

ENCL: 00

SUB CODE: PH, CH

NO REF 30V: 006

OTHER: 001

Card 2/2

FEDOROV, G.G.; ZARIF'YANTS, Yu.A.; KISELEV, V.F.

Properties of the surface of a freshly cleft graphite. Part 3. Zhur.  
fiz.khim. 37 no.10:2344-2346 O '63. (MIRA 17:2)

1. Moskovskiy gosudarstvennyy universitet.

DOBROVOL'SKIY, N.N.; ZARIF'YANTS, Yu.A.; KISELEV, V.F.; LEZHNEV, N.N.;  
FEDOROV, G.G.

Properties of the surface of a freshly left graphite. Part 4.  
Zhur. fiz. khim. 38 no.2:506-509 F '64. (MIRA 17:8)

1. Moskovskiy gosudarstvennyy universitet Nauchno-issledovatel'skiy institut shinnoy promyshlennosti.

FEDOROV, G. I.  
CA

PROCESSES AND PROPERTIES INDEX

10

Preparation of rubicene from fluorenone with the aid of metallic calcium. V. I. Khmelevskii and G. I. Fedorov: *Zh. Fiz. Khim. (U. S. S. R.)* 9, 1423-5 (1939); cf. C. A. 33, 7777. — The substitution of Ca for Mg turnings in the method described in an earlier paper produced 13% rubicene. Similar reaction of 10 g. Ph(C) with 2 g. Ca turnings yielded 20% *m*-diphenylanthracene, m. 239°.  
Chas. Blanc

Lab. Org. Chem., Ural Indust. Inst. in Kirov

A.S. S.A. METALLURGICAL LITERATURE CLASSIFICATION

FEDOROV, G.I.; ZHUKOV, A.I.

We discuss the article "Possibilities for mechanizing the work on communication lines." Avtom., telem. i svias' no.4:37 Ap '57.

(MIRA 11:4)

1. Starshiy elektromekhanik Lisinskoy distantzii signalizatsii i svyazi Yugo-Vostochnoy dorogi (for Fedorov). 2. Zamestitel' nachal'nika sluzhby signalizatsii i svyazi Odesskoy dorogi (for Zhukov).  
(Railroads--Communication systems)

POS'YANSKIY, Aleksandr Davydovich; FEDOROV, G.I., nauchnyy red.;  
SHELKOV, N.I., red. izd-va; YEZHOVA, L.P., tekhn. red.

[Brief course in descriptive geometry] Kratkii kurs nachertatel'noi geometrii. Moskva, Gos. izd-vo "Vysshaya shkola,"  
1961. 270 p. (MIRA 15:3)  
(Geometry, Descriptive)

FEDOROV, G.I.; KONSTANTINOVA, A.F.

Transmission of light through plates made from uniaxial optically active crystals of the acial classes. Part 1: General solution.  
Opt. i spektr. 12 no.3:407-411 Mr '62. (MIRA 15:3)  
(Crystals--Optical properties)



GEL'BERG, L.A., kand.tekhn.nauk; FEDOROV, G.I., kand.tekhn.nauk

Present-day state housing construction and ways of improving it.

Izv.ASIA 4 no.1:47-56 '62.

(MIRA 15:11)

(Building research)

IZMAYLOV, R.I.; OKRUZHNOV, A.M.; FEDOROV, G.I.; VIROBYANTS, R.A.

Thermocatalytic conversions of hydrocarbons of a petroleum  
C<sub>6</sub>-fraction on Al<sub>2</sub>O<sub>3</sub>-Pt catalyst. Neftekhimija 1 no.4:505-  
508 J1-Ag '61. (MIRA 16:11)

1. Institut organicheskoy khimii AN SSSR, Kazan'.

FEDOROV, G.I.

GEL'BERG, I.A., kandidat tekhnicheskoy nauk, starshiy nauchnyy sotrudnik;  
KATS, Ye.A., inzhener; FEDOROV, G.I.; PORFIR'YEV, M.M., kandidat  
tekhnicheskoy nauk; SIGAYEV, A.V., kandidat tekhnicheskoy nauk;  
KRYUCHKOV, N.V., kandidat tekhnicheskoy nauk, redaktor; FEVZNER,  
A.S., redaktor; PERSON, M.N., tekhnicheskoy redaktor

[Comparative technical and economic evaluation of apartment houses  
having different numbers of stories] Sravnitel'naya tekhniko-ekonomicheskaya kharakteristika zhiloi zastroiki razlichnoi etazhnosti.  
Moskva, Gos. izd-vo lit-ry po stroit. i arkhitekture, 1954. 68 p.

(MIRA 8:3)

1. Akademiya arkhitektury SSSR, Moscow. Nauchno-issledovatel'skiy institut arkhitektury zhilishcha. 2. Institut arkhitektury zhilishcha (for Gel'berg, Kats, Fedorov) 3. Institut gradostroitel'stva (for Porfir'yev)  
(Apartment house)

GEL'BERG, I.A., kandidat tekhnicheskikh nauk; KATS, Ye.A.,  
inzhener; KOLOTILKIN, B.M., kandidat tekhnicheskikh nauk;  
FEDOROV, G.I., inzhener; KUTSENOVA, A.A., redaktor izdatel'stva;  
TOKER, A.M., tekhnicheskiiy redaktor

[Designs of four- and five-story apartment houses; a technical  
and economic analysis] Planirovochnye reshenia chetyrekh-  
piatietazhnykh zhilykh domov; tekhnikoekonomicheskii analiz.  
Moskva, Gos. izd-vo lit-ry po stroit. i arkhit., 1956.  
152 p. (MLBA 10:4)

(Apartment houses)

FEDOROV, G.I., kand.tekhn.nauk; ZAREMBA, B.V., inzh., red.; MORSKOY,  
K.L., red.izd-va; OSENKO, L.M., tekhn.red.

[Economic aspects of planning foundation and underground areas of  
2-5-story apartment houses] Ekonomika proektnykh reshenii  
tsokol'nykh i podzemnykh chastei zdanii v 2-5-etazhnoi zhiloi  
sastroike. Sost. G.I.Fedorov. Moskva, Gos.izd-vo lit-ry po  
stroit., arkhit. i stroit. materialam, 1960. 73 p.

(MIRA 14:4)

1. Akademiya stroitel'stva i arkhitektury SSSR. Institut zhi-  
lishcha.

(Underground construction)

FEDOROV, G. I. Cand Sci -- (diss) "Methods of <sup>reducing</sup> ~~cost reduction in~~ the construction <sup>(costs)</sup>  
through the rational layout of ground floors and basements in  
~~of four- or five-story residential buildings by means of rational layouts of~~  
~~ground floors and basements.~~ Mos, 1957. 16 pp 21 cm. (Acad of Construction and  
Architecture USSR), 110 copies  
(KL, 7-57, 107)

46

FEDOROV, G.I., kand.tekhn.nauk

Estimated and actual costs of apartment houses. Zhil. dom no.1:60-  
63 '60. (MIRA 14:1);

(Apartment houses)  
(Construction industry—Costs)

GEL'BERG, L.A., kand.tekhn.nauk; FEDOROV, G.I., kand.tekhn.nauk

Changing the estimated cost of apartment houses when adapting  
them to local building conditions. Izv. ASIA no.2:89-93 '61.  
(MIRA 15:1)

(Apartment houses--Cost of construction)



124-57-2-2356

Translation from: Referativnyy zhurnal, Mekhanika, 1957, Nr 2, p 123 (USSR)

AUTHOR: Fedorov, G. K.

