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THERMODYNAMIC CONSIDERATIONS ON THE CATALYTIC OXIDATION REACTIONS OF CARBON MONOXIDE AND SULFUR DIOXIDE. [1962] 20p. 13 refs.
Order from SLA \$1.60 62-14964

I. Seiyama, T.
II. Suenaga, A.
III. Sakai, W.

Trans. of Nihon Kagaku Zasshi (Japan) 1961, v. 82, no. 3, p. 292-298.

DESCRIPTORS: *Carbon compounds, *Monoxides, *Sulfur compounds, *Dioxides, Oxidation, *Catalysts, *Semiconductors, *Thermodynamics, Reaction kinetics.

With respect to the oxidation reaction of carbon monoxide, the order of activities of various types of oxide catalysts was evaluated from the thermodynamic properties of the catalysts. Similar considerations were made on the oxidation of sulfur dioxide, but in this instance, simple application of the thermodynamic values of the oxide catalysts did not agree with the actual state (Chemistry--Physical, TT, v. 9, no. 5) (over)

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Decomposition Velocity of Potassium
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Vol VII, 1950, pp 350-353.
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RESEARCH INTO MATERIAL FOR DRILL PIPE CAS-
INGS WITH A MINIMUM YIELD STRENGTH OF
98 kg/mm². [1962] [28] p. 22 refs.
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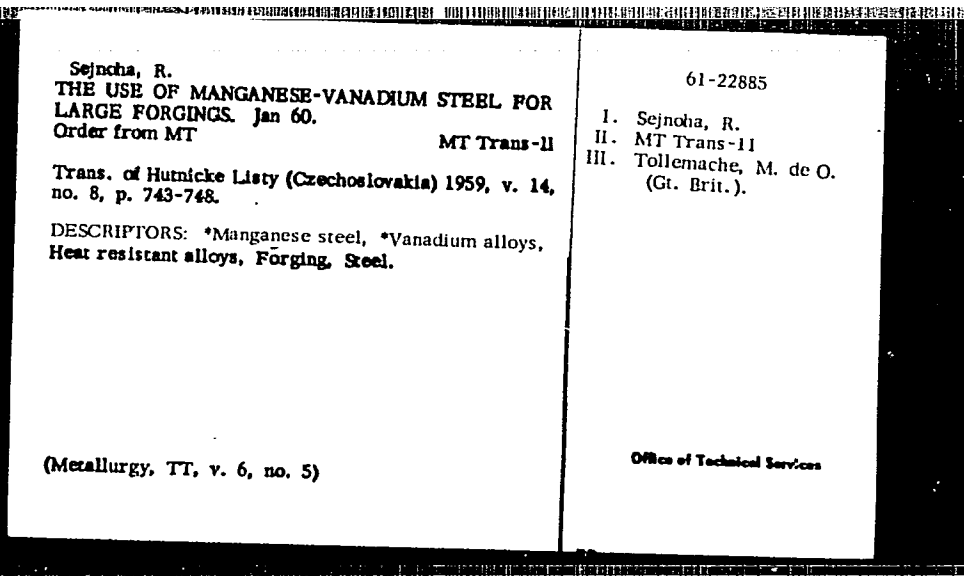
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no. 9, p. 829-840.

DESCRIPTORS: *Steel, *Pipes, *Steel tubing, Produc-
tion, Mechanical properties, Heat treatment.

Sixteen experimental melts were cast with carbon con-
tents from 0.18 to 0.32% and with various combina-
tions of manganese, silicon and chromium, and one
melt was also alloyed with nickel. From the ingots
rods were forged, which were subjected to heat treat-
ment which corresponded to the heat treatment of the
pipes. The results of the mechanical tests were
worked out by means of correlation analysis to pro-
duce equations expressing the relationship between the
(Metallurgy--Ferrous Metals, TT, v. 8, no. 4)(over)

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JPRS 23583

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<p>DESCRIPTORS: Chemical industry, *Additives, *Hardness.</p>		
<p>(Engineering--Chemical, TT, v. 8, no. 10)</p>		<p>0226000</p>
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ACSI I-2392-B
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Acids by Satake, Kazuo, and Seki, Tokuchiro,
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pp 47R-50R.

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Vapor Pressures of Molecular Crystals (Eighth Report). Vapor Pressures of Aromatic Nitro Compounds, by Isamu Hitta, Syuzo Saki, Masanori Konetsani, Kentaro Sato, 13 pp.

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Sekiguchi, H.

PREPARATION OF POLYBUTANAMIDE FROM
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Order from SLA \$1.60

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Trans. of Soc[ieté] Chim[ique de] France. Bull[etin]
1960, p. 1827-1830.

DESCRIPTORS: *Pyrroles, *Polymers, *Canavanine,
*Amides, Polymerization, Condensation reactions.

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(Chemistry--Organic, TT, v. 7, no. 11)

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DESCRIPTORS: *Rubber, Processing.

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KINETIC STUDY OF THE CATALYTIC POLYMERIZATION OF α -PYRROLIDONE (II). [1961] 11p.
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Order from SLA \$1.60

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Trans. of Soc[été] Chim[ique de] France. Bull[etin] 1960, no. 1831-1834.

DESCRIPTORS: *Pyrrolidinone, *Polymerization, Catalysis, *Reaction kinetics, Lactams

Subsequent to the preparation of polybutanamide from α -pyrrolidone (Soc. Chim. France Bull. 1960: 1827-1830; available in translation from SLA \$1.60 as 61-20855) we have studied the kinetics of catalytic polymerization of this pentagonal lactam. Polymerization is initiated by an N-acyl lactam whose acyl residue is incorporated into the polymer; catalysis is by an alkaline salt of the lactam. The reaction is (over)

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PHYSICAL AND MECHANICAL PROPERTIES OF POLYBUTANAMIDE (IV). [1964] 12p. (7 figs. 18 refs. omitted)
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DESCRIPTORS: *Lactams, Polymerization, *Amides, Polymers, Viscosity, Synthetic fibers, Mechanical properties.

208343

A study of the mechanical properties of polybutanamide fiber has demonstrated a remarkable resistance and tenacity with reasonable elasticity (20% elongation and kg/mm² load at the yield point, and 30 to 40% elongation and 50 kg/mm² load at the breaking point, properties comparable to those of natural silk. (Author)

(Chemistry--Organic, TT, v. 7 no. 10)

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REACTION MECHANISM OF THE ALKALINE CATALYTIC POLYMERIZATION OF α -PYRROLIDONE (III). [1961] 13p. (17 refs. omitted)
Order from SLA \$1.60

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Trans. of Soc[ieté] Chim[ique de] France. Bull[etin] 1960, p. 1835-1838.

DESCRIPTORS: Lactams, Amides, Polymerization, Chemical reactions, Catalysis, *Pyrrolidinone.

The kinetic study of the alkaline catalytic polymerization of α -pyrrolidone and the conclusions relative to the reactions of the initiator lead us to visualize for this reaction an anionic "lactamolytic" reaction mechanism like a polyaddition. By assuming three kinds of reaction between the amides and the lactam anion, according to the type of amides: cyclic amide, imide (Chemistry--Organic, TT, v. 7, no. 10) (over)

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Multi-Anode β Gas Flow Counter for Large
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The Operation of Drying Plants for Dewatering Coals
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CONDITIONS FOR PROFITABLE OPERATION AND
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