TITLE: To the Investigation of Three-dimensional Frames (K issledovaniyu prostranstvennykh ferm)

PERIODICAL: Issled. po teorii sooruzheniy, 1954, Nr 6, pp 335-346

ABSTRACT: Presentation of an analytical method for the investigation of three-dimensional systems of bars. An equation is derived to express the projections of a generic vector on four arbitrary radii that are given in terms of their respective direction cosines in some system of coordinates. A number of examples serves to illustrate the application of that equation in the stability analysis and the construction of the influence surfaces for plane and three-dimensional frames.

Yu. P. Grigor'yev

1. Structures--Theory 2. Structures--Stability 3. Mathematics

Card 1/1

FEDOROV, G.K., kandidat tekhnicheskikh nauk.

Kinematic investigation of plane and space mechanisms by the  
projection method. Trudy Khab.IIT no.7:37-97 '54. (MLRA 8:1)  
(Kinematics) (Mathematical physics)

SOV/124-57-7-7465

Translation from: Referativnyy zhurnal. Mekhanika, 1957, Nr 7, p 3 (USSR)

AUTHOR: Fedorov, G. K.

TITLE: Special Cases of the Resolution of a Sliding Vector Into Its Components  
(Osobyye sluchai razlozheniya skol'zyashchego vektora na sostavlyayushchiye)

PERIODICAL: Tr. Khabarov. in-ta inzh. zh.-d. transp., 1956, Nr 9, pp 174-181

ABSTRACT: The author examines the resolution of a sliding vector along n prescribed directions in space.

Reviewer's name not given

Card 1/1

SVICHINSKIY, Nikolay Nikolayevich; YATSENKO, Mikhail Yakovlevich;  
FEDOROV, G.K., red.; FEDOROV, V.P., red.izd-va; LAVRENOVA,  
N.B., tekhn.red.

[Preparation of ships for their inspection by the Register  
of the U.S.S.R.] Podgotovka sudov k osvidetel'stvovaniu  
Registrom SSSR. Moskva, Izd-vo "Morskoi transport," 1960.  
96 p. (MIRA 13:11)  
(Ships--Registration and transfer)

FEDOROV, G.K.

Use of stereomodels in location surveys of railroads. Geod. i kart.  
no.11:41-46 N '61. (MIRA 15:1)

(Railroads--Surveying)

FEDOROV, G.K., assistant

Location of railroads without the use of topographic plans.  
Trudy LIIZHT no.180:12-18 '61. (MIRA 15:7)  
(Railroads--Location)

FEDOROV, G.K., dotsent, kand. tekhn. nauk; DORONIN, V.I., starshiy pre-  
podavatel', kand. tekhn. nauk

Construction of velocity and acceleration plans using a pro-  
jection method. Trudy Khab. IIT no.16:235-249 '64  
(MIRA 18:2)

GEL'BERG, L.A.; FEDOROV, G.I.; ZAL'TSMAN, A.M.; KAPUSTYAN, Ye.D.;  
BAYAR, O.G.; DELLE, V.I.; SHERENTSI, A.A.; MAKLAKOVA, T.G.;  
MONFED, Yu.B.; KOLOTILKIN, B.M.; GLADKOV, B.V.; GAVALOV,  
O.V., red.; GOLOVKINA, A.A., tekhn. red.

[Housing construction in the U.S.S.R.; present state and  
prospects for development] Zhilishchnoe stroitel'stvo v SSSR;  
sostoianie i perspektivy razvitiia. Moskva, Gosstroizdat,  
1962. 202 p. (MIRA 15:11)  
(Apartment houses) (Construction industry)



FEDOROV, G. N.

"Toxic-Hygiene Characteristics in the Manufacture of Lead-Containing Chromatic Chromate Pigments." Cand Med Sci, Odessa State Med Inst, Odessa, 1953. (RZhKhim, No 6, Mar 55)

So: Sum. No 670, 29 Sept 55 - Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (15)

FEDOROV, G.M.

Method for determining porphyrin in urine. Lab. delo 5 no.3:35-36  
My-Je '59. (MIRA 12:6)

1. Kafedra gigiyeny truda (zav. - prof. Ya. B. Reznik) Odesskogo  
meditsinskogo instituta im. N.I. Pirogova.  
(PORPHYRIN AND PORPHYRIN COMPOUNDS)  
(URINE--ANALYSIS)

FEDOROV, G.M.

Modernization of Francke's needle (scarificator needle) by means of  
a movable blade and limitation of the depth of the puncture. Lab.delo  
5 no.4:60-61 JI-Ag '59. (MIRA 12:12)

1. Iz Odesskogo meditsinskogo instituta imeni N.I. Pirogova (dir. -  
prof. I.Ya. Deyneka).  
(SURGICAL INSTRUMENTS AND APPARATUS)

FEDOROV, G.M., kand.med.nauk; LEPYAVKO, A.G., kand.med.nauk

Simple methods of moistening oxygen inhaled from an air pillow.  
Vrach.delo no.7:743 J1 '59. (MIRA 12:12)

1. Gospiatal'naya terapevticheskaya klinika (ispolnyayushchiy obyazannosti zaveduyushchego - dotsent P.L. Ginzburg) Odesskogo meditsinskogo instituta.

(OXYGEN--THERAPEUTIC USE)

SAVEL'YEV, I.A.; FEDOROV, G.M.

Simple method for sharpening hypodermic needles. Med.sestra 19  
no.8:34-35 Ag '60. (MIRA 13:7)

1. Odesskiy meditsinskiy institut imeni N.I. Pirogova.  
(HYPODERMIC NEEDLES)

FEDOROV, G.M., kand.med.nauk

Method of taking air samples for dust determinations in the field.  
Gig.i san. 25 no.8:40-41 Ag '60. (MIRA 13:11)

1. Iz kafedry gigiyeny truda Odesskog meditsinskogo instituta imeni  
N.I.Pirogova.

(AIR—POLLUTION)

FEDOROV, G.M., kand.med.nauk

~~SECRET~~  
Needle-scarificator with removable lances and its correct use. Vrach.  
delo no.10:118-119 0 '60. (MIRA 13:11)

1. Odesskiy meditsinskiy institut.  
(SURGICAL INSTRUMENTS AND APPARATUS)

FEDOROV, G.M.

Vacuum method for taking blood from guinea pigs. Lab. delo 7  
no.1:51 Jm '61. (MIRA 14:1)

1. Odesskiy meditsinskiy institut imeni N.I.Pirogova.  
(BLOOD—COLLECTION AND PRESERVATION)



FEDOROV, G.M.

Rapid fluorescent method for the quantitative determination of  
coproporphyrin in urine. Lab. delo 7 no.3:13-14 Mr '61. (MIRA 14:3)

1. Odesskiy meditsinskiy institut imeni N.I.Pirogova.  
(COPROPORPHYRIN) (FLUORESCENCE)

FEDOROV, G.M.

Effective use of Franke's scarificator. Vrach. delo no.8:135-136  
Ag '61. (MIRA 15:3)

1. Odesskiy meditsinskiy institut.  
(SURGICAL INSTRUMENTS AND APPARATUS)

FEDOROV, G.M., inshener

Efforts made by the testing department to obtain higher productivity of equipment. Tekst.prom.15 no.8:47-48 Ag'55. (MLRA 8:11)

1. Smenny master Glukhovskogo kombinata  
(Textile industry--Equipment and supplies)

FEDOROV, G.M.

Centralized automatic lubrication of spinning and twisting  
spindles. Tekst. prom. 21 no.10:30-81 0 '61. (MIRA 14:10)

1. Nachal'nik pryadil'nogo tsekha bumagopryadil'noy fabriki  
Glukhovskogo khlopchatobumazhnogo kombinata.  
(Spinning machinery--Lubrication)  
(Automatic control)

ZHARZHEVSKIY, I.L.; FEDOROV, G.M.

Assembling the ash collector system. Energ.stroi. no.24:77-83  
'61. (MIRA 15:4)

1. Glavnyy inzh. montazhnogo uchastka tresta "Sevzapenergmontazh"  
(for Zharzhevskiy). 2. Proizvoditel' rabot montazhnogo uchastka  
tresta "Sevzapenergmontazh" (for Fedorov).  
(Narva region--Electric power plants--Design and construction)

FEDOROV, G.M., aspirant kafedry parosilovykh ustanovok.

History of the development of Soviet uniflow boilers. Trudy GIIVT  
10:149-162 '51. (MLRA 10<sup>2</sup>1)

(Boilers)

KHARITONOV, Leopol'd Georgiyevich, dots., kand. tekhn. nauk;  
SHPALENSKIY, M.A., inzh., retsenzent; FEDOROV, G.N.,  
inzh., retsenzent; FRID, L.I., inzh., red.; BODROVA,  
V.A., tekhn. red.

[Shipbuilding materials] Sudostroitel'nye materialy.  
Moskva, Izd-vo "Rechnoi transport," 1963. 260 p.

(MIRA 16:6)

(Shipbuilding materials)

LAKHANIN, Vladimir Vladimirovich; ZAKHAROV, Yuriy Vasil'yevich;  
LEBEDEV, Oleg Nikolayevich; FEDOROV, G.N., retsenzent;  
MIGICHEV, B.S., red.; SHLENNIKOVA, Z.V., red.

[Use of atomic energy in water transport] Ispol'zovanie  
atomnoi energii na vodnom transporte. Moskva, Transport,  
1965. 187 p. (MIRA 18:4)



VILKINTS, Semen Borisovich; FEDOROV, G.P., red.; FEDOROV, B.M., red. izdatel'stva;  
BRATISHKO, L.V., tekhn. red,

[Making woodpulp] Proizvodstvo drevesnoi massy. Moskva, Goslesbun-  
izdat, 1957. 295 p. (MIRA 10:12)

(Wood pulp)

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

1ST AND 2ND CROSS

PROCESSES AND PROPERTIES INDEX

FEDOROV, G. P.

11

30

recycling rubber. G. Ya. Meller, G. P. Fedorov, B. M. Ikski and O. A. Treiman. Russ. 47,070, May 31, 1930. Old rubber is heated under pressure with dil. H<sub>2</sub>SO<sub>4</sub> and steam to decomp. textile material and devulcanize the rubber simultaneously.

AS 0-514 METALLURGICAL LITERATURE CLASSIFICATION

1ST AND 2ND CROSS

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

FBI/COPY, U. S. P.

USSR/Engineering  
Rubber - Reclaiming  
Efficiency, Industrial

Jul 48

"Technological Changes in the Production of Reclaimed Rubber," A. L. Farusov, P. A. Reshetnikov, G. P. Fedorov, Moscow Reclaimed Rubber Factory, 1 p

"Prom Energet" No 7

Suggestion was awarded a third prize in 1947 All-Union Contest. Operation formerly required 250 kw electricity and 1.4 Cal steam per ton of reclaimed rubber. Authors designed special machine (sketch given) for pulverizing rubber.

PA 6/49T28

FEDOROV, G. P.

USSR/Mining - Machines

Card 1/1

Authors : Fedorov, G. P., and Sochinskiy, V. P.

Title : Mechanization of the Excavation of Coal with Cutting and Loading Machines, and Taking Into Account All Basic Processes

Periodical : Mekh. Trud. Rab. Ed. 3, 38 - 39, Apr - May 1954

Abstract : A comprehensive review is presented on problems related to the mechanized excavation of coal with cutting and loading machines, type UKMG. The author takes into account various technological processes involved in the above operations, and suggests improvements which, in his opinion, would result in saving time and increased coal production. Drawings.

Institution : ....

Submitted : ....

FEDOROV, G.P., polkovnik meditsinskoy sluzhby, kandidat meditsinskikh nauk

Problems in medical climatology and climatological therapy in  
sanatoria. Voen.-med. zhur. no.5:45-51 My '56. (MLRA 9:9)  
(WEATHER--MENTAL AND PHYSIOLOGICAL EFFECTS)  
(HEALTH RESORTS, WATERING PLACES, ETC.)

FEDOROV, G.P.

Medical rating of weather. Vop.kur.fizioter. i lech.fiz.kul't. 21 no.  
3:23-27 J1-S '56. (MIRA 9:10)

1. Iz Yaltinskogo klinicheskogo sanatoriya no.1.  
(WEATHER--MENTAL AND PHYSIOLOGICAL EFFECTS)

YEVDOKIMOV, I.I.; ALEKSEYEV, V.D.; ASHIKHMIN, A.K.; BAYEV, N.V.; BEGLAR'YAN, P.A.; BYCHKOV, I.A.; YESLOVA, Ye.T.; VYZHEKHOVSKAYA, M.P.; GURETSKIY, S.A.; DEMIDOV, I.M.; YESIPOV, Ye.P.; ZHUKOV, V.D.; ZELINSKIY, M.G.; ZOL'NIKOV, F.T.; ZOLOTOVA, L.I.; KIVIN, A.N.; KOMARNITSKIY, Yu.A.; KONSTANTINOV, A.N.; KUL'CHITSKAYA, A.K.; MAKSIMENKO, I.I.; MELENT'YEV, A.A.; MOROZOV, I.G.; MURZINOV, M.I.; OZEMBLOVSKIY, Ch.S.; OSTRYAKOV, K.I.; PANINA, A.A.; PAVLOVSKIY, V.V.; PERMINOV, A.S.; PERSHIN, B.F.; PRONIN, S.F.; PSHENNYI, A.I.; POKROVSKIY, M.I.; RASPONOMAREV, Ye.A.; SEMIN, I.N.; SKLYAROV, Yu.N.; TIBABSHEV, A.I.; FARBEROV, Ya.D.; FEDOROV, G.P.; SHUL'GIN, Ya.S.; YAKIMOV, I.A.; VERINA, G.P., tekhn.red.

[Labor feats of railway workers; stories about the innovators]  
Trudovye podvigi zheleznodorozhnikov; rasskazy o novatorakh. Moskva,  
Gos.transp.zhel-dor.izd-vo, 1959. 267 p. (MIRA 12:9)  
(Railroads) (Socialist competition)

FEDOROV, G.P., kand.tekhn.nauk

Use of scrapers for gobbing in wide work entry driving.  
Ugol' Ukr. 6 no.2:29-31 F '62. (MIRA 15:2)

1. Dnepropetrovskiy gornyy institut.  
(Mine filling)



*Fedorov, G.S.*  
AUTHOR: Fedorov, G.S., Candidate of Economic Sciences 3-1-5/32  
TITLE: The Force of Convincing Examples (Sila ubeditel'nykh primerov)  
PERIODICAL: Vestnik Vysshey Shkoly, 1958, # 1, pp 22-26 (USSR)

ABSTRACT: The article states that the capitalistic system is experiencing at the present time a general crisis which does not only extend to economy and politics but also to ideology. It speaks of the means and methods applied by the theorists of capitalism in order to lead astray the masses, to regenerate confidence in capitalism and to calumniate the socialistic order. The so called "people's capitalism", which has been raised in the USA to the level of an official state doctrine, is one of these means. The Political Economy Chair of the Military Academy of the Rear and Transport (Voyennaya akademiya tyla i transporta) is endeavoring to use every kind of instruction - lectures, seminars, consultations, conversations, and reports as well as the press - in order to disclose the theory of "people's capitalism" which aims to retouch the facade of monopolistic capitalism. The article then quotes several examples from lectures of the academy's instructors, naming A.I.Vilionskiy, Candidate of Economic Sciences,

Card 1/2

The Force of Convincing Examples

3-1-5/32

A.V.Meleshchenko and V.F.Stel'mashchuk. The first one spoke on the "Slave Holding Method of Production" and referred to the plantation slavery in the South of the USA, mentioning also the indencencies in Little Rock and other towns, and the statement of James J. Patton, President of the USA National Farmers' Union, that since 1952 about 250,000 farmers have been ruined thereby reinforcing the ranks of the working class and the unemployed. The author further speaks about the noisy propaganda in connection with the "revolution in income" which has allegedly taken place in the USA and of other economic subjects, trying to degrade the American way of life.

There are 2 German and 2 Russian references.

ASSOCIATION: Military Academy of Logistics and Transportation (Voyennaya akademiya tyla i transporta)

AVAILABLE: Library of Congress

Card 2/2

BARSUKOV, N.I., kand.sel'skokhozyaystvennykh nauk; KIZYURIN, A.D., doktor sel'skokhozyaystvennykh nauk; BORINEVICH, V.A., kand.sel'skokhozyaystvennykh nauk; BORMUSOVA, S.N., agronom; VERMNICHEVA, M.D., kand.sel'skokhozyaystvennykh nauk; GESHBLE, E.E., doktor biol. nauk; GOROKHOV, G.I., kand.sel'skokhozyaystvennykh nauk; GUBKIN, S.M., kand. veterinarnykh nauk; YELYKOVA, L.I., kand.sel'skokhozyaystvennykh nauk; KOTT, S.V., doktor biol. nauk; KOCHKINA, V.A., agronom; LAMBIN, A.Z., doktor biol.nauk; LEBEDEVA, Ye.M., agronom; MALAKHOVSKIY, A.Ye., doktor sel'skokhozyaystvennykh nauk; MAYBORODA, N.M., kand. sel'skokhozyaystvennykh nauk; MAYDANYUK, A.E., zootekhnik; OVSYALNIKOV, G.Ye., kand.sel'skokhozyaystvennykh nauk; PETROV, F.A., kand.biol.nauk; POGORELOV, P.F., agronom; POLKOSHNIKOV, M.G., dotsent; RENARD, G.K., kand. sel'skokhozyaystvennykh nauk; RUCHKIN, V.N., prof.; SADYRIN, M.M., kand.sel'skokhozyaystvennykh nauk; TOBOL'SKIY, V.YA., vetvrach; TYAZHEL'NIKOV, S.D., kand.sel'skokhozyaystvennykh nauk; UKHIN, I.I., kand.sel'skokhozyaystvennykh nauk; FEDOROV, G.V., kand.sel'skokhozyaystvennykh nauk; CHIRKOV, D.I., zootekhnik; TSINGOVATOV, V.A., prof.; SHVETSOVA, A.N., kand.sel'skokhozyaystvennykh nauk; SHEVLYAGIN, A.I., kand.sel'skokhozyaystvennykh nauk; SHMENOVSKIY, A.A., red.; GOLUBINSKAYA, Ye.S., red.; NECHAYEVA, Ye.G., red.; PERMSYPKINA, Z.D., tekhnicheskij red.

[Siberian agronomist's reference manual] Spravochnais kniga agronoma Sibiri. Moskva, Gos. izd-vo sel'khoz. lit-ry, Vol.2. 1957. 839 p. (Siberia--Agriculture) (MIRA 11:3)

FE DO ROY, G. V.

112-2-3512

Translation from: Referativnyy Zhurnal, Elektrotekhnik, 1957,  
Nr 2, p.146 (USSR)

AUTHOR: Fedorov, G.V.

TITLE: Experience in Operating Electrical Equipment on the Moscow  
Subway Rolling Stock of the "Г" and "Д" Types and  
Prospects for Modernizing it (Opyt ekspluatatsii  
elektrooborudovaniya podvizhnogo sostava Moskovskogo  
Metropolitena tipa "G" i "D" i perspektivy ego moderni-  
zatsii)

PERIODICAL: In Sbornik: Materialy nauch.-tekhn. soveshchaniya po  
tyagovomu elektrooborudovaniyu. Noyabr' 1953, Riga, 1955,  
pp. 56-60

ABSTRACT: Thirty four train pairs every hour traverse the double-  
track sections of the 45 km long Moscow subway; carrying  
2.3 million passengers daily, the schedule of train move-  
ments is adhered to 99.99 per cent of the time. The high  
quality of the equipment and careful maintenance ensure that  
the schedule is kept. The cars pass a 20 to 40 minute  
inspection every six to eight hours, and a routine inspection  
every three days. Current repairs are carried out every

Card 1/3

112-2-3512

Experience in Operating Electrical Equipment (Cont.)

seven days, and regular overhauling every two and one half to three months. Maintenance repairs are performed after a run of 160,000 to 200,000 kilometers, and average repairs after a run of 320,000 to 400,000 kilometers. The old type A and B car models have a number of defects such as bakelite-shoe braking and other imperfections. The type T cars have been considerably improved. They have electric brakes. Their traction motors ensure rheostatic braking at speeds as high as 60 km/hr without arcing on the commutator and without throw-over on the commutator and without throw-over on the car chassis.

Card 2/3

Experience in Operating Electrical Equipment (Cont.)

112-2-3512

Stamped parts instead of rolled parts are used for the mechanical structure of the car. The development of cracks of the wheel-pair shackles and elsewhere has been arrested. The  $\Delta$  type car is better than other cars in that it has a Cardan shaft, support-chassis traction motor suspension, improved operating conditions of rotors, undercarriage and rails. It also weighs 17 per cent less than the other subway cars. In conclusion, the following ways of improving cars are indicated: further reducing car weight by using alloy steel, etc.; replacing rheostatic braking by regenerative braking and correcting the minor defects which still remain in the  $\Gamma$  and  $\Delta$  cars. [The Moscow subway imeni V.I. Lenin]

I.V.I.

Card 3/3

~~FEDOROV~~ Georgiy Vasil'yevich; MIZIKOV, S.M.; SERGEYEV, D.V.; SOKOLOV, L.S.,  
inzhener, redaktor; BOBROVA, Ye.N., tekhnicheskii redaktor

[Subway rolling stock; type D cars] Podvizhnoi sostav metropolitena  
vagonov tipa D. Moskva, Gos.transp.shel-dor.izd-vo, 1957. 251 p.  
(Subways--Rolling stock) (MLRA 10:9)

RODIGIN, Nikolay Mikhaylovich; RODIGINA, Emiliya Nikolayevna; FEDOROV,  
G.V., otv.red.; BANKVITSER, A.L., red.isd-va; LEBEDEVVA, A.A.,  
tekhn.red.

[Chemical step reactions; mathematical analysis and computations]  
Posledovatel'nye khimicheskie reaktsii; matematicheskiy analiz  
i raschet. Moskva, Izd-vo Akad.nauk SSSR, 1960. 137 p.

(MIRA 14:2)

(Chemical reaction, Rate of)



Fedorov, G.V.

USSR/Magnetism - Ferromagnetism

F-4

Abstr Jour : Referat Zhur - Fizika, No 5, 1957, 11997

Author : Volkenshteyn, N.V., Fedorov, G.V.

Inst : Institute of Physics of Metals, Ural' Branch, Academy of Sciences, Sverdlovsk.

Title : Measurement of the Hall-Kikoin Effect.

Orig Pub : Fiz. metallov i metallovedeniye, 1956, 2, No 2, 377-378

Abstract : Description of a new method for measuring the Hall effect in ferromagnets, a method that does not require the preparation of a specimen in the form of a long rod, and consequently, which permits a measurement in an electromagnet. The specimen, made in the form of a thin plate, is compressed between two halves of an ellipsoid, made of the investigated substances, with insulation of mica. The ellipsoid and the specimen are placed in the magnetic

Card 1/2

Card 2/2

FEDOROV, G.V.

USSR/Nuclear Physics - Penetration of Charged and Neutral  
Particles Through Matter.

C-6

Abs Jour : Ref Zhur - Fizika, No 4, 1957, 8850

Author : Orlov, A.N., Fedorov, G.V.

Inst : Ural' Branch, Academy of Sciences, USSR.

Title : Absorption of Bremsstrahlung of the Betatron in a  
Two-Layer Absorber.

Orig Pub : Zh. tekhn. fiziki, 1956, 26, No 9, 1991-1993

Abstract : It is shown that when radiation passes through a couple-  
layer absorber the intensity of the transmitted radiation  
depends on the selected sequence of absorbers: the atten-  
uation will be greater if the heavy absorber is ahead of  
the light one. Experiments carried out with a betatron  
with a maximum energy of 20 Mev and absorbers made of  
steel and water confirm this conclusion.

Card 1/1

FEDOROV, G.V.

**AUTHORS:** Palatnik, L. S. and Fedorov, G.V. (Khar'kov). 24-9-18/33

**TITLE:** Graphical calculation of the local composition in three-component alloys of variable concentration. (Graficheskiy raschet lokal'nogo sostava v trekhkomponentnykh splavakh peremennoy kontsentratsii).

**PERIODICAL:** Izvestiya Akademii Nauk SSSR, Otdeleniye Tekhnicheskikh Nauk, 1957, No.9, pp. 119-120 (USSR)

**ABSTRACT:** A description is given of a graphical method of calculation of the local composition in three-component alloys of variable concentration using experimental distribution curves for each of the components. The method is applied for calculating the concentrations of systems produced by means of a method developed by S. A. Vekshinskiy (Ref.1). An alloy of variable composition obtained by evaporating various metals from two evaporators represents a binary field of concentrations and in such a field it is possible to calculate a family of circles along which the composition of the alloy remains constant. A three-component system forms a ternary field of concentration and the calculation is correspondingly more complex; families of curves of equal composition no longer exist but families of curves of another type do exist, one representing the lines along

Card 1/3

24-9-18/33

Graphical calculation of the local composition in three-component alloys of various concentration.

which the concentration ratio of two components remains constant and the other lines along which the concentration of one component remains constant. In a real collector these families of curves correspond to a family of analogous straight lines on the concentration triangle of Gibbs. By photometering of a semi-transparent layer of metal deposited on a glass collector which is perpendicular to the crucible axis, the distribution was studied of the density as a function of various parameters (speed of evaporation of the metal etc.) and, following that, the most suitable conditions of evaporation were chosen. The distribution of the condensate represents a mound with a circular symmetry whereby the symmetry axis of the mound coincides with the crucible axis; Fig.1 shows the distribution of a condensate on the collector for Sb, Bi and Ag. From these distribution curves, curves of equal thicknesses for each metal are plotted, finally, a coordinate network is obtained which permits solving the following two basic problems: to determine the composition of the condensate at a given point of the collector and to determine on the collector a point having

Card 2/3

24-9-18/33

Graphical calculation of the local composition in three-component alloys of various concentration.

a given composition. For illustrating the method, the results are included of the calculation of a collector of a real system of Ag-Bi-Sb, Fig.2. The here described method is applicable for any evaporator with a known distribution.

There are 2 figures and 2 Slavic references.

SUBMITTED: April 19, 1957.

AVAILABLE: Library of Congress.

Card 3/3

FEDOROV, G. V.

24-11-29/31

AUTHORS: Komnik, Yu. F., Palatnik, L.S. and Fedorov, G. V. (Khar'kov)

TITLE: Distribution of the condensate on a plane in the case of evaporation of metal from a cylindrical crucible. (Raspredeleniye kondensata na ploskosti pri isparenii metalla iz tsilindrisheskogo tiglya).

PERIODICAL: Izvestiya Akademii Nauk SSSR, Otdeleniye Tekhnicheskikh Nauk, 1957, No.11, pp. 193-196 (USSR)

ABSTRACT: The authors assume that when using a cylindrical crucible as an evaporator, distribution of the condensate of the molecular flow does not necessarily depend on the level of the metal in the crucible, provided this level is sufficiently distant from the opening of the crucible. They studied the dependence on the metal level in the cylindrical crucible of the condensate distribution on a plane collector for the purpose of verifying their assumptions. They also studied the influence of the temperature on the distribution of the condensate and the influence of the location of the metal level in the crucible on the speed of evaporation of the metal. The condensate distribution on a plane transparent base was studied by photometering of semi-transparent metallic layers. The experiments were effected by means of a

Card 1/2

24-11-29/31

Distribution of the condensate on a plane in the case of evaporation metal from a cylindrical crucible.

vacuum set-up containing apparatus as shown in Fig.1, p.194. Cylindrical porcelain crucibles 4.5 and 8 mm dia. and about 30 mm deep were used as evaporators; these were heated by 1 mm dia. nichrome wire. The metal was fed into the system in the form of balls, whereby the charging apparatus was inside the vacuum system. Fig.2 shows the obtained curves of condensate distribution for various levels of metal (tin) in the crucible. The dependence was studied of the distribution of the condensed metal on the evaporation temperature as well as on the metal level in the crucible. The existence was detected of a range of depths of the metal in the crucible within which the distribution of the condensate on the collector and the speed of evaporation of the metal remained constant. This indicates that cylindrical crucibles can be applied as evaporators for making preparations of three component systems of thicknesses up to 0.1 mm according to the method of S. A. Vekshinskiy. There are 4 figures and 7 references, all of which are Slavic.

Card 2/2

SUBMITTED: April 19, 1957.

ASSOCIATION: Khar'kov Polytechnical Institute. (Khar'kovskiy Politekhnikheskiy Institut)

AVAILABLE: Library of Congress.

FEDOROV, G.V.

PA - 2658

AUTHOR  
TITLE

PALATNIK L.S., FEDOROV G.V.

A photometric determination of the thickness of semitransparent metal layers.

PERIODICAL

(Fotometrisheskoye opredeleniye tolshchiny poluposrachnykh metalliches-kikh sloyev, -Russian)

Doklady Akademii Nauk SSSR, 1957, Vol 113, Nr 1, pp 100-103, (U.S.S.R.)

Received 5/1957

Reviewed 6/1957

ABSTRACT

The present work discusses a microphotometric method for the measuring of the absolute density of semitransparent metal layers obtained by evaporation in the vacuum and the density distribution of the condensate of the transparent vitreous base on the occasion of the employing of vaporizers of any type.

The method of photometrizing semitransparent layers:

The symmetry axis of the evaporator must be placed vertical to the collector plane. The density distribution of the condensed layers then forms a circular mound with the maximum of density and thickness on the symmetry axis. Apparently the height of the maximum is proportional to the duration of precipitation. If the metal layers are precipitated on several collectors, the dependence of the blackening  $S$  of the collector can be obtained from the duration  $\tau$  of precipitation. Thus, also the dependence  $S = S(h)$  is obtained, because for the thickness of the layer  $h = k\tau$ , applies, where  $k$  denotes a proportionality factor. If now one of the collectors is photometrized along a straight line passing through the maximum, the dependence of the blackening  $S$  on the distance up to a given

Card 1/2



A Photometric Determination of the Thickness of  
Semitransparent Metal Layers.

PA - 2658

point of the collector can be determined. The corresponding dependences  $S(\mathcal{T})$  and  $S(\rho)$  are shown in form of a diagram. From these two dependences the dependence  $\mathcal{T}(\rho)$  is then determined, which is equivalent to the dependence  $h_{rel}(\rho)$  in relative unities on the ordinate axis. Experimental Part: Experiments were carried out in a vacuum chamber. A cylindrical ferfort crucible served as evaporator and heating took place by means of a nichromium spiral. At a distance of about 40mm from the crucible an immobile metal screen with a long narrow gap was fitted vertical to the crucible axis. Then a glass collector was fitted on to this metal screen. The here discussed photometric method was tested on bismuth films. Bismuth was vaporized on at 600° and 680°, and vaporizing-velocity was changed 6 to 7 times. On this occasion the authors noticed no influence exercised by condensation velocity on the dependence of the blackening of the thickness of the film. From data obtained from some experiments a diagram was determined for the dependence of  $S$  on the thickness  $h$  of the bismuth layer; this method is well suited for the determination of the thickness of semitransparent metal layers. (2ill.)

ASSOCIATION  
PRESENTED BY  
SUBMITTED  
AVAILABLE  
Card 2/2

Polytechnical Institute KHAR'KOV  
Member of the Academy S.A. VEKSHINSKIY  
24.11.1956  
Library of Congress

FEDOROV, G.V., Cand Tech Sci -- (diss) "Application  
of the method of ~~patterns of changing~~ <sup>models varying</sup> composition to  
the study of ~~two~~ <sup>tri-</sup> component alloys." Khar'kov, 1958,  
16 pp with diagrams (Min of higher Education UkSSR.  
Khar'kov Polytechnic Inst im V.I. Lenin) 150 copies  
(KL, 29-58, 133)

- 77 -

AUTHORS: Palatnik, L. S., Fedorov, G. V., Kosevich, V. M. <sup>32-24-6-31/44</sup>

TITLE: On Methods of Measuring the Microhardness in Thin Layers  
(K metodike izmereniya mikrotverdosti v tonkikh sloyakh)

PERIODICAL: Zavodskaya Laboratoriya, 1958, Vol. 24, Nr 6, pp 759 - 761  
(USSR)

ABSTRACT: Metallic layers of various thickness were investigated which had been evaporated on various bases in a vacuum. The thickness of the layers was measured by a microinterferometer according to Linnik, while the microhardness was determined by means of the apparatus FMT-3. The latter had been equipped with two special devices by the mechanic V. V. Gordienko: an automatic load mechanism, and a coordinate table which makes possible the observation of the exact position of the investigated point at the collector in polar coordinates. The apparatus was standardized by means of rock salt; the selection of the metals to be investigated was made in such a way that various combinations occurred: soft layer-hard basis, hard layer-soft basis. The results obtained are given in a table. From the mode of operation mentioned may be

Card 1/2

On Methods of Measuring the Microhardness in Thin Layers

32-24-6-31/44

seen that the metal layer was condensed on glass, that the structure was fine disperse and the surface completely plane. Graphical representations with the corresponding explanations are given. From the results obtained may be seen that with soft layers on hard bases the value for  $n$  is lower than for hard layers on soft bases which must be taken into account in the determination of the microhardness of various coatings; besides it was observed that the value  $n$  is greater with small load. Therefore the microhardness can be determined at the samples obtained according to the method by S. A. Vekshinskiy (Ref 4); the thickness of the sample should be  $10 \mu$  and the material of the bases should be harder than that to be investigated. There are 2 figures, 2 tables, and 8 references, 8 of which are Soviet.

ASSOCIATION: Khar'kovskiy politekhnicheskij institut im. V. I. Lenina  
(Khar'kov Polytechnical Institute imeni V. I. Lenin)

1. Metal films--Mechanical properties
2. Metal films--Test methods
3. Metal films--Physical properties
4. Interferometers--Performance

Card 2/2

AUTHORS: Volkenshteyn, N. V., Fedorov, G. V., Vonsovski, S. V. SOV/56-35-1-11/59

TITLE: The Hall Effect of Pure Nickel Within the Range of Helium Temperatures (Effekt Kholla chistogo nikelya v oblasti geliyevykh temperatur)

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1958, Vol 35, Nr 1, pp 85 - 88 (USSR)

ABSTRACT: The present paper deals with the investigation of the temperature dependence of  $R_1$  and  $R_0$  in the temperature range of between room temperature and  $4,2^\circ\text{K}$ ; according to reference 1 the following holds for the Hall field:  
$$e = R_0 H_2 + R_1 J$$
 ( $J$  = magnetization,  $R_0$  ordinary Hall constant),  
and  $R = A \xi^2$  (Ref 6) ( $A$  = constant,  $\xi$  specific electric resistance).  
The first data concerning the temperature dependence of Ni within the range of from room temperature to Curie (Kyuri) point were supplied by Kikoin (Ref 2); Jan and Gijzman (Jan, Gijzman) (Ref 3) investigated  $R_0$  and  $R_1$  for Ni and Fe, and found an unsharp minimum in the ranges of  $30-50^\circ$  (Ni) and  $50-70^\circ$  (Fe).  $R_1$  decreased from  $T = 300^\circ\text{K} \rightarrow T = 14^\circ\text{K}$  to a twentieth part of its value.

Card 1/3

The Hall Effect of Pure Nickel Within the Range of  
Helium Temperatures

SOV/56-35-1-11/59

The authors of this paper investigated the Hall effect in pure Ni(99,99%). Size of sample: 9.4.0,3 mm,  $H = 5000 \text{ Oe}$ ,  $B$  in the sample: 22 000 G; sensitivity of the potentiometer  $2 \cdot 10^{-8} \text{ V}$ ; measurements were carried out at room temperature,  $0^\circ \text{C}$ , as well as in baths of liquid N, liquid H, and liquid He; specific resistances:  $\rho_{20,4^\circ} / \rho_{293^\circ} = 12,36 \cdot 10^{-3}$  and  $\rho_{4,2^\circ} / \rho_{293^\circ} = 10,28 \cdot 10^{-3}$ .

The measuring results are given in figures 1-4 in form of diagrams.  $R_1$  decreases sharply with reduced temperature and has a minimum at 20 -  $30^\circ \text{K}$ ;  $R_1(T=300^\circ \text{K}) \sim 100 \cdot 10^{-12} \text{ V.cm/A.G}$ ,  $R_1(T=14^\circ \text{K}) \sim 5 \cdot 10^{-12} \text{ V.cm/A.G}$ .  $R_0$  decreases from  $300^\circ$  to  $4,2^\circ \text{K}$  to about  $1/3 (0,6 \rightarrow 0,2 \cdot 10^{-12} \text{ V.cm/A.G})$  and has no minimum. In conclusion the authors (Refs 10-16) discuss the theory of the Hall effect and the possibility of calculating  $R_1$  according to Patrakhin (Ref 15) within the framework of the (s-d) exchange model of ferromagnetism (Vonsovskiy, Ref 16). There are 4 figures and 17 references, 10 of which are Soviet.

Card 2/3

The Hall Effect of Pure Nickel Within the Range of  
Helium Temperatures

SOV/56-35-1-11/59

ASSOCIATION: Institut fiziki metallov Ural'skogo filiala Akademii nauk SSSR  
(Institute of Metal Physics, Ural Branch AS USSR)

SUBMITTED: February 11, 1958

Card 3/3

AUTHORS: Palatnik, L. S., Fedorov, G. V. SOY/20-120-1-20/63

TITLE: The Application of the Method of Samples of Varying Composition to the Investigation of Three-Component Alloys (Primeneniye metoda obraztsov peremennogo sostava dlya issledovaniya trekh-komponentnykh splavov)

PERIODICAL: Doklady Akademii nauk SSSR, 1958, Vol. 120, Nr 1, pp. 79 - 82 (USSR)

ABSTRACT: A particular feature of this method is the considerable deviation of the samples under investigation from equilibrium because of the rapid dissipation of the heat of condensation by a collector. The thinner the layer and the higher its thermal diffusivity, the more will the system depart from the thermodynamical equilibrium. At certain conditions a protracted stabilization of non-equilibrium conditions takes place. This is based upon the maintenance of the internal stress in the condensed layer, on the degree of colloidal dispersion of the monocrystals and of the metastable structures, on the considerable oversaturation and inhomogeneity of the solid solutions etc. The following alloys of a varying

Card 1/3



The Application of the Method of Samples of Varying Composition to the Investigation of Three-Component Alloys SOV/20-120-1-20/63

composition were investigated in connection with the development of a method for the computation of the concentrations of ternary systems: Pb- Bi- Cu, Pb- Bi- Ag, Pb- Bi - Sb. The metals were evaporated from crucibles and condensed on glass plates, which were heated up to 100 - 120°. The denotations used in the diagram are explained. A common basis of all three diagrams given is the binary system Pb - Bi, in which an intermetallic  $\beta$  phase is formed. Lead and bismuth are practically insoluble in copper. The authors give some numerical data on the mutual solubility of the aforementioned metals. The experimental results obtained show that the method of preparing and investigating samples of varying composition employed is fully efficacious, at least for a preliminary examination of the phase diagrams of ternary systems. An advantage offered by the method developed is the considerable reduction of the work necessary for the determination of the topological (semi-quantitative) structure of the phase diagrams. There are 4 figures and 8 references, 6 of which are Soviet.

Card 2/3

The Application of the Method of Samples of Varying      SOV/20-120-1-20/63  
Composition to the Investigation of Three-Component Alloys

ASSOCIATION: Khar'kovskiy politekhnicheskiy institut im. V.I.Lenina (Khar'kov  
Polytechnical Institute imeni V.I.Lenin)

PRESENTED: January 27, 1958, by S.A.Vekshinskiy, Member, Academy of  
Sciences, USSR

SUBMITTED: January 24, 1958

1. Alloys--Sampling
2. Alloys--Thermodynamic properties
3. Alloys--Composition
4. Alloys--Test results

Card 3/3

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ACC NR: AP6018556 SOURCE CODE: UR/0181/66/008/006/1895/1898

AUTHOR: Volkenshteyn, N. V.; Fedorov, G. V. 47  
B

ORG: Institute of Physics of Metals, AN SSSR, Sverdlovsk (Institut fiziki metallov AN SSSR)

TITLE: Hall effect in holmium 11

SOURCE: Fizika tverdogo tela, v. 8, no. 6, 1966, 1895-1898

TOPIC TAGS: holmium, Hall effect, temperature dependence, rare earth metal

ABSTRACT: In view of the lack of data on the temperature dependence of the Hall effect in holmium, the authors measured it in the interval 4.2 - 300K. The holmium purity was 99.9% ( $\rho_{300K}/\rho_{4.2K} = 12$ ). The inductions used in the sample reached 20 kG. The procedure used for the measurements was the same as in an earlier paper (FMM v. 2, 377, 1956). The data-reduction procedure was also described elsewhere (FMM v. 18, 26, 1963). The dependence of the specific Hall emf on the induction for different temperatures and on the temperature for different inductions are plotted, and the effective Hall coefficient is calculated. The temperature dependence of the specific Hall emf exhibits a minimum at 100C, and the Hall coefficient consists of two components, the ordinary one, which does not depend on the temperature (found to be  $-2.8 \times 10^{-12}$  ohm-cm/G), and a component inversely proportional to the temperature, which includes the anomalous Hall coefficient connected with the paramagnetic magnetization. The latter is likewise independent of the temperature and is found to be  $-2.2 \times 10^{-12}$  ohm-cm/G. The results are compared with those obtained for dysprosium and erbium,

Card 1/2

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and are found to be quite similar. It is concluded that measurements with polycrystal-  
line samples disclose the connection between the singularities of the Hall effect and  
the characteristics of the magnetic structures, but do not yield complete information  
on the connection between them. Orig. art. has: 1 formula.

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18:1210

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SOV/126-8-3-9/33

**AUTHORS:** Palatnik, L.S., Fedorov, G.V. and Gladkikh, N.T.

**TITLE:** Study of Aluminium Alloys of the System Al-Cu-Mg on Specimens of Variable Composition

**PERIODICAL:** Fizika metallov i metallovedeniye, 1959, Vol 8, Nr 3, pp 378-386 (USSR)

**ABSTRACT:** Specimens of variable composition were prepared in vacuum by simultaneous evaporation and condensation of the constituents from three cylindrical evaporators (Ref 9 and 10) situated at distances of 70 mm from the collector at the corners of an equilateral triangle inscribed in a circle of 60 mm diameter. The processes of evaporation of the constituents were chosen so that a summary concentration range of copper and magnesium between 0.5 and 16% were ensured on the collector. By means of a photometric method (Ref 11) distribution functions for each of the metals were determined for the chosen evaporation process on the basis of which a calculation of the concentration by graphic methods was carried out (Ref 12). After establishing the required vacuum in the apparatus (approximately  $10^{-5}$  mm Hg) for the removal of adsorbed gases, the collector was heated to a temperature of

Card 1/5

24

66226

SOV/126-8-3-9/33

Study of Aluminium Alloys of the System Al-Cu-Mg on Specimens of Variable Composition

approximately 100°C, then cooled by running water and condensation of the film was brought about. During the condensation of this system on a polished steel collector, which was previously cleaned by treatment with boiling alkali, the specimen was seen to fracture as a certain thickness was attained, which was accompanied by exfoliation due to internal stresses which arise during condensation. At a sufficiently deep etch of the collector with concentrated nitric acid, specimens were obtained in which the junction between the film and the base was increased (probably due to condensation developed as a result of surface etching) and was sufficient for measuring the microhardness. However, the relatively uneven surface made microhardness testing somewhat difficult. In order to rectify this shortcoming a number of experiments was carried out on the application of polished and passivated aluminium as collector. Aluminium was chosen as base for the following reasons: the closeness of the coefficients of thermal expansion of aluminium and the condensate should bring about a decrease

Card 2/5

4

66226

SOV/126-8-3-9/33

Study of Aluminium Alloys of the System Al-Cu-Mg on Specimens of Variable Composition

in internal stresses in the layer and a thick aluminium oxide layer ought to resist mutual diffusion between the condensate and the base. As aluminium oxide is closely adherent to the metal it can be expected that the condensed layer will also be firmly adherent to the aluminium oxide. Specimens of variable composition were investigated by microhardness and X-ray structural phase analysis methods. The microhardness was measured with the PMT-3 instrument at loads of 20 and 40 g. X-ray pictures were taken in the irradiation of an iron anode in a 85 mm diameter camera. The variable composition specimens were annealed in vacuum at various temperatures. Specimens obtained by deposition on the cooled steel collector were investigated by the microhardness method in the concentration range of 0.5 to 6% Mg and 0.5 to 6% Cu through 0.3 to 1% Cu and Mg along the  $C_{31}$  sections (Fig 1), along which the ratio between the concentration of copper and that of aluminium was kept constant (0.005, 0.010, 0.015 etc). Parallel with the microhardness testing

Card 3/5

66226

SOV/126-8-3-9/33

Study of Aluminium Alloys of the System Al-Cu-Mg on Specimens of Variable Composition

a qualitative X-ray phase analysis was carried out. Microhardness was tested two weeks after preparing the specimens. X-ray structural data for an alloy containing 3.3% Cu and 2.4% Mg are shown in Table 1. In Fig 6, comparative curves for the microstructure along the section  $Cu + Mg = 2\%$  show: (1) literature data (Ref 7); (2) experimental results. Fig 7 shows similar curves along the section  $Cu + Mg = 5\%$ . X-ray data for alloys condensed on a hot collector ( $200^{\circ}C$ ) are given in Table 2. Similar data for alloys condensed on a hot collector at  $400^{\circ}C$  are given in Table 3. The authors arrive at the following conclusions: A condition close to equilibrium for aluminium alloys of the system Al-Cu-Mg is attained either by annealing at approximately  $250^{\circ}C$  by mutual heterodiffusion or in the preparation process - condensation on a hot base (approximately  $200^{\circ}C$ ) by surface heterodiffusion. Diagrams for the microhardness of the aluminium corner of the three-constituent system Al-Cu-Mg, after condensation and natural ageing, have been constructed; microhardness

Card 4/5



66226  
SOV/126-8-3-9/33

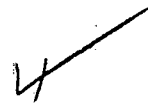
Study of Aluminium Alloys of the System Al-Cu-Mg on Specimens of  
Variable Composition

curves after annealing at 150 and 200°C for the sections  $C_{31} = 0.025$  and  $0.035$ , and after annealing at 250°C for the sections  $C_{31} = 0.015$ ,  $0.025$ ,  $0.035$  and  $0.050$ , have been plotted. By means of the microhardness methods metastable compounds of the  $Al_xCuMg_2$  and  $Al_xCuMg$  type have been found to exist which are responsible for the great hardness of the condensed alloys. There are 7 figures, 3 tables and 14 references, 12 of which are Soviet, 1 English and 1 German.

ASSOCIATION: Khar'kovskiy gosudarstvennyy universitet;  
Khar'kovskiy politekhnicheskiy institut (Khar'kov  
State University; Khar'kov Polytechnic Institute)

SUBMITTED: June 28, 1958

Card 5/5





14 (3), 18 (7)

AUTHORS: Komar, A. P., Academician, AS UkrSSR, SOI/20-125-3-17/66  
Volkonshteyn, N. V., Fedorov, G. V.

TITLE: The Change of the Sign of the Constant of Hall in the  
Ordering of Atoms in an Alloy (Izmeneniye znaka postoyannoy  
Zholla pri uporyadochenii atomov v splave)

PERIODICAL: Doklady Akademii nauk SSSR, 1959, Vol 125, Nr 5,  
pp 530-531 (USSR)

ABSTRACT: The authors first mention some previous papers on this  
subject. The alloy  $Ni_3Mn$  is characterized by a dependence of  
its electric and galvano-magnetic properties on the  
spontaneous magnetization  $I_s$  and on the degree of the long-  
range order  $\eta$ . This dependence discerns this alloy from  
pure ferromagnetic metals and also from binary alloys of  
similar structure and composition. The Hall electromotive  
force  $E_H$  of the alloys  $Ni_3Mn$  was investigated for the cases  
of different heat treatment of the used samples in a wide  
temperature range down to the temperature of liquid helium.  
According to these investigations,  $E_H$  strongly depends

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The Change of the Sign of the Constant of Hall in the Ordering of Atoms in an Alloy BM/20-105-3-17/55

on the manner of fixing the investigated state of the alloy. Even in the case of fixing the non-ordered state (in ordering from high temperatures), the different rate of hardening exerts a great influence. If the alloy is quickly cooled from 800° to room temperature, the alloy is paramagnetic at this temperature. The Hall constant of this sample was positive and equal to  $R_0 = + 0.09 \cdot 10^{-12}$  v. cm/o. gauss.

In the case of ferromagnetics,  $v_H$  (it seems to denote the Hall potential) is calculated according to the formula

$$v_H = \frac{R_0 B_i}{d} + \frac{R_s \cdot 4\pi I_s}{d}$$

$R_0$  denotes the ordinary Hall constant and  $R_s$  the Hall constant connected with the spontaneous magnetization  $I_s$ .  $R_s$  has an unusual, non-classical, positive sign. This fact agrees also with the results of American authors (Ref 7). The Hall electromotive force, which is due to the existence of  $I_s$ , was found only at low temperatures.

Cont 2/1